

X-ray Absorption Spectroscopy applied to Energy-related Materials

Valérie Briois

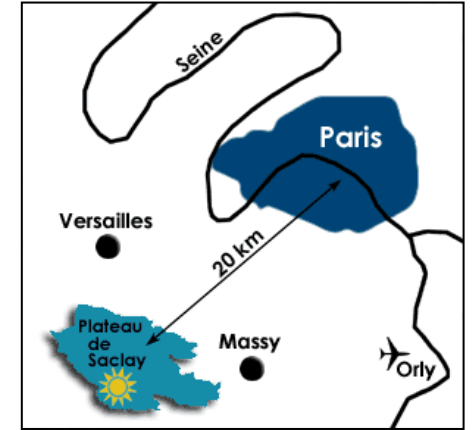
briois@synchrotron-soleil.fr



SOLEIL Synchrotron



Storage ring nominal energy: 2.75 GeV
 (354 m de circumference)
 Stored Current: 450 mA/500 mA
 In operation since 2007



18/12/2015
 14:59:43
 Function Mode: TOP-UP
 Filling Mode: 4/4
 Lifetime: 14.28 h
 Integrated Current: 14645.2 A.h
 Average Pressure: 5.1e-10 mbar

Bending Magnet		Insertion Devices		
ODE	I02_C	PSICHE	PLEIADES	
MARS	DESIRS	PUMA	CRISTAL	
DISCO	DEIMOS	GALAXIES	TEMPO	
METRO	I09_L	HERMES	PX1	
SAMBA	PX2	SWING	ANTARES	
ROCK	ANATOMIX	NANOSCOPIUM	SEXTANTS	
DIFFABS	SIXS	CASSIOPEE	SIRIUS	
	LUCIA			

Infrared: SMIS, AILES

	Delivery Since	End Of Beam	Remaining Time	Orbit(RMS)	Emittance	Tune
	Tue Dec 15 07:00	Dec-21 07:00	64:00:18	H: 46.6 μm	4.46 nm.rad	0.1578
				V: 69.9 μm	47.5 pm.rad	0.2280

SOLEIL SYNCHROTRON

Shift Lignes

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Outline

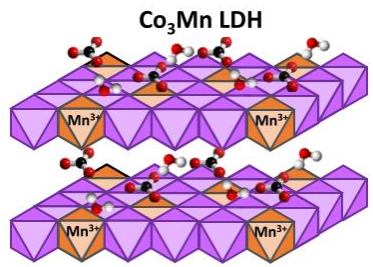
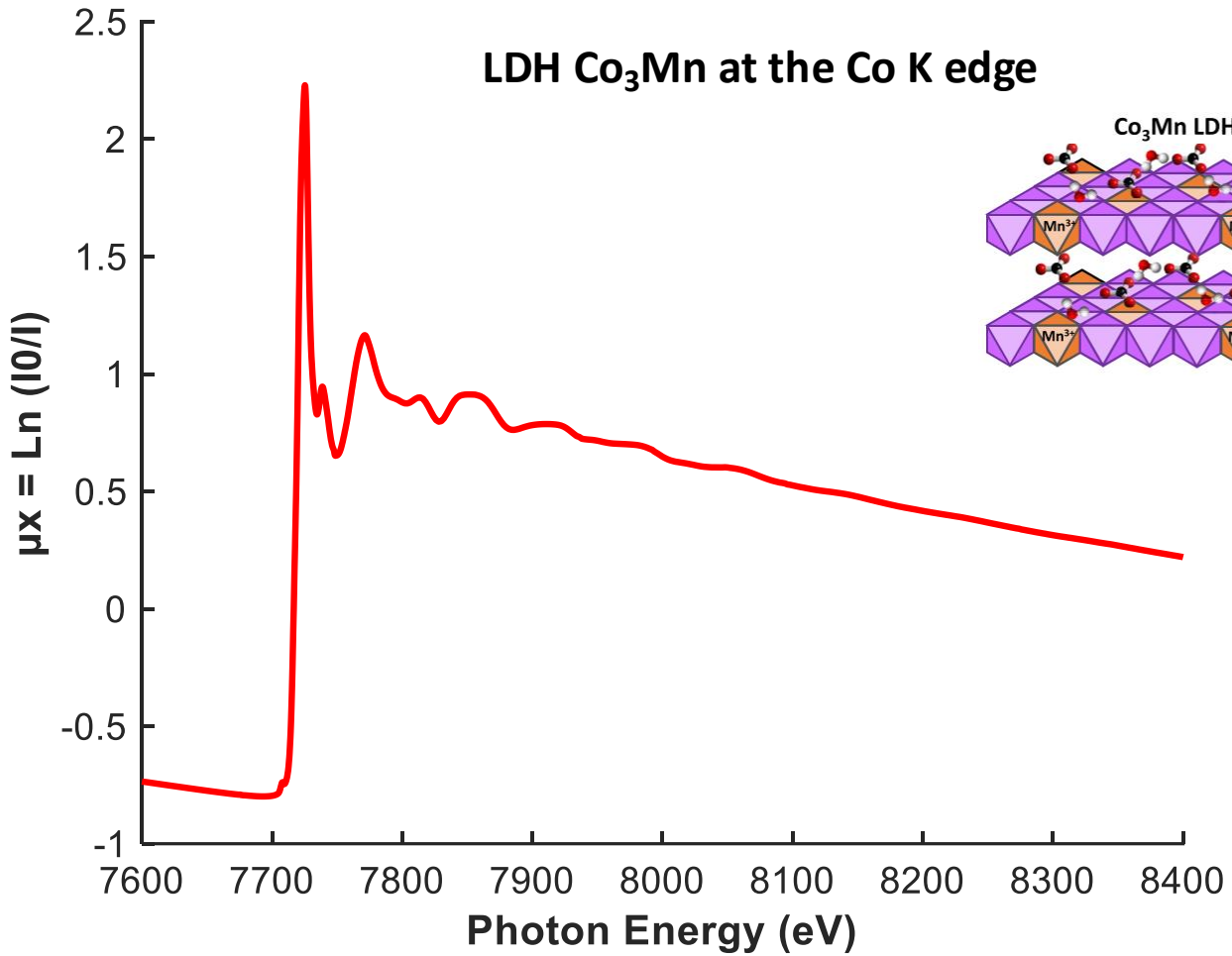
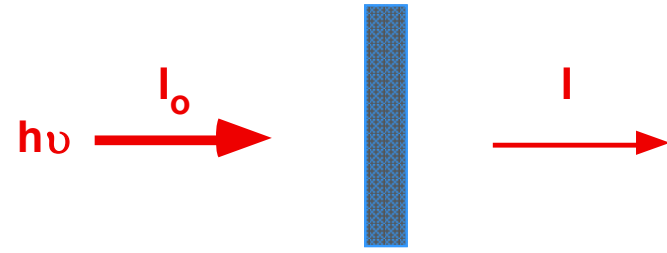
- 1) XAS Introduction
- 2) ROCK: A Quick-EXAFS Beamline for Energy-related Materials
 - Sample Environment
 - Multivariate Analysis
- 3) From Quick-EXAFS to Full-Field Hyperspectral Quick-EXAFS Imaging



X-Ray Absorption Spectroscopy

$$\mu_x = \ln I_0 / I \text{ versus } h\nu$$

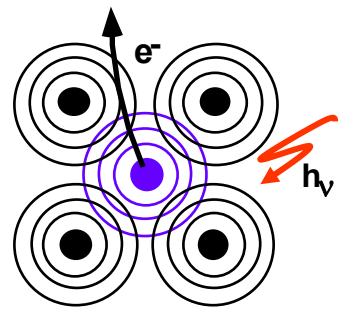
Beer-Lambert law



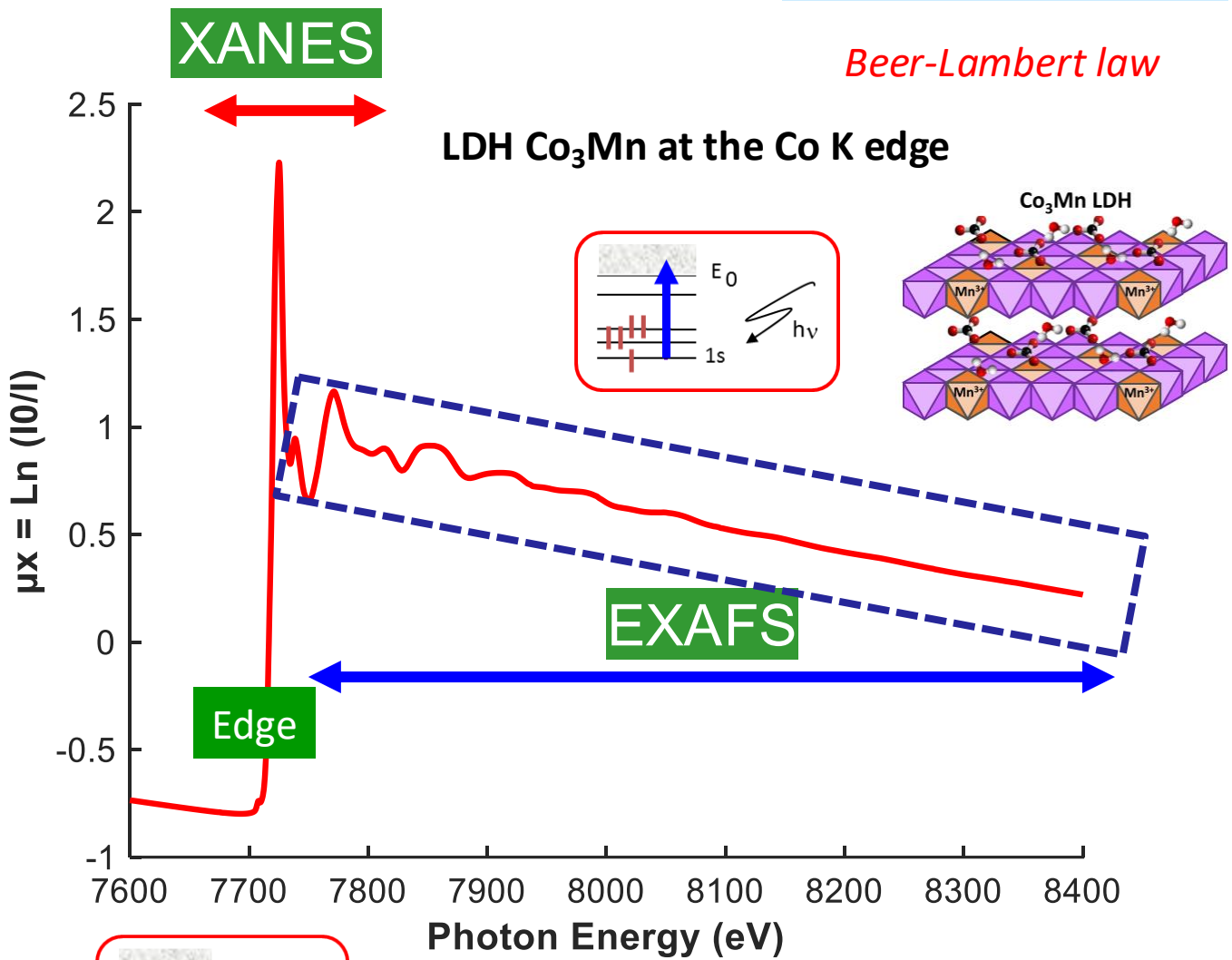
X-Ray Absorption Spectroscopy

$\mu x = \ln I_0 / I$ versus $h\nu$

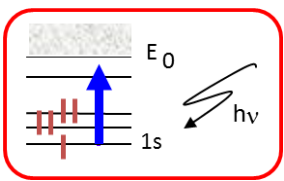
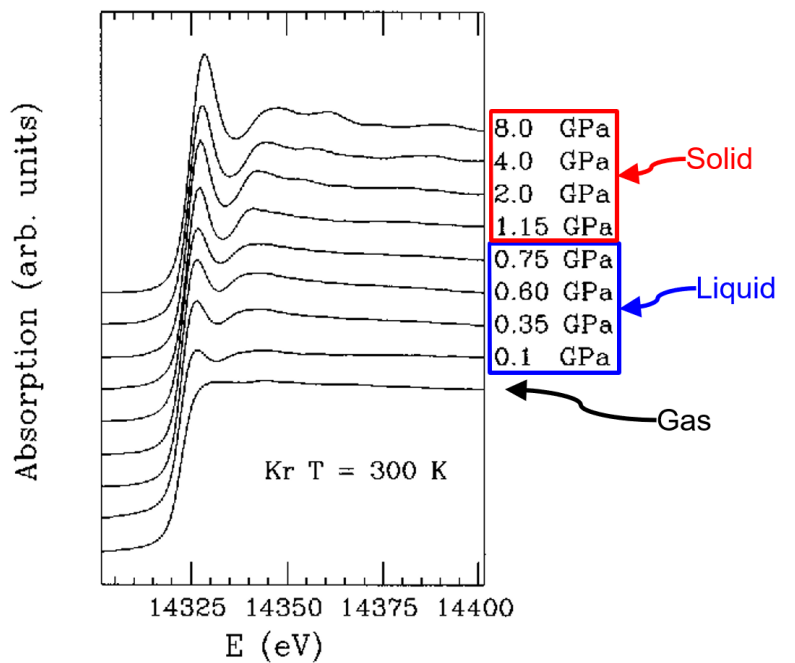
Beer-Lambert law



