

From Total Scattering to Total Understanding: Structure-Functionality Relationships in Scheelite-Type Oxides

Presented by:

Mr Bryce Mullens

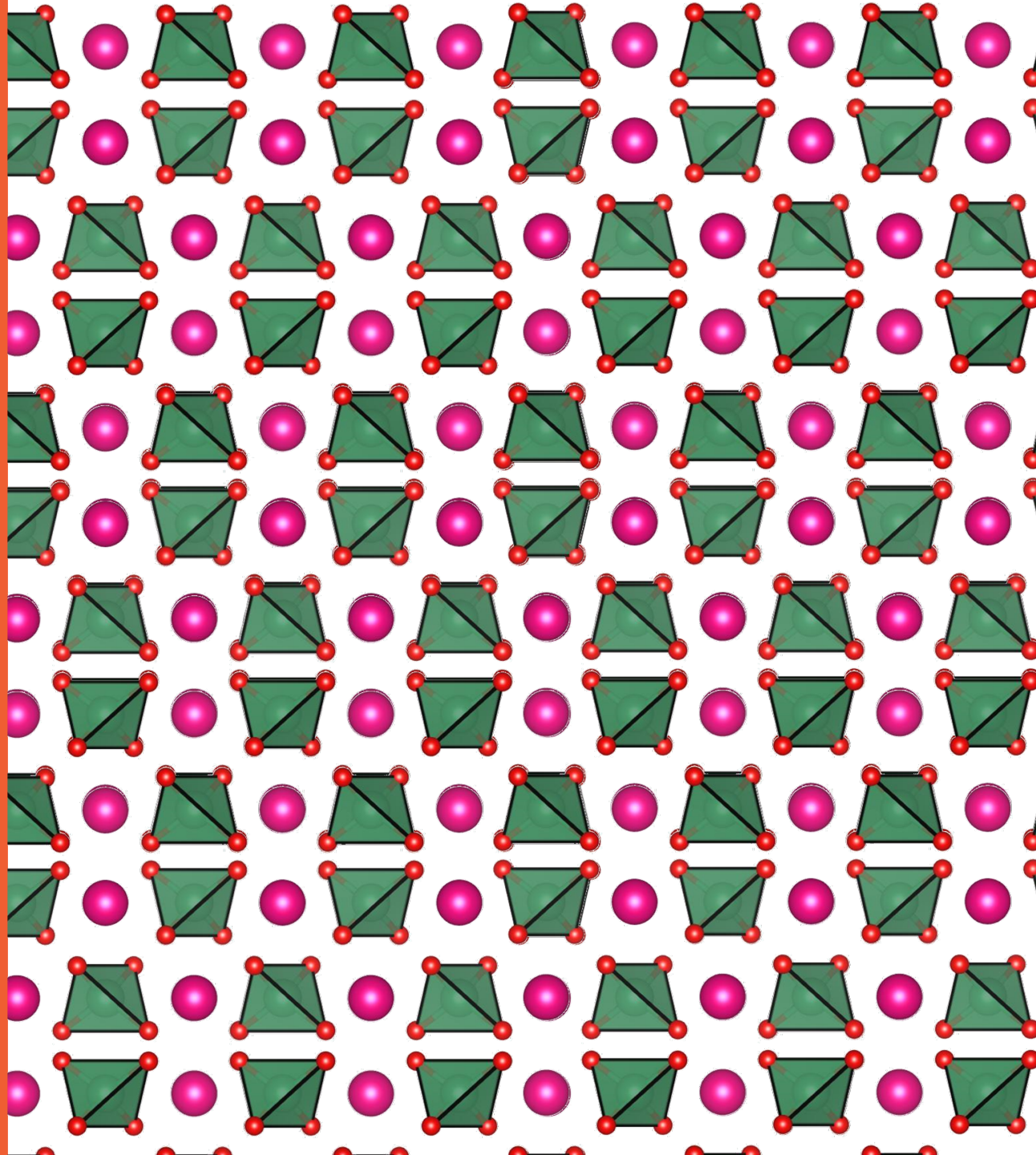
University of Sydney / Université Paris-Saclay

PhD Candidate in Solid-State / Materials Chemistry

ADD2022 – Wednesday 19th October 2022



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PARIS-SACLAY



Supervisors

- › Prof Brendan J. Kennedy
- › Prof Chris Ling
- › Dr Zhaoming Zhang
- › Prof Gianguido Baldinozzi
- › Prof Karena Chapman
- › Prof Maik Lang

Postdocs

- › Dr Matilde Saura Múzquiz
- › Dr Fred Marlton



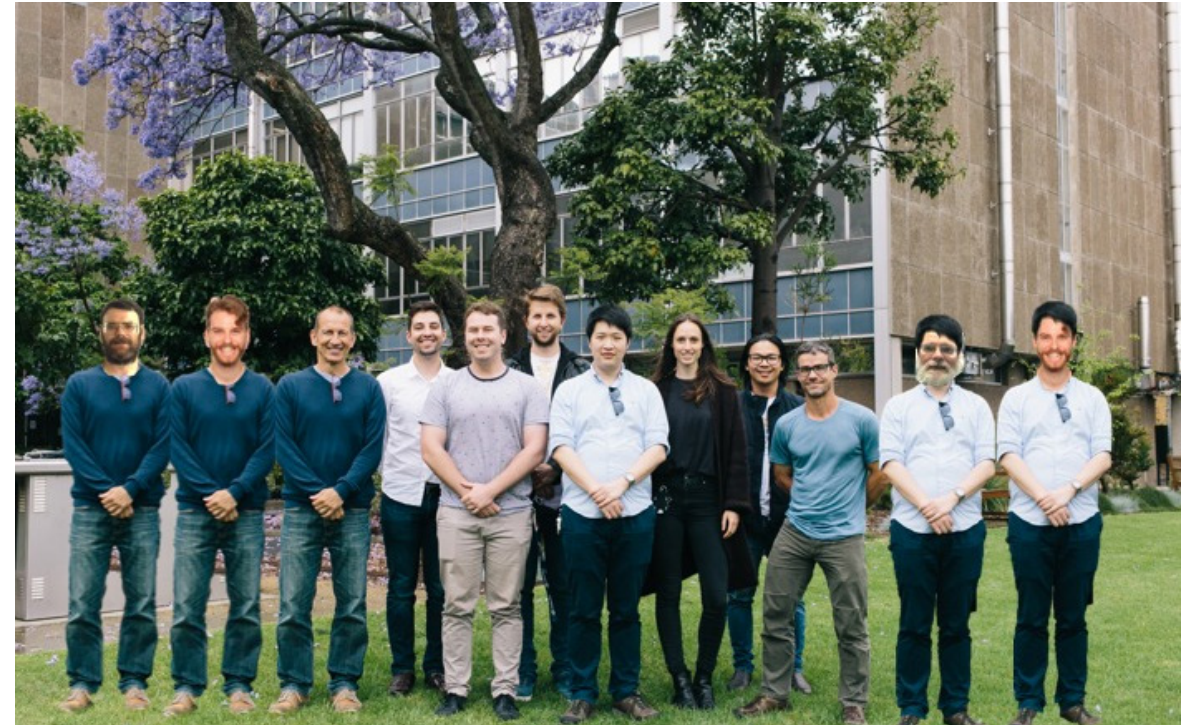
Oak Ridge National Laboratory

- › Dr Alicia Manjon-Sanz
- › Dr Joerg Neuefeind
- › Dr Michelle Everett

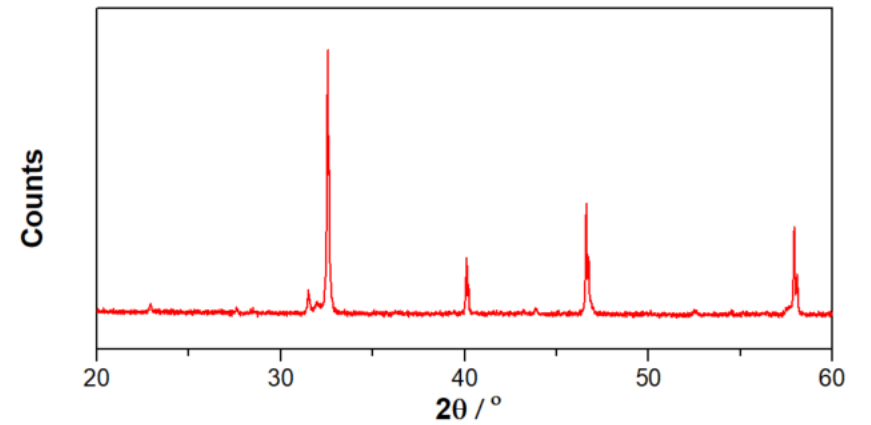
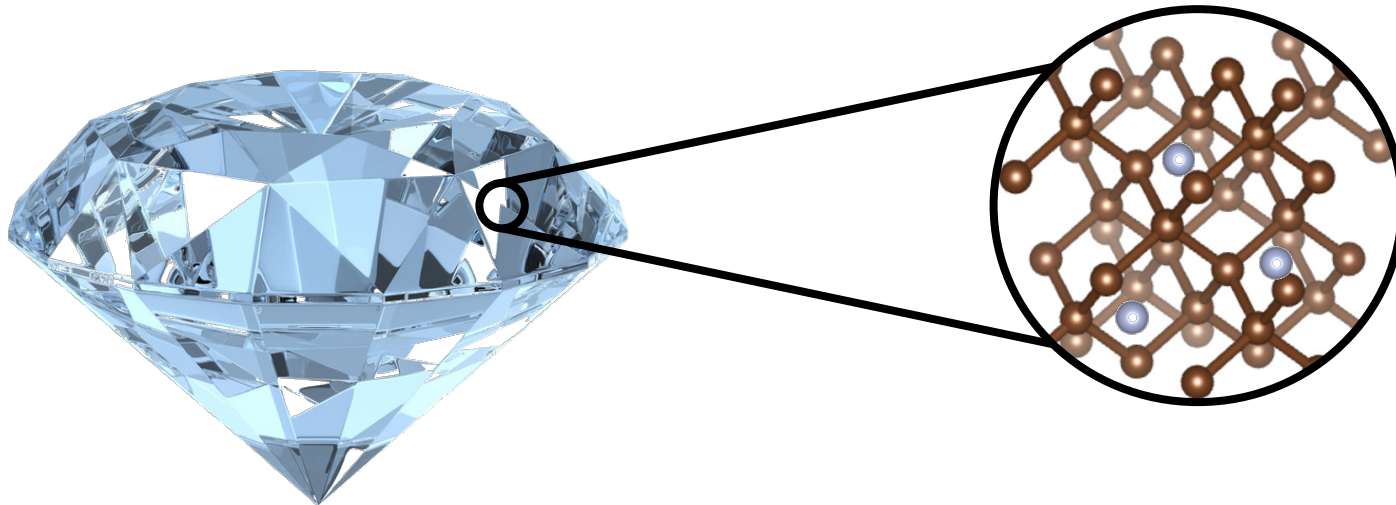
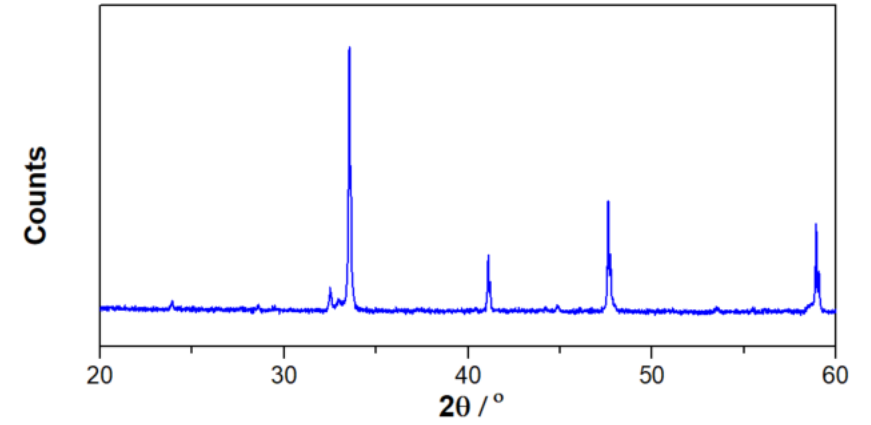
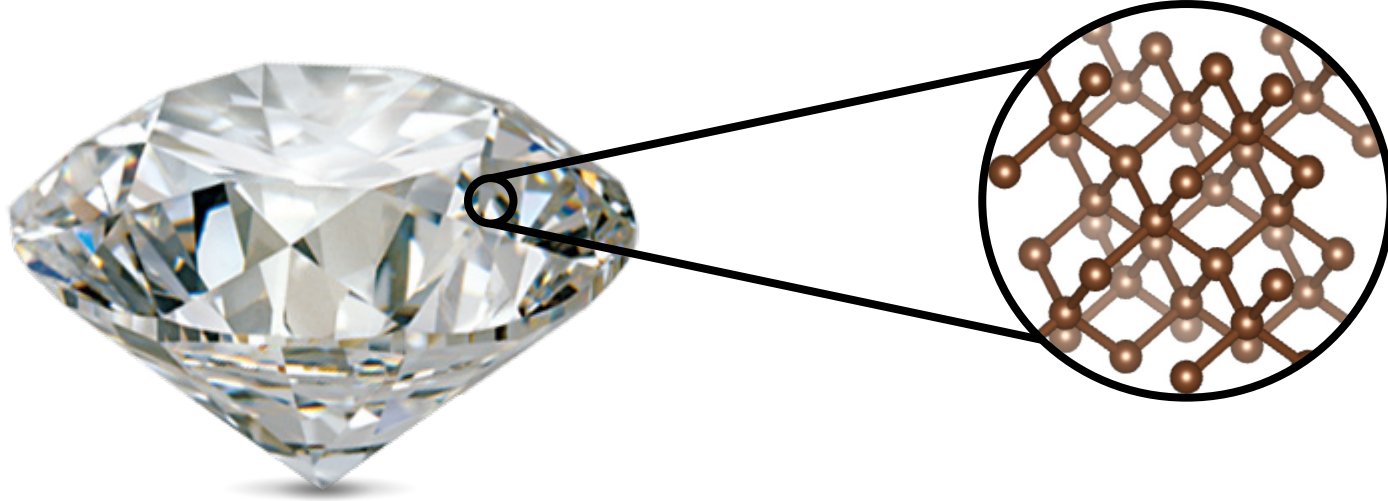
Diamond Light Source

