

# First FullProf School 2025: Diffraction data analysis of energy materials



## The Rietveld Method

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# Outline



**Structural models: the expressions of structure factors in the case of conventional description of crystal structures.**

**Brief introduction to the Rietveld method. R-factors. Strategy for setting up a refinement process.**

**Quantitative phase analysis.**

# The calculated profile of powder diffraction patterns

$$y_{ci} = \sum_{\{h\}} I_h \Omega(T_i - T_h) + b_i$$

$I_h = I_h(\beta_I)$  Contains structural information:  
atom positions, magnetic moments, etc

$\Omega = \Omega(x_{hi}, \beta_P)$  Contains micro-structural information:  
instr. resolution, defects,  $\int_{-\infty}^{+\infty} \Omega(x) dx = 1$

$b_i = b_i(\beta_B)$  Background: noise, incoherent scattering

$$\Omega(x) = g(x) \otimes f(x) = \text{instrumental} \otimes \text{intrinsic profile}$$

$$y_{ci} = \sum_{\{h\}} I_h \Omega(T_i - T_h) + b_i$$

$$I_h = S \left\{ L p O A C F^2 \right\}_h$$

Integrated intensities are proportional to the square of the structure factor  $F$ . The factors are:  
 Scale Factor ( $S$ ), Lorentz-polarization ( $Lp$ ), preferred orientation ( $O$ ), absorption ( $A$ ), other “corrections” ( $C$ )

# The Structure Factor contains the structural parameters (isotropic case)

$$F(\mathbf{h}) = \sum_{j=1}^n O_j f_j(h) T_j \sum_s \exp \left\{ 2\pi i \left[ \mathbf{h} \{S|\mathbf{t}\}_s \mathbf{r}_j \right] \right\}$$

$$\mathbf{r}_j = (x_j, y_j, z_j) \quad (j = 1, 2, \dots, n)$$

$$T_j = \exp\left(-B_j \frac{\sin^2 \theta}{\lambda^2}\right)$$

# Structural Parameters (simplest case)

$$\mathbf{r}_j = (x_j, y_j, z_j)$$

**Atom positions (up to  $3n$  parameters)**

$$O_j = k \frac{m_j}{M}$$

**Occupation factors (up to  $n-1$  parameters)**

$$B_j$$

**Isotropic displacement (temperature) factors (up to  $n$  parameters)**

# Structural Parameters (complex cases)

**As in the simplest case plus additional (or alternative) parameters:**

- **Anisotropic temperature (displacement) factors**
- **Anharmonic temperature factors**
- **Special form-factors (Symmetry adapted spherical harmonics ), TLS for rigid molecules, etc.**
- **Magnetic moments, coefficients of Fourier components of magnetic moments , basis functions, etc.**

# The Structure Factor in complex cases

$$F(\mathbf{h}) = \sum_{j=1}^n O_j f_j(h) T_j \sum_s g_j(\mathbf{h}_s) \exp \left\{ 2\pi i \left[ \mathbf{h} \{S|\mathbf{t}\}_s \mathbf{r}_j \right] \right\}$$

$$\mathbf{h}_s = \begin{pmatrix} h \\ k \\ l \end{pmatrix}_s = S_s^T \begin{pmatrix} h \\ k \\ l \end{pmatrix} \quad (s = 1, 2, \dots, N_G)$$

$g_j(\mathbf{h}_s)$

Complex form factor of object  $j$   
Anisotropic DPs  
Anharmonic DPs



## Example: General $2\theta$ dependence of the instrumental broadening (determined by a standard sample)

$$H_{hG}^2 = (U_f + (1 - \xi_f)^2 D_{fST}^2(\alpha_D)) \tan^2 \theta + \frac{I_{fG}}{\cos^2 \theta} + H_{gG}^2$$
$$H_{hL} = (X_f + \xi_f D_{fST}(\alpha_D)) \tan \theta + \frac{[Y_f + F_f(\alpha_S)]}{\cos \theta} + H_{gL}$$

The Gaussian and Lorentzian components of the instrumental Voigt function are interpolated between empirically determined values.