XAFS oscillations
Interferences between
Outgoing wave and
Backscattered waves



High-pressure EXAFS of solid and liquid Kr



XANES : X-ray Absorption Near Edge Structure

EXAFS: Extended X-ray Absorption Fine Structures

EXAFS signal = Sum of damped sine waves, each being related to Contributions of neighbours around the absorbing atom at a given distance

EXAFS Formula : One-electron and single scattering

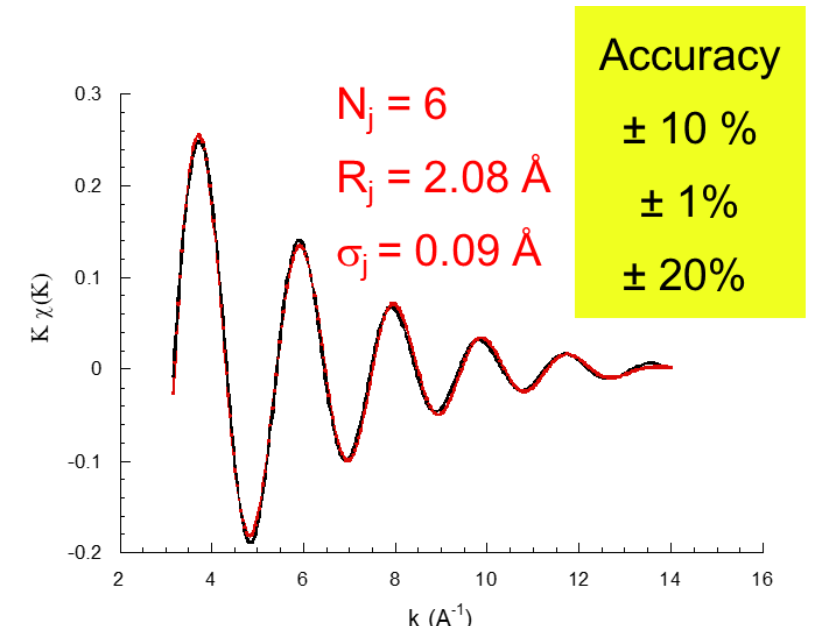
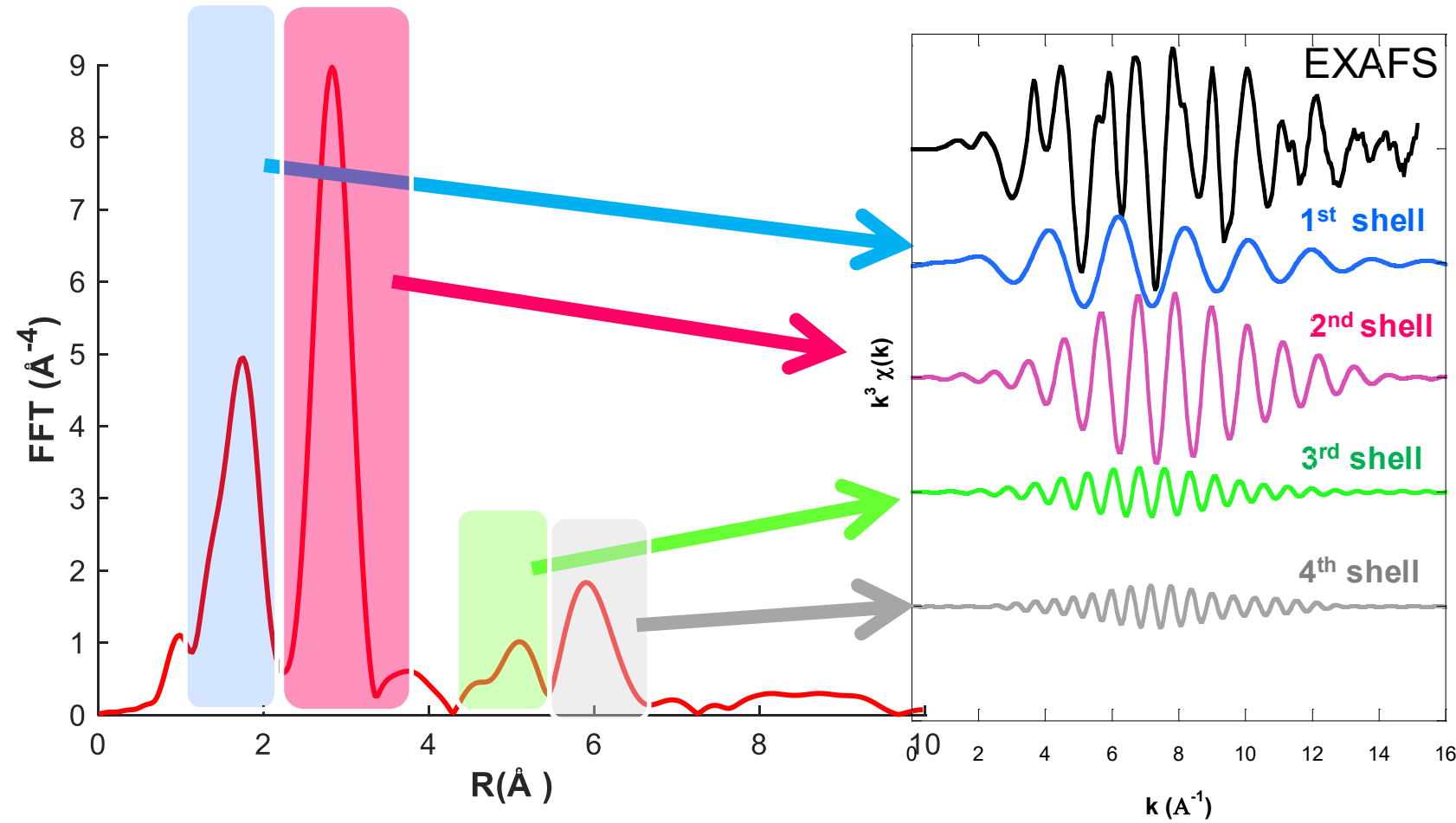
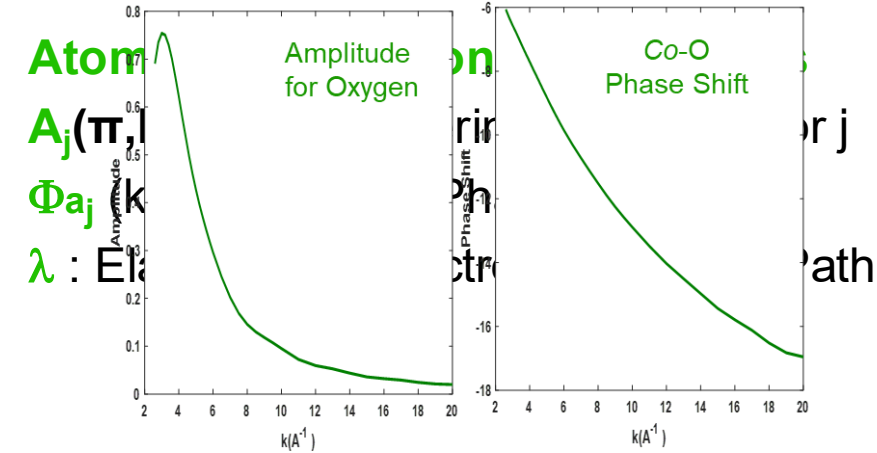
$$\chi(k) = - S_0^2 \sum_j \frac{N_j}{kR_j^2} \frac{\exp(-2R_j)}{\lambda} \exp(-2\sigma_j^2 k^2) A_j(\pi, k) \sin(2kR_j + \Phi_{aj}(k))$$

Structural Parameters related to the j^{th} shell

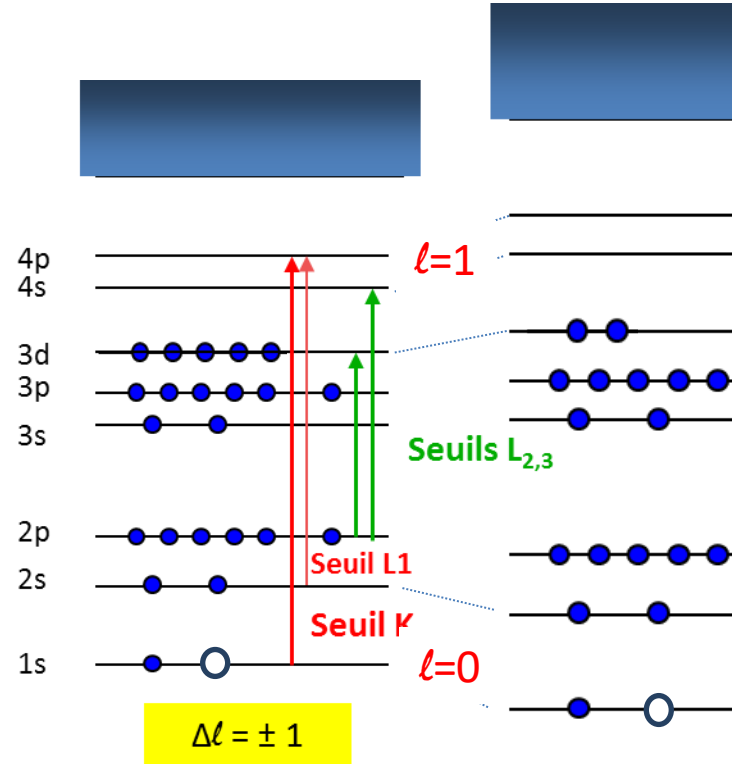
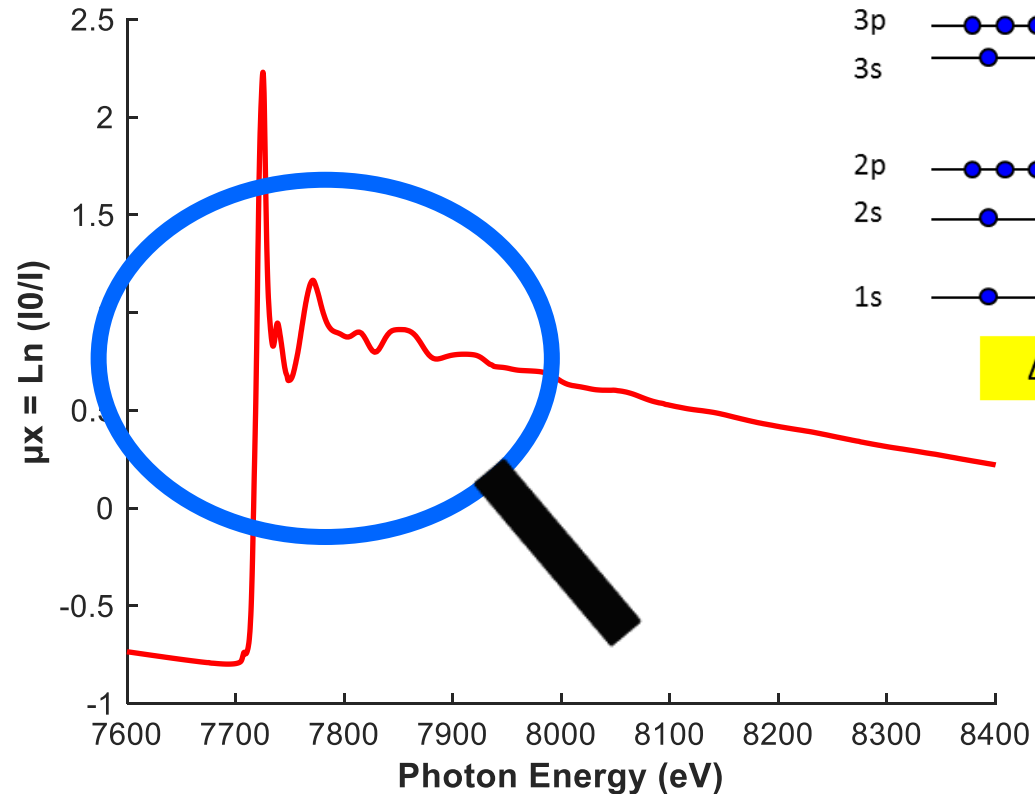
N_j : Coordination number