- › Dr Phil Chater

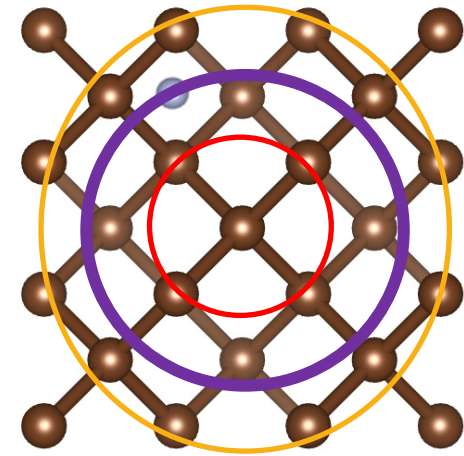
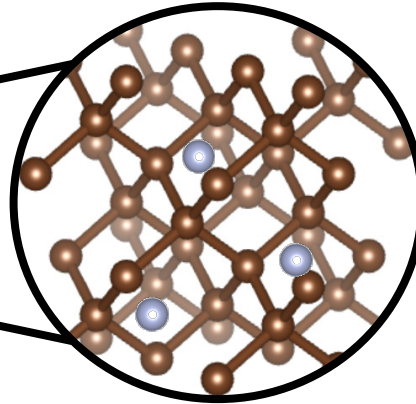
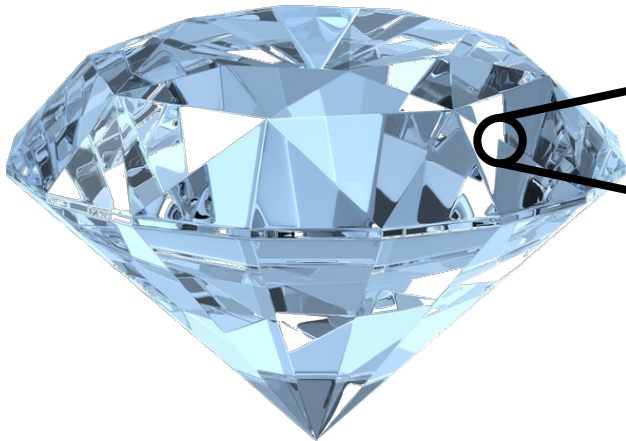
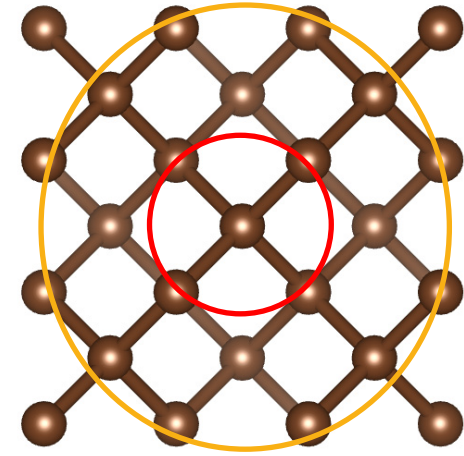
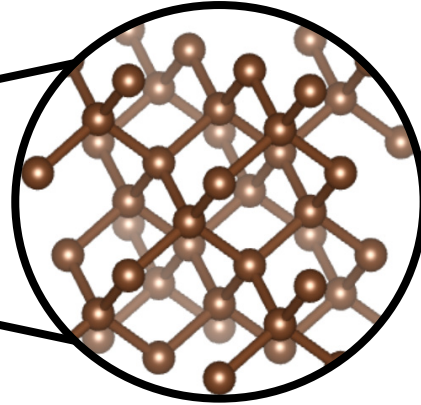
The University of Sydney Solid-State Research Group



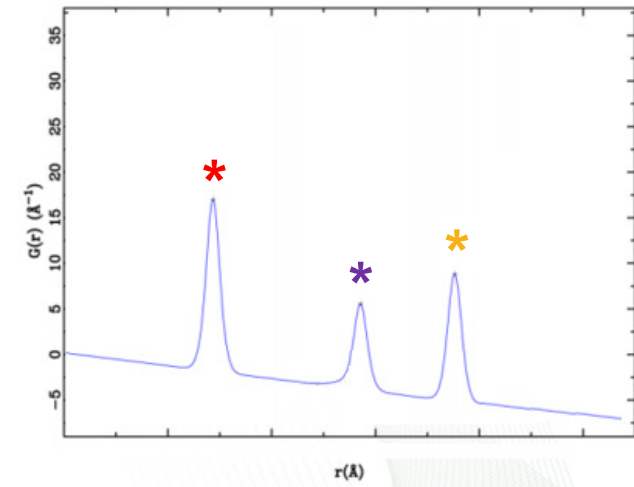
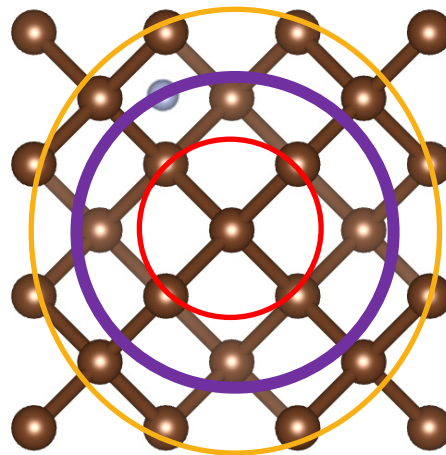
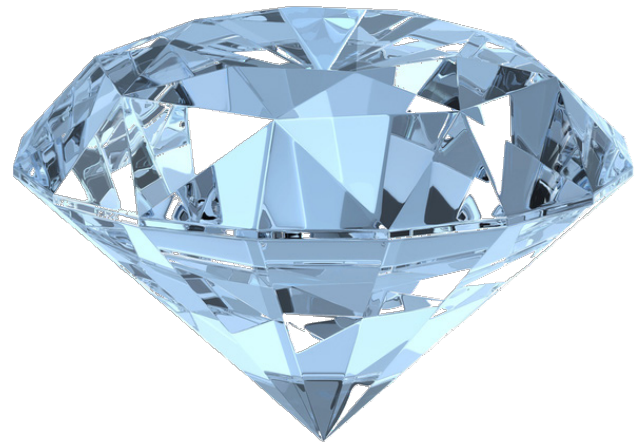
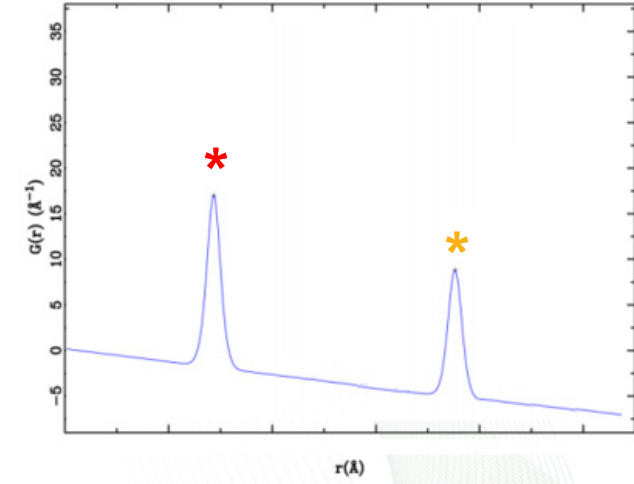
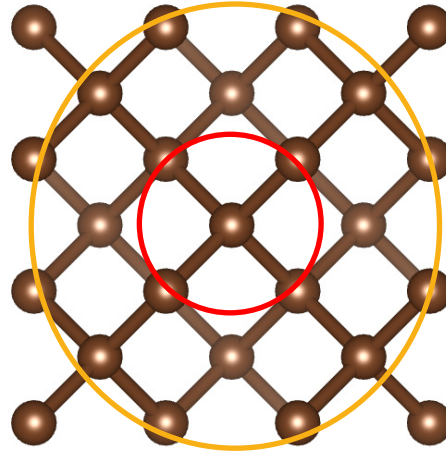
Structures and Properties of Diamond



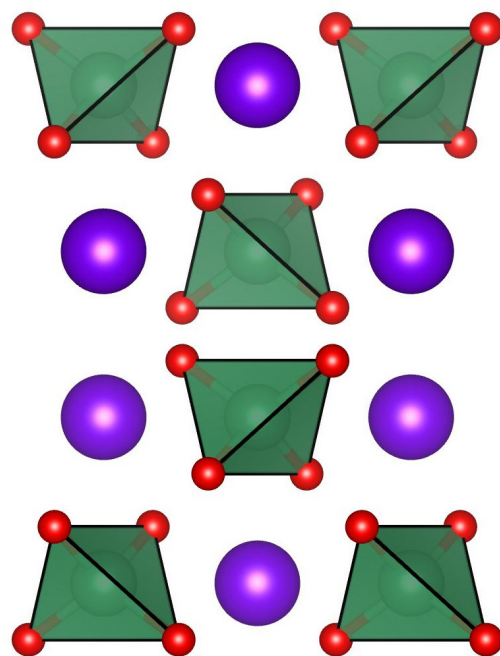
Structures and Properties of Diamond



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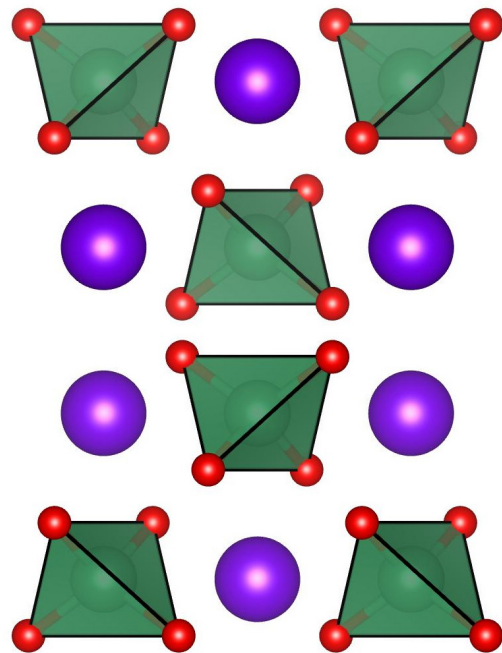
The Wonderful ABO_4 Structure



ABO_4
Tetragonal Scheelite

1 H																	2 He
3 Li	4 Be											5 B	6 C	7 N	8 O	9 F	10 Ne
11 Na	12 Mg											13 Al	14 Si	15 P	16 S	17 Cl	18 Ar
19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr
37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe
55 Cs	56 Ba *	71 Lu	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn
87 Fr	88 Ra *	103 Lr	104 Rf	105 Db	106 Sg	107 Bh	108 Hs	109 Mt	110 Ds	111 Rg	112 Cn	113 Nh	114 Fl	115 Mc	116 Lv	117 Ts	118 Og
		* 57 La	58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb		
		* 89 Ac	90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No		

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55 Cs	56 Ba	* 71 Lu	* 72 Hf	* 73 Ta	* 74 W	* 75 Re	* 76 Os	* 77 Ir	* 78 Pt	* 79 Au	* 80 Hg	* 81 Tl	* 82 Pb	* 83 Bi	* 84 Po	* 85 At	* 86 Rn
87 Fr	88 Ra	* 103 Lr	* 104 Rf	* 105 Db	* 106 Sg	* 107 Bh	* 108 Hs	* 109 Mt	* 110 Ds	* 111 Rg	* 112 Cn	* 113 Nh	* 114 Fl	* 115 Mc	* 116 Lv	* 117 Ts	* 118 Og
		57 La	58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb		
		* 89 Ac	* 90 Th	* 91 Pa	* 92 U	* 93 Np	* 94 Pu	* 95 Am	* 96 Cm	* 97 Bk	* 98 Cf	* 99 Es	* 100 Fm	* 101 Md	* 102 No		

Solid State Ionics 262 (2014) 530–535

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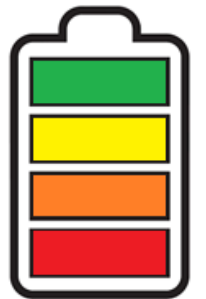
Solid State Ionics

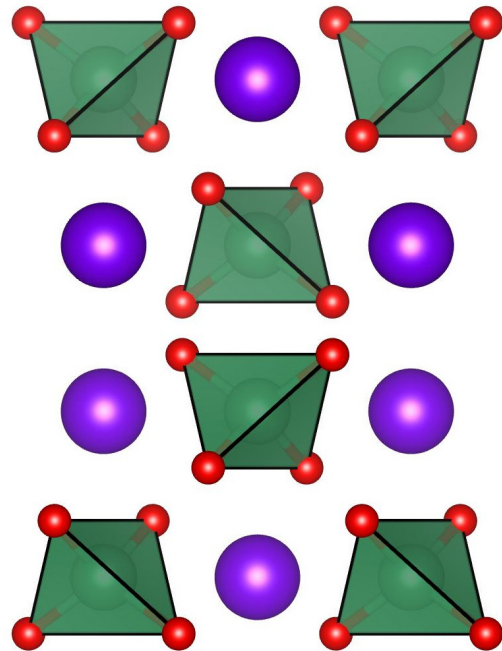
journal homepage: www.elsevier.com/locate/ssi

Crystal structure and potential interstitial oxide ion conductivity of LnNbO_4 and $\text{LnNb}_{0.92}\text{W}_{0.08}\text{O}_{4.04}$ (Ln = La, Pr, Nd)

Cheng Li, Ryan D. Bayliss, Stephen J. Skinner*

Department of Materials, Imperial College London, Exhibition Road, London, SW7 2AZ, UK



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Materials Research Bulletin 61 (2015) 422–432


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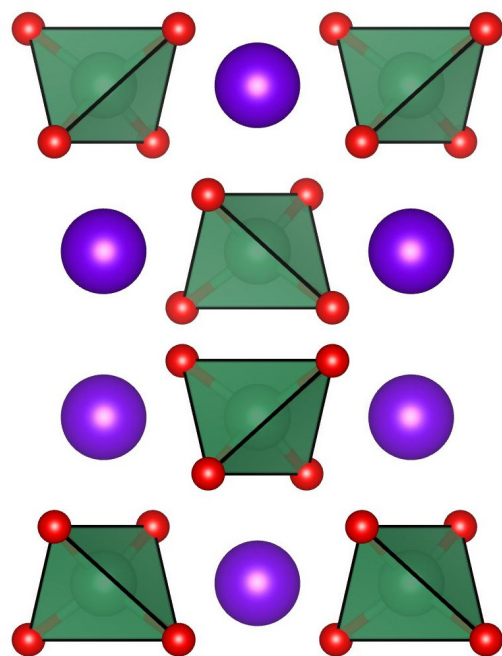
Scheelite-type MWO_4 ($M = Ca, Sr, \text{ and } Ba$) nanophosphors: Facile synthesis, structural characterization, photoluminescence, and photocatalytic properties

C. Shivakumara ^{a,*}, Rohit Saraf ^b, Sukanti Behera ^a, N. Dhananjaya ^c, H. Nagabhushana ^d






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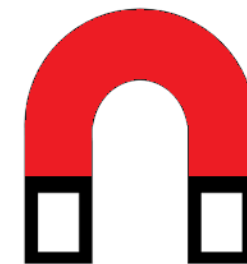
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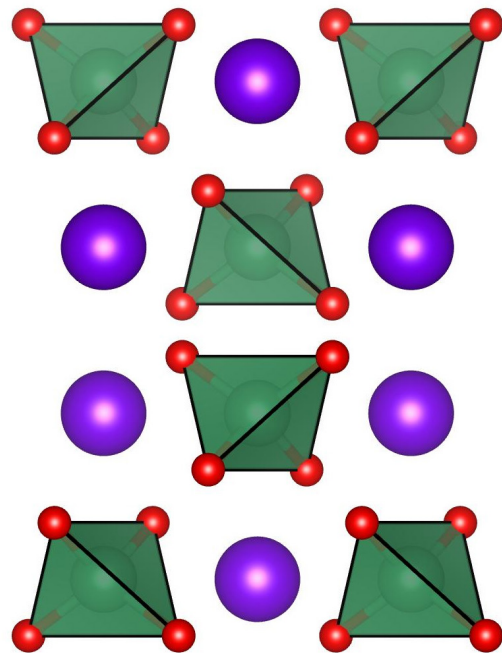
DOI: [10.1039/C9CP00448C](https://doi.org/10.1039/C9CP00448C) (Communication) *Phys. Chem. Chem. Phys.*, 2019, 21, 7261-7264

Structural and magnetic studies of $KOsO_4$, a $5d^1$ quantum magnet oxide[†]

Sean Injac ^a, Alexander K. L. Yuen ^{ab}, Maxim Avdeev ^{ab}, Fabio Orlandi ^c and Brendan J. Kennedy ^{ab*}



