

From Total Scattering to Total Understanding: Structure-Functionality Relationships in Scheelite-Type Oxides

Presented by:

Mr Bryce Mullens

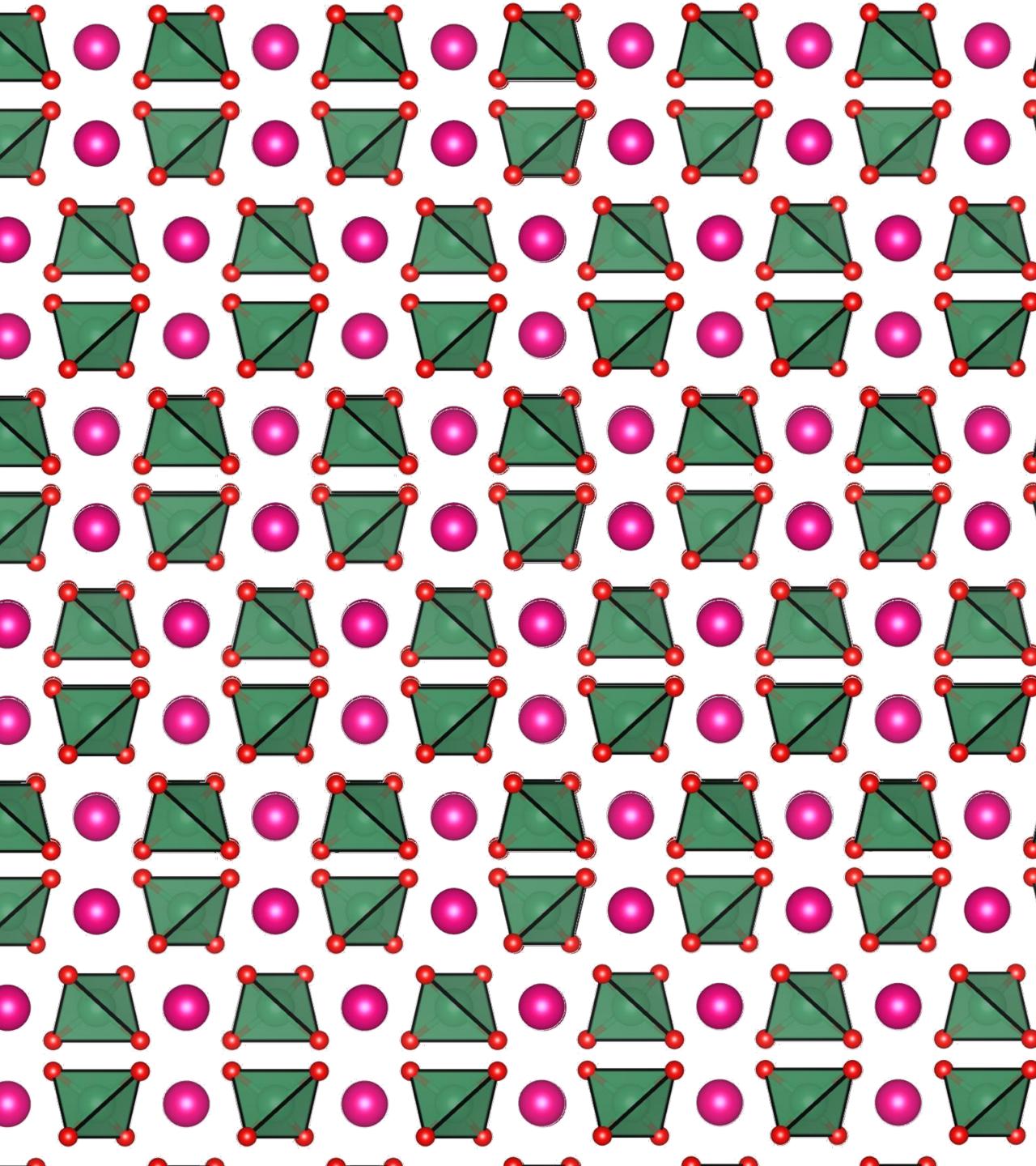
University of Sydney / Université Paris-Saclay

PhD Candidate in Solid-State / Materials Chemistry

ADD2022 – Wednesday 19th October 2022



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Thank You!

Supervisors

- › Prof Brendan J. Kennedy
- › Prof Chris Ling
- › Dr Zhaoming Zhang
- › Prof Gianguido Baldinozzi
- › Prof Karena Chapman
- › Prof Maik Lang

Postdocs

- › Dr Matilde Saura Múzquiz
- › Dr Fred Marlton

 **OAK RIDGE**
National Laboratory

 **diamond**

Oak Ridge National Laboratory

- › Dr Alicia Manjon-Sanz
- › Dr Joerg Neufeind
- › Dr Michelle Everett

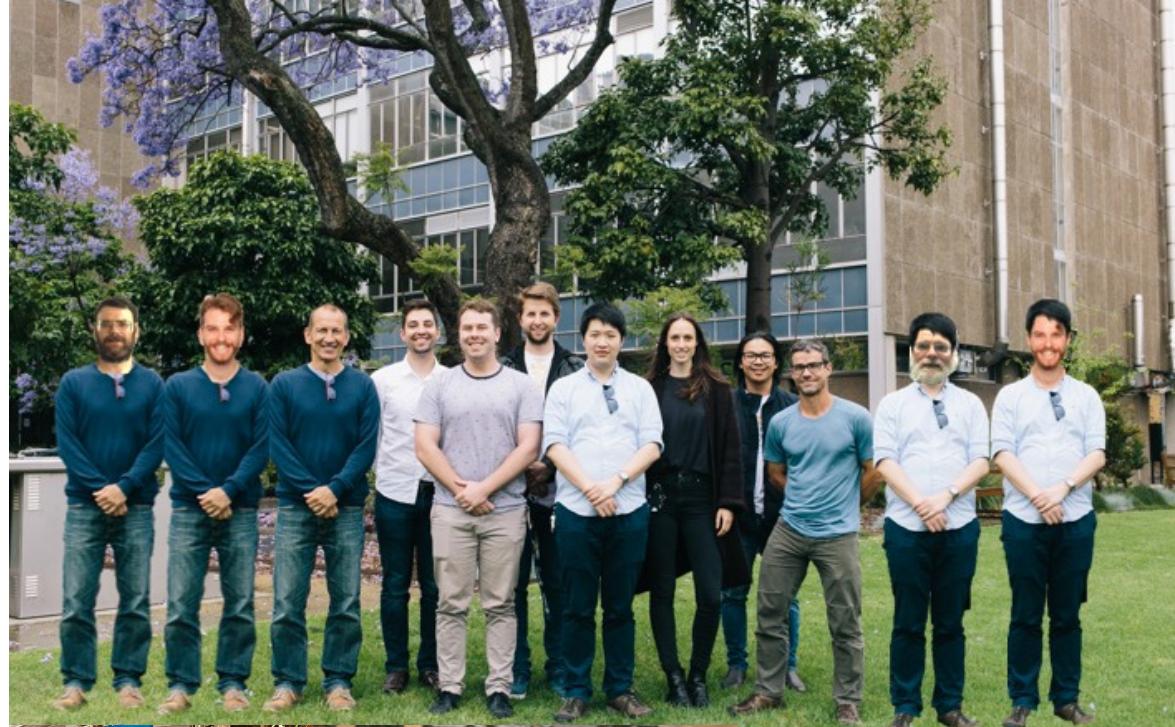
Diamond Light Source

- › Dr Phil Chater

The University of Sydney Solid-State Research Group

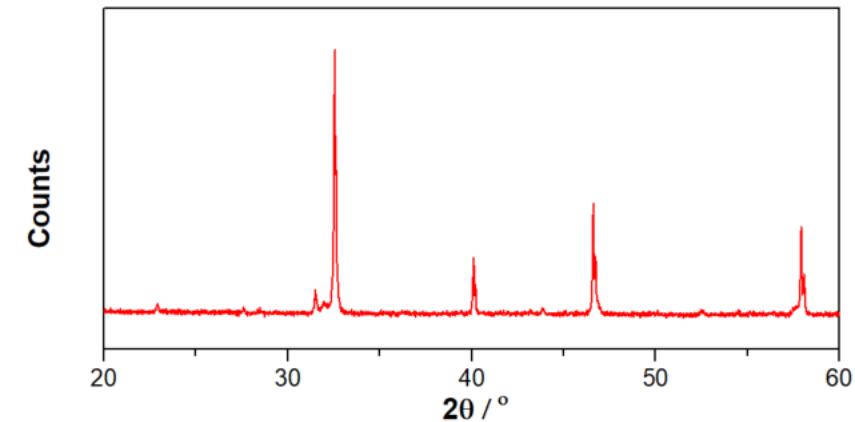
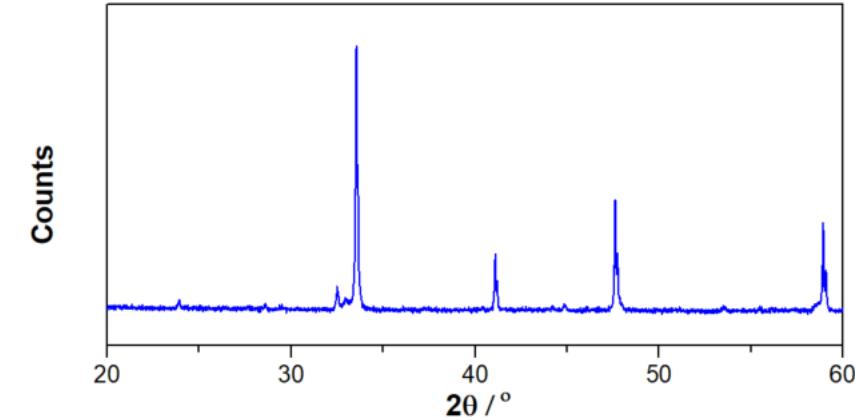
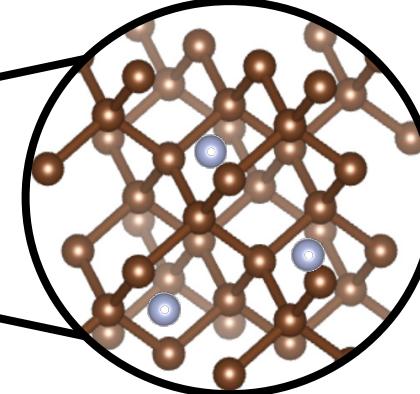
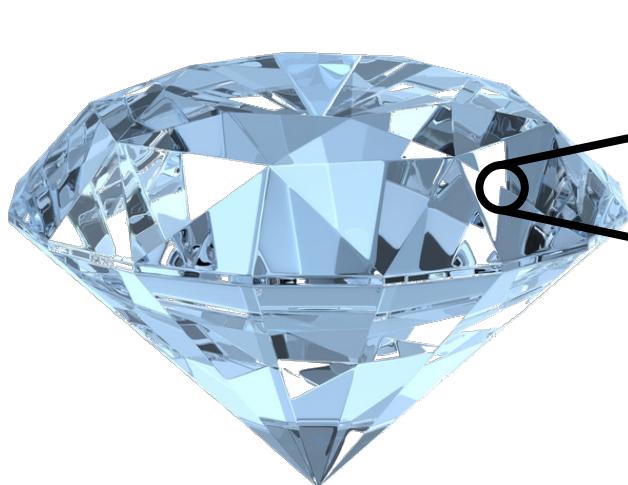
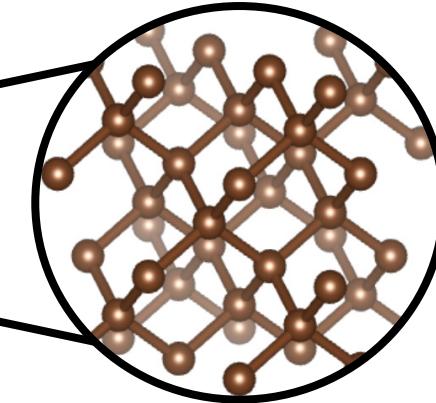
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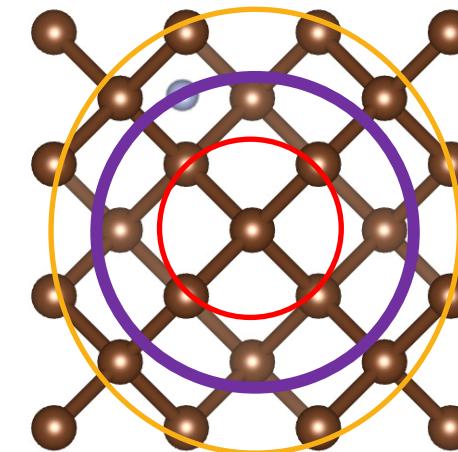
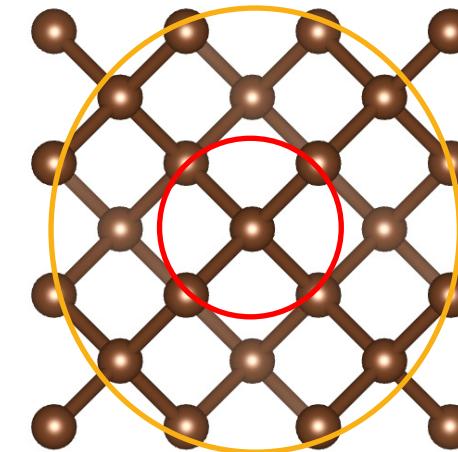
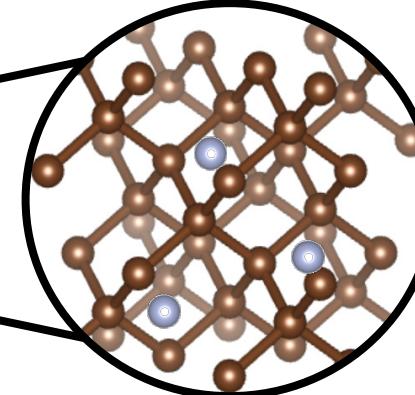
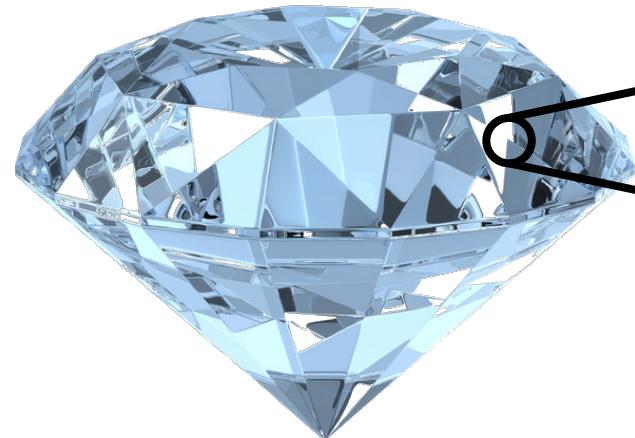
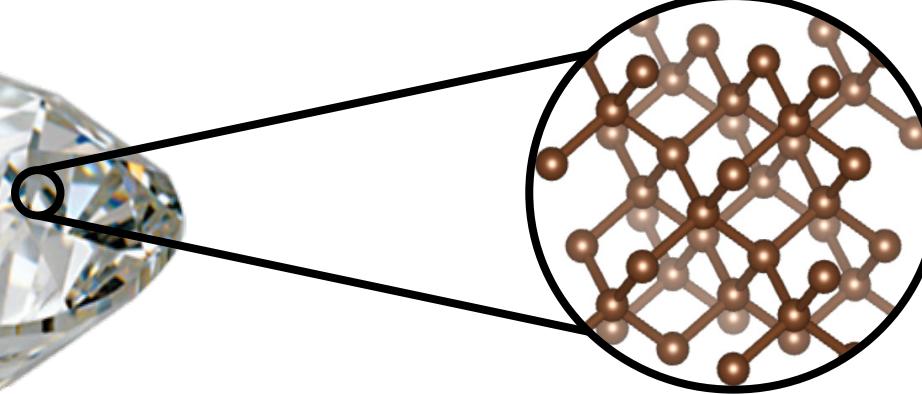