R_j : Absorber-Neighbour Distance

σ_j : Debye-Waller Factor

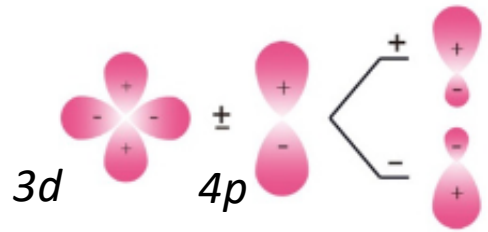


A few words about the XANES

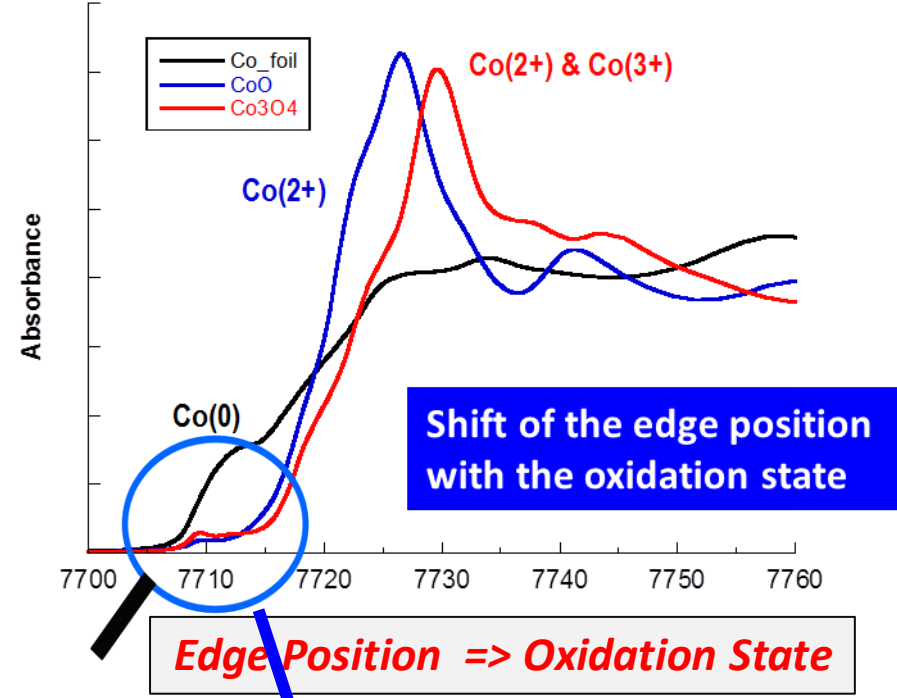
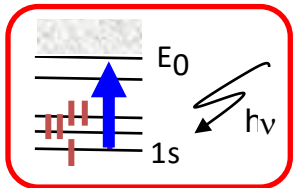


OXIDATION →

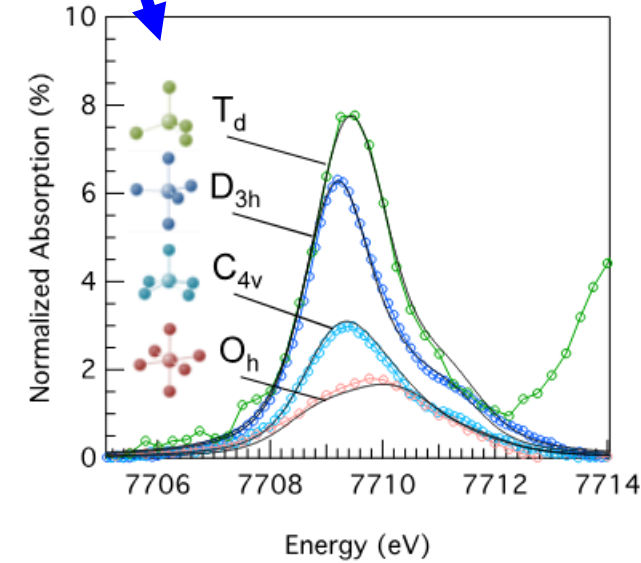
d-p hybridization



Non-Centrosymmetric Group



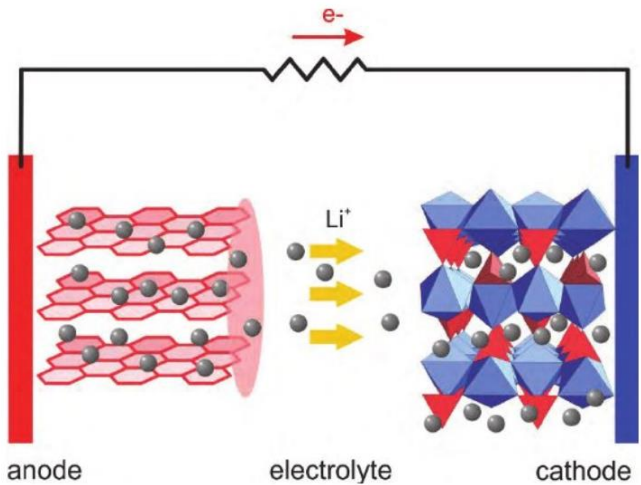
M. Hunault et al. J. Phys.: Conf. Series 712 (2016) 012005



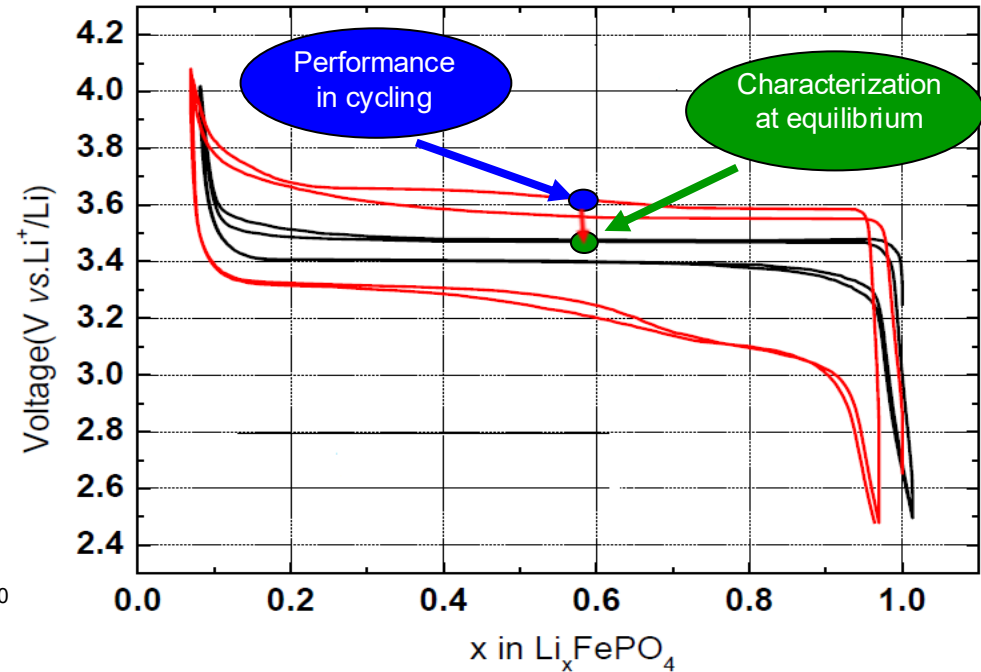
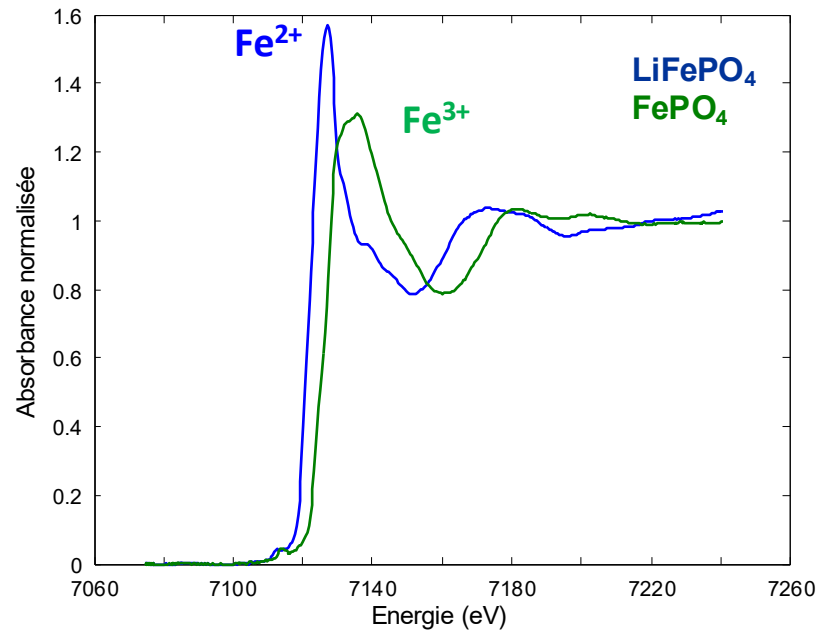
Intensity of Pre-Edge => Symmetry

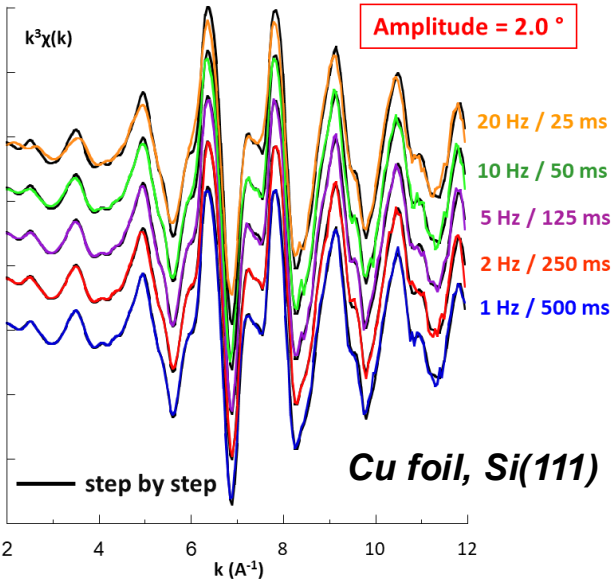
X-ray Absorption Spectroscopy is an element-selective technique providing powerful information on :

- the local order around the absorbing atom (distance, N, Ligand...)
- oxidation state
- local point group symmetry

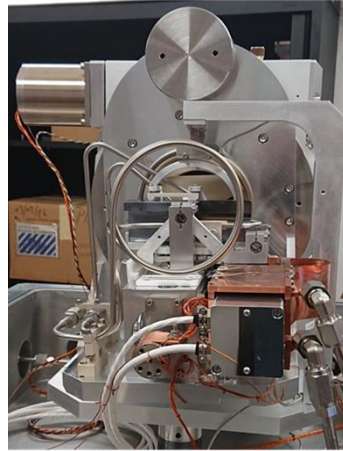


DOI: 10.5772/21635



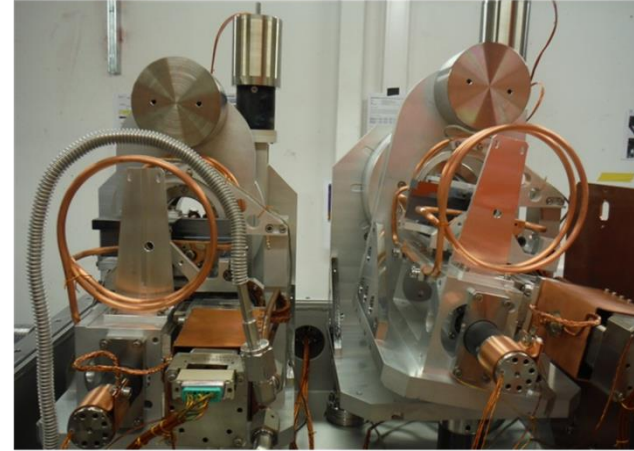


1 LN₂-cooled monochromator



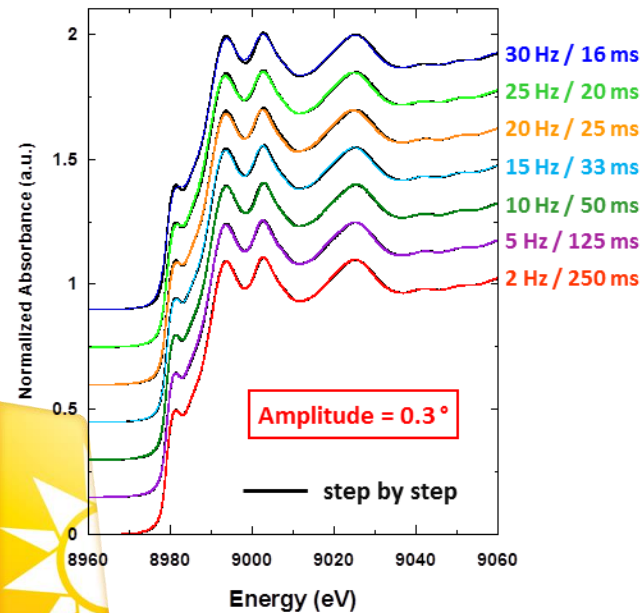
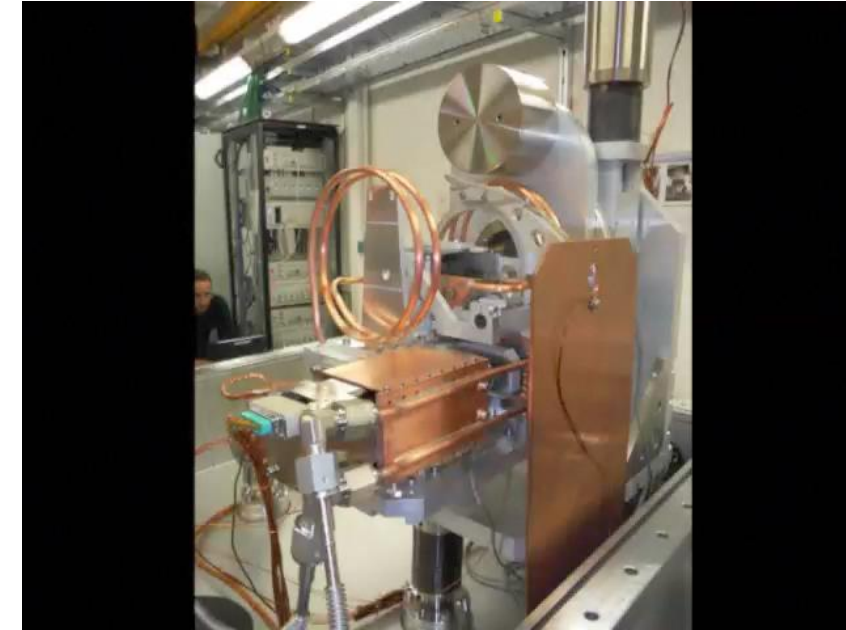
Si(220)
6 - 43 keV