If needed, axial divergence is convoluted numerically with the resulting profile.

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**Quantitative phase analysis.**

# The Rietveld Method

The **Rietveld Method** consist of refining a crystal (and/or magnetic) structure by minimising the weighted squared difference between the observed and the calculated pattern against the parameter vector:  $\beta$

$$\chi^2 = \sum_{i=1}^n w_i \{y_i - y_{ci}(\beta)\}^2$$

$$w_i = \frac{1}{\sigma_i^2}$$

$\sigma_i^2$  : is the variance of the "observation"  $y_i$

# Least squares: Gauss-Newton (1)

Minimum necessary condition:  $\frac{\partial \chi^2}{\partial \beta} = 0$

A Taylor expansion of  $y_{ic}(\beta)$  around  $\beta_0$  allows the application of an iterative process. The shifts to be applied to the parameters at each cycle for improving  $\chi^2$  are obtained by solving a linear system of equations (normal equations)

$$\mathbf{A} \delta_{\beta_0} = \mathbf{b}$$
$$A_{kl} = \sum_i w_i \frac{\partial y_{ic}(\beta_0)}{\partial \beta_k} \frac{\partial y_{ic}(\beta_0)}{\partial \beta_l}$$
$$b_k = \sum_i w_i (y_i - y_{ic}) \frac{\partial y_{ic}(\beta_0)}{\partial \beta_k}$$

# Least squares: Gauss-Newton (2)

The shifts of the parameters obtained by solving the normal equations are added to the starting parameters giving rise to a new set

$$\beta_1 = \beta_0 + m.\delta_{\beta_0}$$

The new parameters are considered as the starting ones in the next cycle and the process is repeated until a convergence criterion is satisfied. The variances of the adjusted parameters are calculated by the expression:

$$\sigma^2(\beta_k) = (\mathbf{A}^{-1})_{kk} \chi_v^2$$

$$\chi_v^2 = \frac{\chi^2}{N - P + C}$$

# Least squares: a local optimisation method

- The least squares procedure provides (when it converges) the value of the parameters constituting the local minimum closest to the starting point
- A set of good starting values for all parameters is needed
- If the initial model is bad for some reasons the LSQ procedure will not converge, it may diverge.

# R-factors and Rietveld Refinement (1)

$$R_p = 100 \frac{\sum_i |y_{obs,i} - y_{calc,i}|}{\sum_i |y_{obs,i}|}$$

**R-pattern**

$$R_{wp} = 100 \left[ \frac{\sum_i w_i |y_{obs,i} - y_{calc,i}|^2}{\sum_i w_i |y_{obs,i}|^2} \right]^{1/2}$$

**R-weighted pattern**

$$R_{exp} = 100 \left[ \frac{(N - P + C)}{\sum_i w_i y_{obs,i}^2} \right]^{1/2}$$

**Expected R-weighted pattern**

# R-factors and Rietveld Refinement (2)

$$\chi^2_v = \left[ \frac{R_{wp}}{R_{exp}} \right]^2$$

**Reduced Chi-square**

$$S = \frac{R_{wp}}{R_{exp}}$$

**Goodness of Fit indicator**



# R-factors and Rietveld Refinement (3)

## Two important things:

- The sums over “ $i$ ” may be extended only to the regions where Bragg reflections contribute
- The denominators in  $R_P$  and  $R_{WP}$  may or not contain the background contribution

# Crystallographic R-factors used in Rietveld Refinement

$$R_B = 100 \frac{\sum_k |I_{obs,k} - I_{calc,k}|}{\sum_k |I_{obs,k}|}$$

Bragg R-factor

$$R_F = 100 \frac{\sum_k |F_{obs,k} - F_{calc,k}|}{\sum_k |F_{obs,k}|}$$

Crystallographic  $R_F$ -factor.