Structures and Properties of Diamond



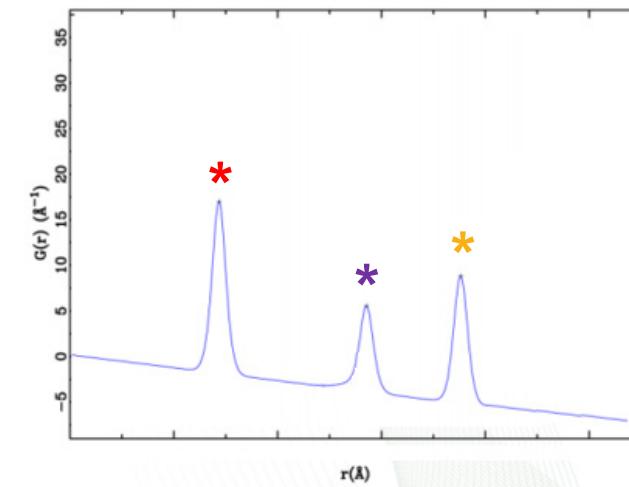
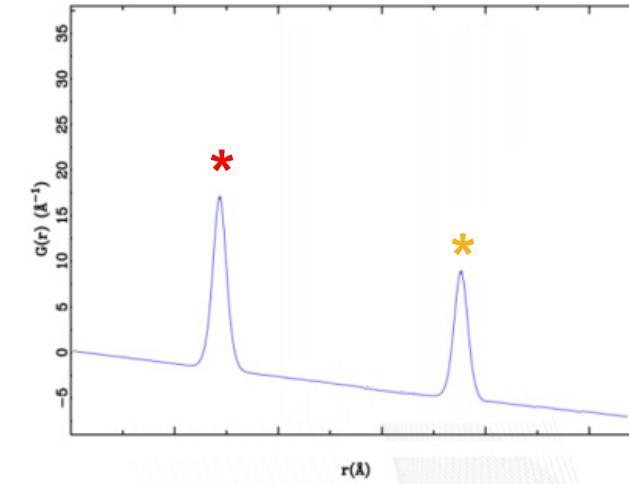
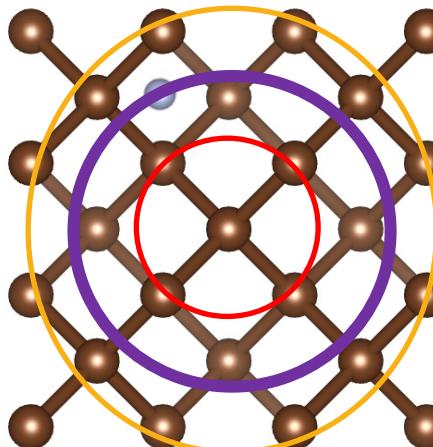
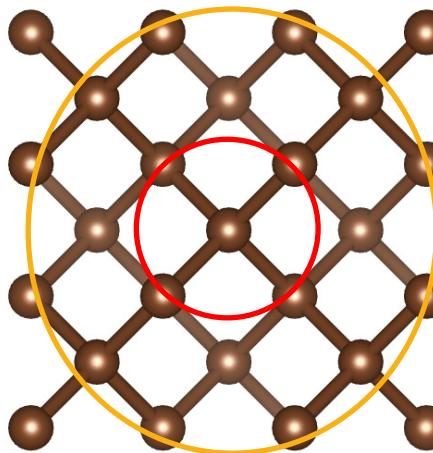
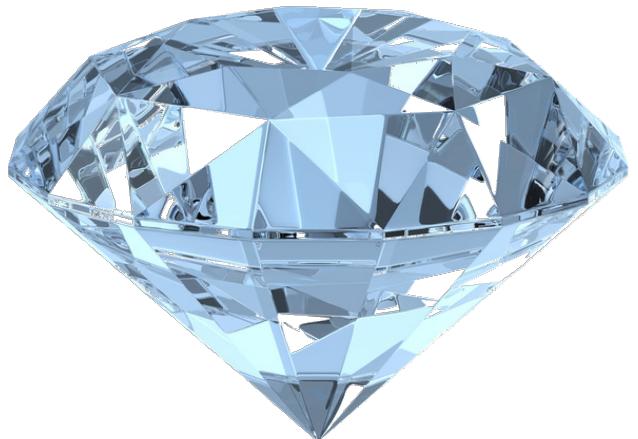


Structures and Properties of Diamond



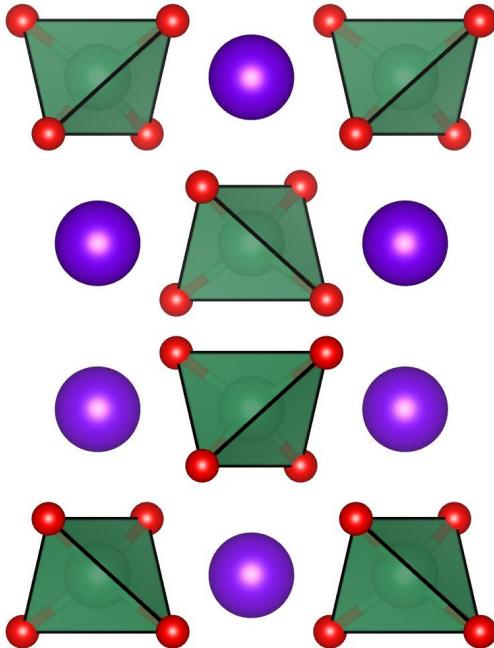


Structures and Properties of Diamond





The Wonderous ABO_4 Structure

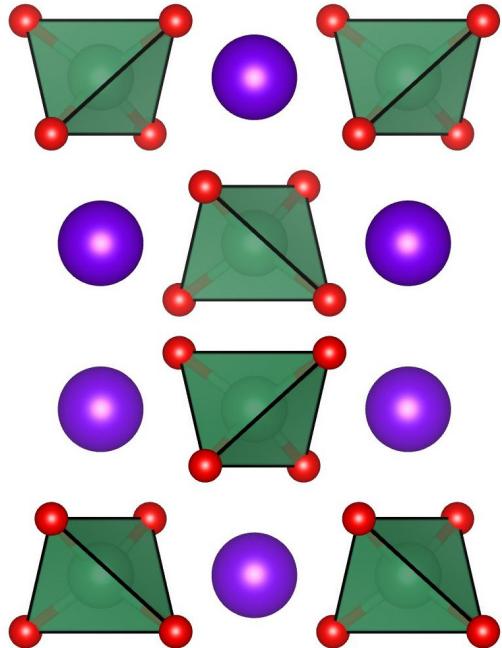


ABO_4
Tetragonal Scheelite

1	H	2	He
3	Li	4	Be
11	Na	12	Mg
19	K	20	Ca
37	Rb	38	Sr
55	Cs	56	Ba
87	Fr	88	Ra
*		71	Lu
*		72	Hf
*		73	Ta
*		74	W
*		75	Re
*		76	Os
*		77	Ir
*		78	Pt
*		79	Au
*		80	Hg
*		81	Tl
*		82	Pb
*		83	Bi
*		84	Po
*		85	At
*		86	Rn
*		103	Lr
*		104	Rf
*		105	Db
*		106	Sg
*		107	Bh
*		108	Hs
*		109	Mt
*		110	Ds
*		111	Rg
*		112	Cn
*		113	Nh
*		114	Fm
*		115	Mc
*		116	Lv
*		117	Ts
*		118	Og
*		57	La
*		58	Ce
*		59	Pr
*		60	Nd
*		61	Pm
*		62	Sm
*		63	Eu
*		64	Gd
*		65	Tb
*		66	Dy
*		67	Ho
*		68	Er
*		69	Tm
*		70	Yb
*		89	Ac
*		90	Th
*		91	Pa
*		92	U
*		93	Np
*		94	Pu
*		95	Am
*		96	Cm
*		97	Bk
*		98	Cf
*		99	Es
*		100	Fm
*		101	Md
*		102	No



The Wonderous ABO_4 Structure



ABO_4
Tetragonal Scheelite

A standard periodic table of elements is shown, highlighting specific groups related to the ABO_4 structure. The Lanthanide series (Ce to Lu) is highlighted in green, the Actinide series (Ac to No) is highlighted in green, and the transition metals (Sc, Ti, V, Cr, Mn, Fe, Co, Ni, Cu, Zn, Ta, Hf, W, Re, Os, Ir, Pt, Au, Hg, Tl, Pb, Bi, Po, At, Rn) are highlighted in blue. The A-site cations (Sc, Y, La, Ce, Pr, Nd, Pm, Sm, Eu, Tb, Dy, Ho, Er, Tm, Yb, Ac, Th, Pa, U, Np, Pu, Am, Cm, Bk, Cf, Es, Fm, Md, No) are grouped together.

Solid State Ionics 262 (2014) 530–535

Contents lists available at ScienceDirect

Solid State Ionics

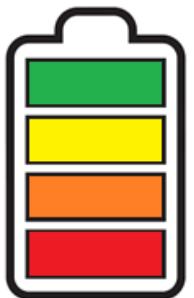
journal homepage: www.elsevier.com/locate/ssi

Crystal structure and potential interstitial oxide ion conductivity of LnNbO_4 and $\text{LnNb}_{0.92} \text{W}_{0.08} \text{O}_{4.04}$ ($\text{Ln} = \text{La, Pr, Nd}$)

Cheng Li, Ryan D. Bayliss, Stephen J. Skinner *

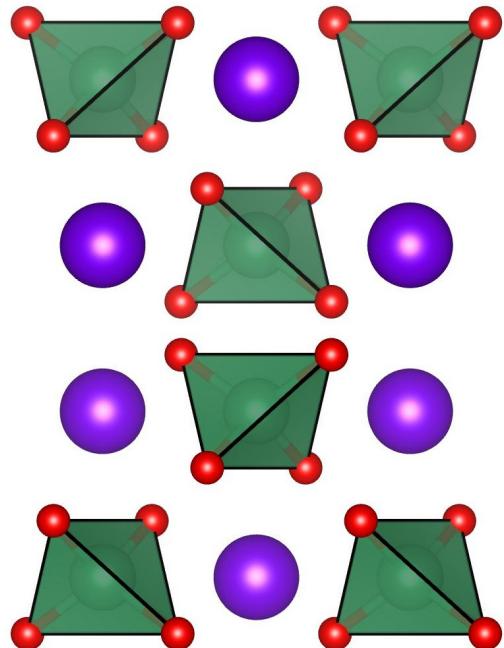
Department of Materials, Imperial College London, Exhibition Road, London, SW7 2AZ, UK

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The Wonderous ABO_4 Structure



ABO_4
Tetragonal Scheelite

1	H	2	He
3	Li	4	Be
11	Na	12	Mg
19	K	20	Ca
37	Rb	38	Sr
55	Cs	56	Ba
87	Fr	88	Ra
*	103	104	Lr
*	105	106	Rf
	Db	Sg	
	Bh	Hs	
	Mt		
	Ds		
	Rg		
	Cn		
	Nh		
	Fl		
	Mc		
	Lv		
	Ts		
	Og		
	57	58	59
	La	Ce	Pr
	60	61	62
	Nd	Pm	Sm
	63	64	65
	Eu	Gd	Tb
	66	67	68
	Dy	Ho	Er
	69	70	71
	Tm	Yb	
	89	90	91
	Ac	Th	Pa
	92	93	94
	U	Np	Pu
	95	96	97
	Am	Cm	Bk
	98	99	99
	Cf	Es	Fm
	100	101	102
	Fm	Md	No

Materials Research Bulletin 61 (2015) 422–432

Contents lists available at ScienceDirect

Materials Research Bulletin

journal homepage: www.elsevier.com/locate/matesbu

Scheelite-type MWO_4 ($M = Ca, Sr$, and Ba) nanophosphors: Facile synthesis, structural characterization, photoluminescence, and photocatalytic properties

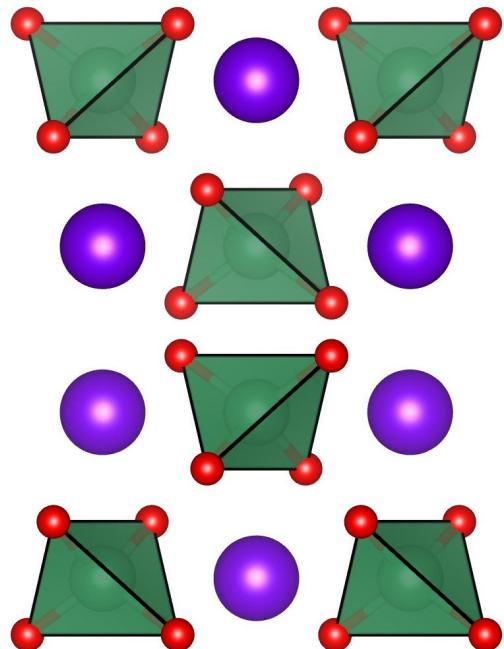
C. Shivakumara ^{a,*}, Rohit Saraf ^b, Sukanti Behera ^a, N. Dhananjaya ^c, H. Nagabhushana ^d

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The Wonderous ABO_4 Structure



ABO_4
Tetragonal Scheelite

1 H	2 He
3 Li	4 Be
11 Na	12 Mg
19 K	20 Ca
37 Rb	38 Sr
55 Cs	56 Ba
87 Fr	88 Ra
* 103 Lr	* 104 Rf
* 105 Db	* 106 Sg
* 107 Bh	* 108 Hs
* 109 Mt	
21 Sc	22 Ti
23 V	24 Cr
25 Mn	26 Fe
27 Co	28 Ni
29 Cu	30 Zn
31 Ga	32 Ge
33 As	34 Se
35 Br	36 Kr
13 Al	14 Si
15 P	16 S
17 Cl	18 Ar
5 B	6 C
7 N	8 O
9 F	10 Ne
14 Si	15 P
16 S	17 Cl
32 Ge	33 As
34 Se	35 Br
50 In	51 Sn
51 Sb	52 Te
52 Te	53 I
54 Xe	
79 Au	80 Hg
81 Tl	82 Pb
83 Bi	84 Po
85 At	86 Rn
111 Rg	112 Cn
113 Nh	114 Fl
115 Mc	116 Lv
117 Ts	118 Og
* 57 La	* 58 Ce
* 59 Pr	* 60 Nd
* 61 Pm	* 62 Sm
* 63 Eu	* 64 Gd
* 65 Tb	* 66 Dy
* 67 Ho	* 68 Er
* 69 Tm	* 70 Yb
* 89 Ac	* 90 Th
* 91 Pa	* 92 U
* 93 Np	* 94 Pu
* 95 Am	* 96 Cm
* 97 Bk	* 98 Cf
* 99 Es	* 100 Fm
* 101 Md	* 102 No



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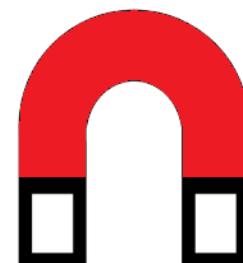
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DOI: [10.1039/C9CP00448C](https://doi.org/10.1039/C9CP00448C) (Communication) *Phys. Chem. Chem. Phys.*, 2019, **21**, 7261-7264