2 water-cooled monochromators



Si(111)
4 - 26 keV

Si(111)
4 - 21 keV

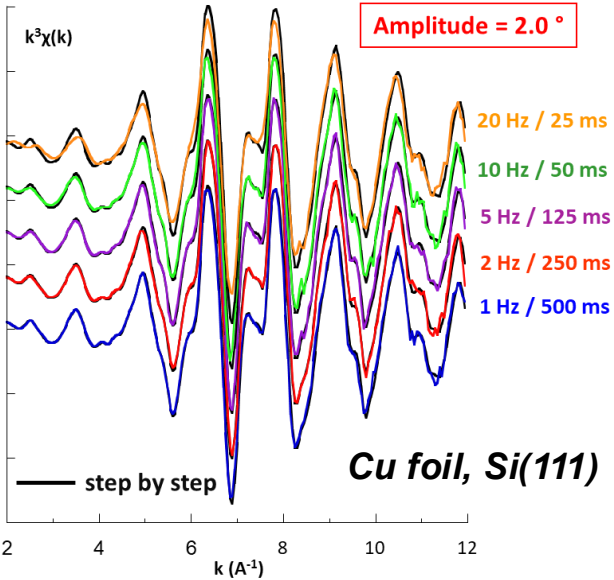


Source: 2.81 T Super Bend
Energy Range: 4 – 42 keV
Flux @ 8.5 keV: Si(111) : $1.7 \cdot 10^{12}$ ph/s
Flux @ 20 keV: Si(111) : $1.0 \cdot 10^{12}$ ph/s
Flux @ 20 keV: Si(220) : $5.0 \cdot 10^{11}$ ph/s
Flux @ 35 keV: Si(220) : $1.5 \cdot 10^{11}$ ph/s
Time resolution in the ms range

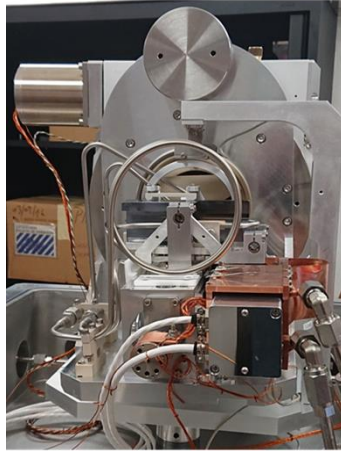
Briois et al., J. Phys.: Conf. Ser. 712 (2016) 012149

Oscillation frequency : 0.1 Hz to 30 Hz
 for one spectrum => 16 ms to 5s

Oscillation amplitude range: 0.3 to 3.9°
 - at least 1000 eV spectra (EXAFS)

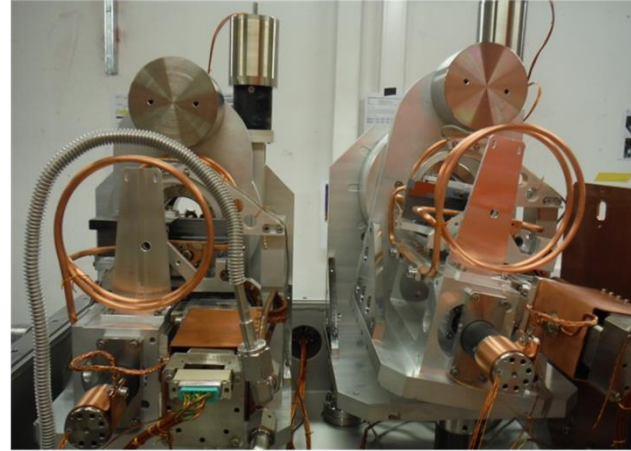


1 LN₂-cooled monochromator



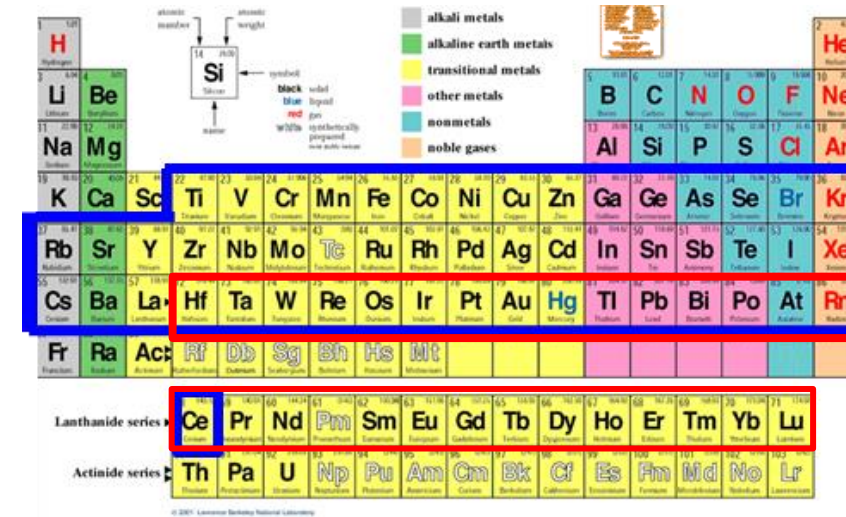
Si(220)
6 - 43 keV

2 water-cooled monochromators

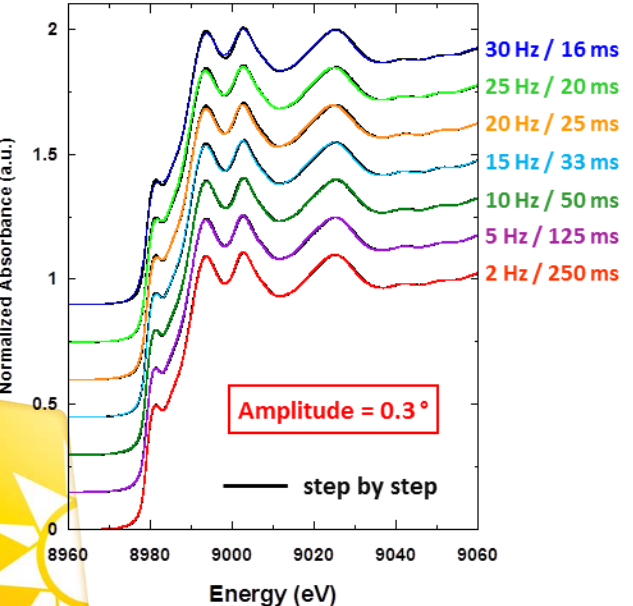


Si(111)
4 - 26 keV

Si(111)
4 - 21 keV



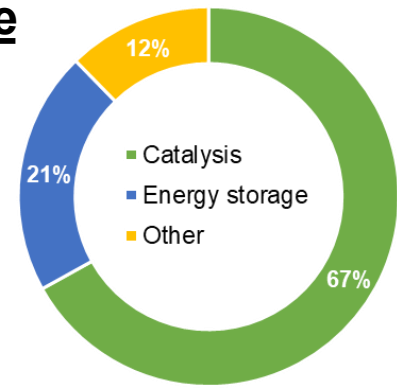
65 elements can be measured ...



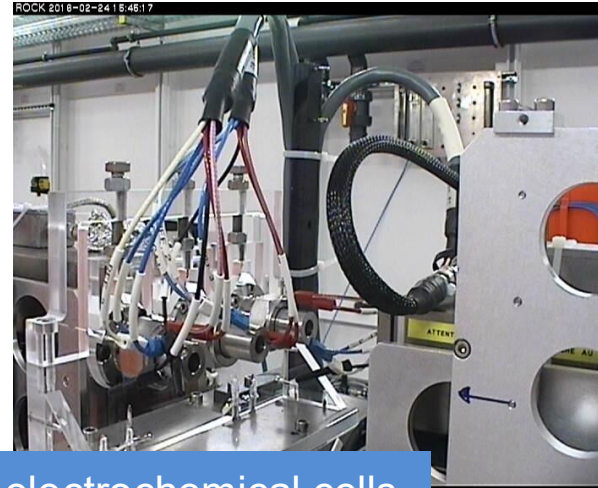
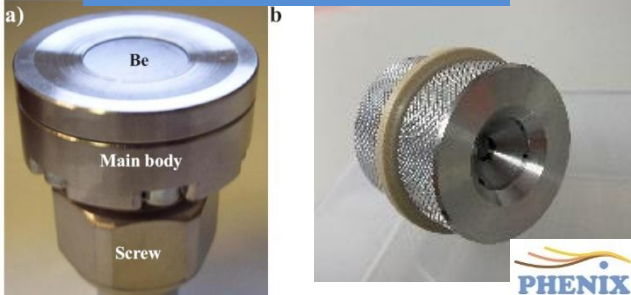
Source: 2.81 T Super Bend
Energy Range: 4 – 42 keV
Flux @ 8.5 keV: Si(111) : 1.7 10¹² ph/s
Flux @ 20 keV: Si(111) : 1.0 10¹² ph/s
Flux @ 20 keV: Si(220) : 5.0 10¹¹ ph/s
Flux @ 35 keV: Si(220) : 1.5 10¹¹ ph/s
Time resolution in the ms range

Briois et al., J. Phys.: Conf. Ser. 712 (2016) 012149

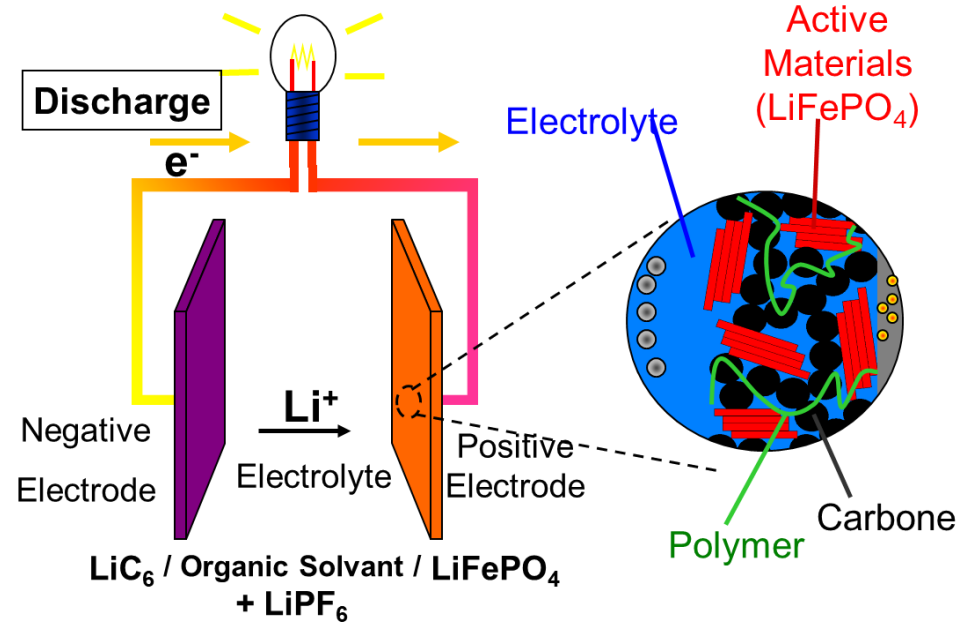
Science



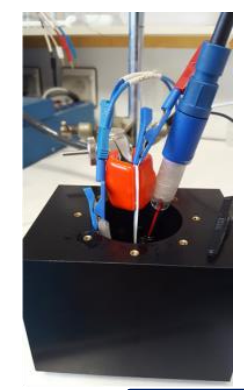
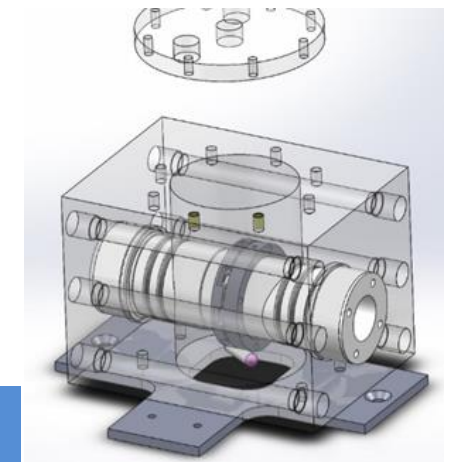
Half Cell Geometry



Multi-electrochemical cells

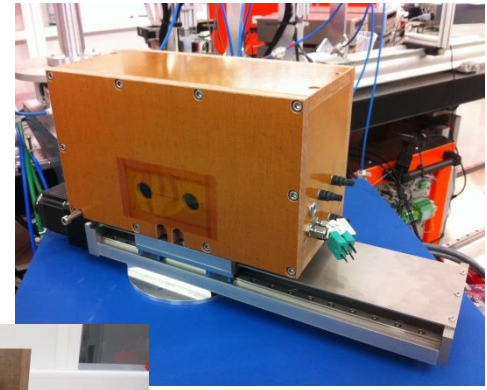
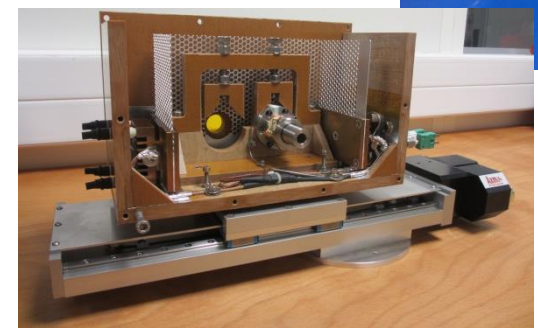