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Materials Research Bulletin

Volume 12, Issue 1, January 1977, Pages 25-33

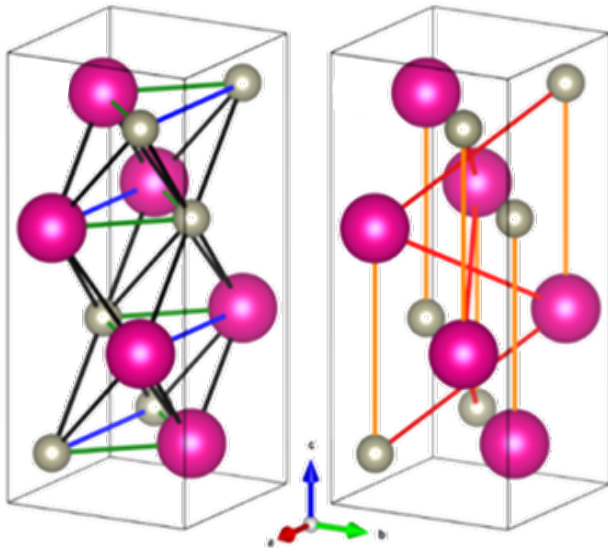


Les niobates $CaUNb_2O_8$ et $MThNb_2O_8$ ($M = Ca, Sr, Cd$). Etude des transformations monoclinique - quadratique

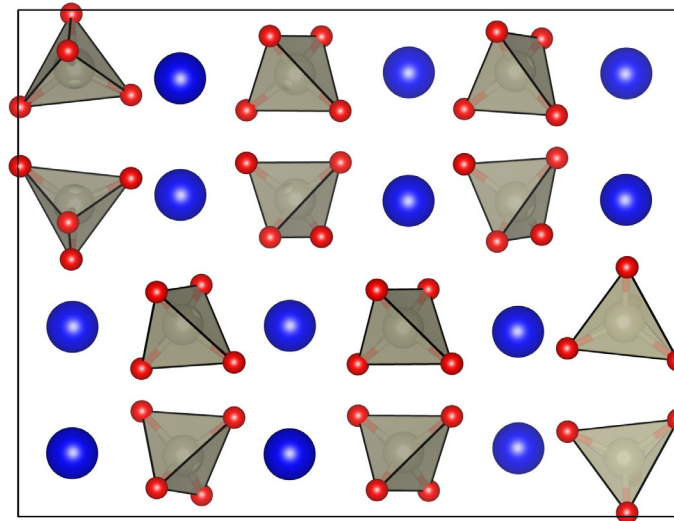
Gilles Fonteneau, Hervé L'Helgoualch, Jacques Lucas



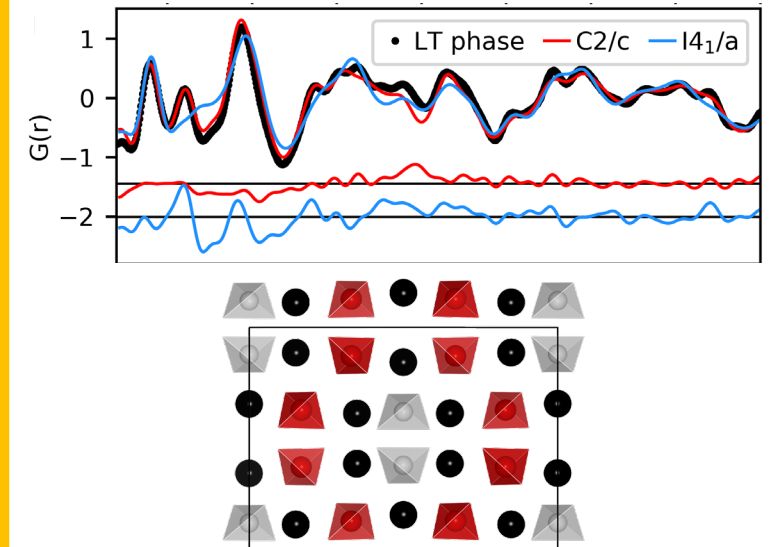
High-Temperature Phase Transition in RbReO_4



Re-Entrant Phase Transition in TlReO_4

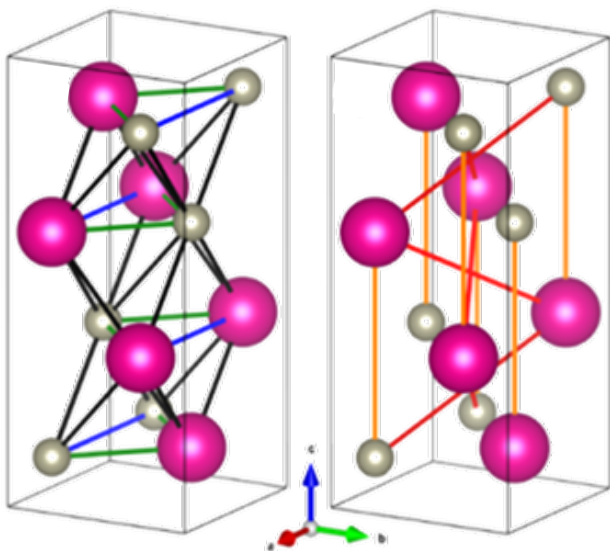


Symmetry-Lowering in $\text{Bi}_3\text{FeMo}_2\text{O}_{12}$

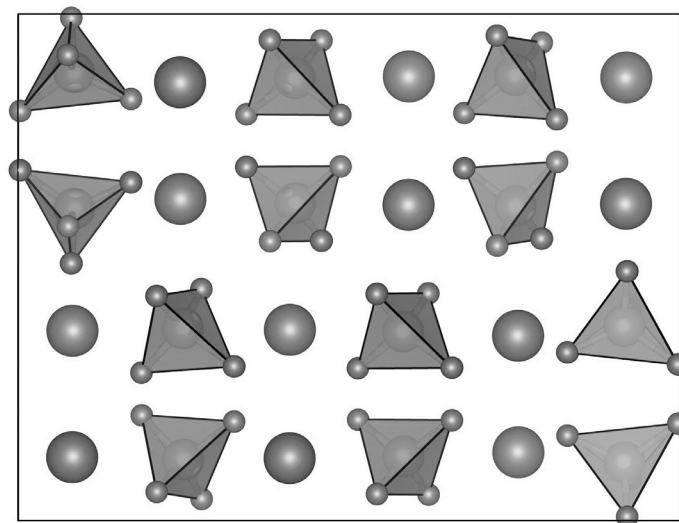


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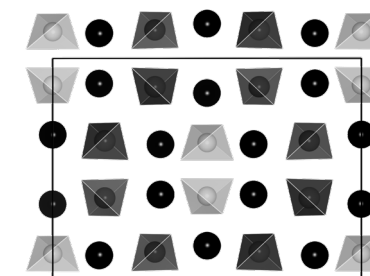
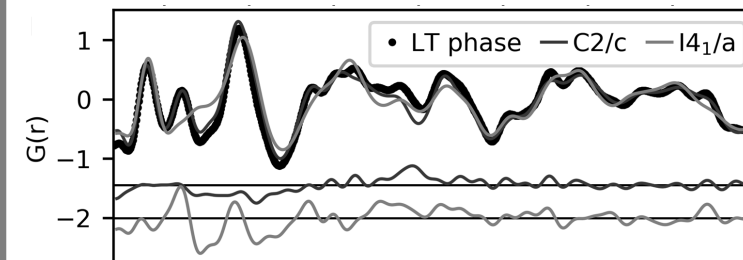
Why do the Re-O bonds decrease upon heating?



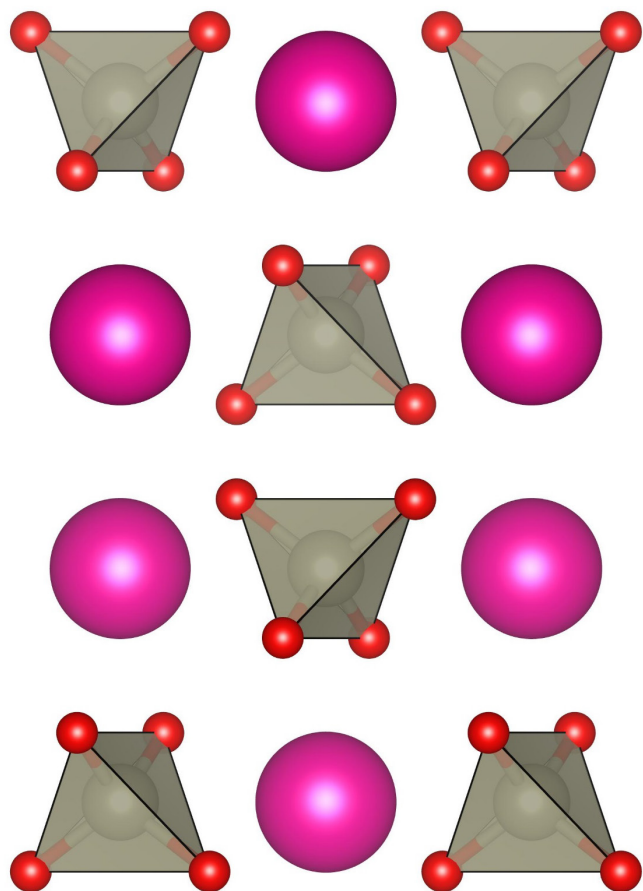
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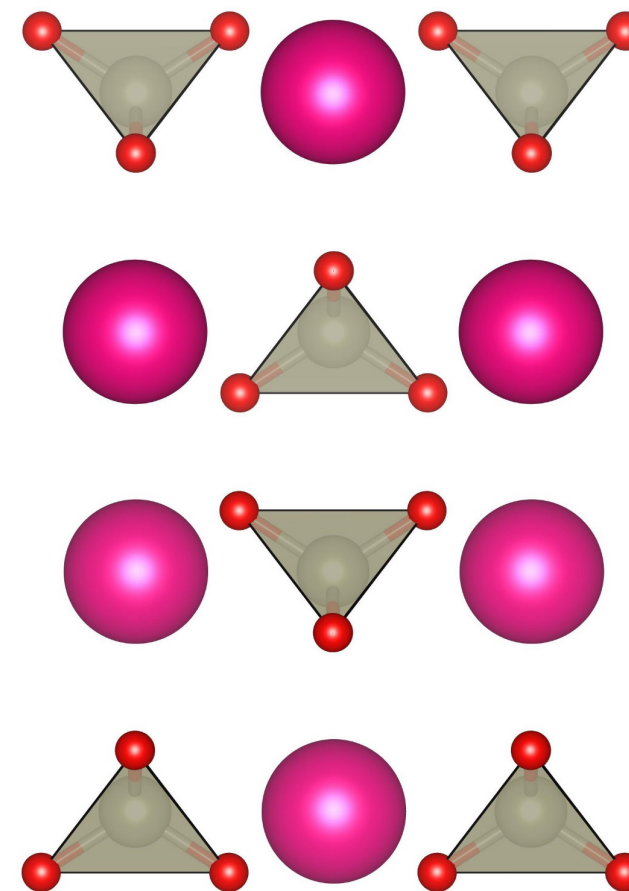
Phase Transition in RbReO_4



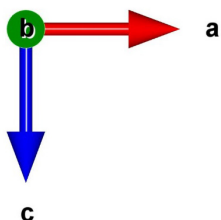
$I4_1/a$
Tetragonal Scheelite



$\sim 615 \text{ K}$



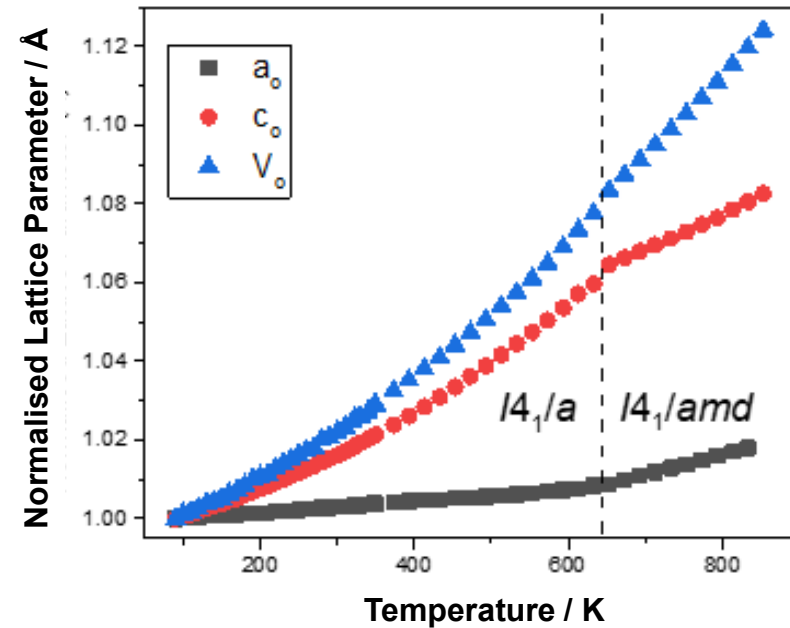
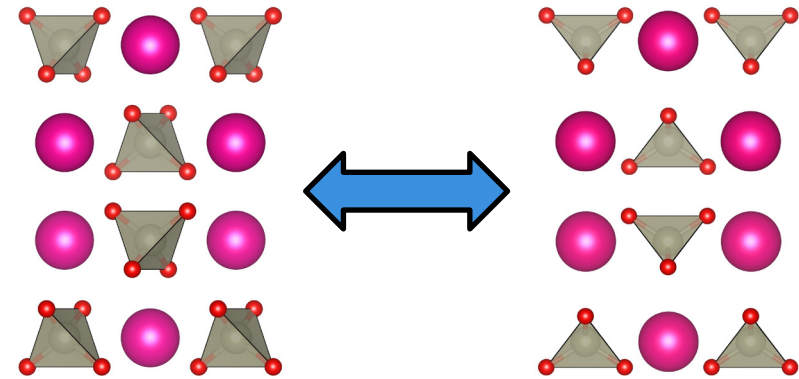
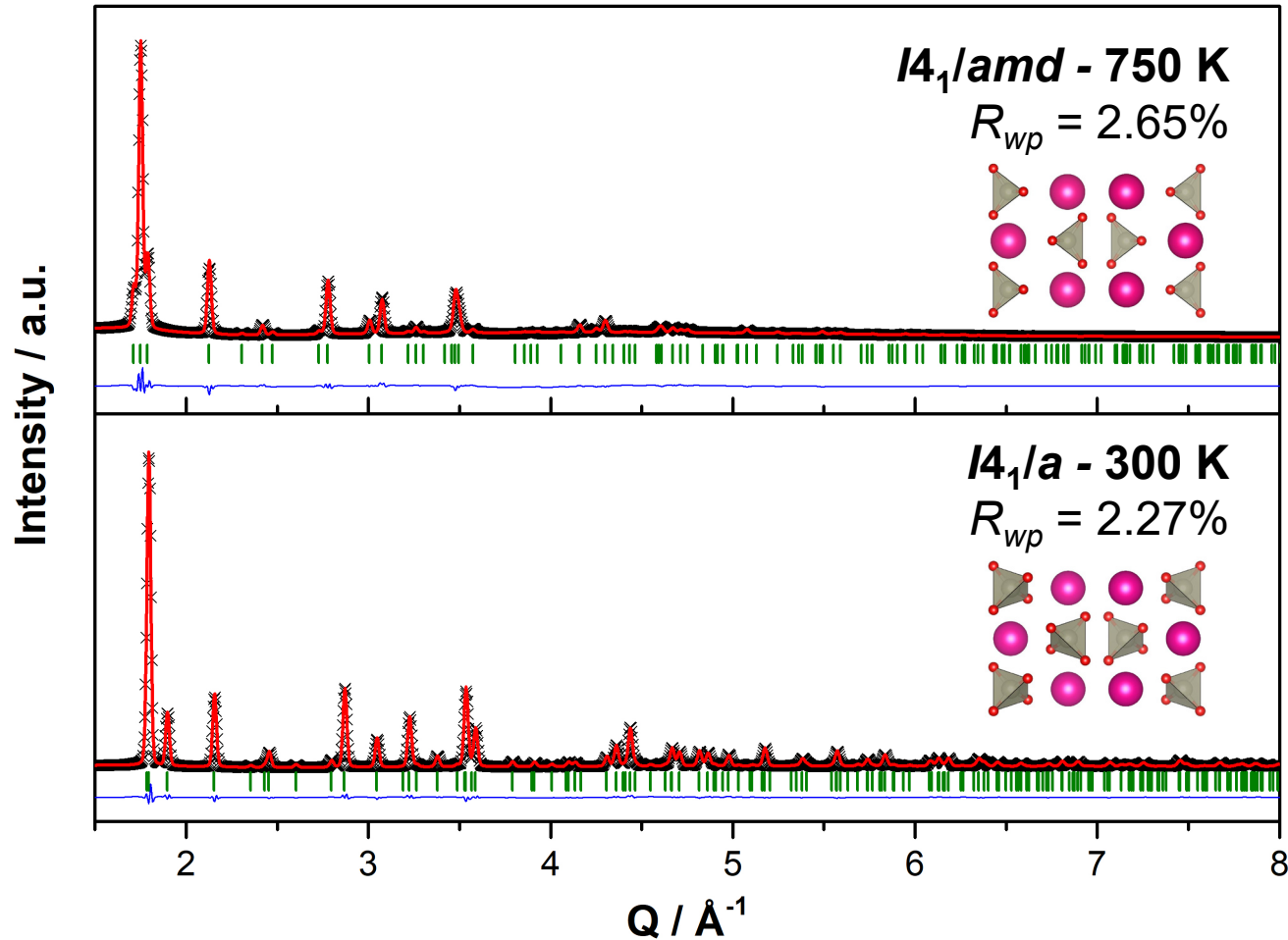
$I4_1/amd$
Scheelite Aristotype