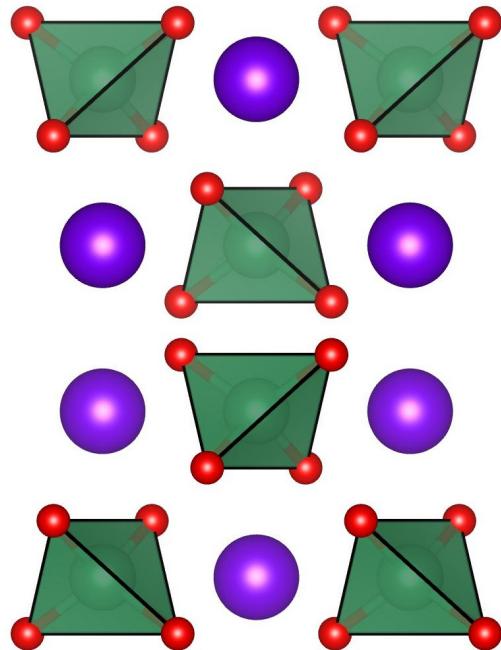
Structural and magnetic studies of $KOsO_4$, a 5d¹ quantum magnet oxide[†]

Sean Injac ^a, Alexander K. L. Yuen ^{ab}, Maxim Avdeev ^{ab}, Fabio Orlandi ^c and Brendan J. Kennedy ^{ab} *^a





The Wonderous ABO_4 Structure



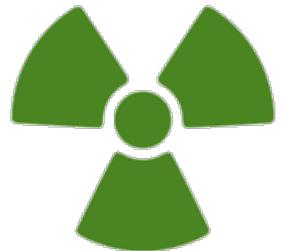
ABO_4
Tetragonal Scheelite

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11 Na	12 Mg
19 K	20 Ca
37 Rb	38 Sr
55 Cs	56 Ba
87 Fr	88 Ra
* 103 Lr	* 104 Rf
21 Sc	22 Ti
39 Y	40 Zr
71 Lu	72 Hf
41 Nb	42 Mo
73 Ta	74 W
43 Tc	44 Ru
75 Re	76 Os
77 Ir	78 Pt
79 Au	80 Hg
81 Tl	82 Pb
83 Bi	84 Po
85 At	86 Rn
114 Fl	115 Mc
116 Lv	117 Ts
118 Og	
* 57 La	58 Ce
* 89 Ac	59 Pr
	60 Nd
	61 Pm
	62 Sm
	63 Eu
	64 Gd
	65 Tb
	66 Dy
	67 Ho
	68 Er
	69 Tm
	70 Yb

Materials Research Bulletin
Volume 12, Issue 1, January 1977, Pages 25-33

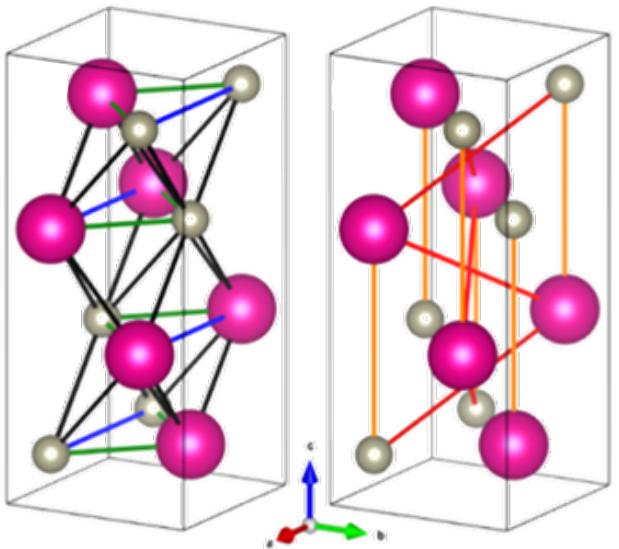
Les niobates $CaUNb_2O_8$ et $MThNb_2O_8$ ($M = Ca, Sr, Cd$). Etude des transformations monoclinique - quadratique

Gilles Fonteneau, Hervé L'Helgoualch, Jacques Lucas

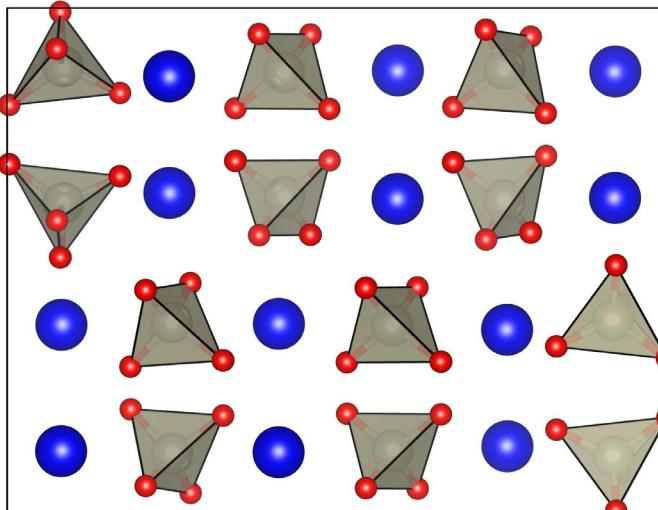




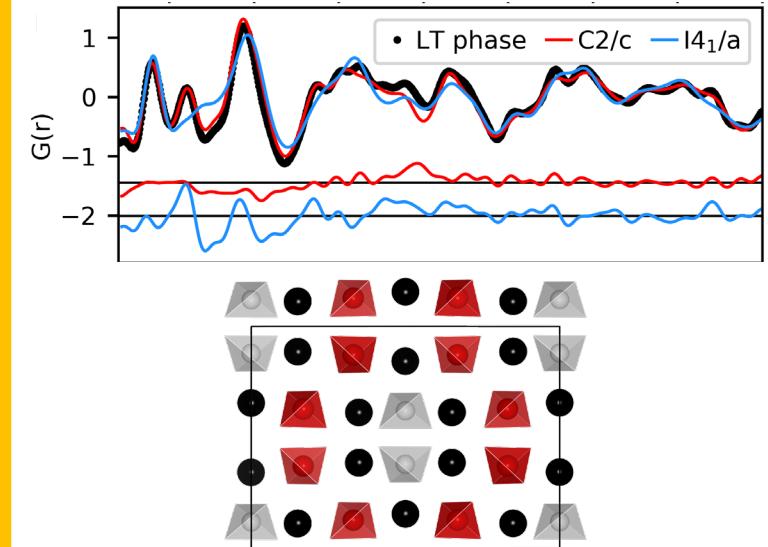
High-Temperature Phase Transition in RbReO₄



Re-Entrant Phase Transition in TiReO₄



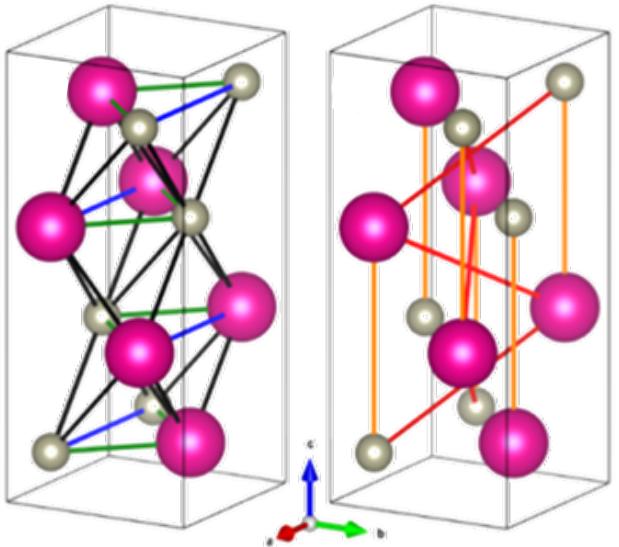
Symmetry-Lowering in Bi₃FeMo₂O₁₂



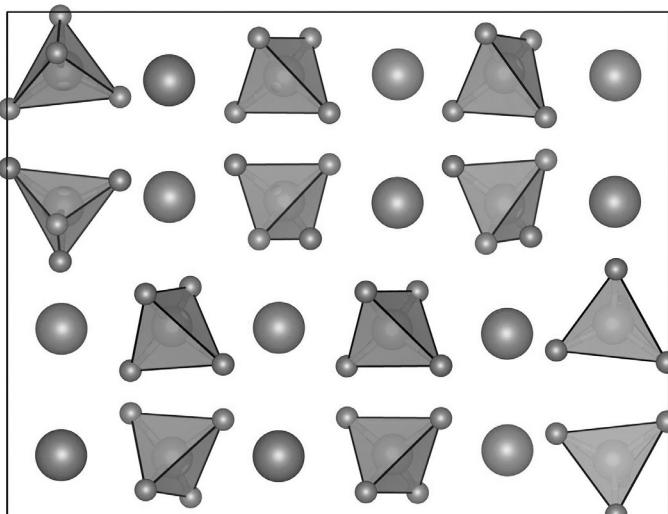


High-Temperature Phase Transition in RbReO₄

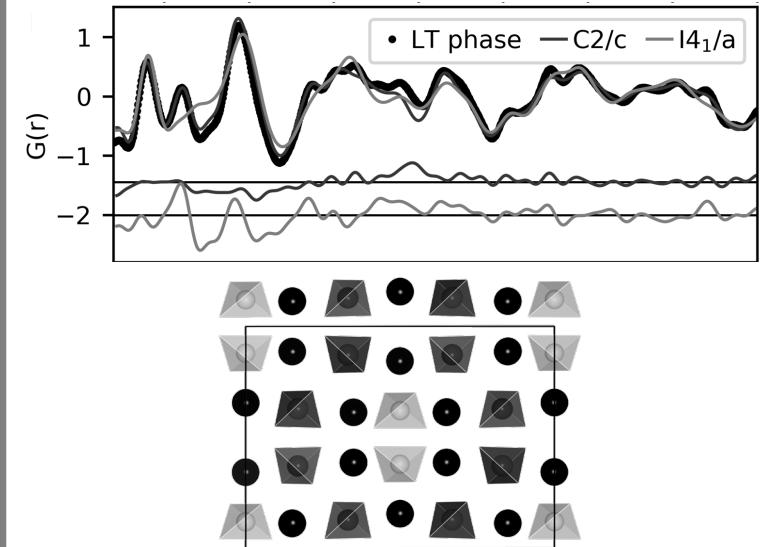
Why do the Re-O bonds decrease upon heating?