Electrochemical cell



Super-Capacitors

Thermostated box for electrochemical cells



JES, 157, (2010) A606-A610

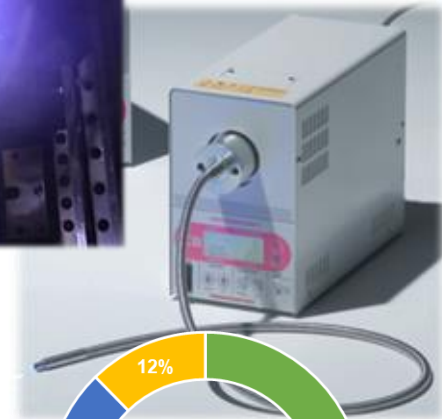


VMP3 Bio-Logic multi-channel potentiostat

Li-Battery
Na-Battery
Mg-Battery

Electrocatalysis

Photo-catalysis



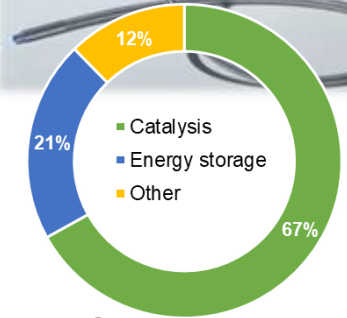
HER & OER
CO₂ photoconversion
Ethanol photoconversion

(c) Gas delivery systems

Thermal Catalysis

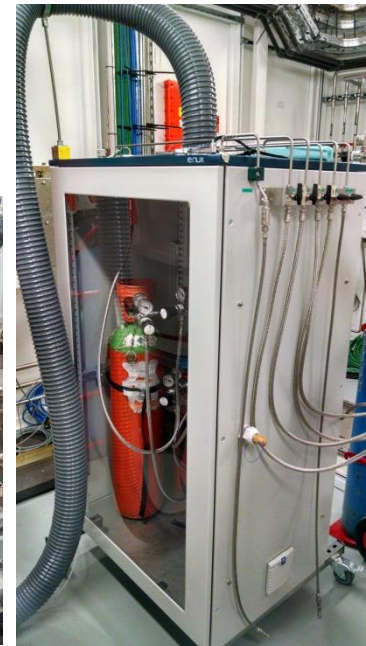
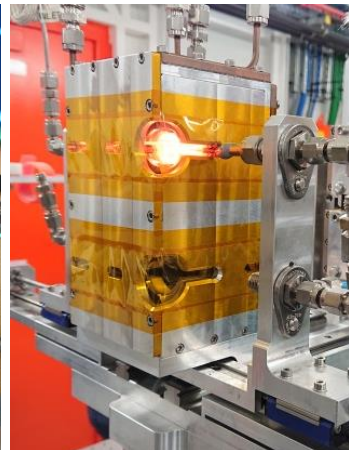
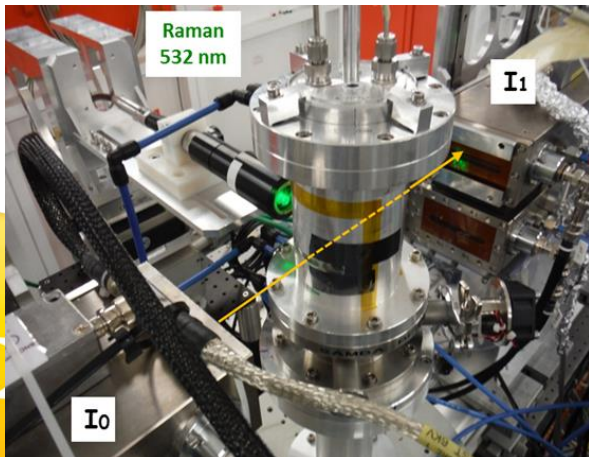
(a) Lytle-type cells
1 bar, RT – 600°C

(b) Capillary cells
20 bar, RT – 1000°C



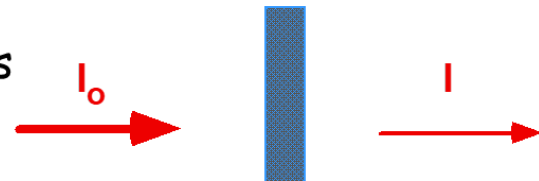
Science

Ethanol Steam Reforming
Dry Methane reforming
Fischer Tropsch
HDS & HDO
CO₂ – CO hydrogenation
Chemical Looping
Combustion

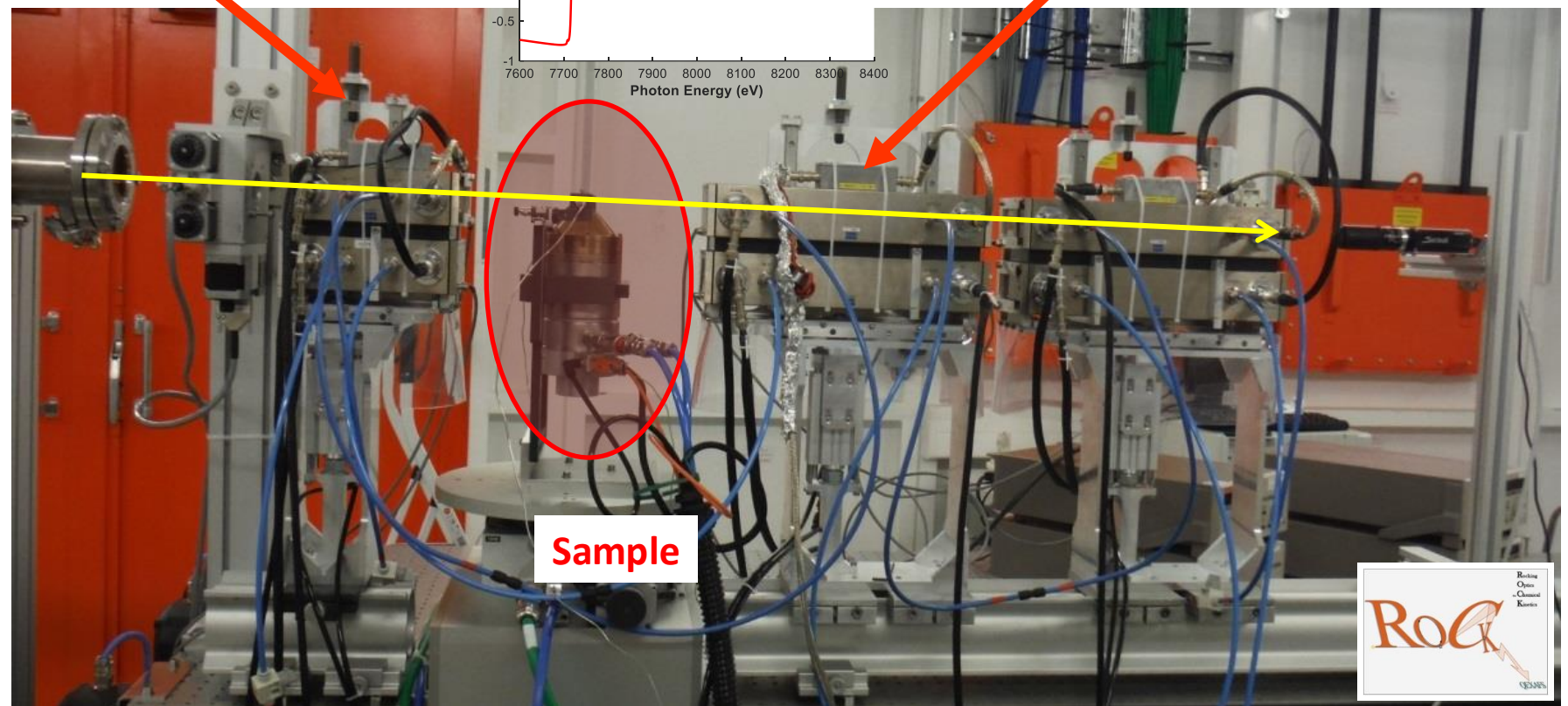
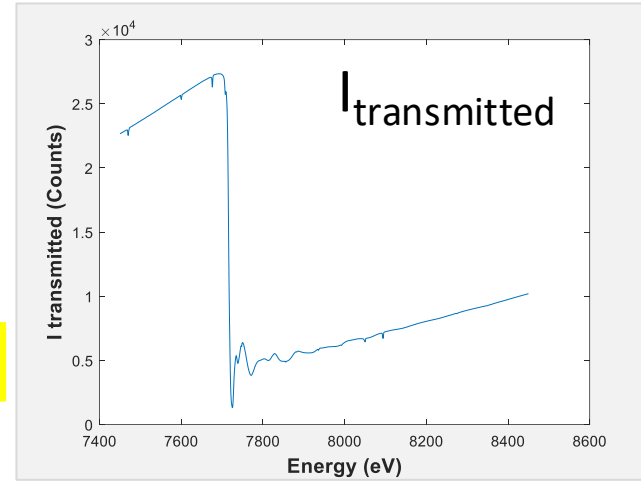
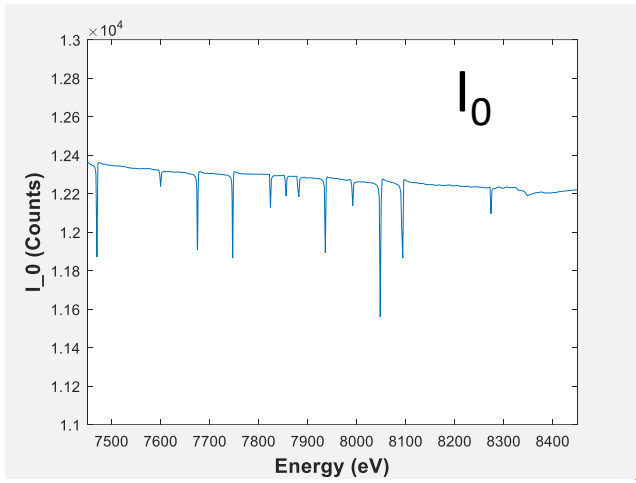
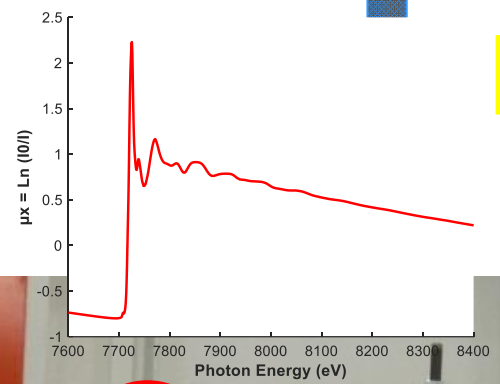


Detection Modes for XAS

Transmission Mode
Concentrated Systems



$$\mu x = \text{Ln}(I_0/I)$$



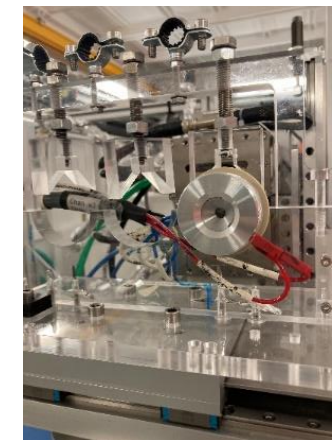
$\mu x = \text{Ln } I_0 / I$
versus $h\nu$
Beer-Lambert Law



G. Moëhl, L. Pérez Ramirez, A. Iadecola and S. Belin.

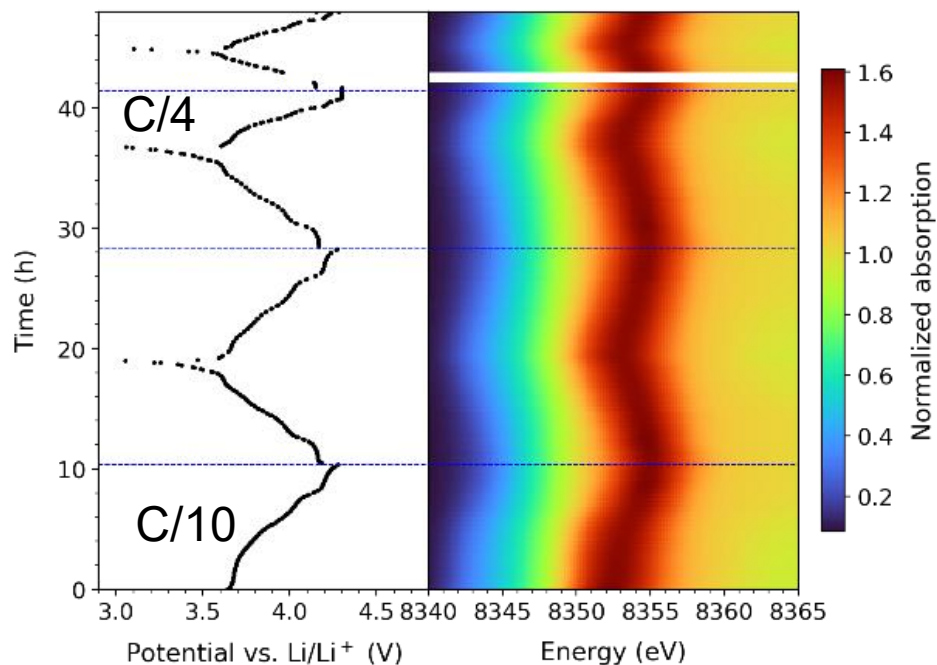
New cell design :