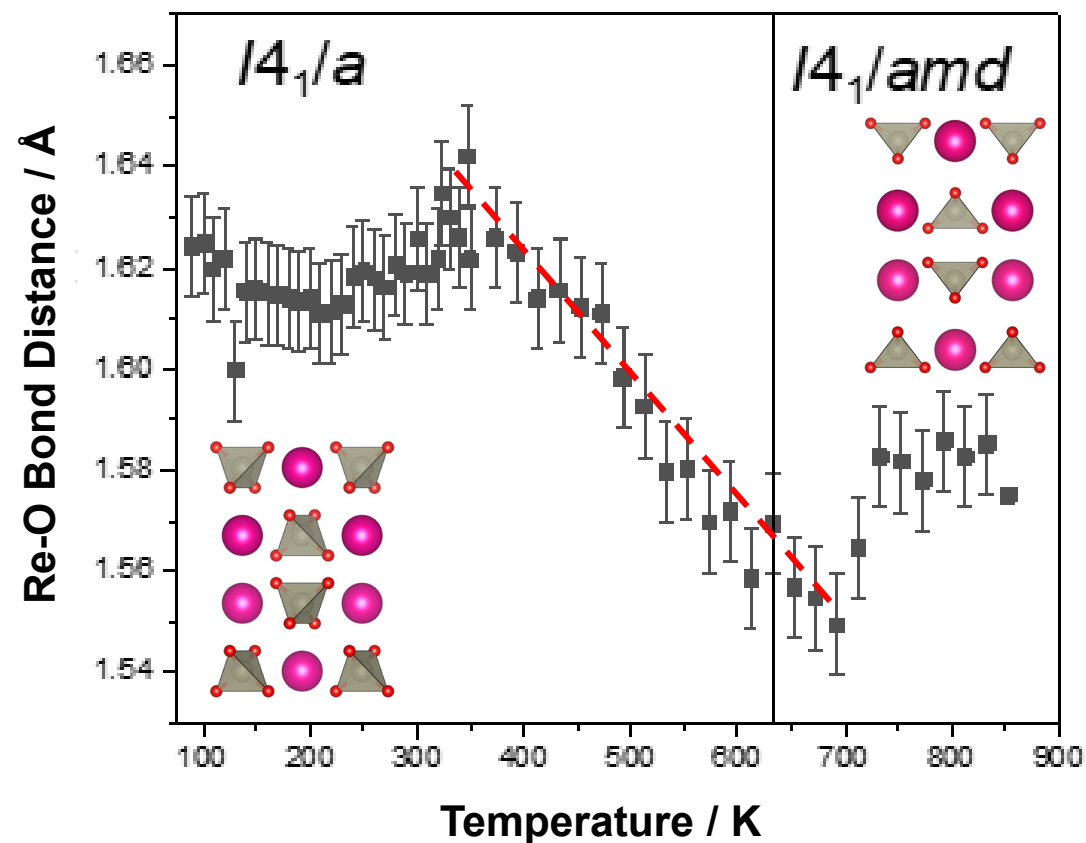
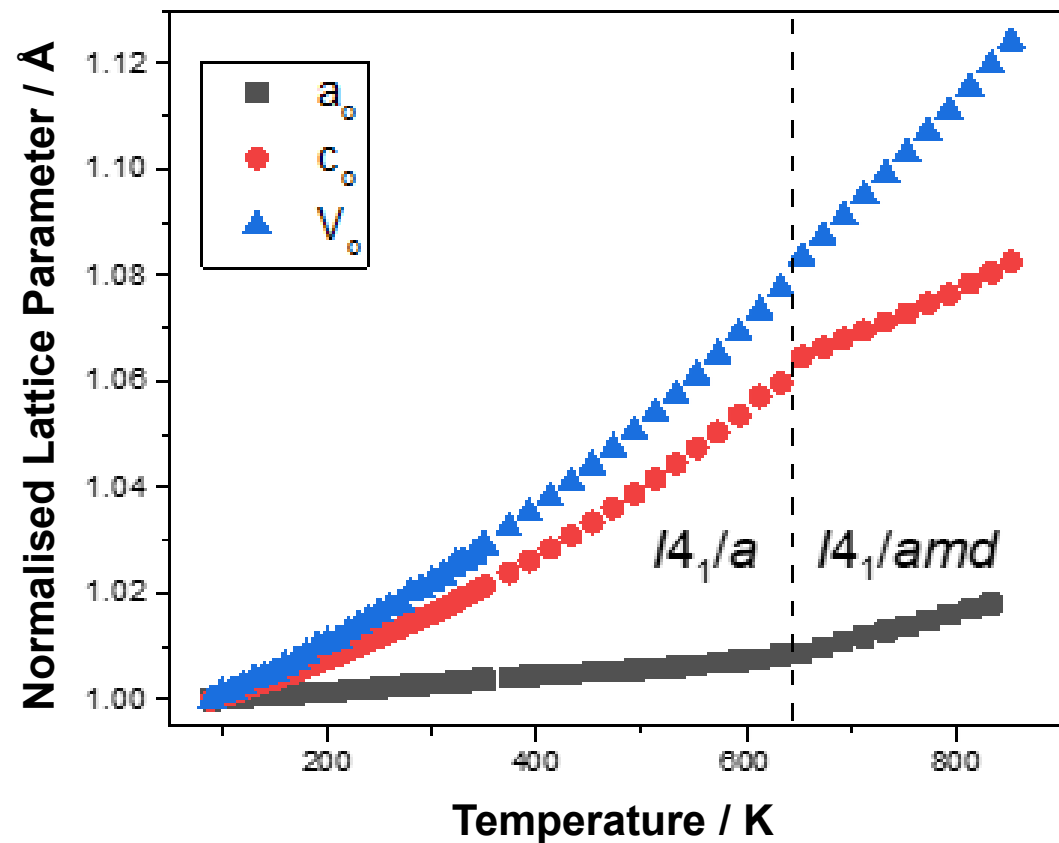
Phase Transition in RbReO_4



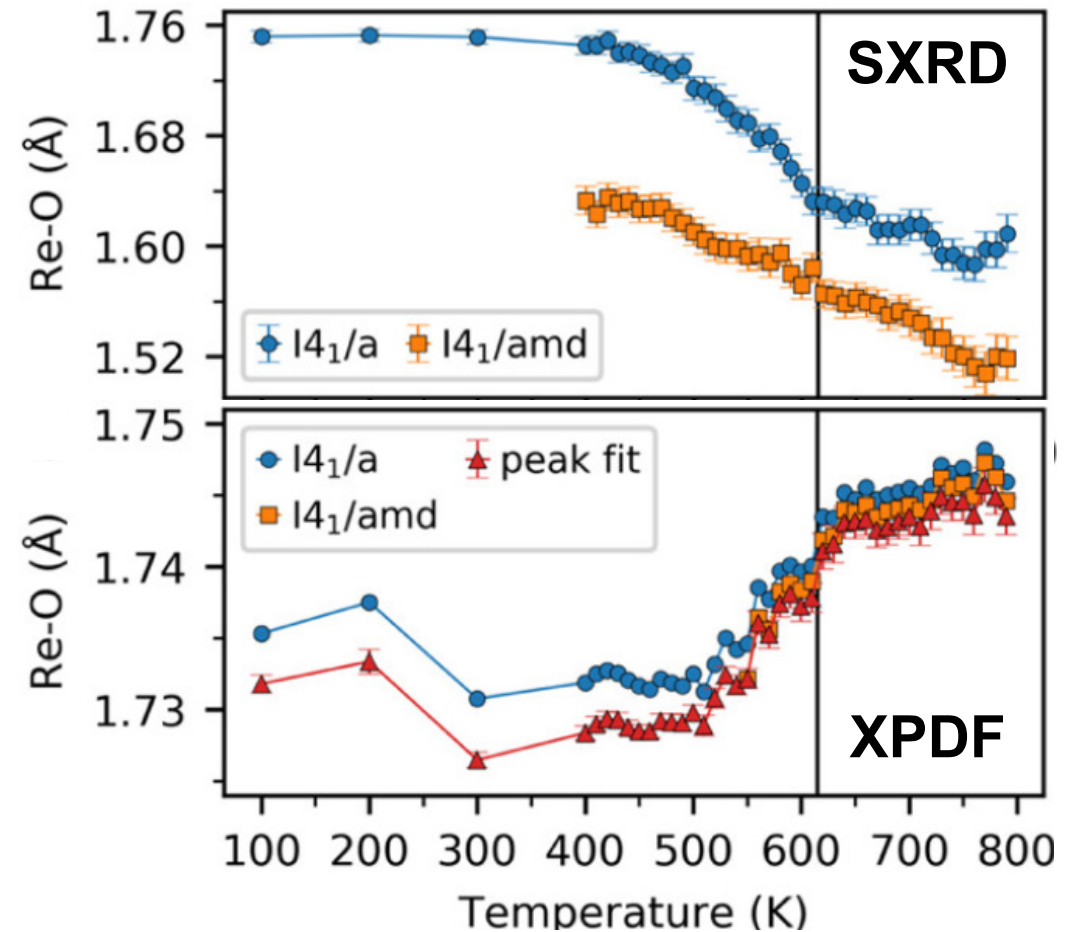
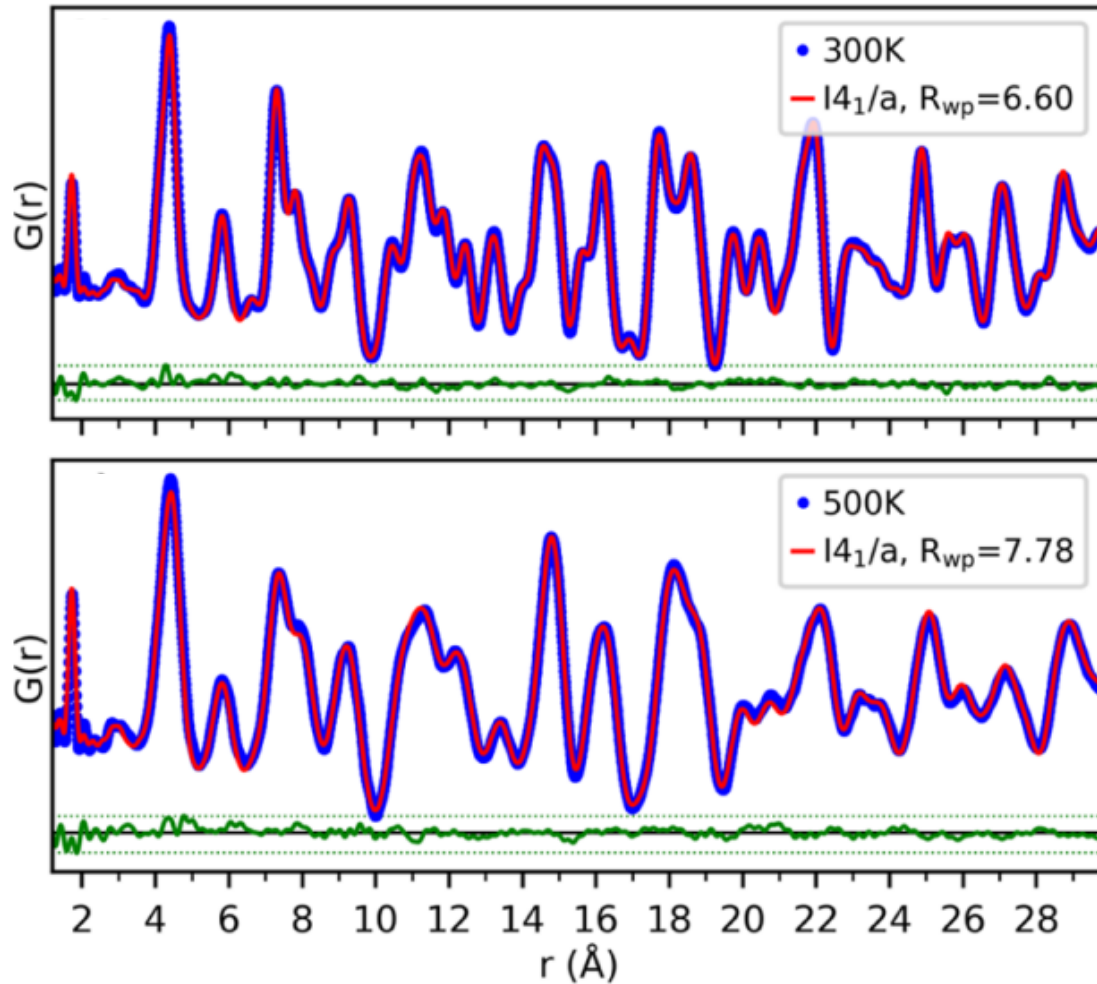
SXRD, $\lambda = 0.16166 \text{ \AA}$