Re-Entrant Phase Transition in TlReO₄

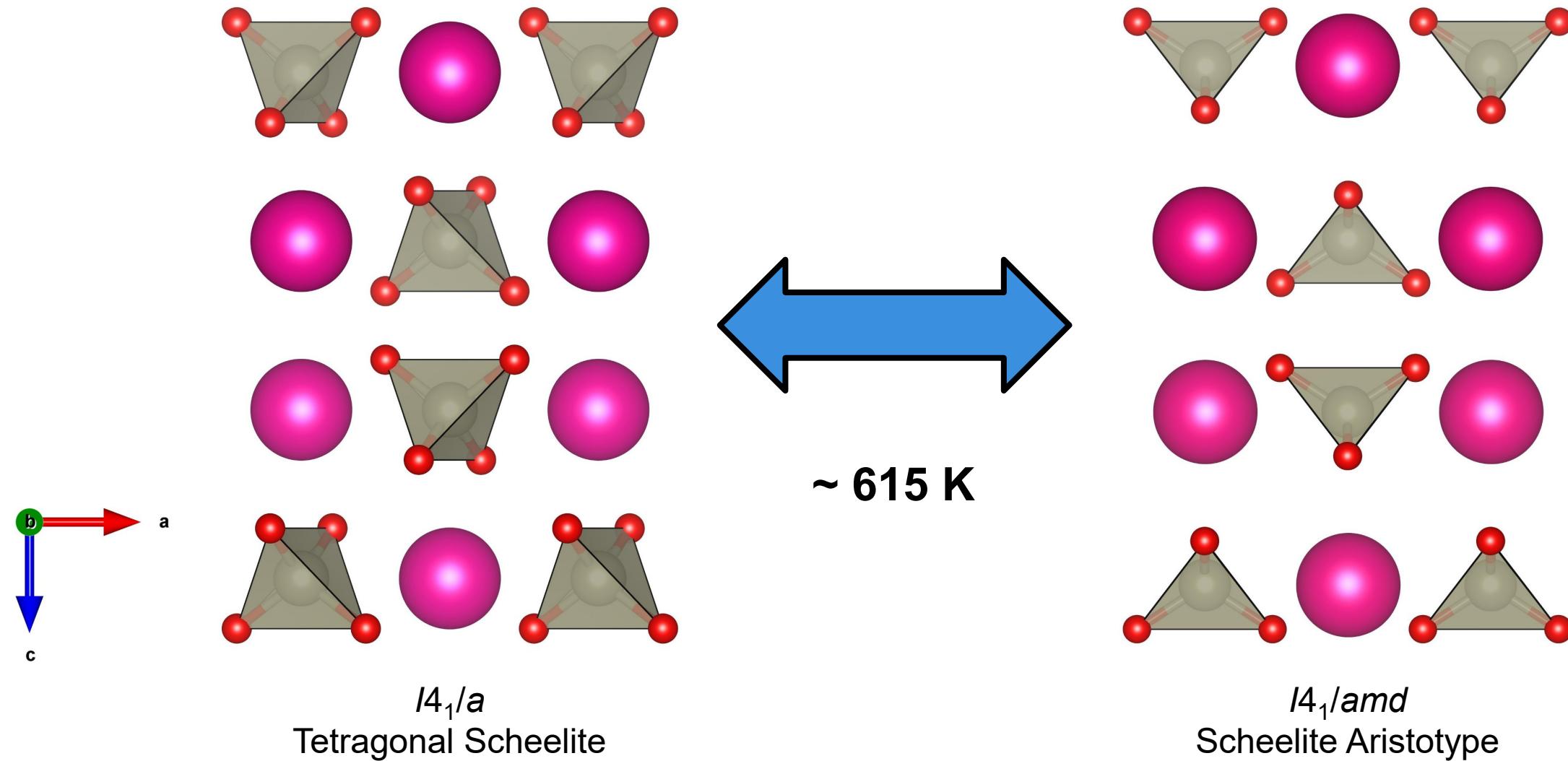


Symmetry-Lowering in Bi₃FeMo₂O₁₂





Phase Transition in RbReO₄

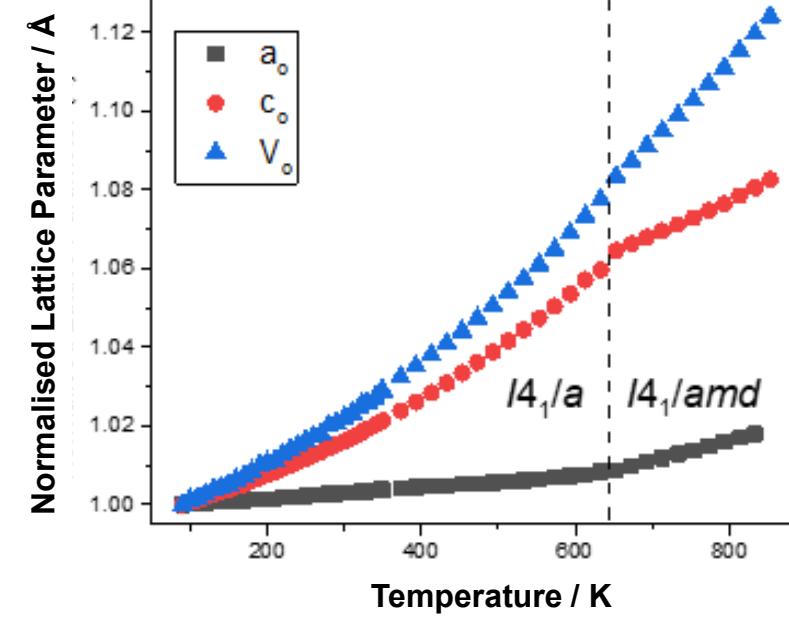
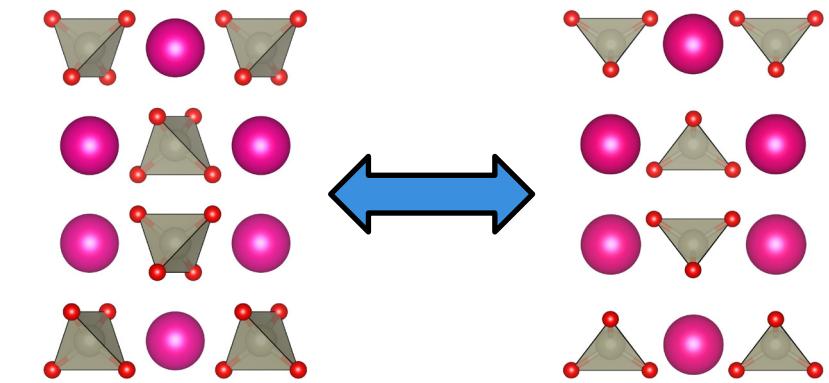
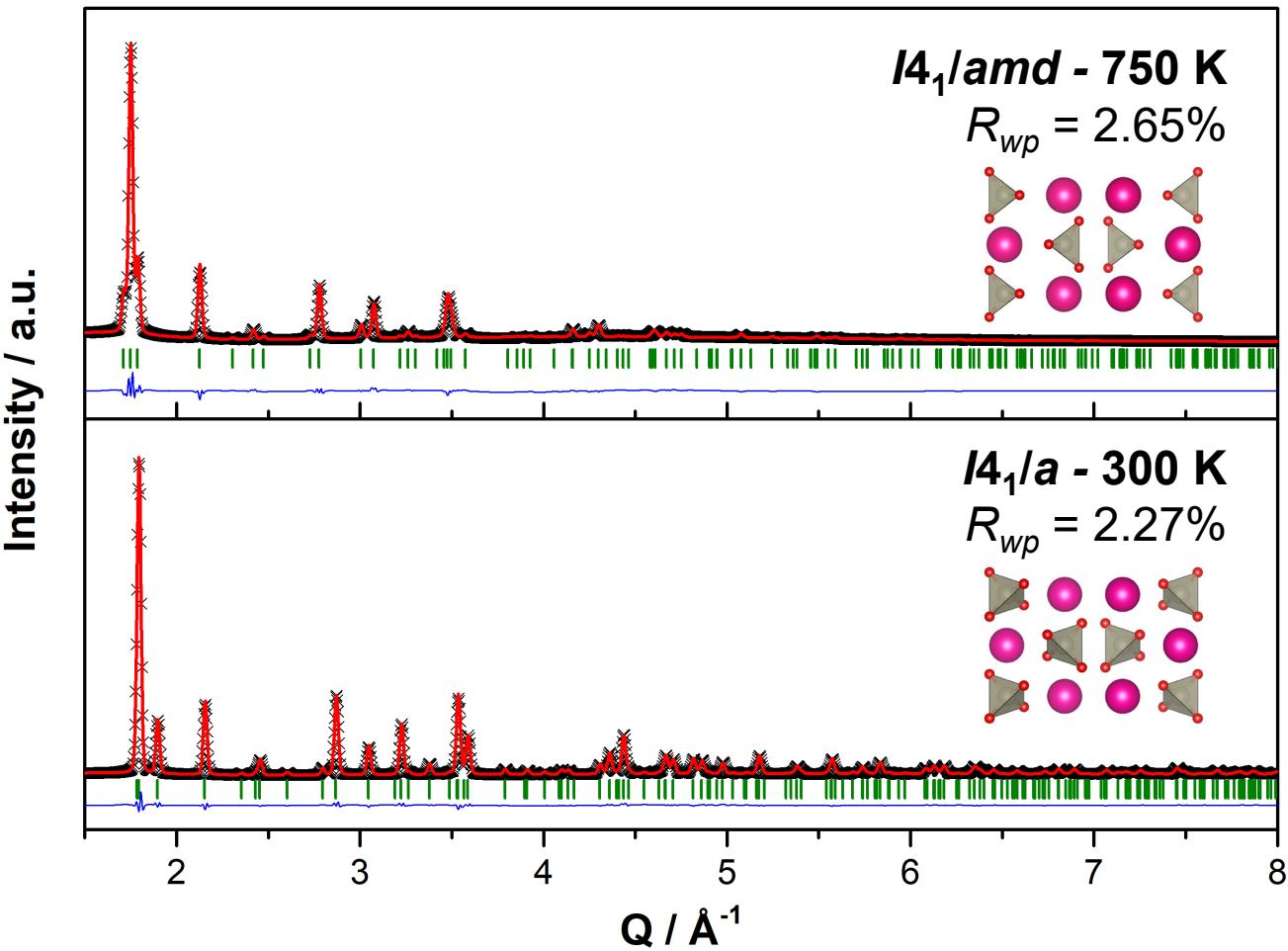




Phase Transition in RbReO₄

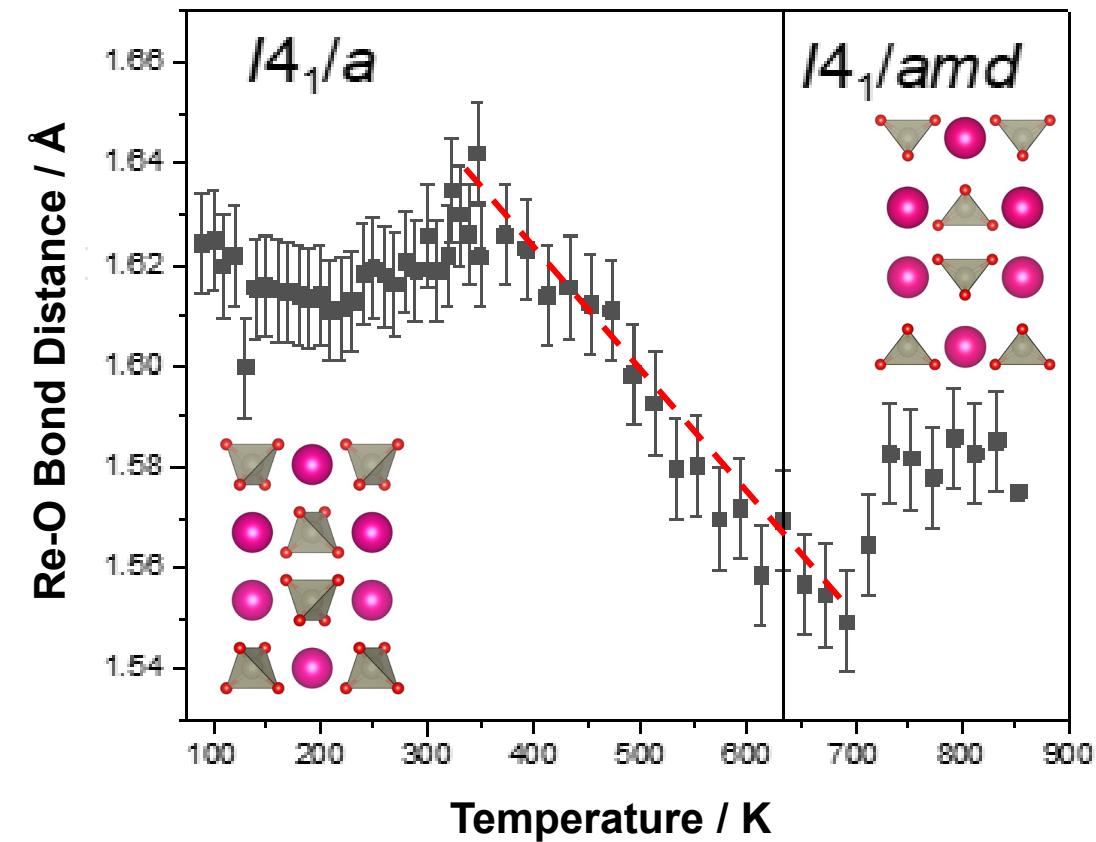
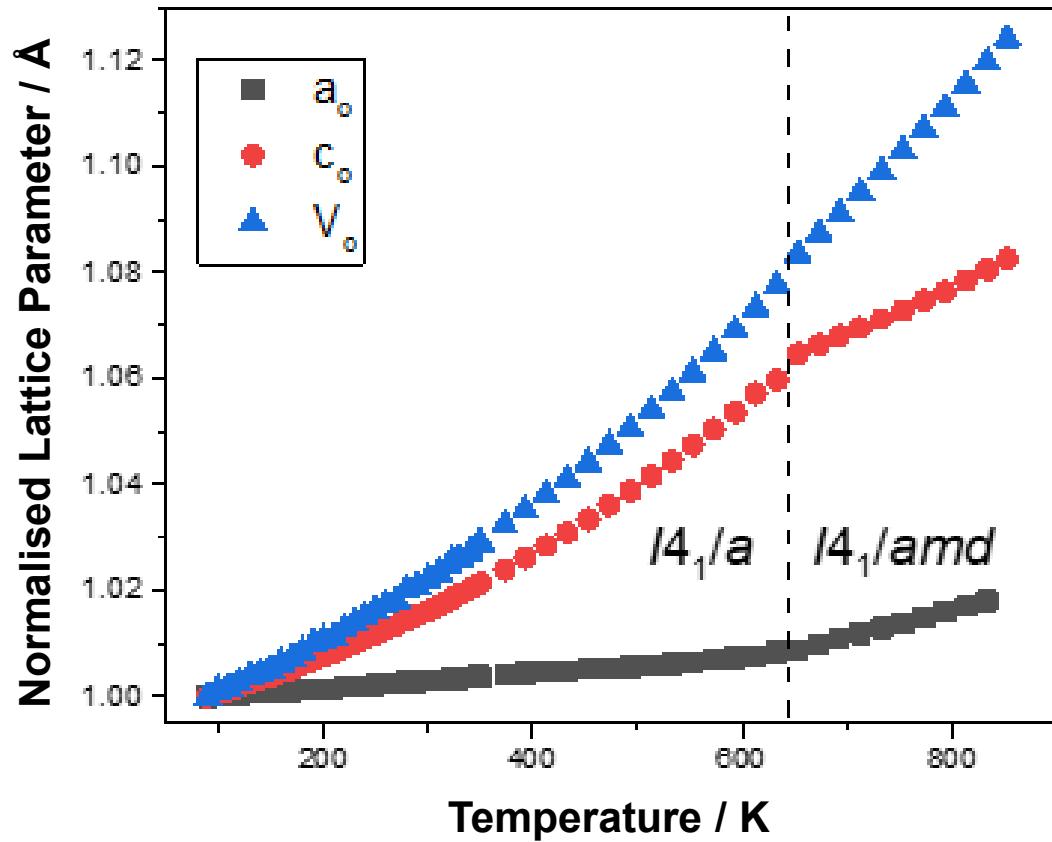


SXRD, $\lambda = 0.16166 \text{ \AA}$



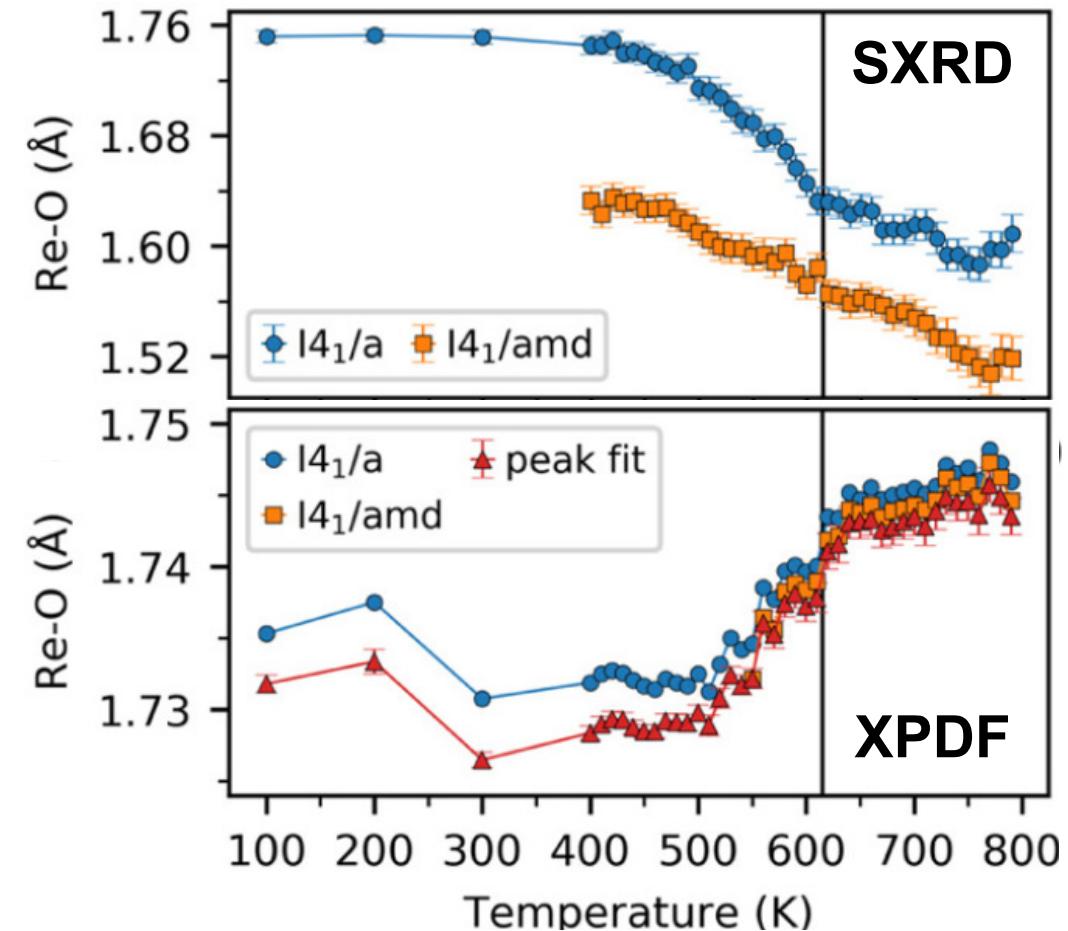
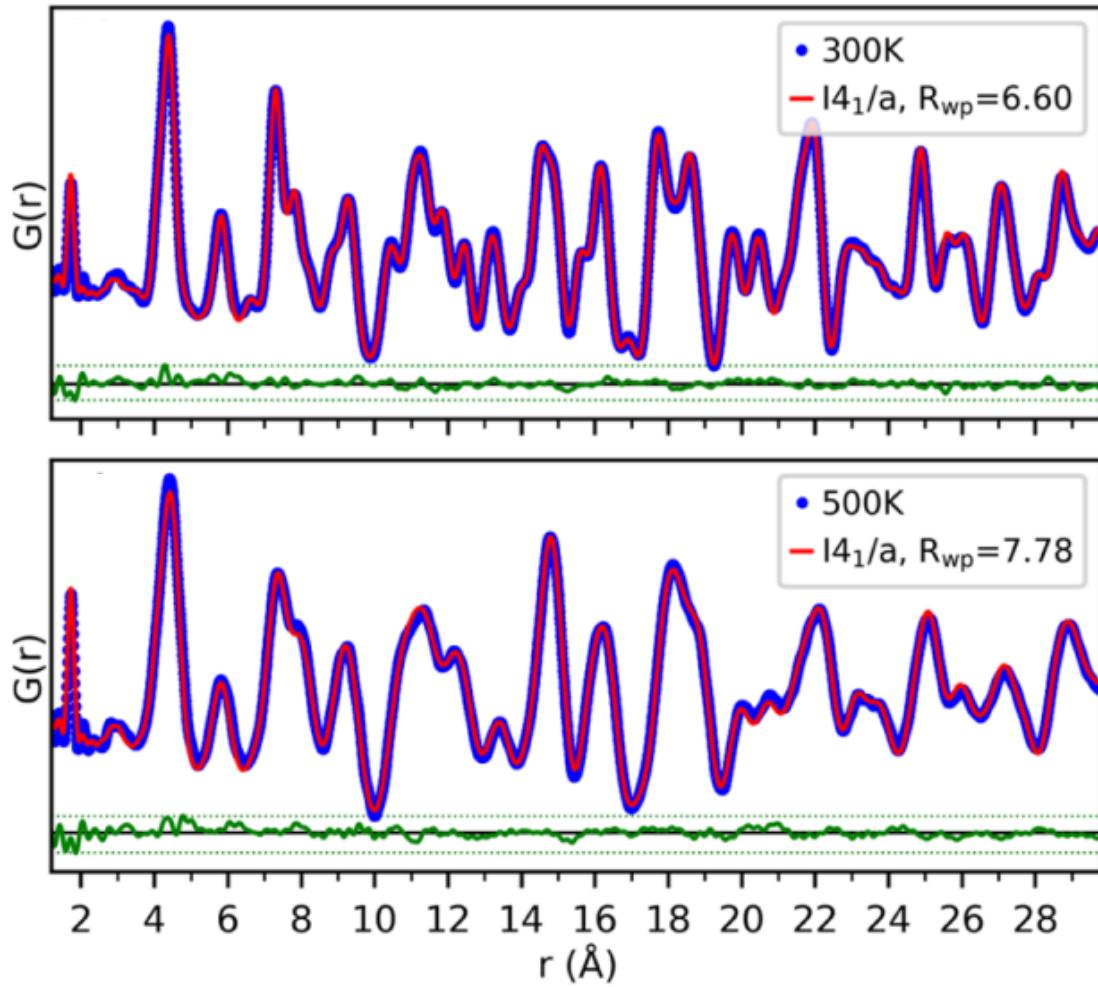