Gilles Moëhl & Laurent Barthe

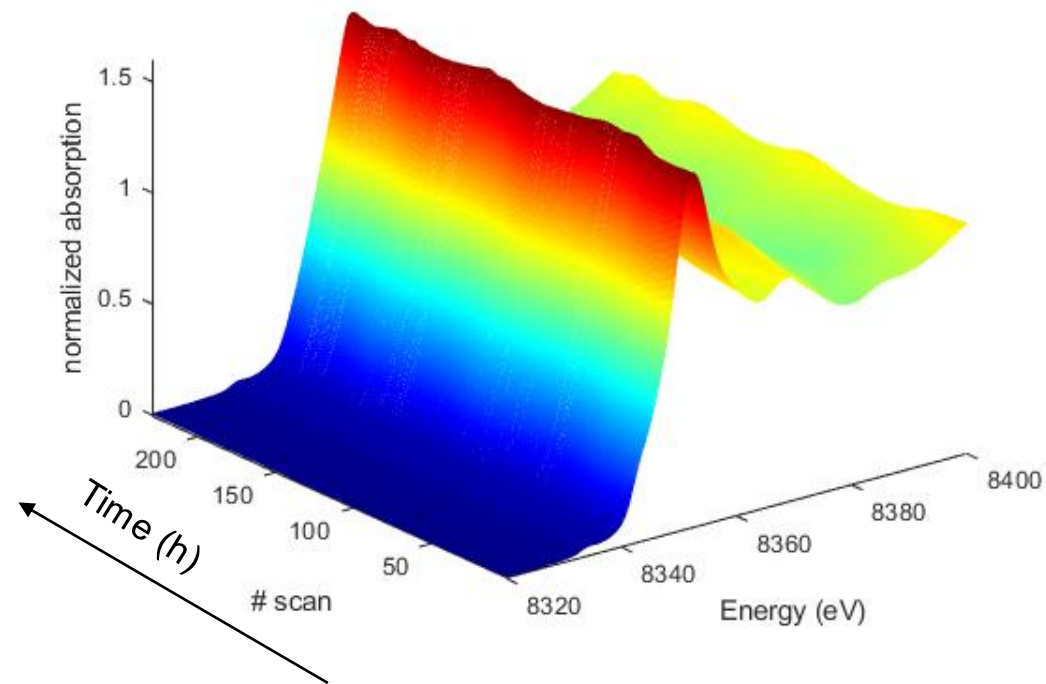


BIG-MAP

- Diamond peek-mak: LNO BASF/Celgard/Lithium, 28 μ L LP57 electrolyte, 4.5 bar pressure
- C=225 mAh 3-4.3 V vs lithium 2 cycles at C/10 + 2 cycles at C/4
- Speed 2Hz (250ms) 2 positions recorded, average time = 2min



229 Ni K-edge XAS spectra

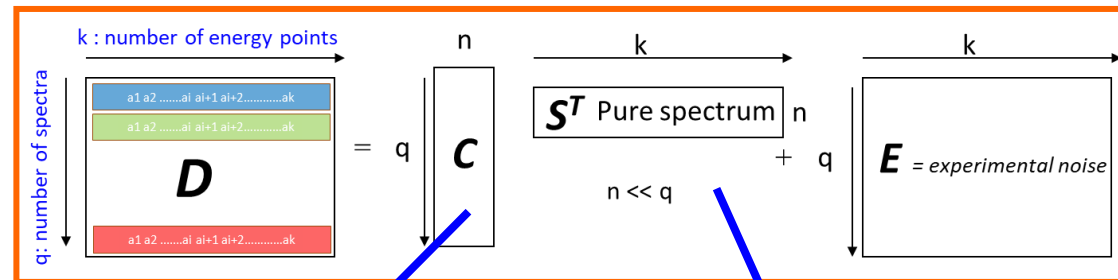
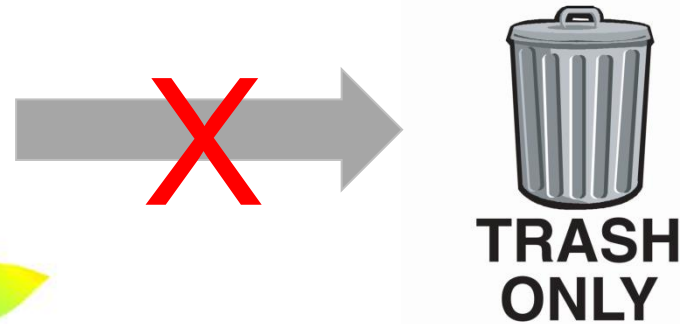
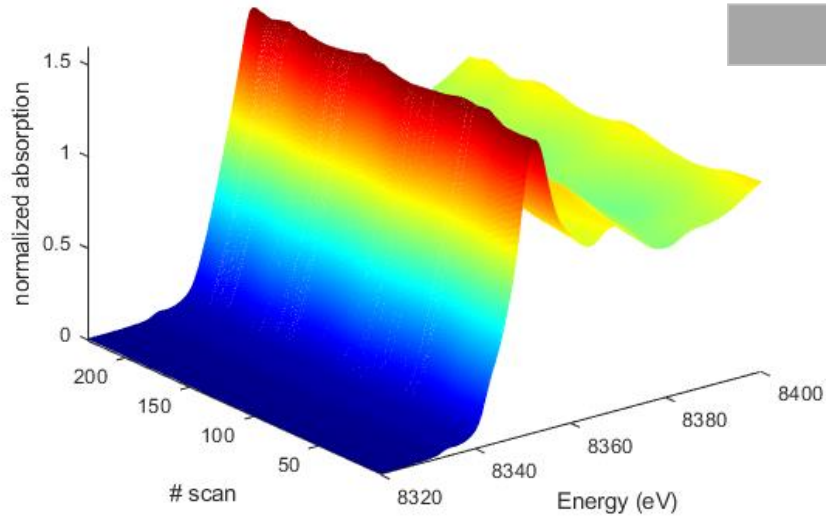


Galvanostatic Cycling curve

Bilinear Decomposition is fully verified for XAS
(Beer-Lambert Law $Abs = \sum C_i Abs_i$)

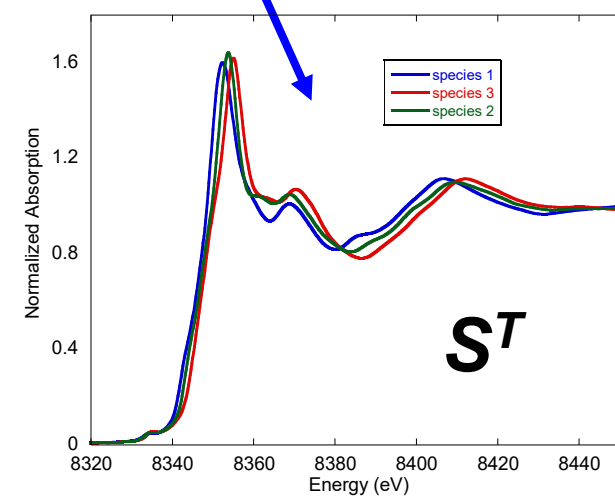
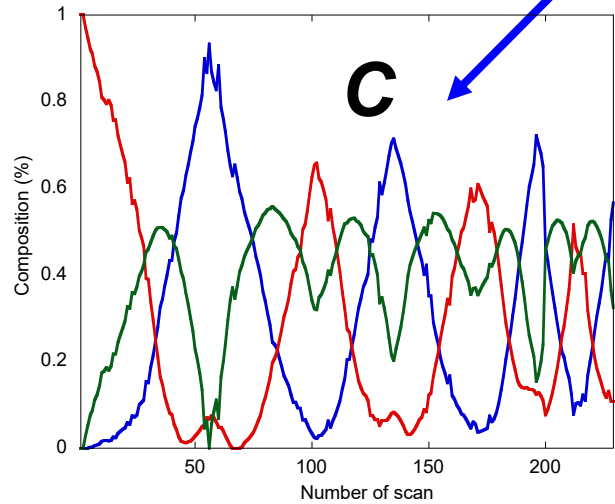
MCR-ALS

$$D = CS^T + E$$



Matlab toolbox
R. Tauler et al. Chemom.
Intell.Lab. Syst. (2015) 140, 1

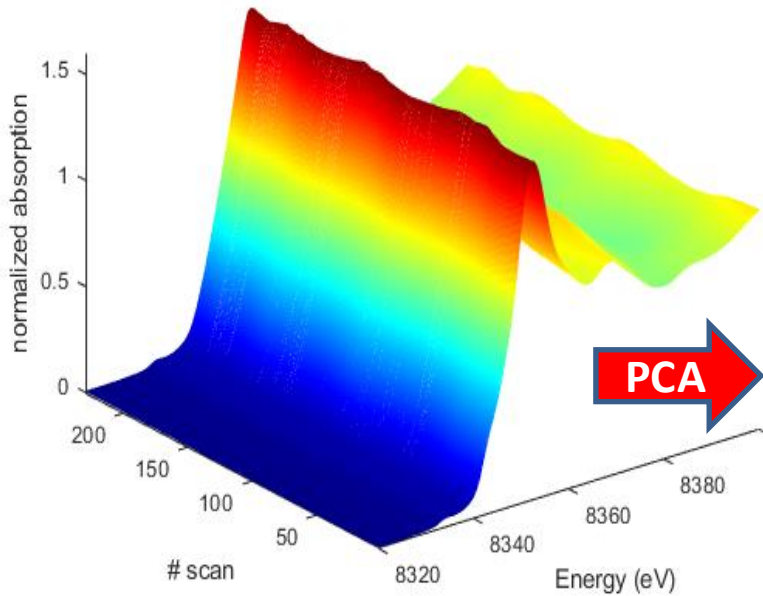
Applications for XAS:
Catalysis Today (2014) 229, 114-122
Comptes Rendus Chimie (2016) 19, 1337-1351
J. Phys. Chem. C (2017) 121, 18544-18556
Catalysis Today (2019) 336, 63-73
PCCP (2020) 22, 18835-18848
Rad. Phys. Chem. (2026)



MCR-ALS : How does it work ?

$$D = CS^T + E$$

Initial estimate of S^T
i.e. Guess of 3 spectra



Estimation of number of Components
e.g. 3 components
SIMPLISMA, EFA



$$\min \|D - CS^T\|^2$$

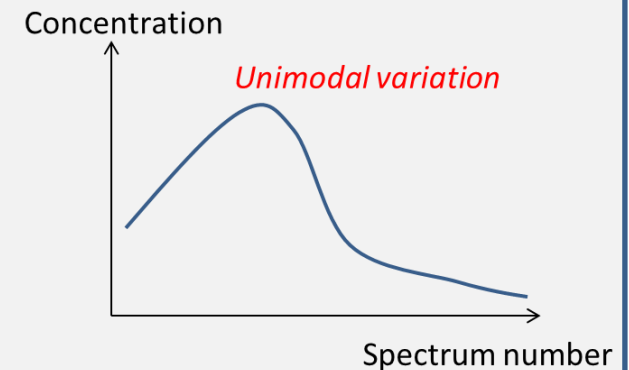
Determination of concentration matrix C
which is used as a new guess for the minimization

CONSTRAINTS to help CONVERGENCE of MCR-ALS fitting

- Non-negativity of C and S
- Closure relation for the concentration of the n components:

$$\sum_{i=1}^n C_i = 100\%$$

- Unimodality of concentration profiles

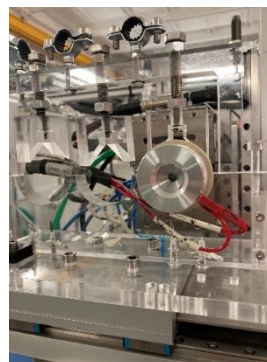
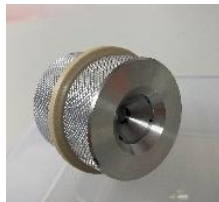


Iteration is stopped when convergence is achieved.