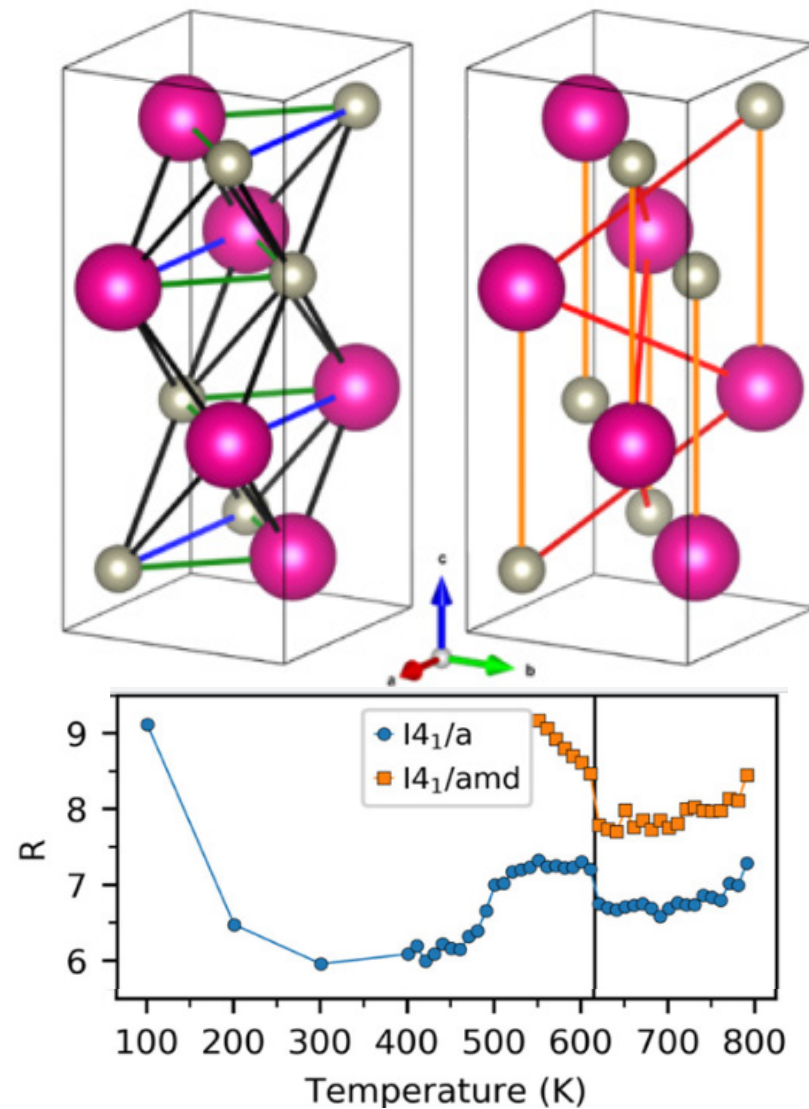
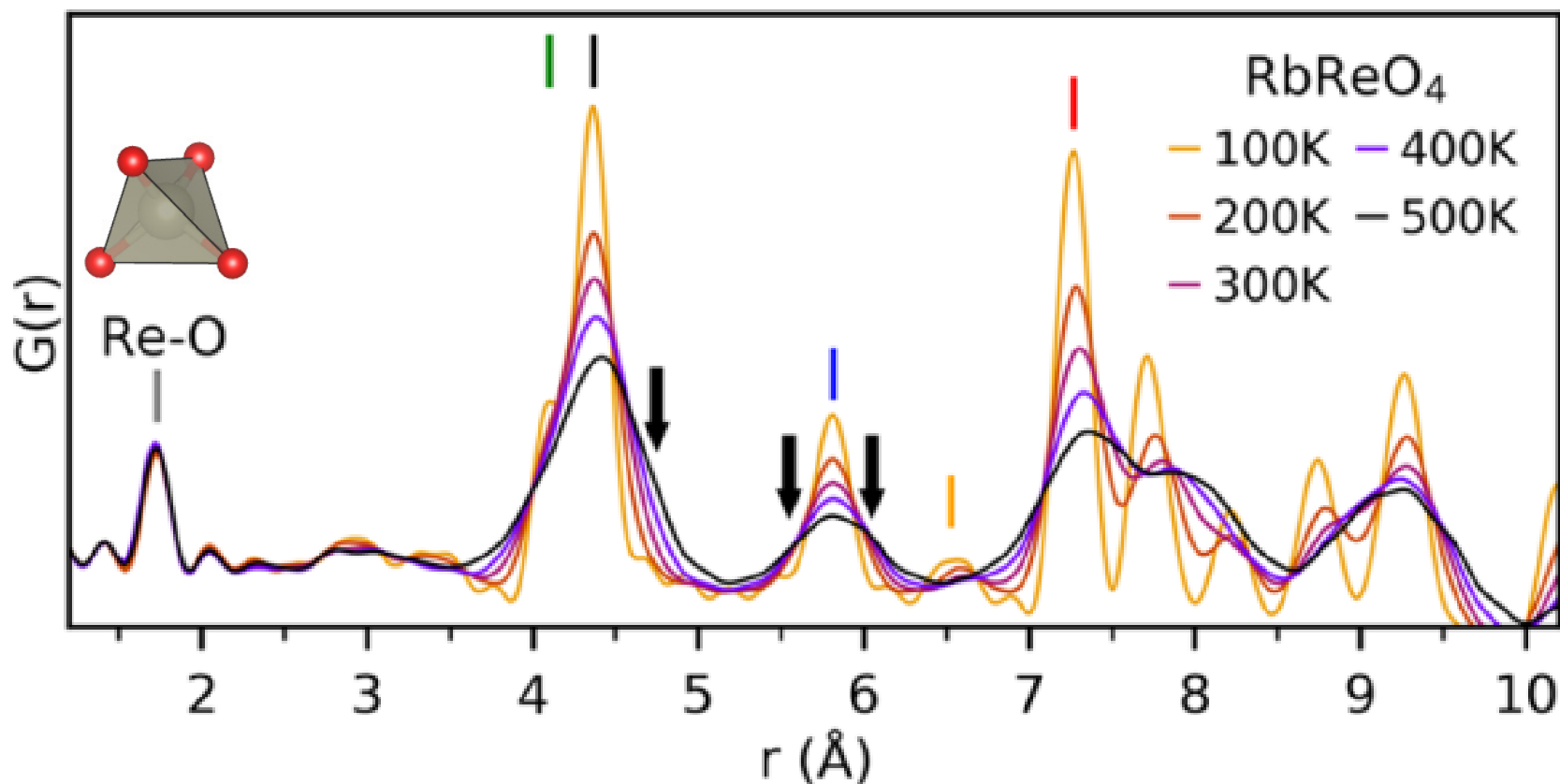
Irregularities in the Long-Range Structure



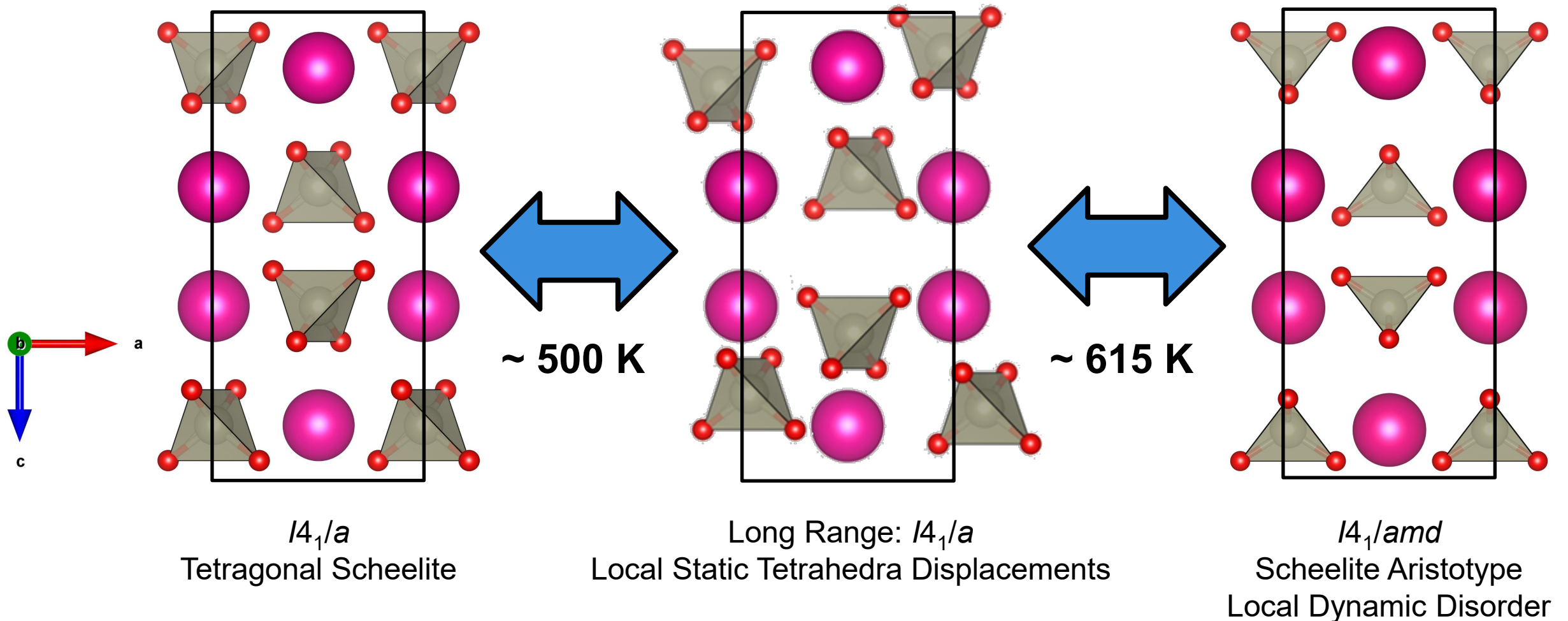
X-Ray Pair Distribution Function Analysis



X-Ray Pair Distribution Function Analysis

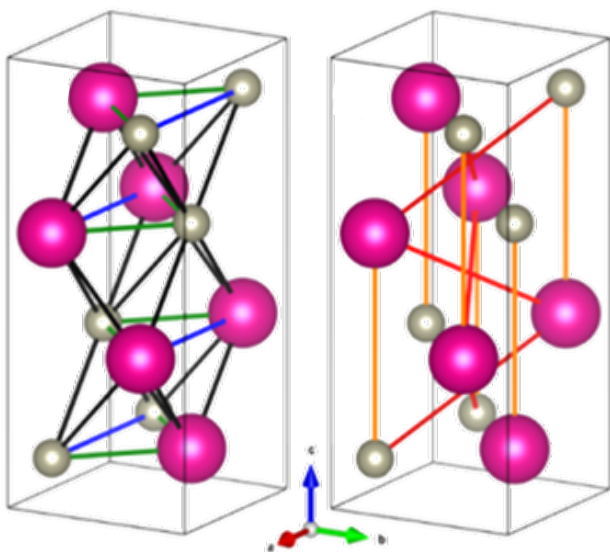


Local Displacements of ReO_4 Tetrahedra



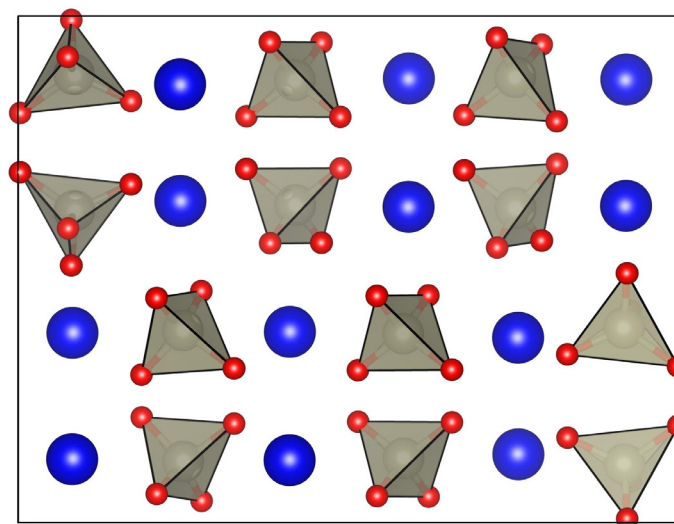
High-Temperature Phase Transition in RbReO_4

Why do the Re-O bonds decrease upon heating?

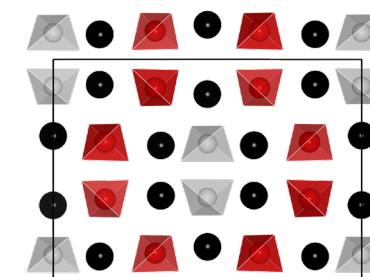
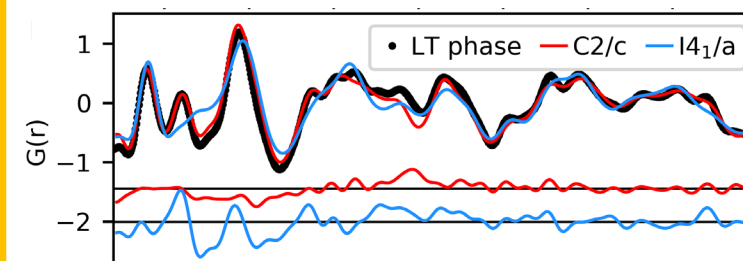


Local displacements of the ReO_4 tetrahedra.

Re-Entrant Phase Transition in TlReO_4

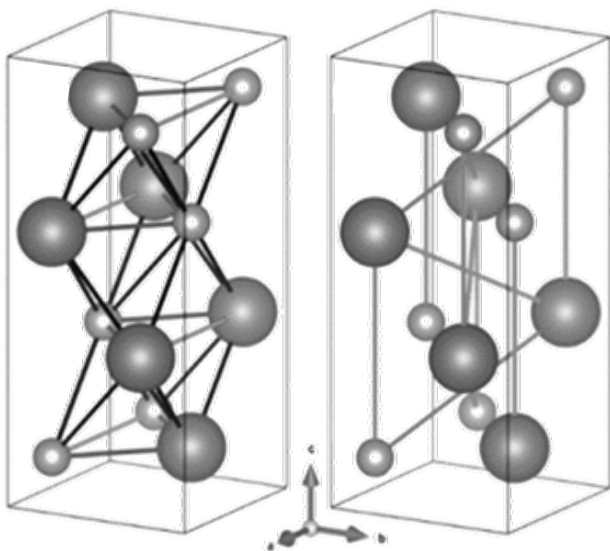


Symmetry-Lowering in $\text{Bi}_3\text{FeMo}_2\text{O}_{12}$



High-Temperature Phase Transition in RbReO_4

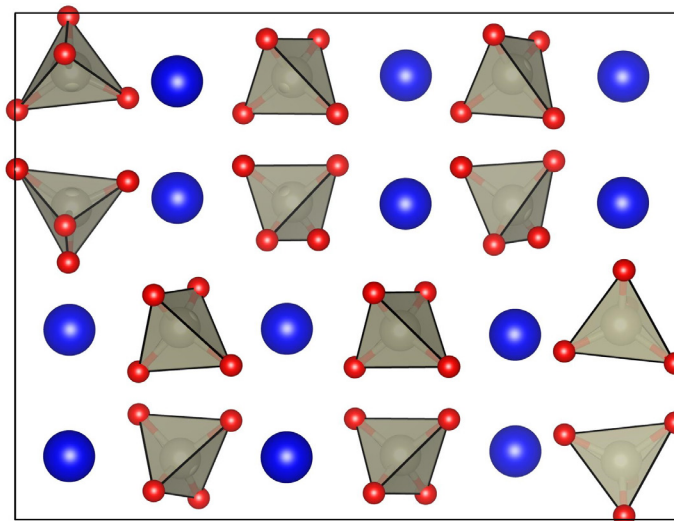
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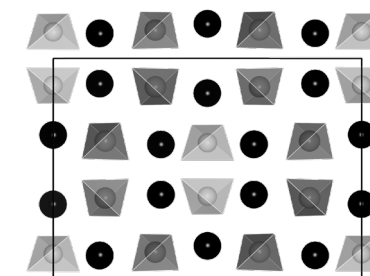
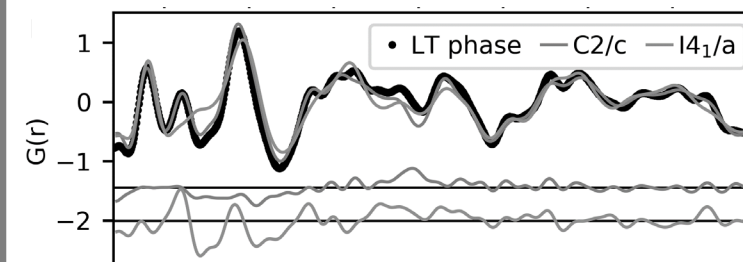
Local displacements of the ReO_4 tetrahedra.