Irregularities in the Long-Range Structure



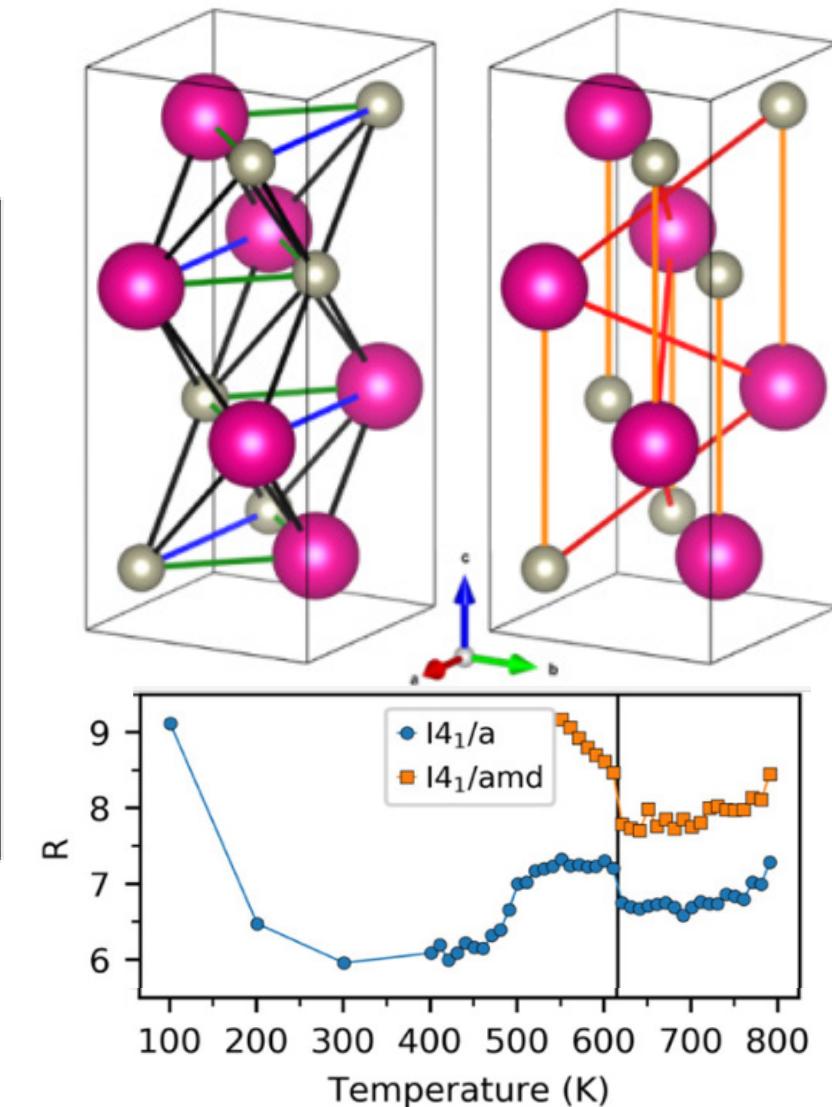
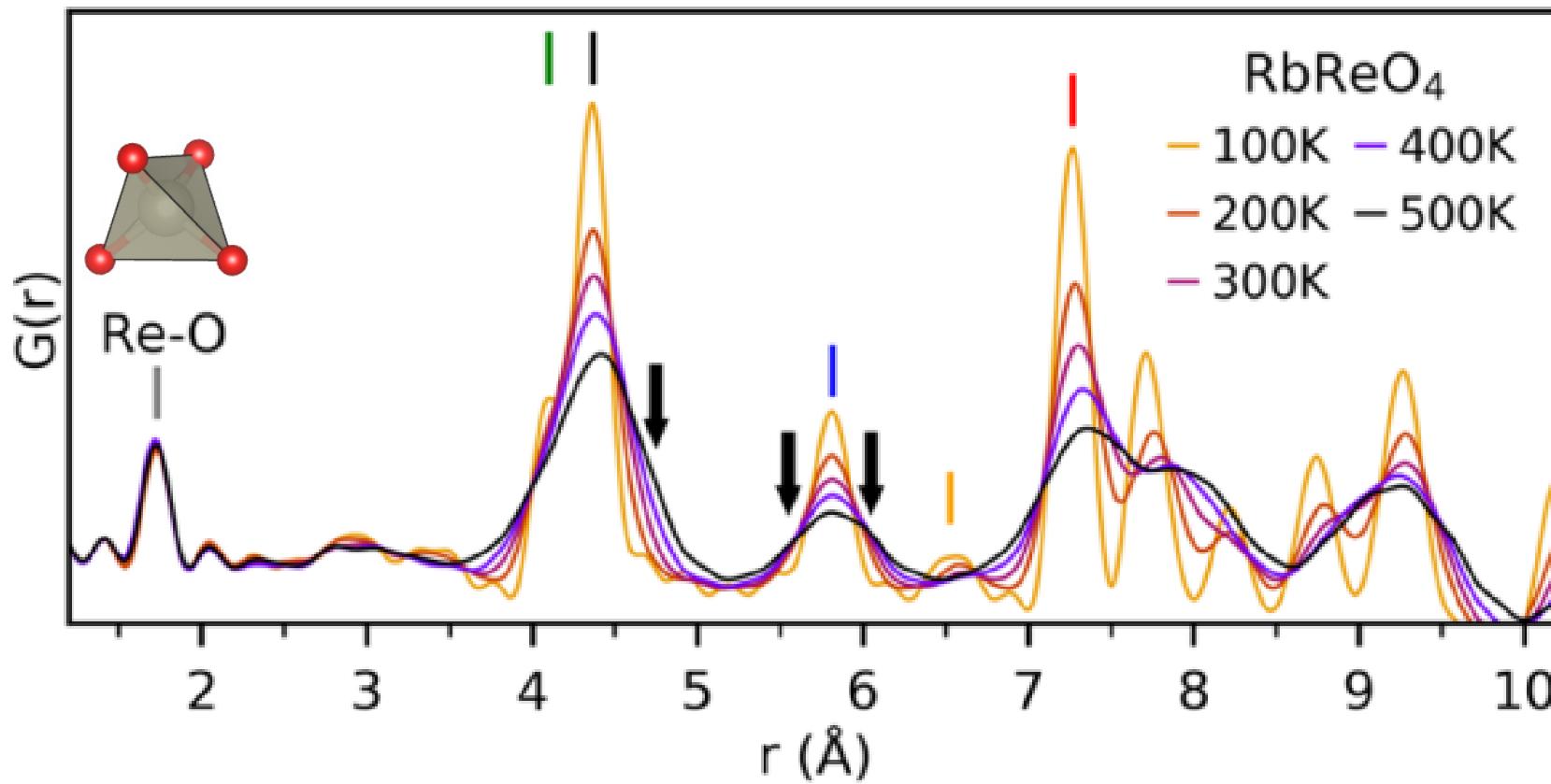


X-Ray Pair Distribution Function Analysis



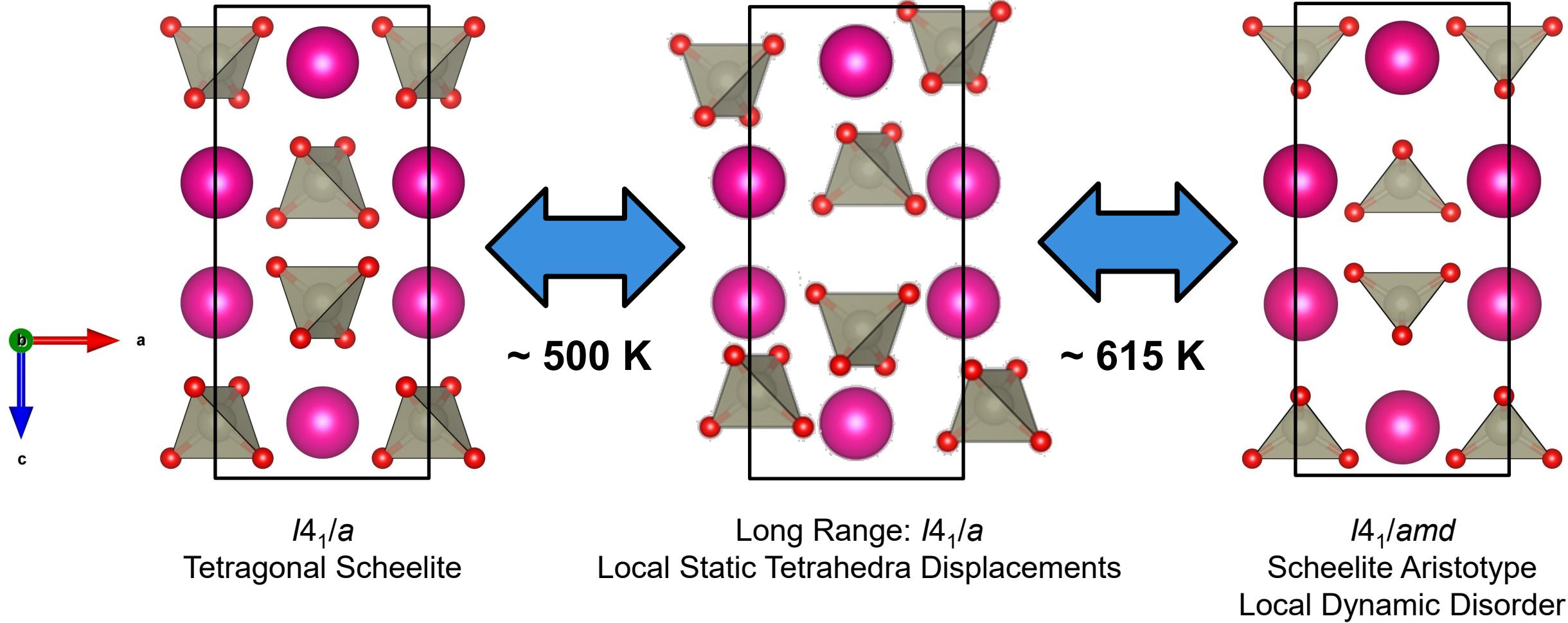


X-Ray Pair Distribution Function Analysis





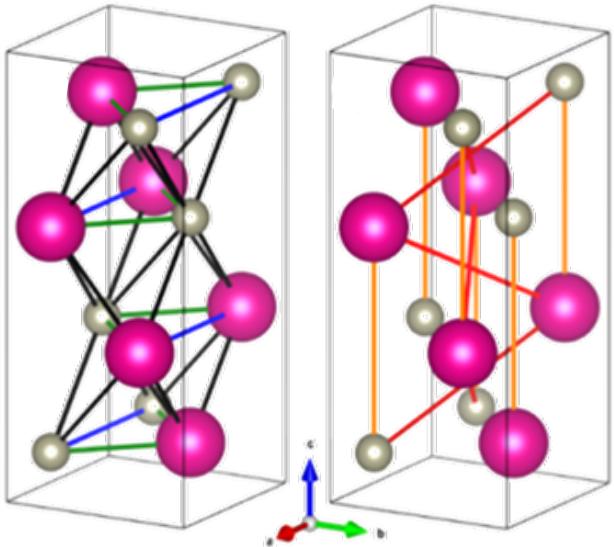
Local Displacements of ReO_4 Tetrahedra





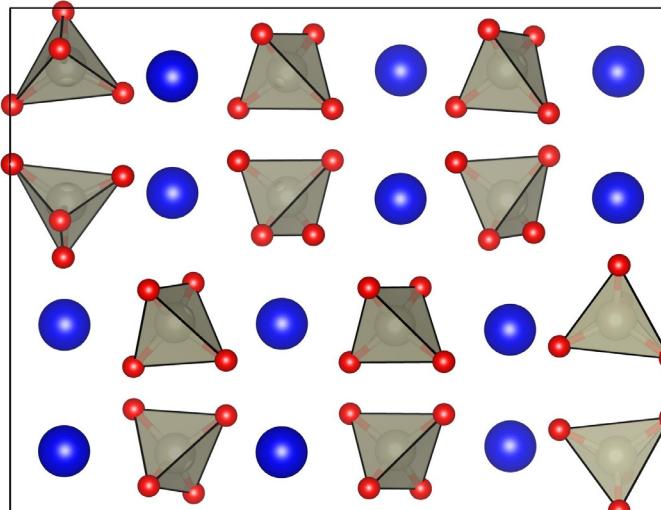
High-Temperature Phase Transition in RbReO₄

Why do the Re-O bonds decrease upon heating?

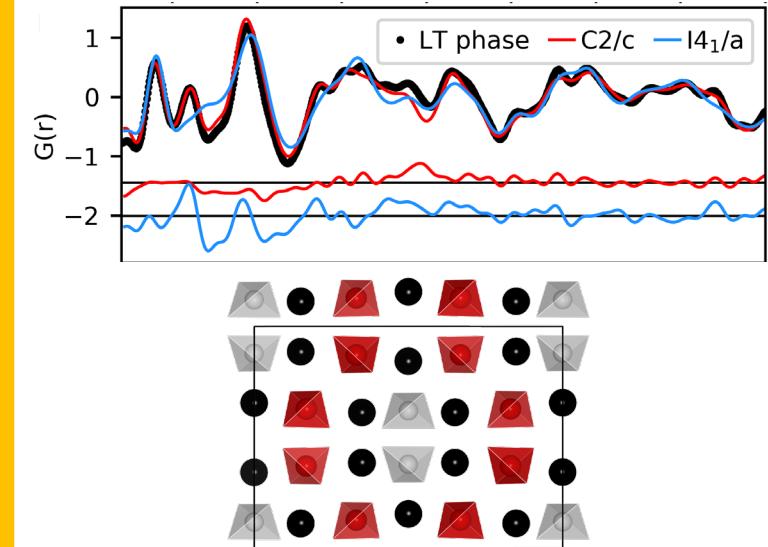


Local displacements of the ReO₄ tetrahedra.