# Crystallographic R-factors used in Rietveld Refinement

$$'I_{obs,k}' = I_{calc,k} \sum_i \left\{ \frac{\Omega(T_i - T_k)(y_{obs,i} - B_i)}{(y_{calc,i} - B_i)} \right\}$$

Provides 'observed'  
integrates intensities for  
calculating Bragg R-factor

$$'F_{obs,k}' = \sqrt{\frac{'I_{obs,k}'}{jLp}}$$

In some programs the crystallographic  
 $R_F$ -factor is calculated using just the  
square root of ' $I_{obs,k}$ '

# Strategy for setting up a Rietveld refinement (1)

**Use the best possible starting model: this can be easily done for background parameters and lattice constants**

**Collect all the information available both on your sample (approximate cell parameters and atomic positions) and on the diffractometer and experimental conditions**

**Do not start by refining all structural parameters at the same time. Some of them affect strongly the residuals (they must be refined first) while others produce only little improvement.**

# Strategy for setting up a Rietveld refinement (2)

**A sensible sequence for the refinement of a crystal structure:**

Scale factor

Zero point, background parameters (if appropriate) and lattice constants.

Atomic positions and displacement parameters

Peak shape and asymmetry parameters.

Atom occupancies (if required).

Microstructural parameters: size and strain effects.

**It is essential to plot frequently the observed and experimental patterns.**

The examination of the difference pattern is a quick and efficient method to detect blunders in the model or in the input file controlling the refinement process. I may also provide useful hints on the best sequence to refine the whole set of model parameters for each particular case.

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**Quantitative phase analysis.**

# Quantitative Phase analysis with the Rietveld Method

The scale factor used in the Rietveld method is proportional to the quantity of corresponding crystalline phase

$$y_i = \sum_{\phi} S_{\phi} \left( \sum_{\mathbf{h}} \mathbf{I}_{\mathbf{h}} \Omega (T_{\mathbf{h}} - T_i) \right)_{\phi} + b_i$$
$$S_{\phi} = \frac{C}{\bar{\mu}} \frac{W_{\phi}}{(ZMV)_{\phi}}$$

D.L.Bish & S.A.Howard, J.Appl.Cryst. **21**, 86 (1988)

# Scale Factors

Experimental constant

Weight fraction of phase  $\phi$

$$S_{\phi} = \frac{C}{\bar{\mu}} \frac{W_{\phi}}{(ZMV)_{\phi}}$$

Average absorption coefficient

Number of formula units of phase  $\phi$

Molecular weight of phase  $\phi$

Unit cell volume of phase  $\phi$

D.L.Bish & S.A.Howard, J.Appl.Cryst. **21**, 86 (1988)



# Quantitative Phase analysis with the Rietveld Method

If all phases are well crystallized one can constraint the sum of the weight fractions to 1, so that:

$$W_{\phi} = \frac{S_{\phi}(ZMV)_{\phi}}{\sum_{i=1,...n} S_i(ZMV)_i}$$

$$W_{\phi} = \frac{S_{\phi}(ZMV)_{\phi} / \tau_{\phi}}{\left[ \sum_{i=1,...n} S_i(ZMV)_i / \tau_i \right]}$$

Micro-absorption  
Brindley coefficients

D.L.Bish & S.A.Howard, J.Appl.Cryst. **21**, 86 (1988)

# Quantitative Phase analysis with the Rietveld Method

**Micro-absorption phenomena can be accounted through Brindley considerations:**

**Classification of powders according to the value of  $\mu r$  product  
( $\mu$ : linear absorption coefficient;  $r$ : linear size of particles)**

- . Fine powders:  $\mu r < 0.01$
- . Medium powders:  $0.01 < \mu r < 0.1$
- . Coarse powders:  $0.1 < \mu r < 1.0$
- . Very coarse powders:  $\mu r > 1.0$

# Brindley coefficients

**The Brindley coefficients can be calculated iteratively by starting with the weight fractions obtained when all  $\tau = 1$  and using the empirical formula:**

$$\tau_{\phi} = 1 - 1.450(\mu_{\phi} - \bar{\mu})r + 1.426[(\mu_{\phi} - \bar{\mu})r]^2$$