Re-Entrant Phase Transition in TlReO_4

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Symmetry-Lowering in $\text{Bi}_3\text{FeMo}_2\text{O}_{12}$



Structures with Post-Transition Metals

83 208.980
Bi
Bismuth
[Xe] 4f¹⁴5d¹⁰6s²6p³
Post-Transition Metal



82 207.2
Pb
Lead
[Xe] 4f¹⁴5d¹⁰6s²6p²
Post-Transition Metal

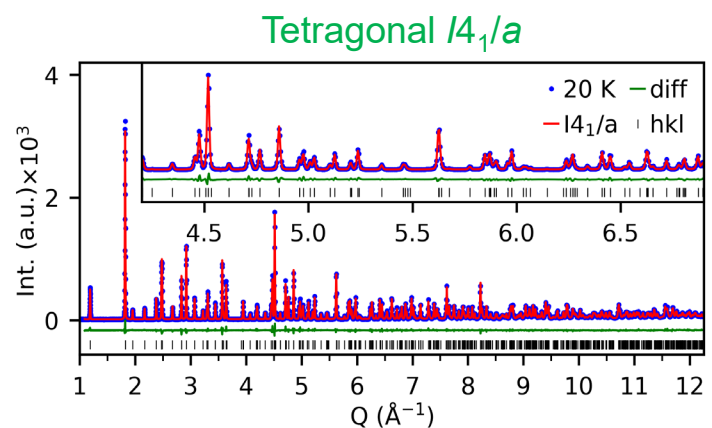
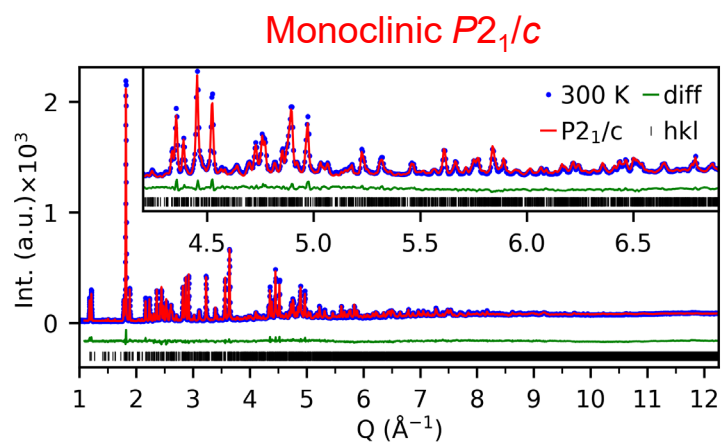


81 204.383
Tl
Thallium
[Xe] 4f¹⁴5d¹⁰6s²6p¹
Post-Transition Metal



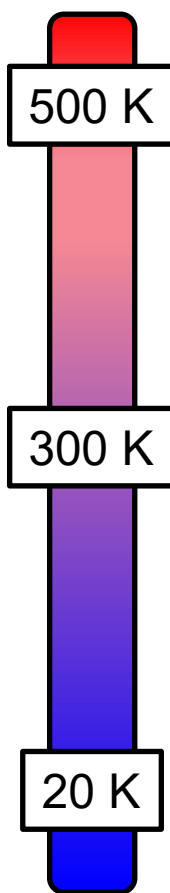
1 H																	2 He
3 Li	4 Be											5 B	6 C	7 N	8 O	9 F	10 Ne
11 Na	12 Mg											13 Al	14 Si	15 P	16 S	17 Cl	18 Ar
19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr
37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe
55 Cs	56 Ba	* 71 Lu	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn
87 Fr	88 Ra	* 103 Lr	104 Rf	105 Db	106 Sg	107 Bh	108 Hs	109 Mt	110 Ds	111 Rg	112 Cn	113 Nh	114 Fl	115 Mc	116 Lv	117 Ts	118 Og
		* 57 La	58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb		
		* 89 Ac	90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No		

Long-Range (POWGEN)



Temperature

Tetragonal
 $I4_1/a$



Monoclinic
 $P2_1/c$

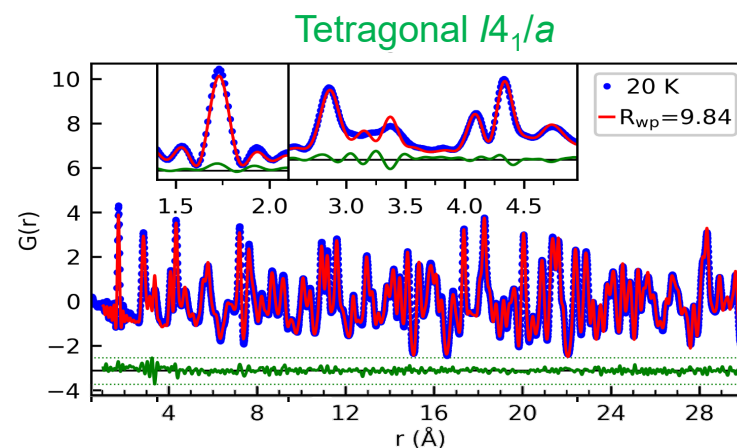
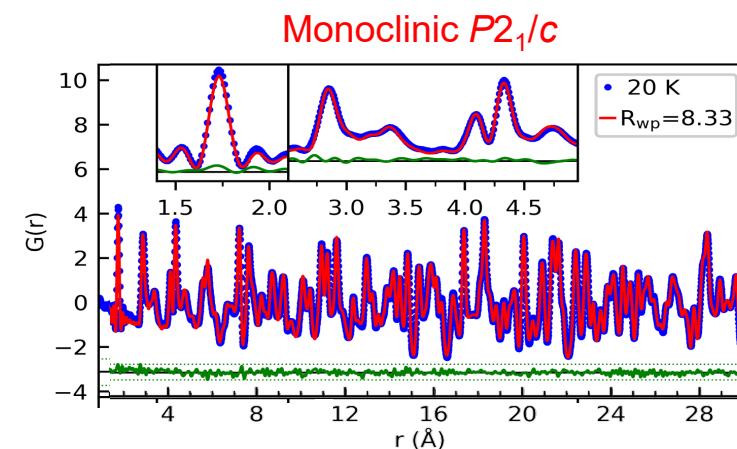
Monoclinic
 $P2_1/c$

Monoclinic
 $P2_1/c$

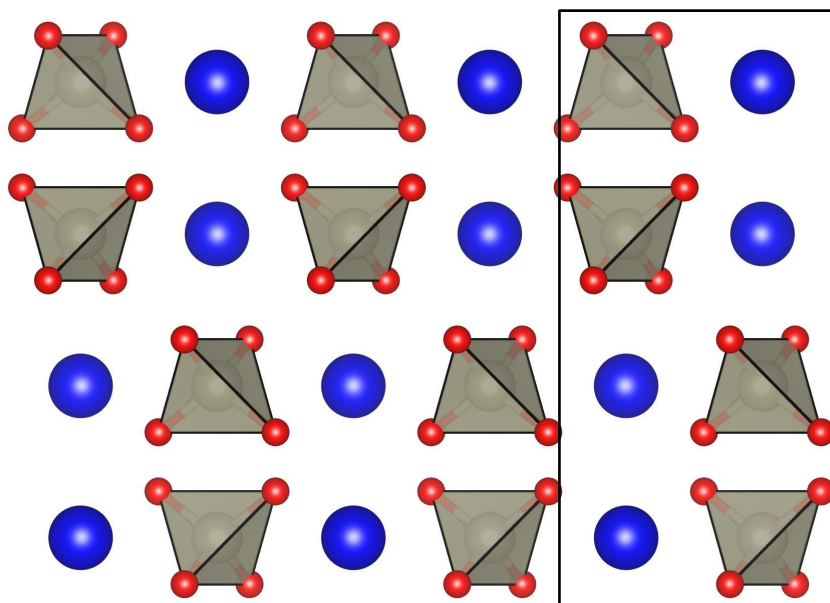
Monoclinic
 $P2_1/c$

Tetragonal
 $I4_1/a$

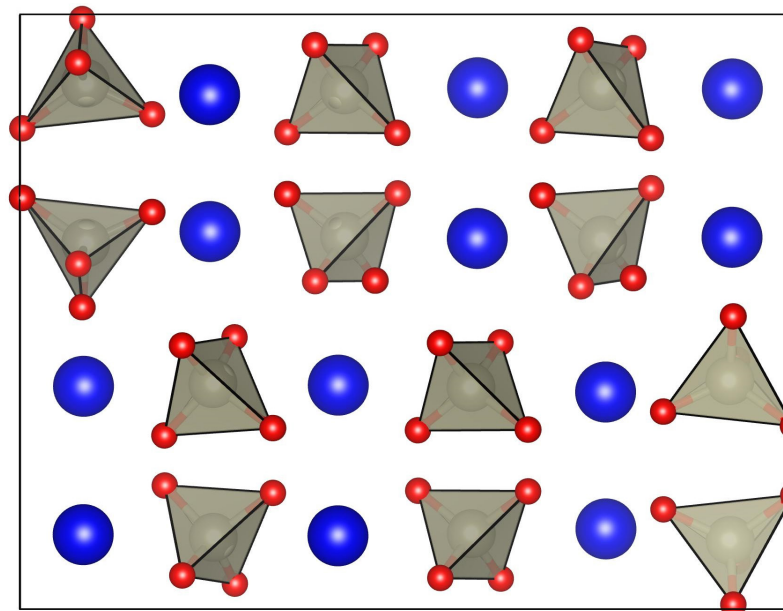
Short-Range (NOMAD)



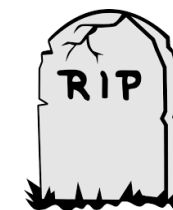
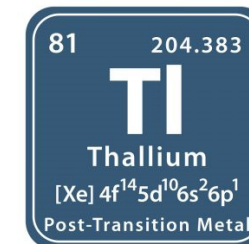
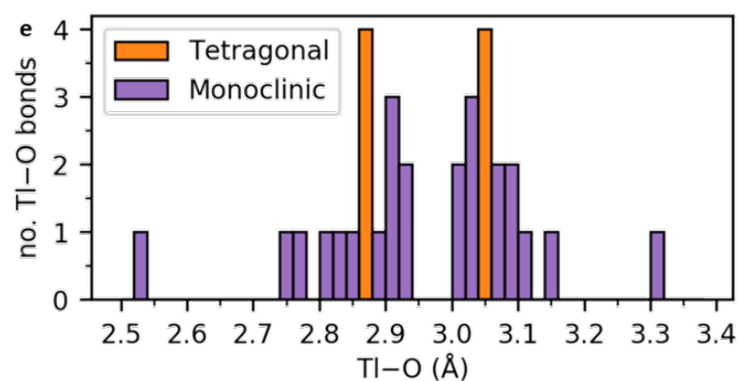
The Hidden Structure of TlReO_4



Tetragonal Long-Range
 $I4_1/a$



Monoclinic Short-Range
 $P2_1/c$

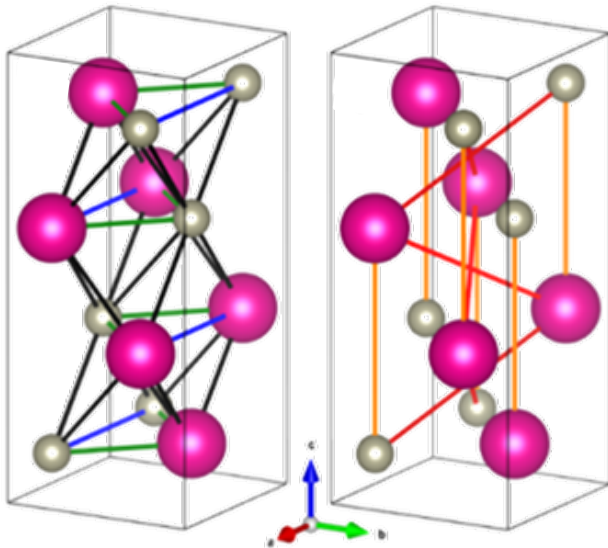


Dr Matilde Saura Muzquiz



High-Temperature Phase Transition in RbReO_4

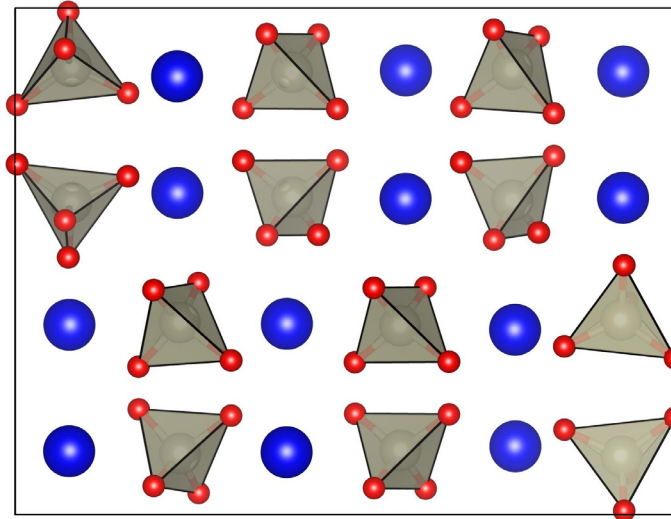
Why do the Re-O bonds decrease upon heating?



Local displacements of the ReO_4 tetrahedra.

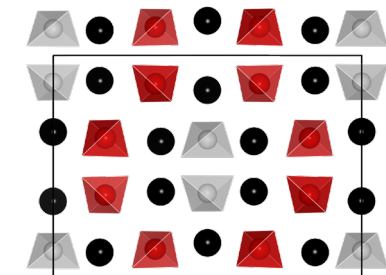
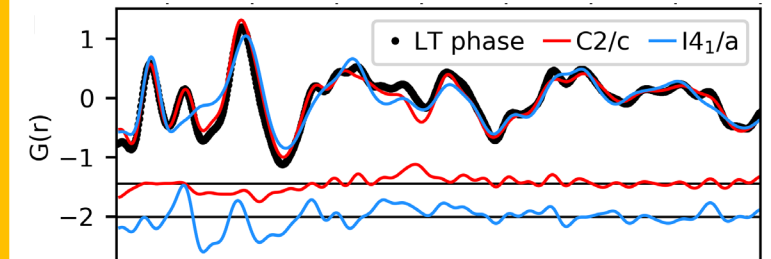
Re-Entrant Phase Transition in TlReO_4

Why do the symmetry lower upon heating?



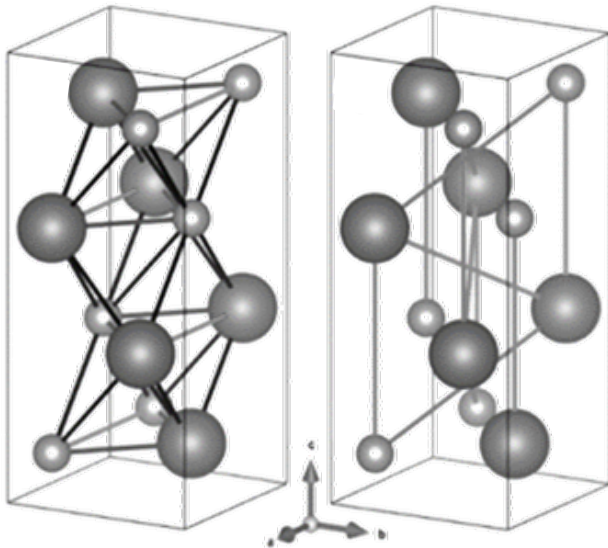
TlReO_4 is locally monoclinic.