Re-Entrant Phase Transition in TlReO₄



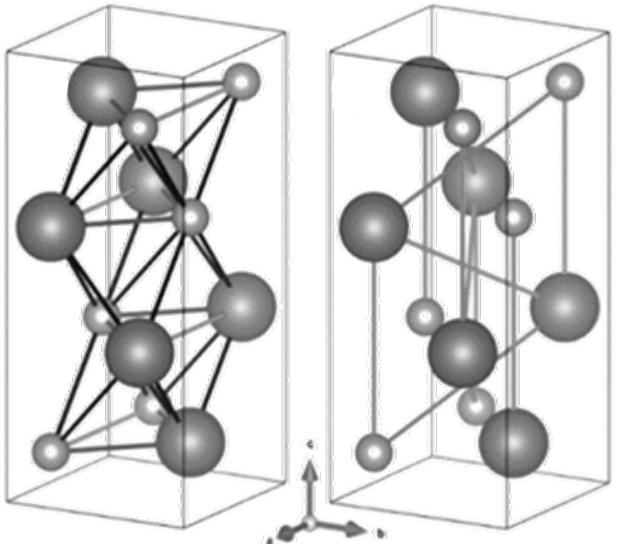
Symmetry-Lowering in Bi₃FeMo₂O₁₂





High-Temperature Phase Transition in RbReO_4

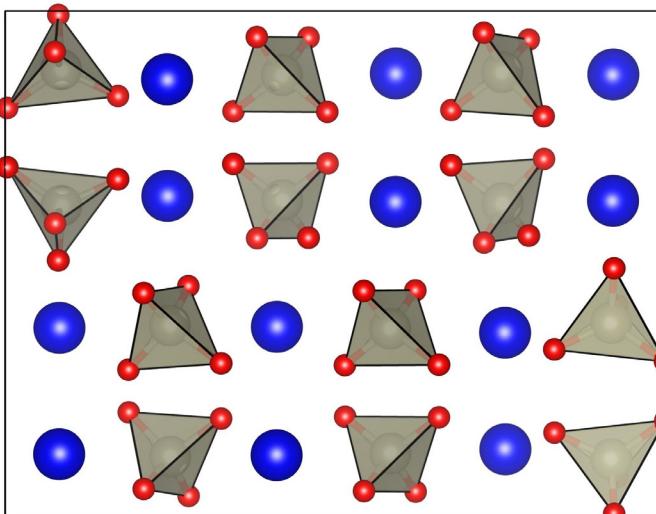
Why do the Re-O bonds decrease upon heating?



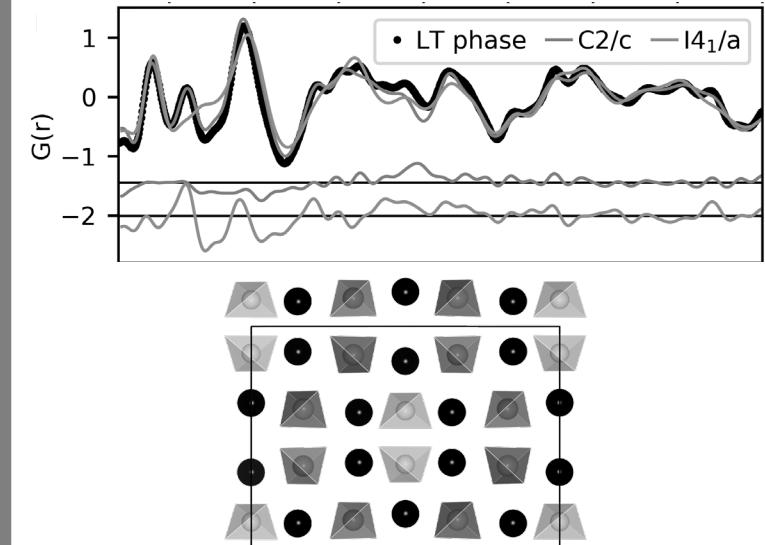
Local displacements of the ReO_4 tetrahedra.

Re-Entrant Phase Transition in TlReO_4

Why do the symmetry lower upon heating?



Symmetry-Lowering in $\text{Bi}_3\text{FeMo}_2\text{O}_{12}$





Structures with Post-Transition Metals

83	208.980
Bi	
Bismuth	
[Xe] 4f ¹⁴ 5d ¹⁰ 6s ² 6p ³	
Post-Transition Metal	

ChemistryTeacher.com



82	207.2
Pb	
Lead	
[Xe] 4f ¹⁴ 5d ¹⁰ 6s ² 6p ²	
Post-Transition Metal	

ChemistryTeacher.com



81	204.383
Tl	
Thallium	
[Xe] 4f ¹⁴ 5d ¹⁰ 6s ² 6p ¹	
Post-Transition Metal	

ChemistryTeacher.com

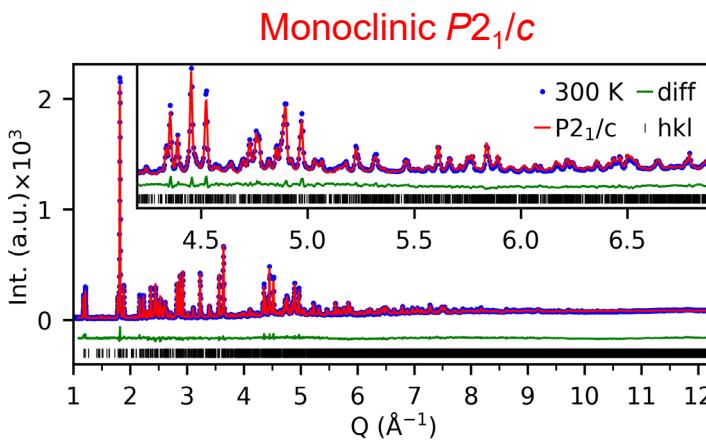


1	H																								2	He
3	Li	4	Be																							
11	Na	12	Mg																							
19	K	20	Ca	21	Sc	22	Ti	23	V	24	Cr	25	Mn	26	Fe	27	Co	28	Ni	29	Cu	30	Zn			
37	Rb	38	Sr	39	Y	40	Zr	41	Nb	42	Mo	43	Tc	44	Ru	45	Rh	46	Pd	47	Ag	48	Cd			
55	Cs	56	Ba	*	71	Lu	72	Hf	73	Ta	74	W	75	Re	76	Os	77	Ir	78	Pt	79	Au	80	Hg	49	In
87	Fr	88	Ra	*	103	Lr	104	Rf	105	Db	106	Sg	107	Bh	108	Hs	109	Mt	110	Ds	111	Rg	112	Cn	50	Sn
*	*																								51	Sb
*	*																								52	Te
*	*																								53	I
*	*																								54	Xe
*	*																								84	Po
*	*																								85	At
*	*																								86	Rn
*	*																								116	Lv
*	*																								117	Ts
*	*																								118	Og
*	*																								67	Ho
*	*																								68	Er
*	*																								69	Tm
*	*																								70	Yb
*	*																								101	Md
*	*																								102	No



Variable Temperature Study of TiReO_4

Long-Range (POWGEN)



Temperature

Tetragonal
 $I4_1/a$

Monoclinic
 $P2_1/c$

Tetragonal
 $I4_1/a$

500 K

300 K

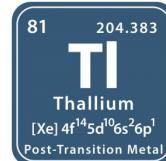
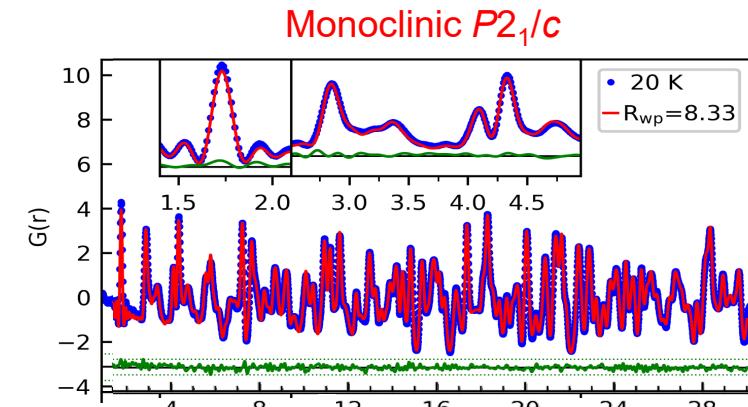
20 K

Monoclinic
 $P2_1/c$

Monoclinic
 $P2_1/c$

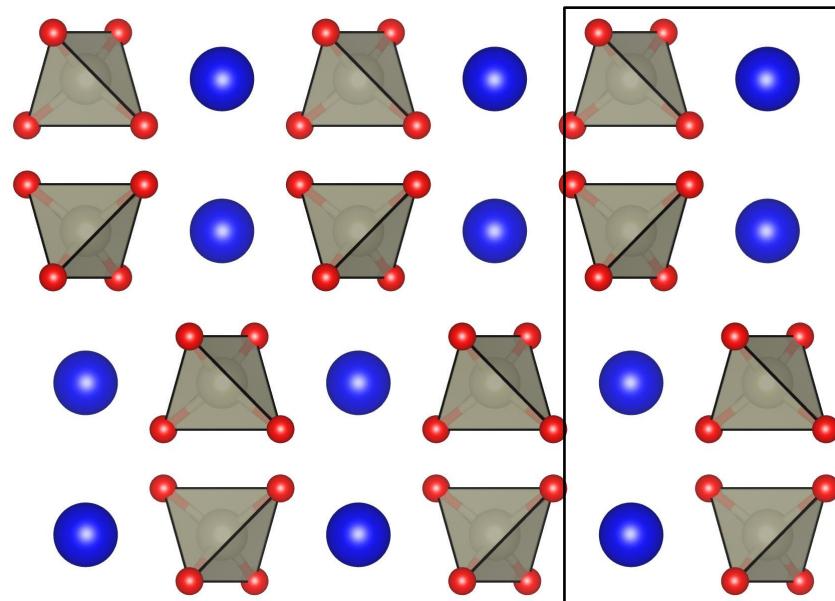
Monoclinic
 $P2_1/c$

Short-Range (NOMAD)

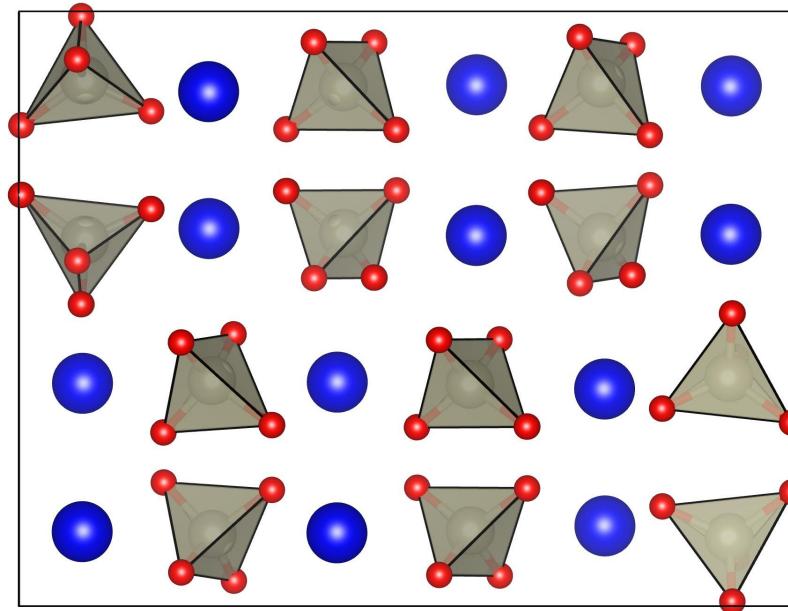




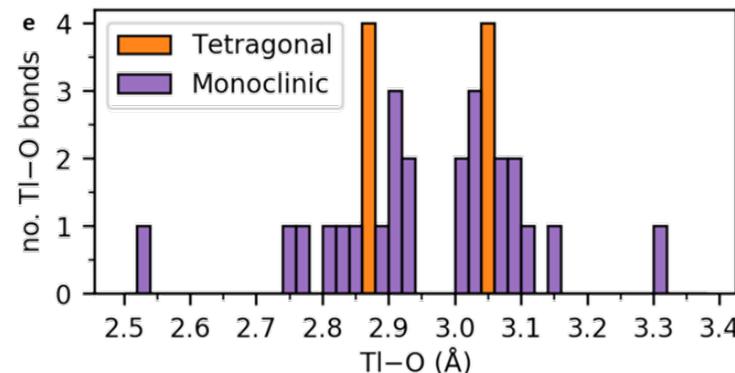
The Hidden Structure of TiReO_4



Tetragonal Long-Range
 $I4_1/a$



Monoclinic Short-Range
 $P2_1/c$



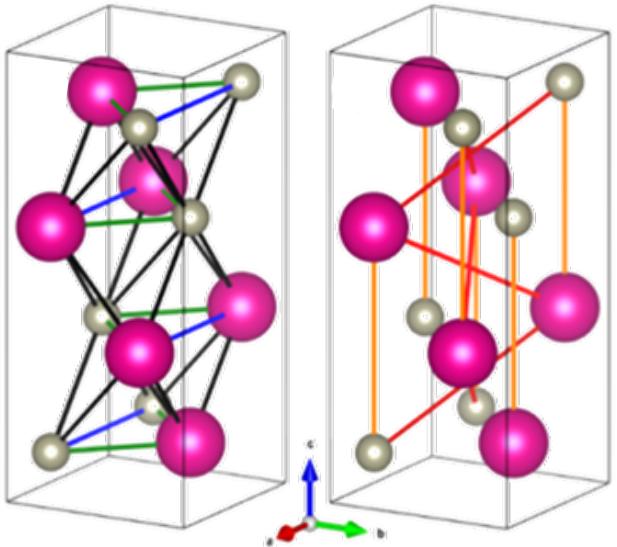
Dr Matilde Saura Muzquiz





High-Temperature Phase Transition in RbReO₄

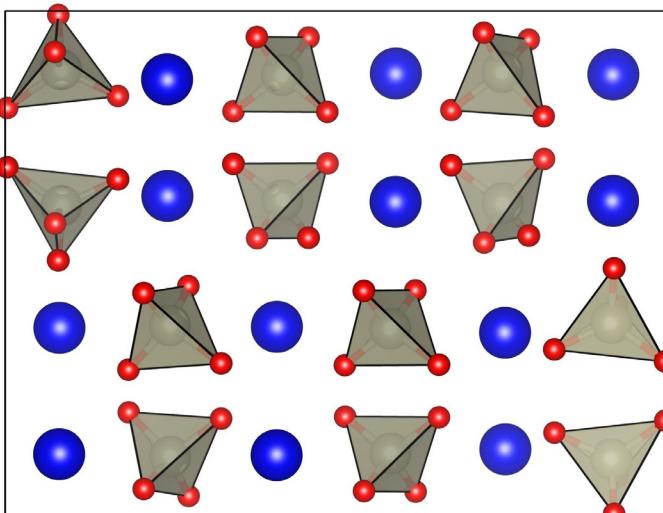
Why do the Re-O bonds decrease upon heating?



Local displacements of the ReO₄ tetrahedra.

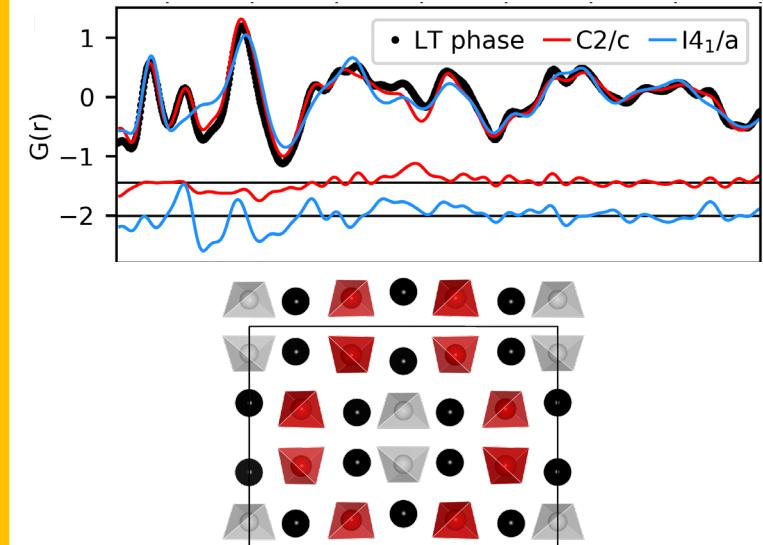
Re-Entrant Phase Transition in TiReO₄

Why do the symmetry lower upon heating?