**Expression valid for low absorption contrast**  $-0.1 \leq (\mu_{\phi} - \bar{\mu})r \leq 0.1$

**Mean Powders (Brindley):**  $0.01 \leq 2r\mu_{\phi} \leq 0.1$

**$r$  is the mean crystallite radius and  $\mu_{\phi}$  the linear absorption coefficient**

G.W. Brindley, Philosophical Magazine. **36**, 347 (1945)

# Quantitative Phase analysis with the Rietveld Method

with

$$W_{\phi} = \frac{S_{\phi}(ZMV)_{\phi} \cdot f_{\phi}^2 / \tau_{\phi}}{\sum_{i=1}^{N_{\phi}} S_i \cdot (ZMV)_i \cdot f_i^2 / \tau_i} = \frac{S_{\phi} ATZ_{\phi} \cdot V_{\phi}}{\sum_{i=1}^{N_{\phi}} S_i ATZ_i \cdot V_i}$$

$S_{\phi}$

Scale factor in FullProf (refinable variable)

$$ATZ_i = Z_i M_i f_i^2 / \tau_i$$

FullProf parameter

$\tau_i$

Brindley factor (particle absorption contrast factor).

$\tau$  is tabulated as a function of  $(\mu_i - \mu) \cdot r$   
FullProf parameter

# Quantitative Phase analysis with the Rietveld Method

$f_i$

Used to transform the site multiplicities in PCR FullProf input file, to their real values. For a stoichiometric phase,  $f = 1$  if these multiplicities are calculated by dividing the Wyckoff multiplicity  $m$  of the site by the general multiplicity  $M$  of the space group. Otherwise,  $f = occ.M/m$ , where  $occ.$  is the occupation number in the PCR file



In order to GET PROPER VALUES OF WEIGHT FRACTIONS LET THE PROGRAM RE-CALCULATE **ATZ** by putting them to ZERO.

The correct **ATZ** value is rewritten in the PCR file.

# Quantitative Phase analysis with the Rietveld Method

1. Crystal structure has to be refined:

`JBT=0 (IRF=0)`

→ Refine the structural parameters as usually

2. **Crystal structure is well known:**

2.1 Create hkl file containing hkl list with corresponding  $F^2$  (`JLKH=5`)

2.2 Refine the pattern without entering atomic positions  
`JBT=-3, IRF=2` (Le Bail fit mode with constant relative intensities for the current phase, but refinable scale factor)

# Quantitative Phase analysis with the Rietveld Method

- 😊 ✓ easy to operate (automatic analysis in FullProf)
  - ✓ no internal standard
  - ✓ non destructive method
  - ✓ up to 16 phases in FullProf
  - ✓ polymorphism, microstructure
  - ✓ neutron case: large amounts of powder analysis (real samples)
  - ✓ industrial applications (cements, clays ...)
- 😞 ✓ structure model dependent:  $\{F_{hkl}\}$  have to be known
  - ✓ beware of preferred orientation

# Some references on Q.P.A.by Rietveld method

➤ R.J. Hill & C.J. Howard, *J. Appl. Cryst.* 20, 467-476 (1987)

Quantitative phase analysis from neutron powder diffraction data using the Rietveld method

➤ G.W. Brindley, *Phil. Mag.* 36, 347-369 (1945)

The effect of grain or particle size on X-ray reflections from mixed powders and alloys considered in relation to the quantitative determination of crystalline substances by X-ray methods

➤ D.L. Bish & S.A. Howard, *J. Appl. Cryst.* 21, 86-91 (1988)

Quantitative phase analysis using the Rietveld method

➤ J.C. Taylor, *Powder Diffraction* 6, 2-9 (1991)

Computer programs for standardless quantitative analysis of minerals using the full powder diffraction profile

➤ R.J. Hill, *Powder Diffraction* 6, 74-77 (1991)

Expanded use of Rietveld method in studies of phase abundance in multiphase mixtures