Symmetry-Lowering in $\text{Bi}_3\text{FeMo}_2\text{O}_{12}$



High-Temperature Phase Transition in RbReO_4

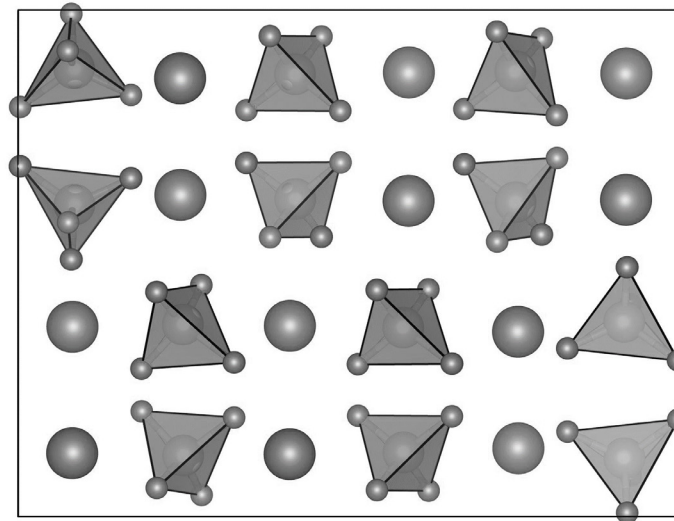
Why do the Re-O bonds decrease upon heating?



Local displacements of the ReO_4 tetrahedra.

Re-Entrant Phase Transition in TlReO_4

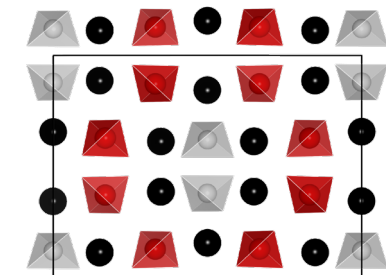
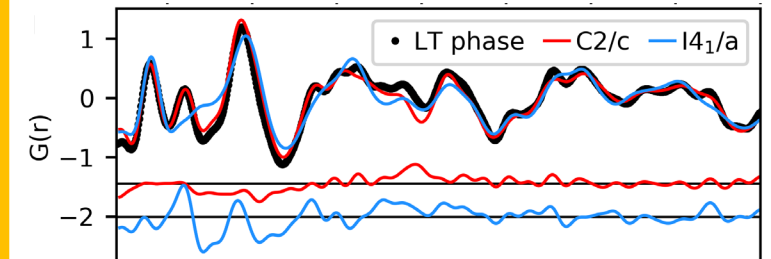
Why do the symmetry lower upon heating?



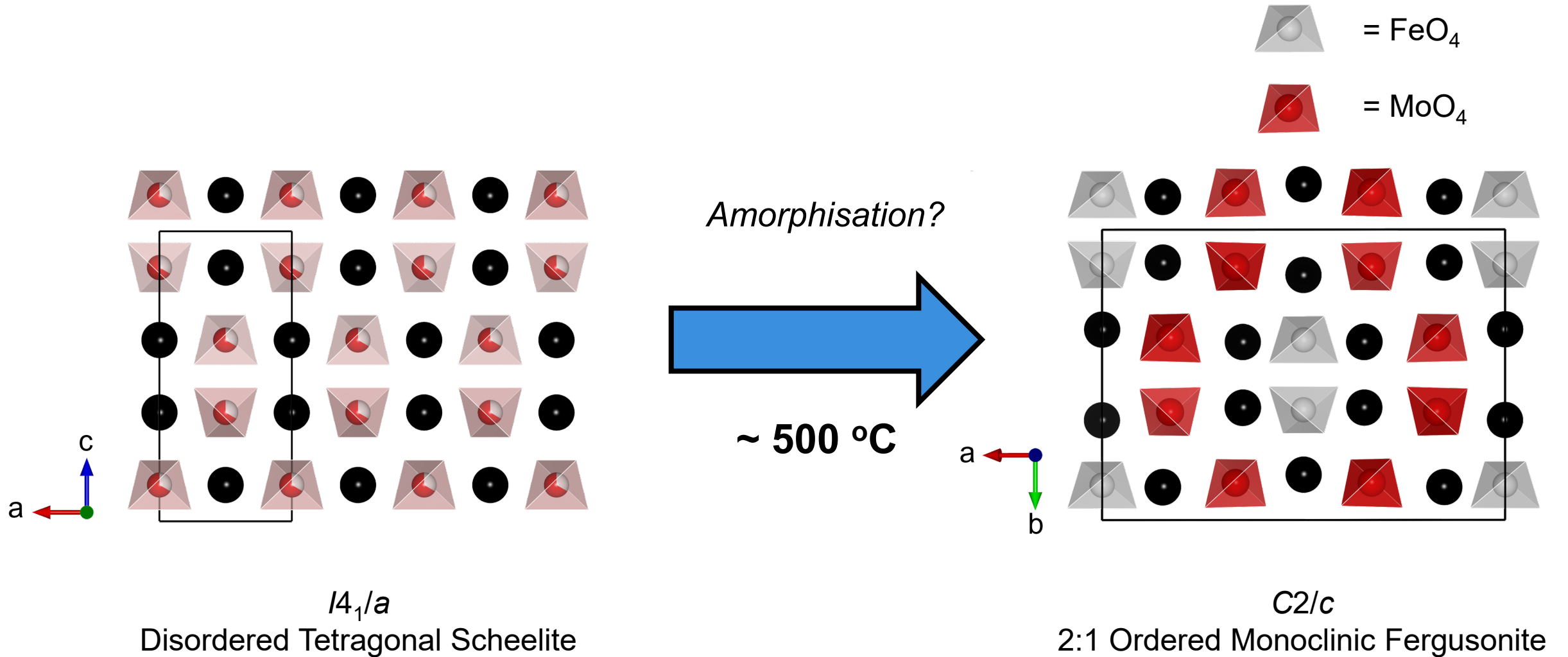
TlReO_4 is locally monoclinic.

Symmetry-Lowering in $\text{Bi}_3\text{FeMo}_2\text{O}_{12}$

Do the cations order upon heating?

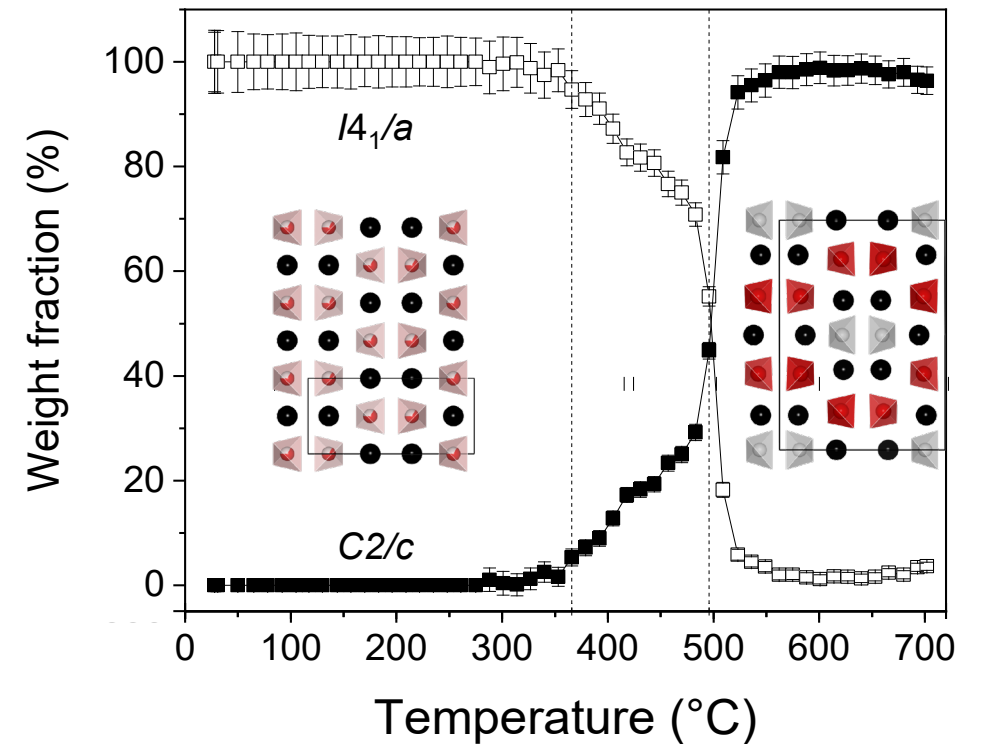
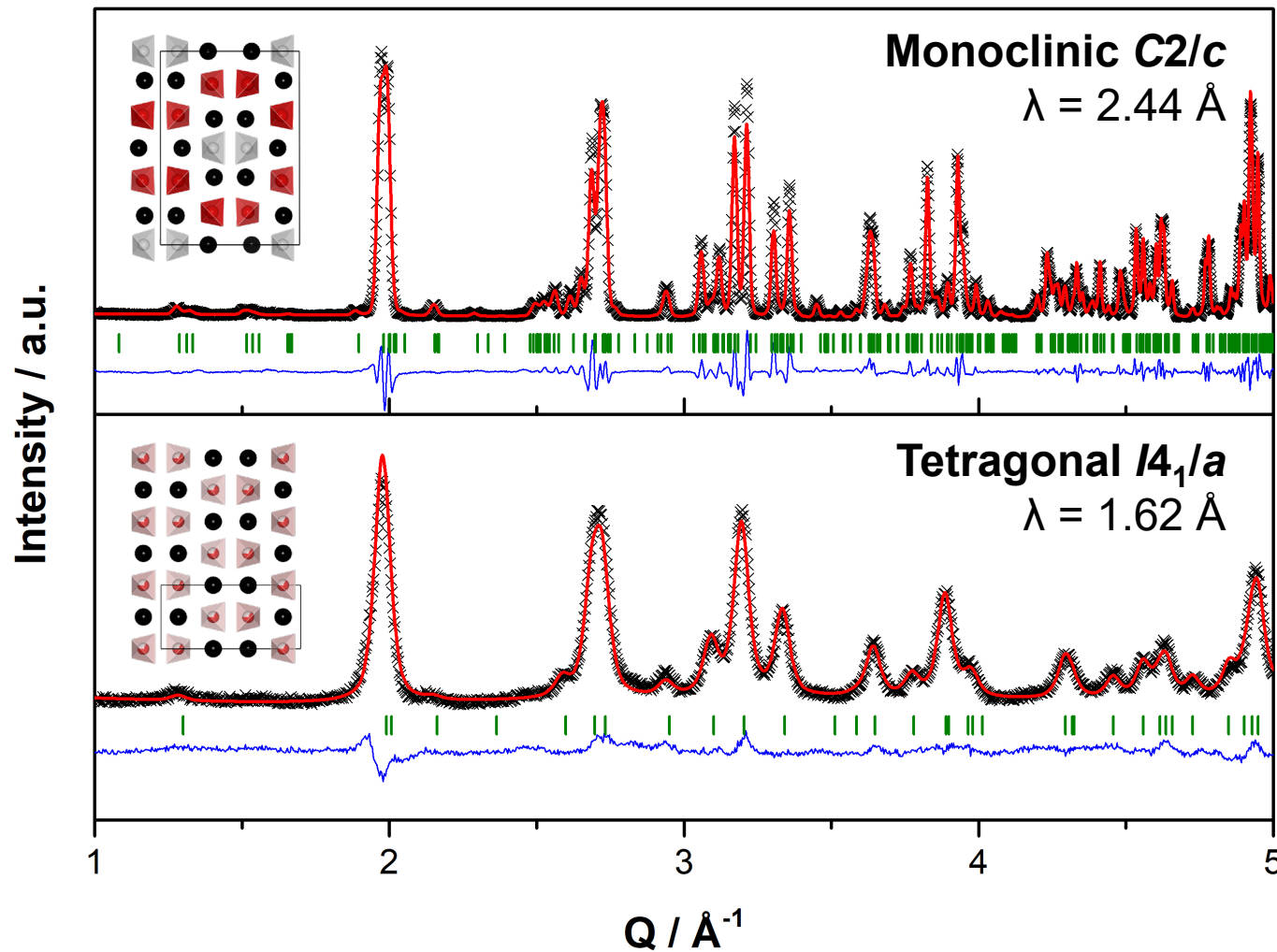