G. Moëhl, L. Pérez Ramirez, A. Iadecola and S. Belin.

New cell design :

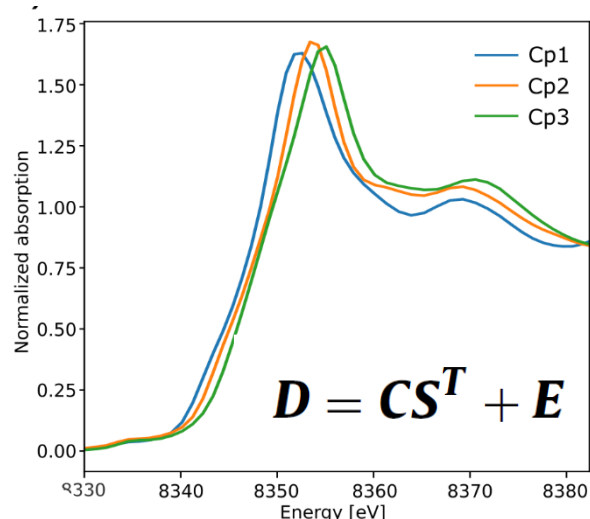
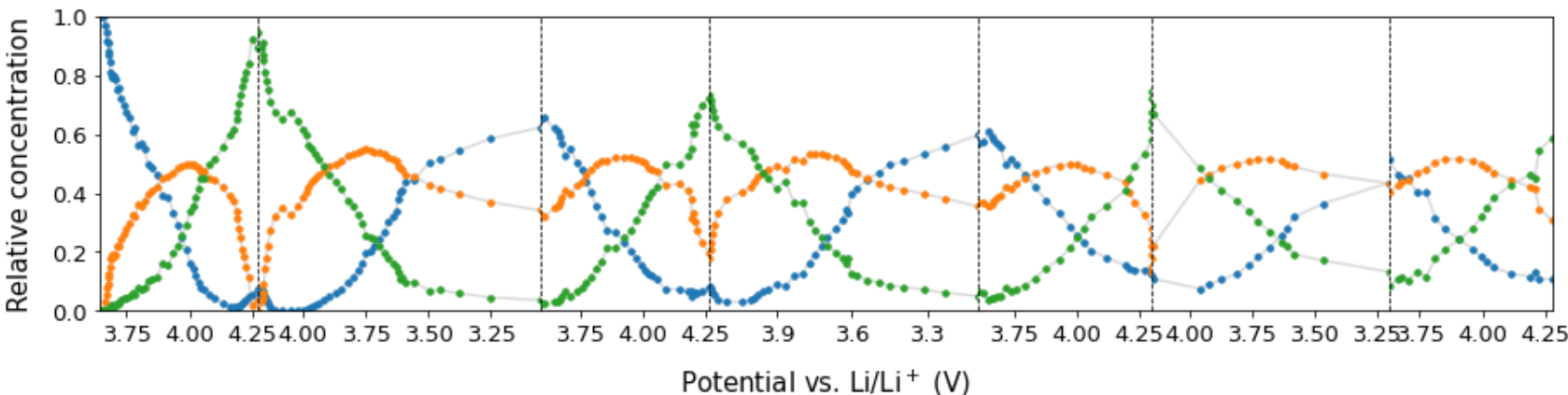
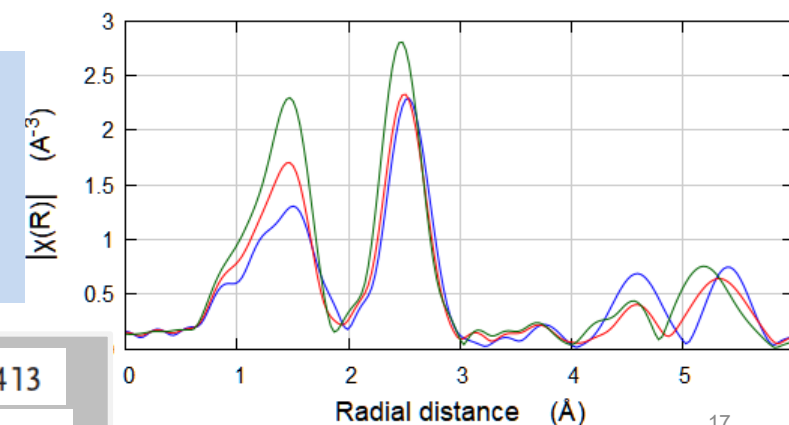
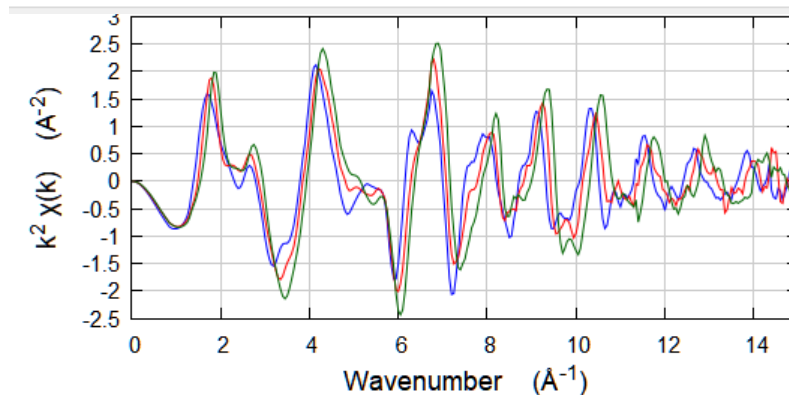
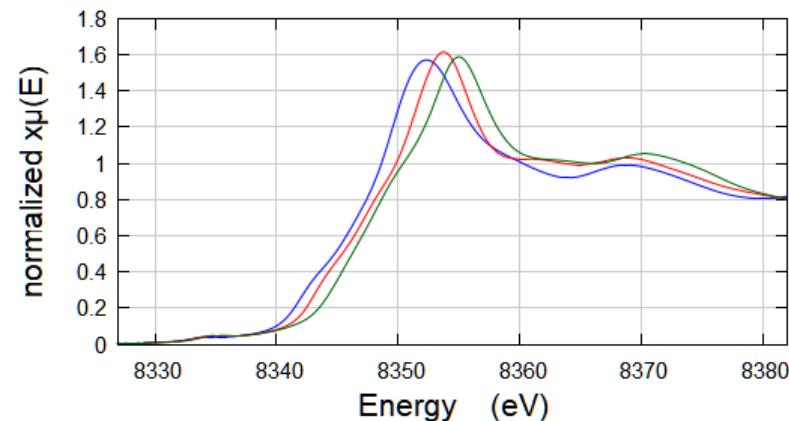
Gilles Moëhl & Laurent Barthe



BIG-MAP



Concentration profiles vs. Potential

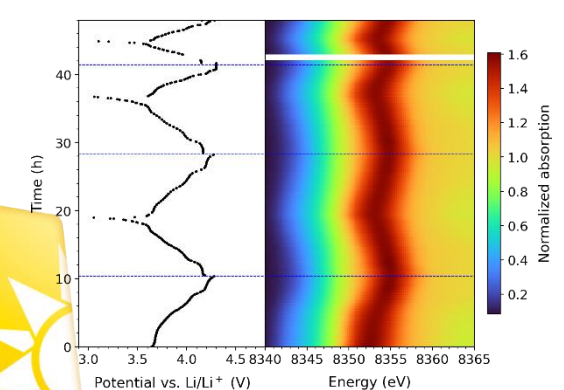


$$D = CS^T + E$$

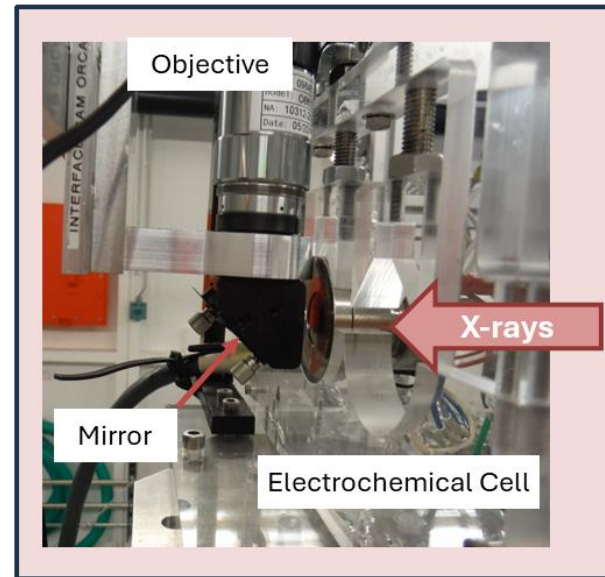
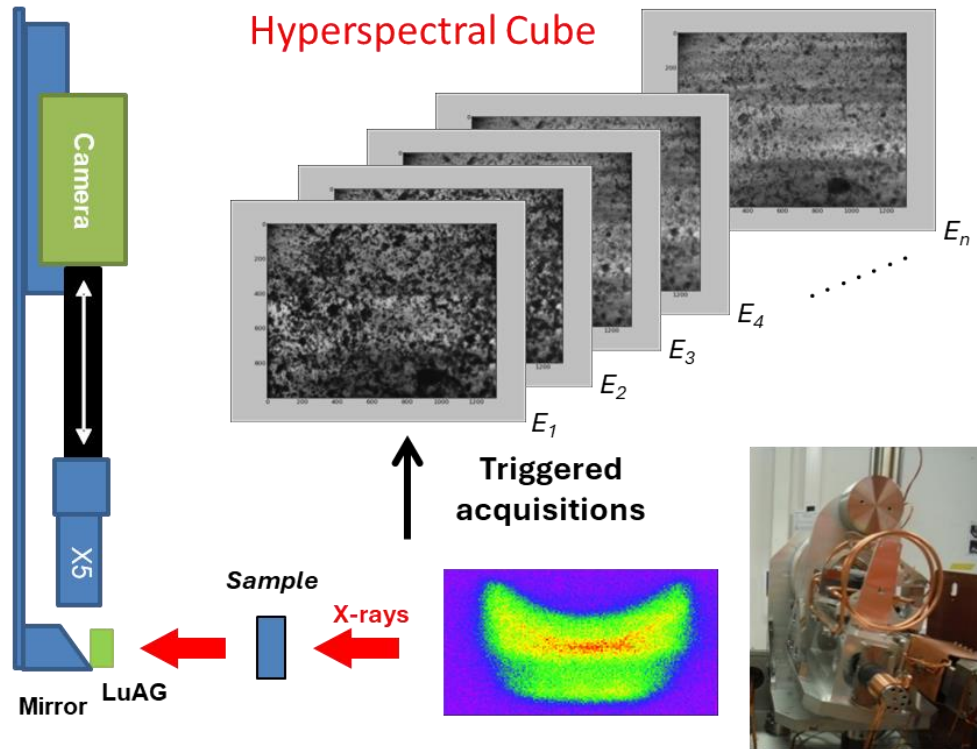
Phase Identification
 ⇒ fit of the EXAFS spectra of the 3 Cps instead of 229 spectra

Adv. Energy Mater. 2024, 14, 2401413

J. Mater. Chem. A, 2025, 13, 28305-28317



Hyperspectral Cube



Field of View

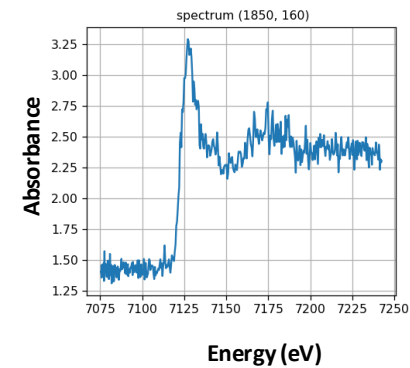
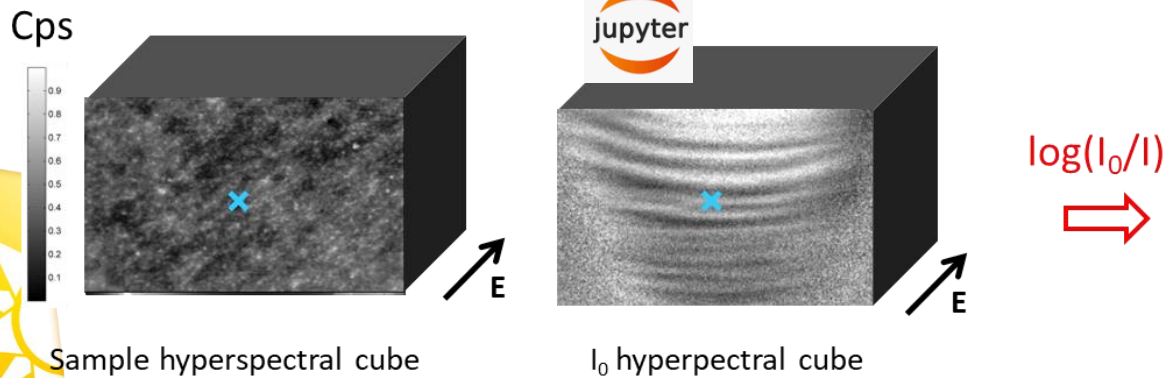
3.3 mm x 1.2 mm
(Navitar x4)