TiReO₄ is locally monoclinic.

Symmetry-Lowering in Bi₃FeMo₂O₁₂

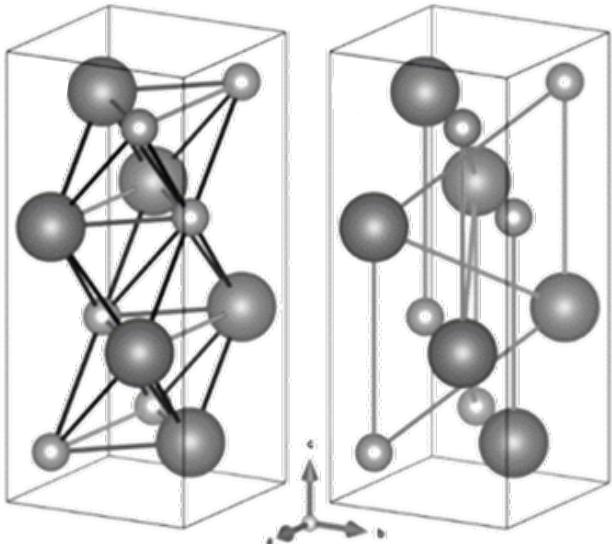




Today's Outline

High-Temperature Phase Transition in RbReO₄

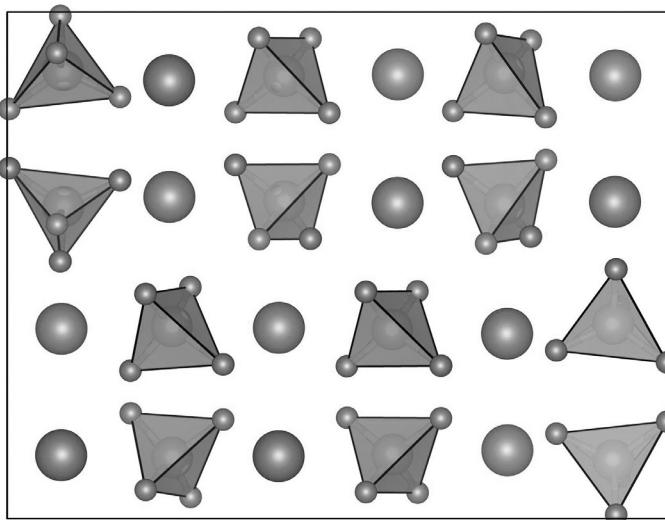
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Local displacements of the ReO₄ tetrahedra.

Re-Entrant Phase Transition in TiReO₄

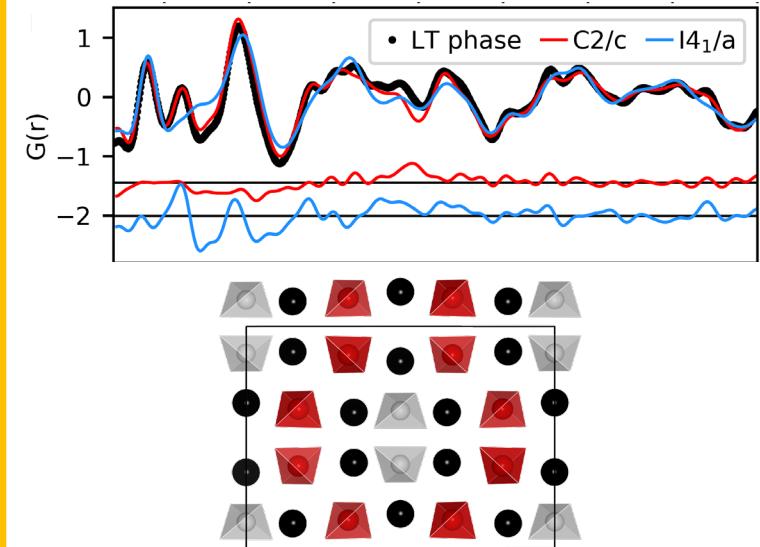
Why do the symmetry lower upon heating?



TiReO₄ is locally monoclinic.

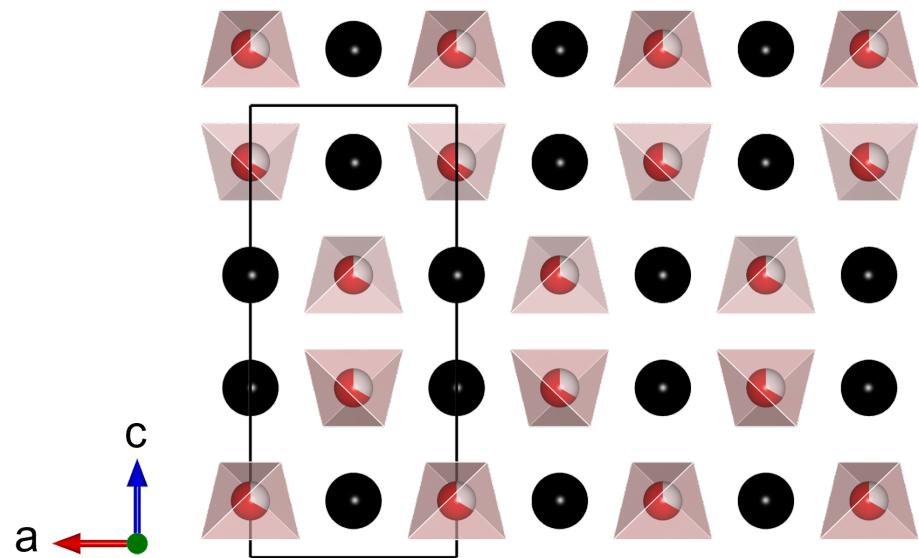
Symmetry-Lowering in Bi₃FeMo₂O₁₂

Do the cations order upon heating?





Phase Transformation in $\text{Bi}_3\text{FeMo}_2\text{O}_{12}$



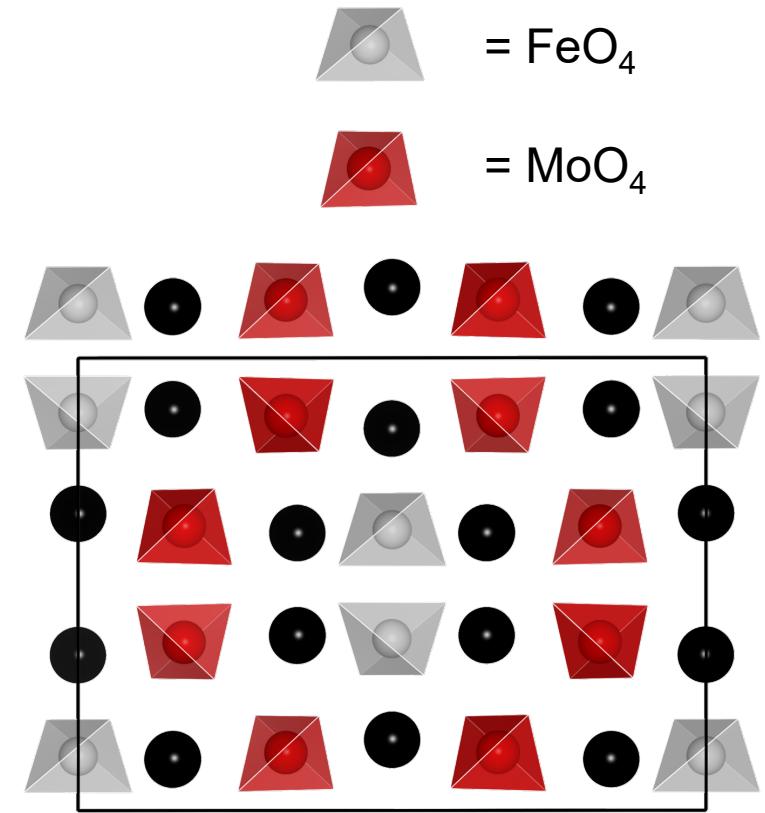
$I4_1/a$
Disordered Tetragonal Scheelite

Amorphisation?



$\sim 500 \text{ }^\circ\text{C}$

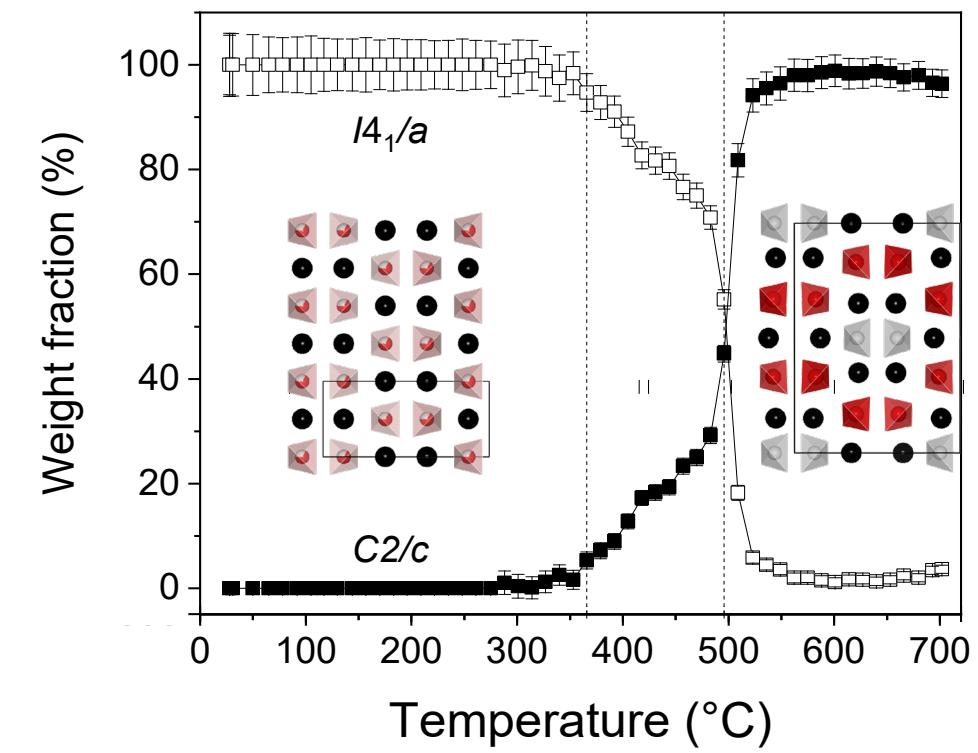
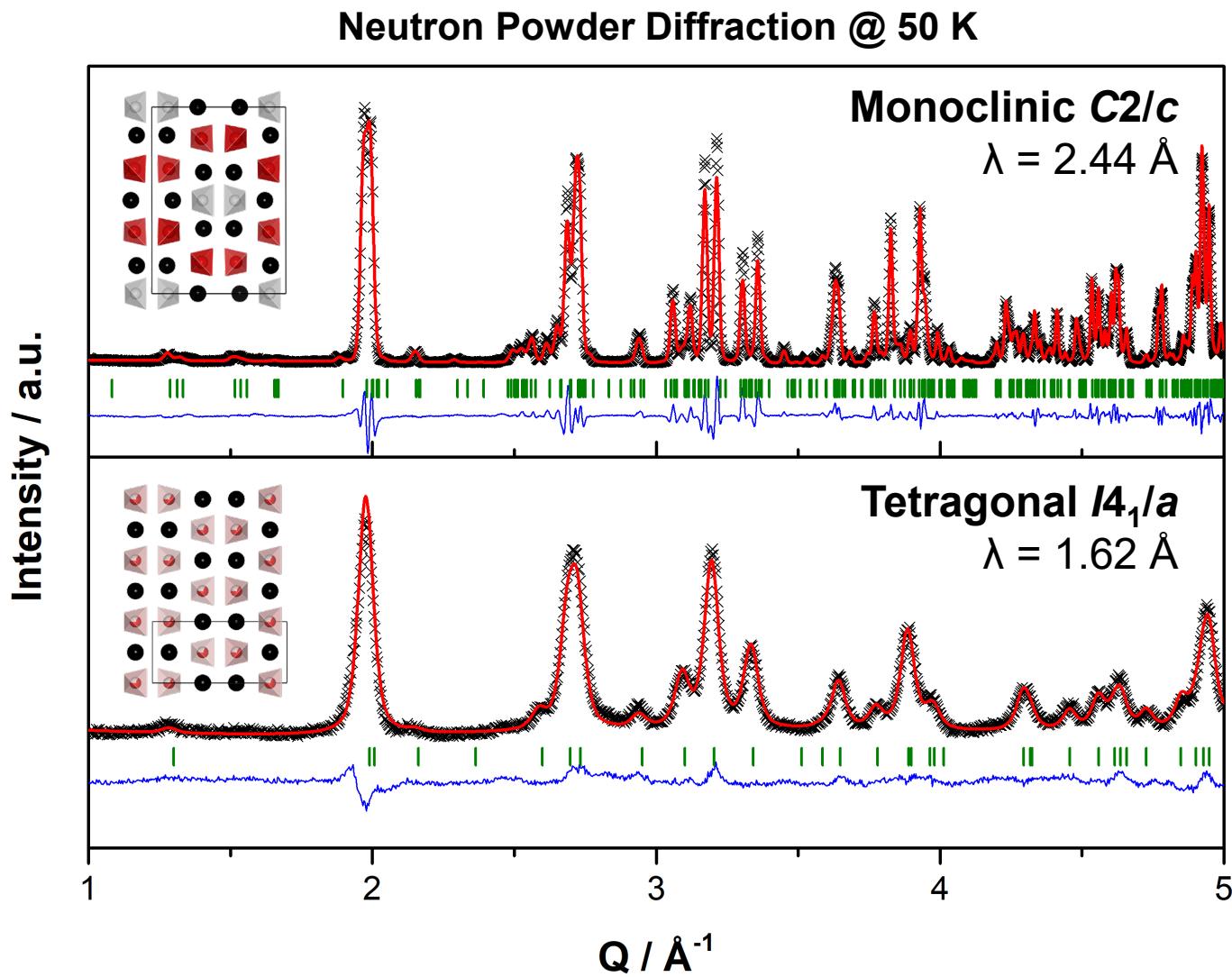
a ← → b