Phase Transformation in $\text{Bi}_3\text{FeMo}_2\text{O}_{12}$

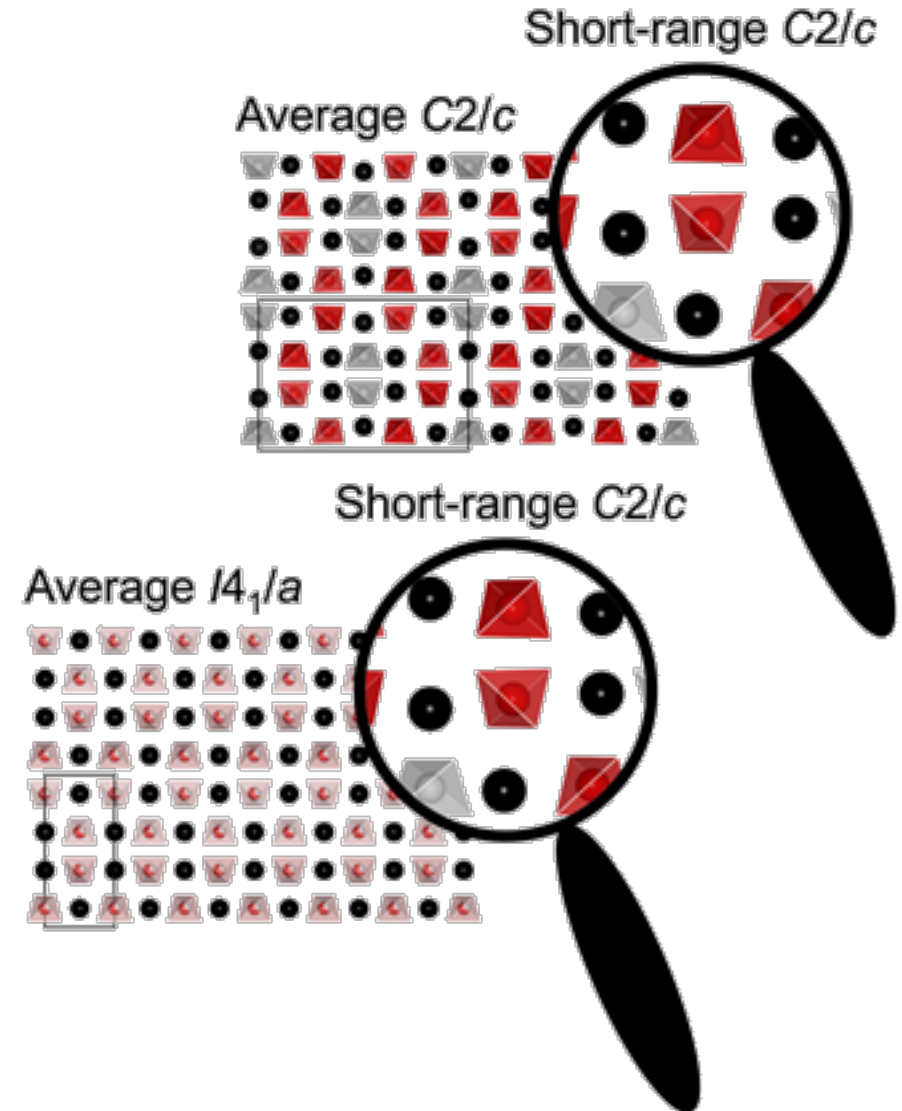
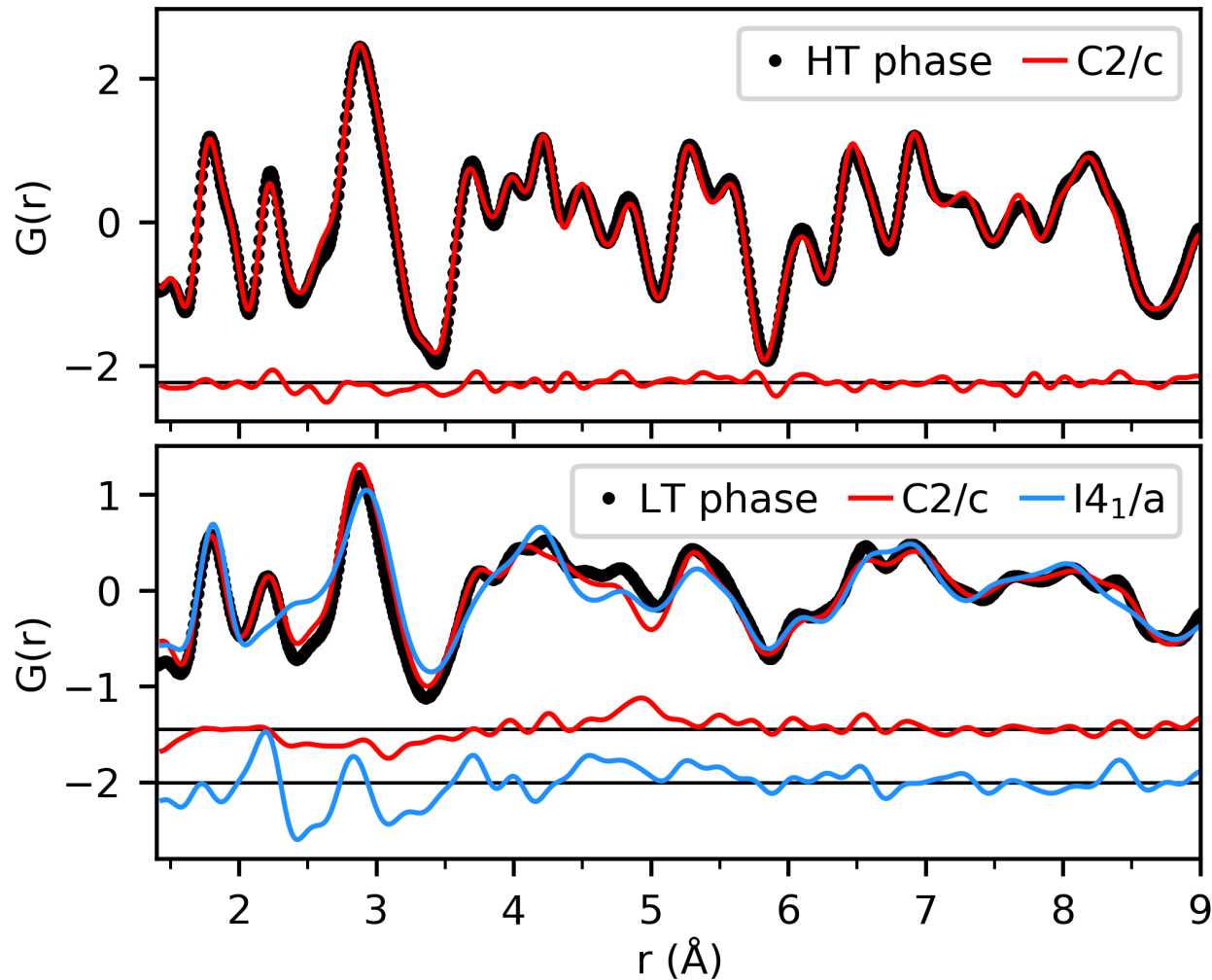


Variable Temperature Neutron Powder Diffraction

Neutron Powder Diffraction @ 50 K

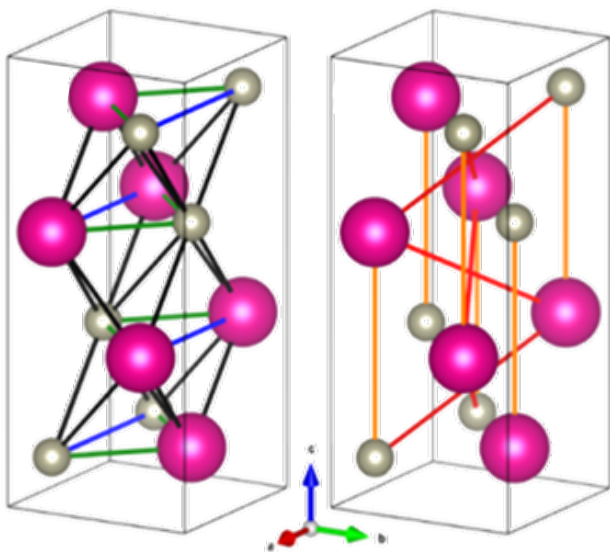


Neutron Pair Distribution Function Analysis @ 100 K



High-Temperature Phase Transition in RbReO_4

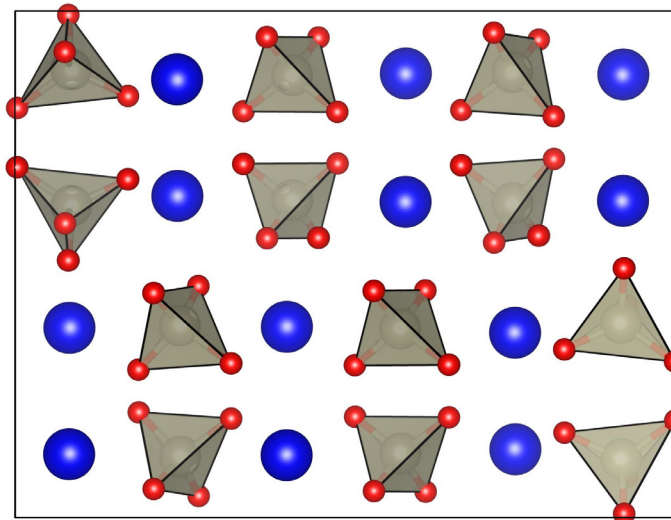
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Local displacements of the ReO_4 tetrahedra.

Re-Entrant Phase Transition in TlReO_4

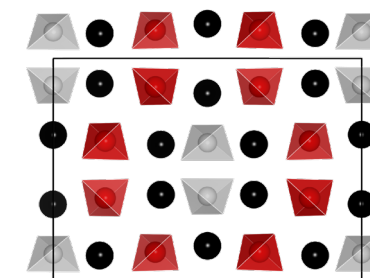
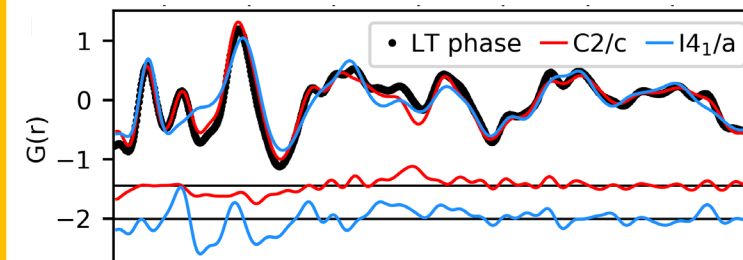
Why do the symmetry lower upon heating?



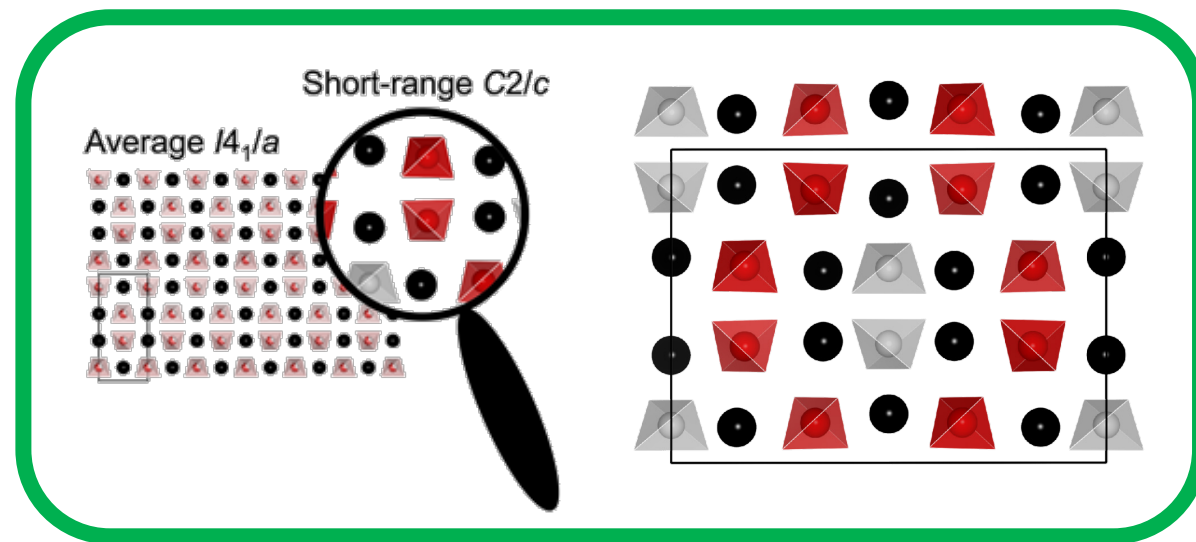
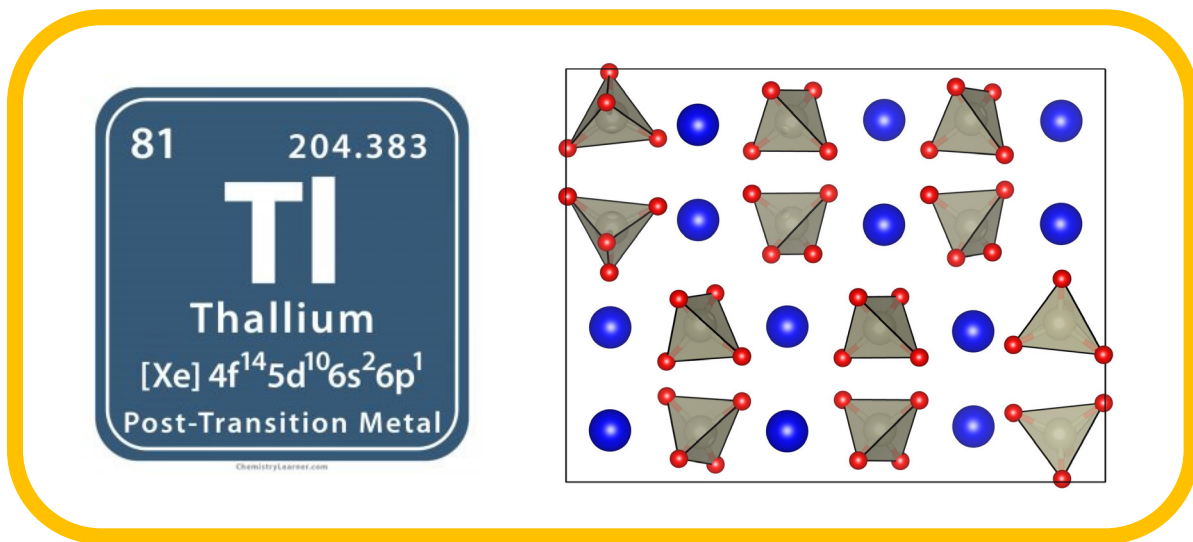
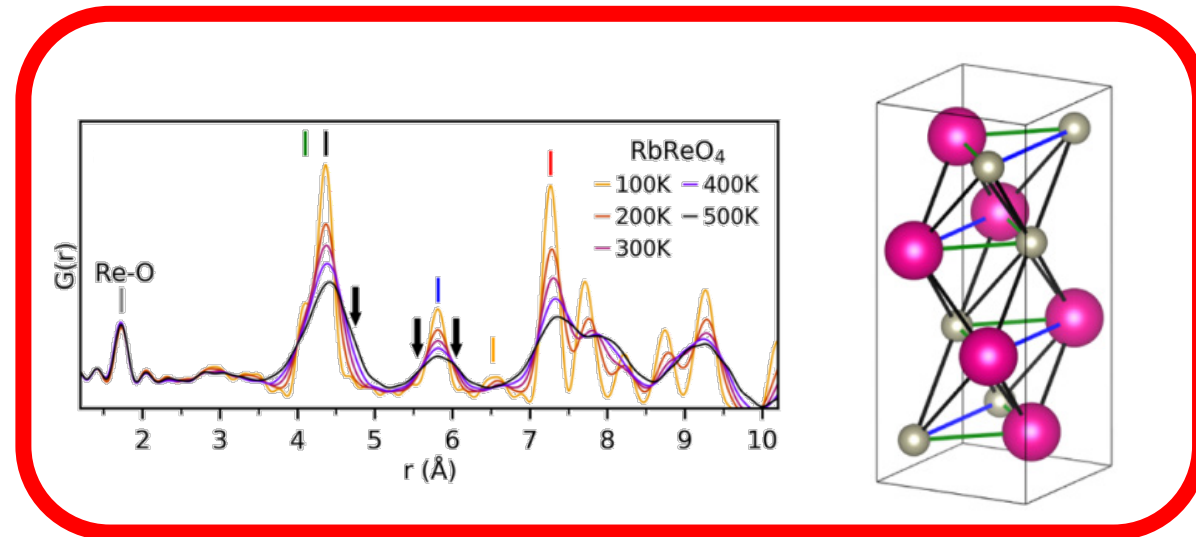
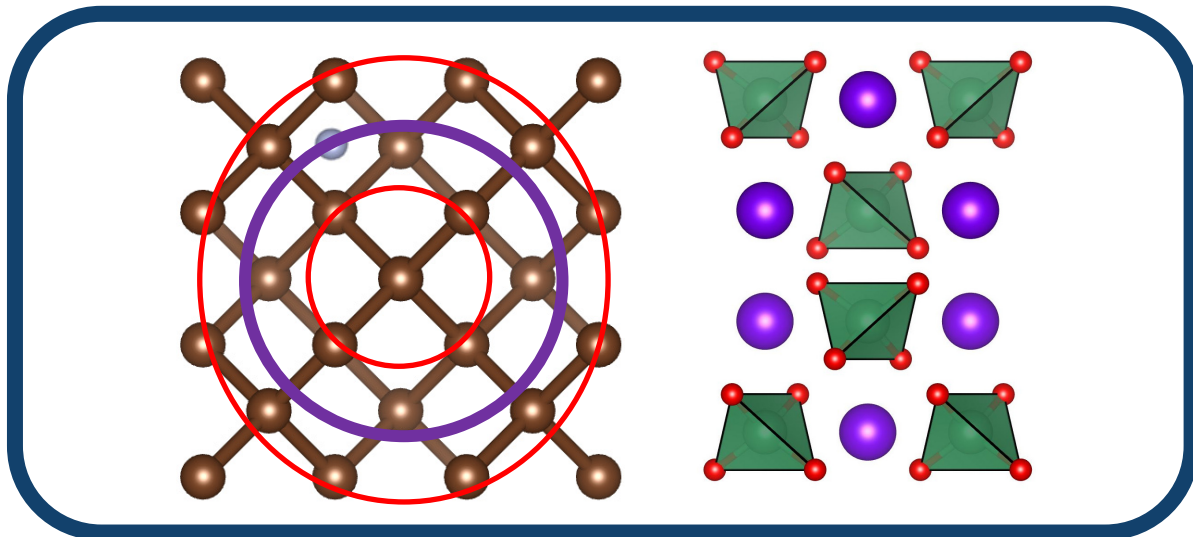
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Symmetry-Lowering in $\text{Bi}_3\text{FeMo}_2\text{O}_{12}$

Do the cations order upon heating?



Difference in tetrahedra environments.





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