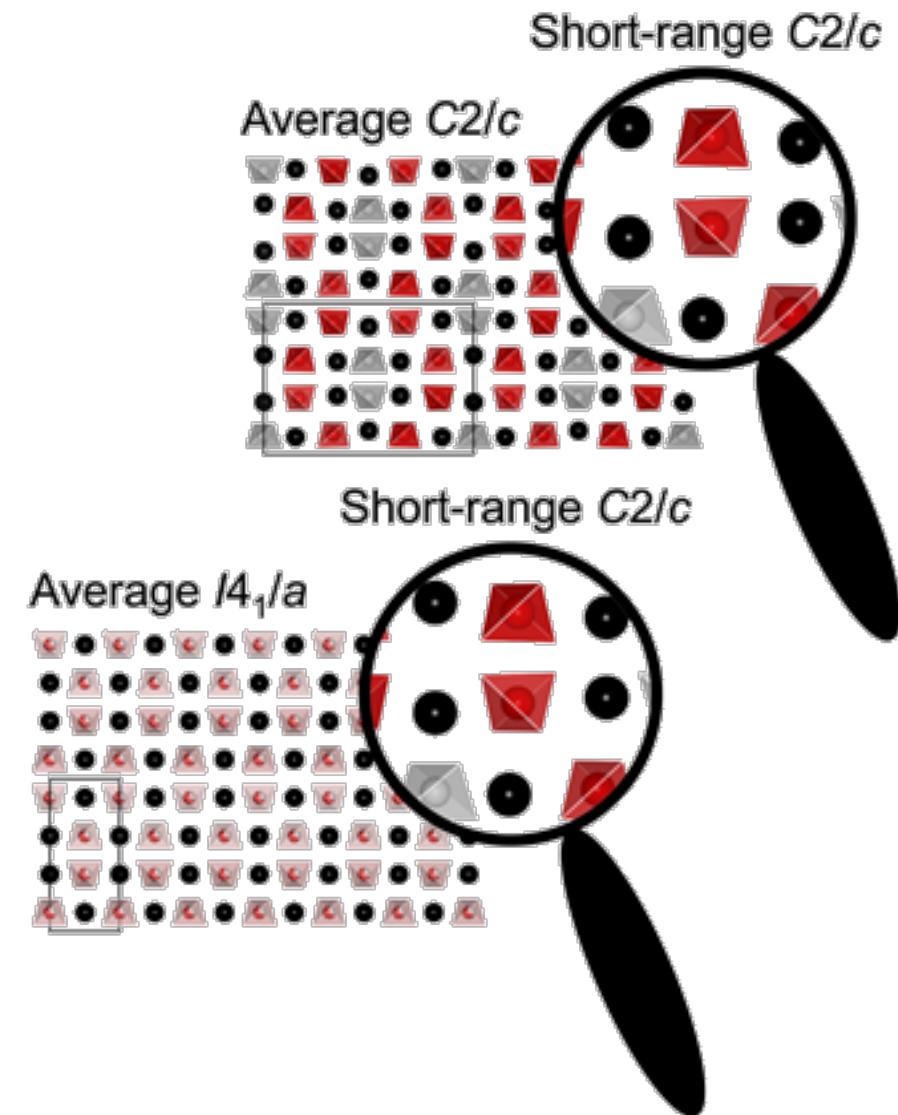
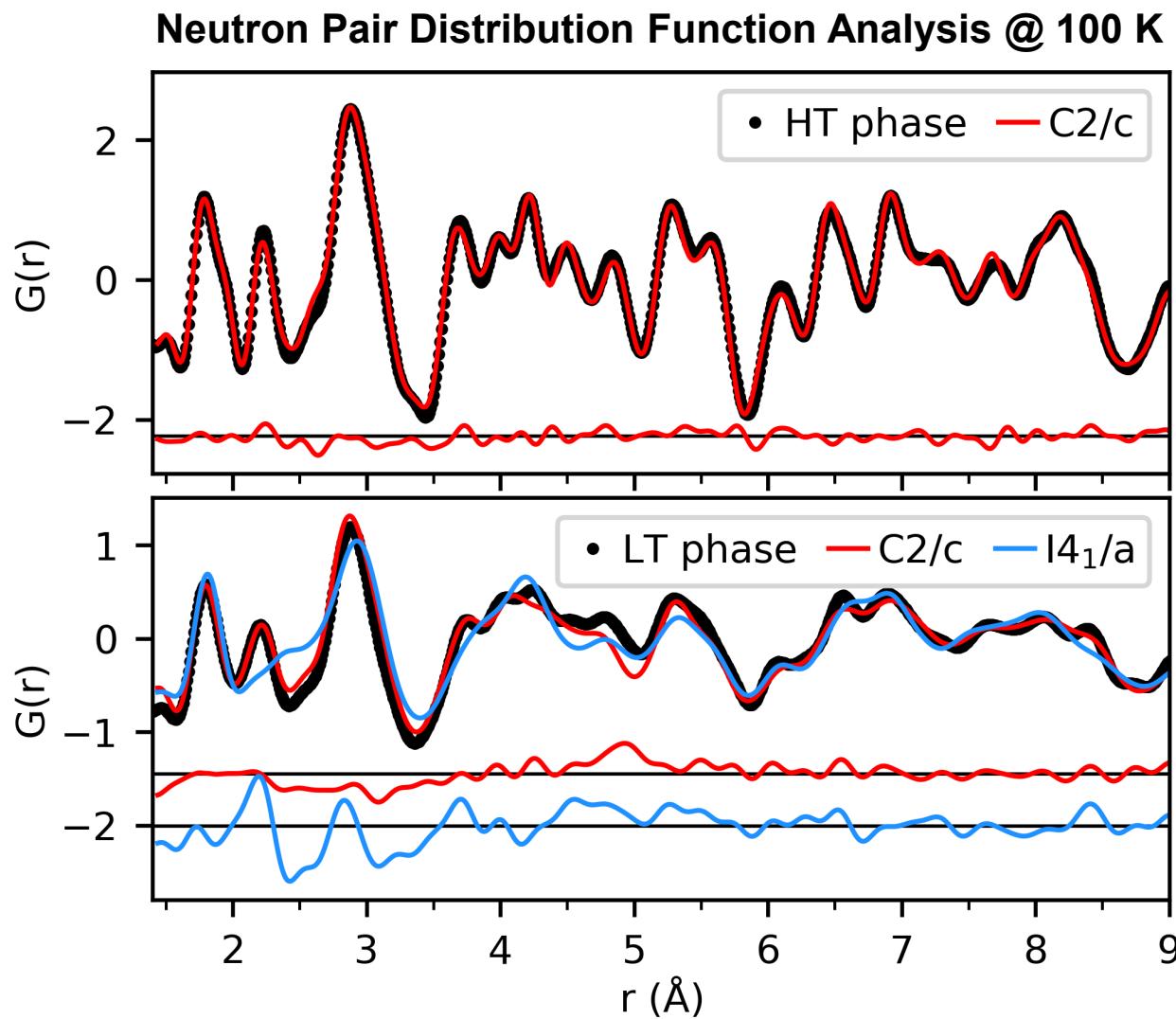


$C2/c$
2:1 Ordered Monoclinic Fergusonite



Variable Temperature Neutron Powder Diffraction



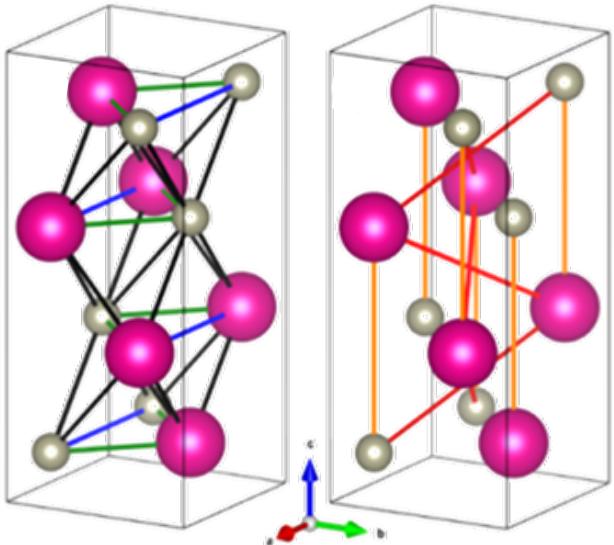




Today's Outline

High-Temperature Phase Transition in RbReO₄

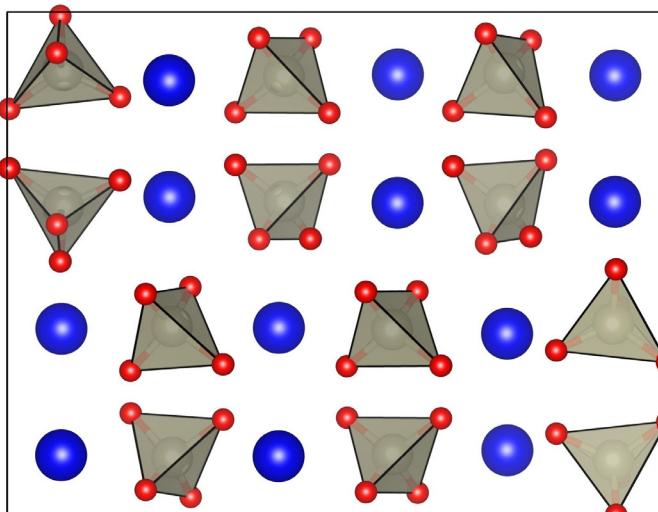
Why do the Re-O bonds decrease upon heating?



Local displacements of the ReO₄ tetrahedra.

Re-Entrant Phase Transition in TiReO₄

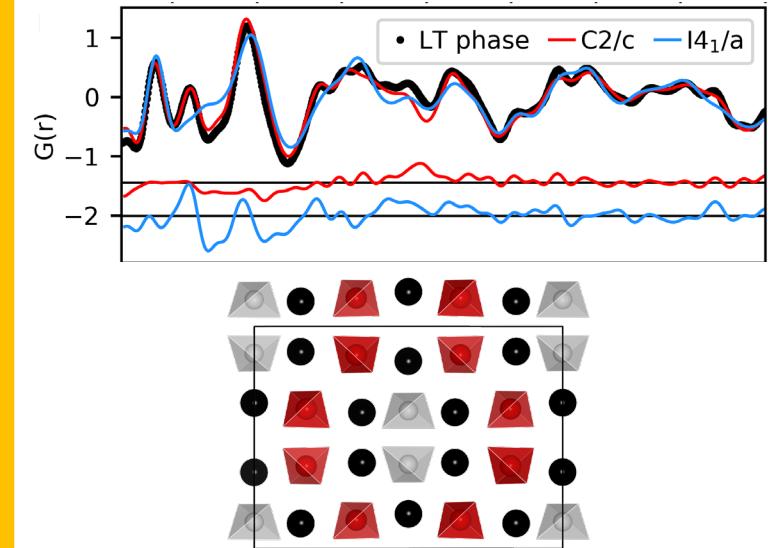
Why do the symmetry lower upon heating?



TiReO₄ is locally monoclinic.

Symmetry-Lowering in Bi₃FeMo₂O₁₂

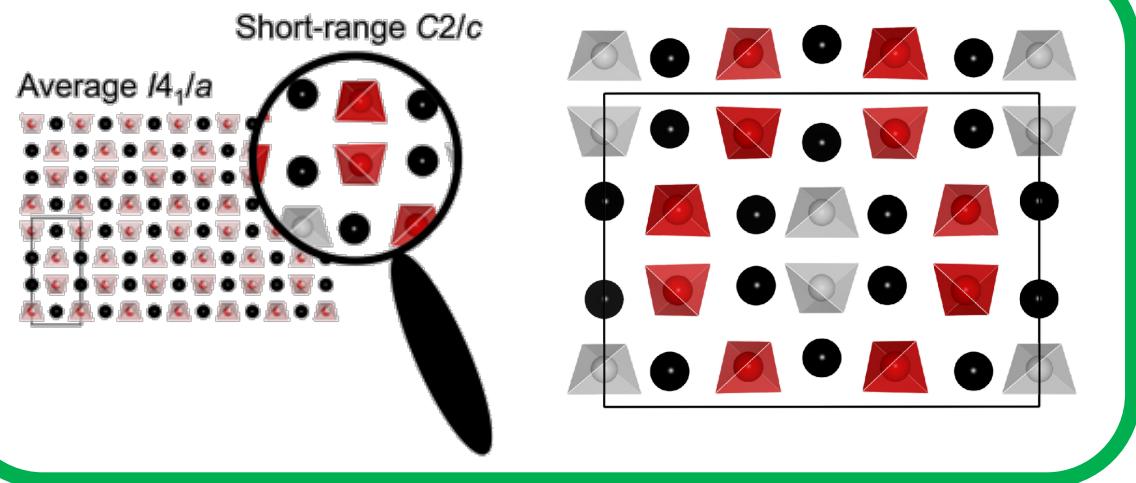
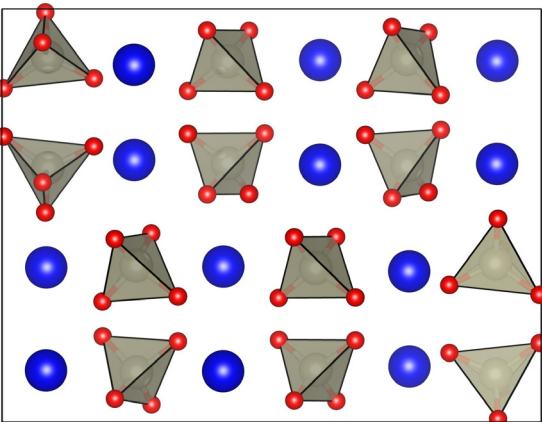
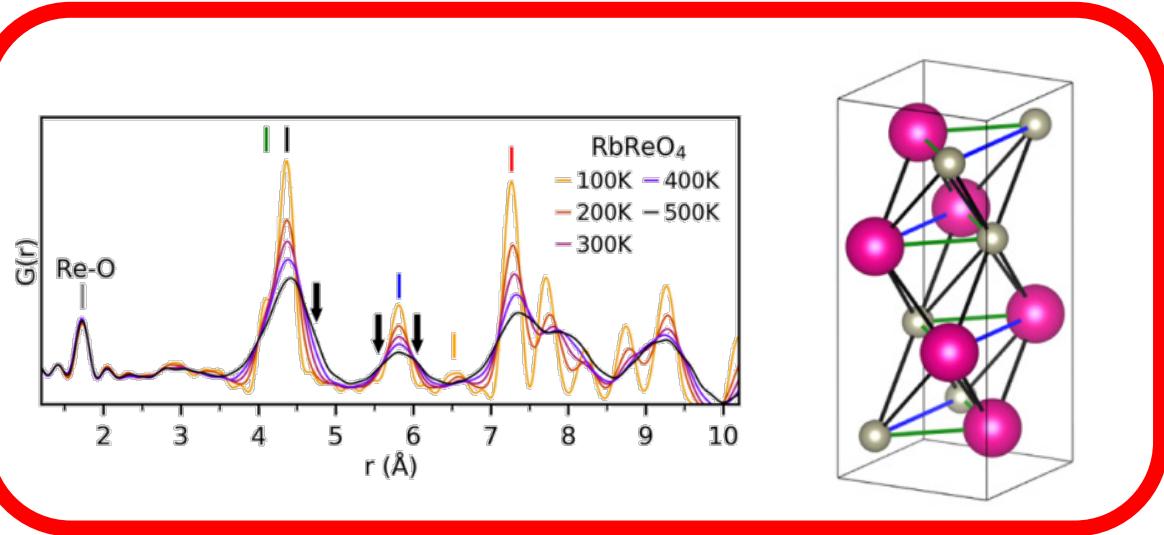
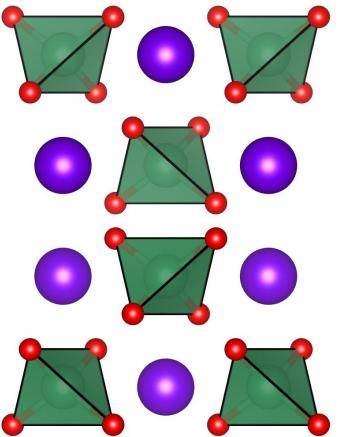
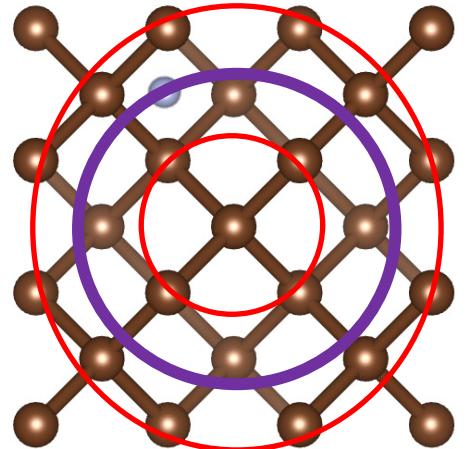
Do the cations order upon heating?



Difference in tetrahedra environments.



Conclusions





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