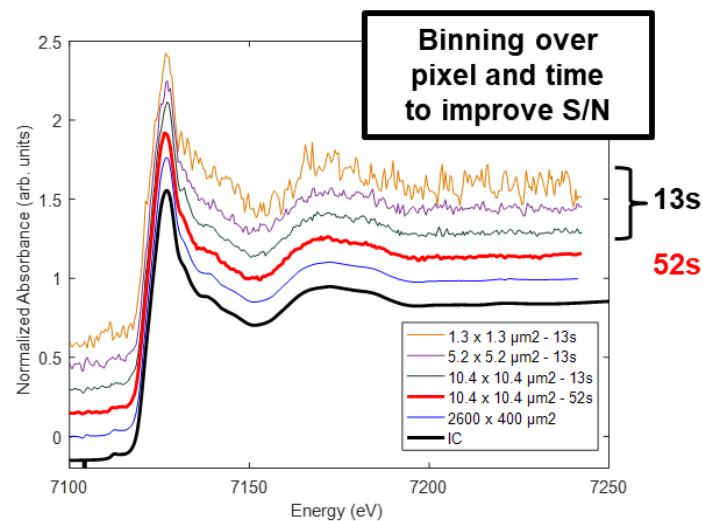
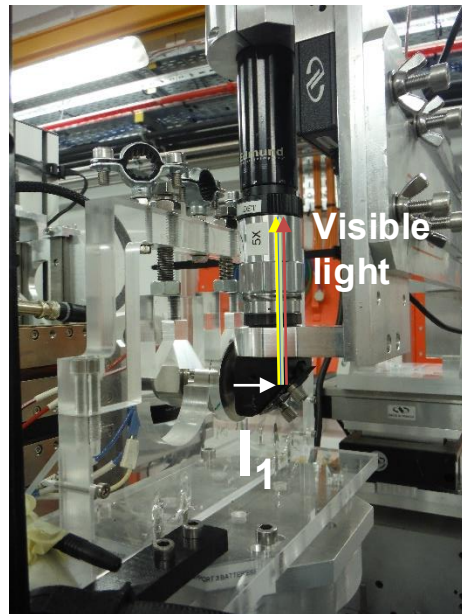
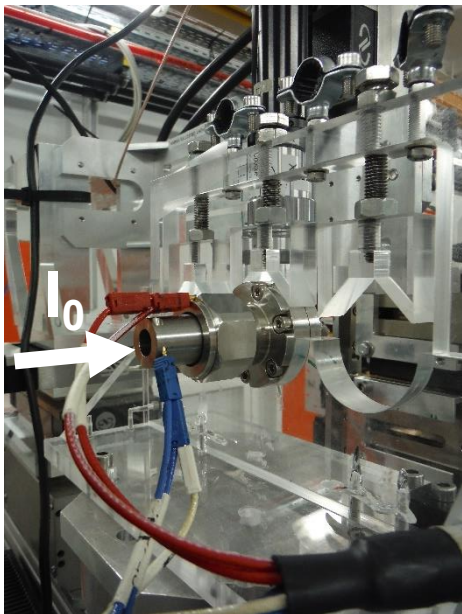
1.3 mm x 0,5 mm
(Mitutuyo x10)

**Spectrum on
1 single pixel
1 single cube**

Spatial Resolution
3 – 4 μ m

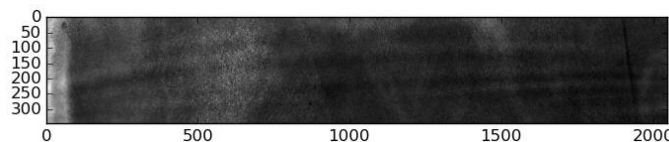
Time/cube
2 s (XANES)
4 to 11 s (XANES + EXAFS)



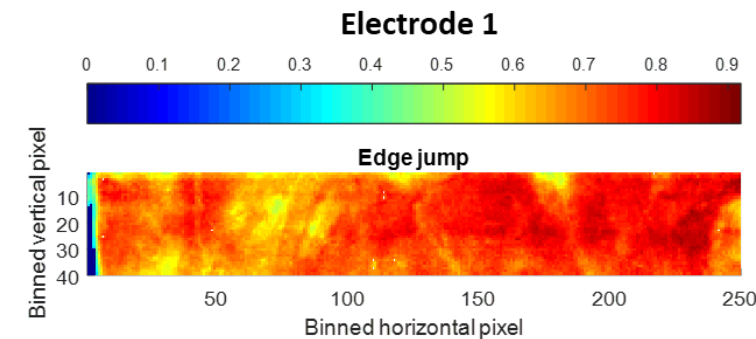


SPATIAL RESOLUTION

ROI : 2048 (H) x 348 (V) pixels
 Field of View : 2.66 (H) x 0.45 (V) mm²
 8x8 Pixel-binning : 10.4 x 10.4 μm²

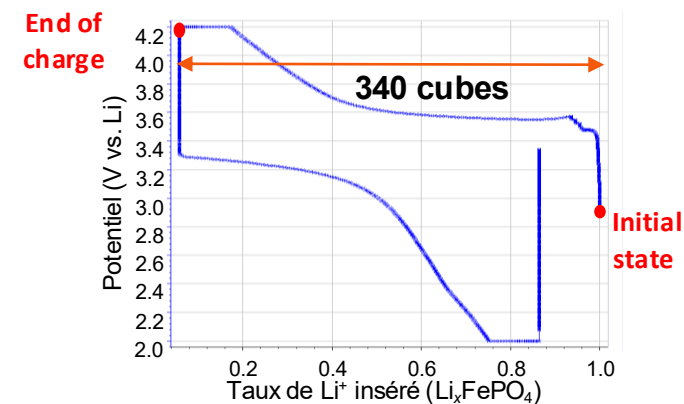


10200 (255 x 40) pixels/spectra per Cube after Pixel-Binning

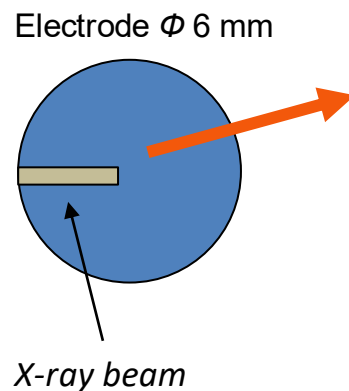
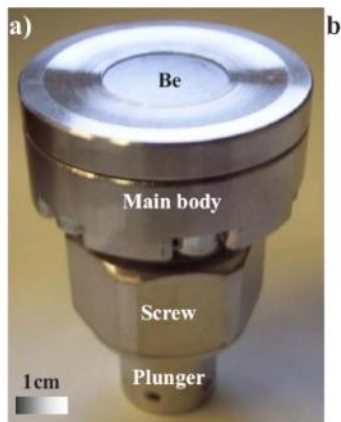


TIME RESOLUTION

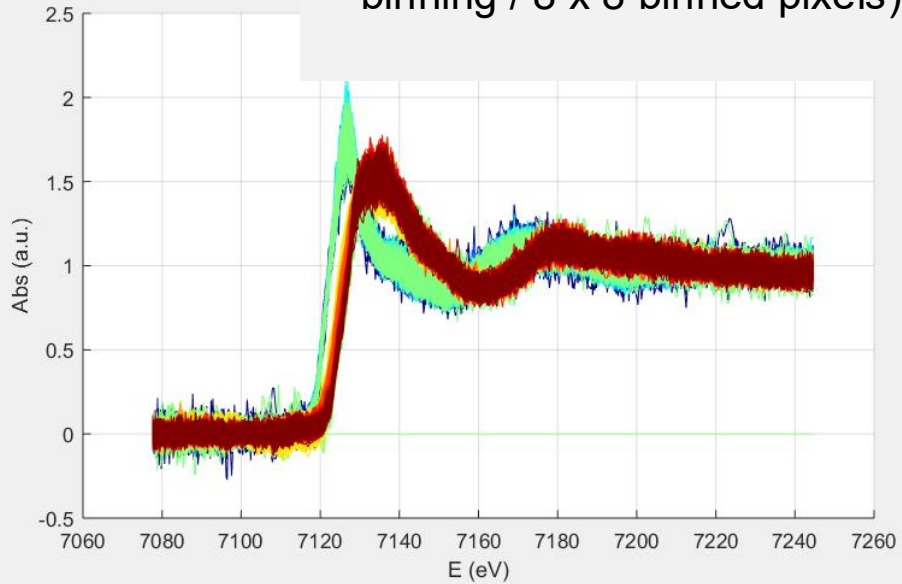
Monitoring of 1C charge
 1 cube every 13 s => -0,0036 Li



10200 pixels x 340 cubes
 = **3 468 000 Spectra**



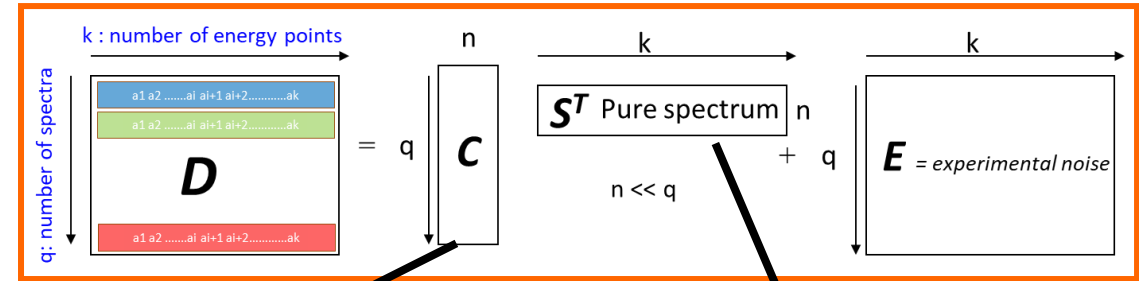
20000 spectra from flattened cube 1 + cube 340 (no-cube binning / 8 x 8 binned pixels)



MCR-ALS

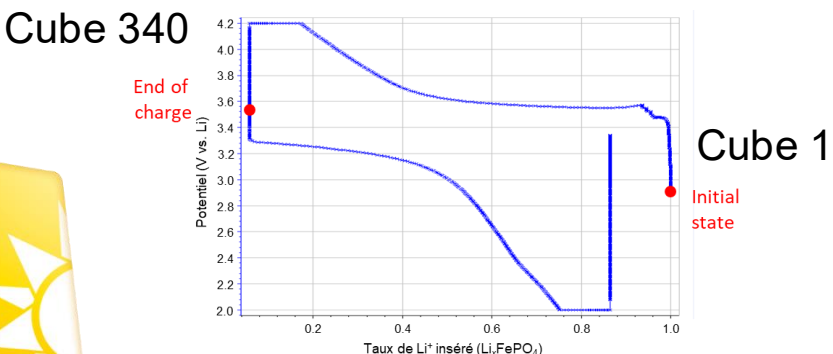
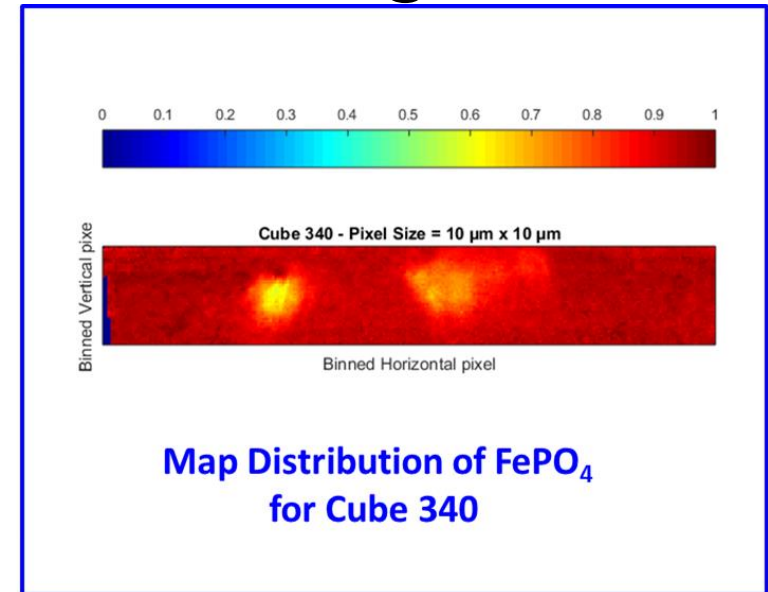
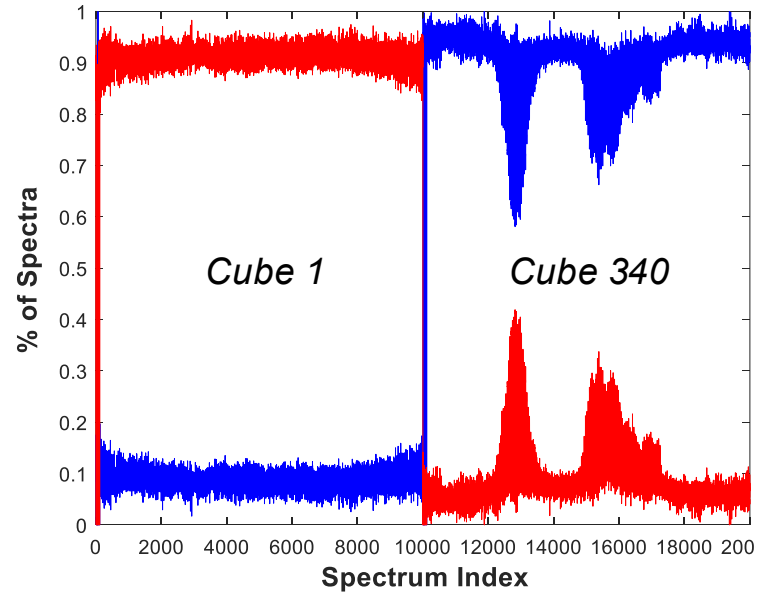
$$D = CS^T + E$$

Constraints Used : Non negativity C and S^T + Closure relation



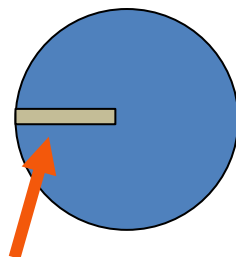
C

S^T

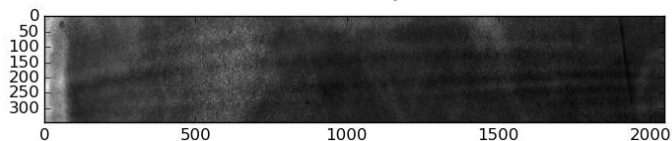




Electrode ϕ 6 mm



Use of Pure Species Spectra
for rebuilding by MCR-ALS the Speciation of Cubes at different stages of the charge
But with SELECTIVITY constraints



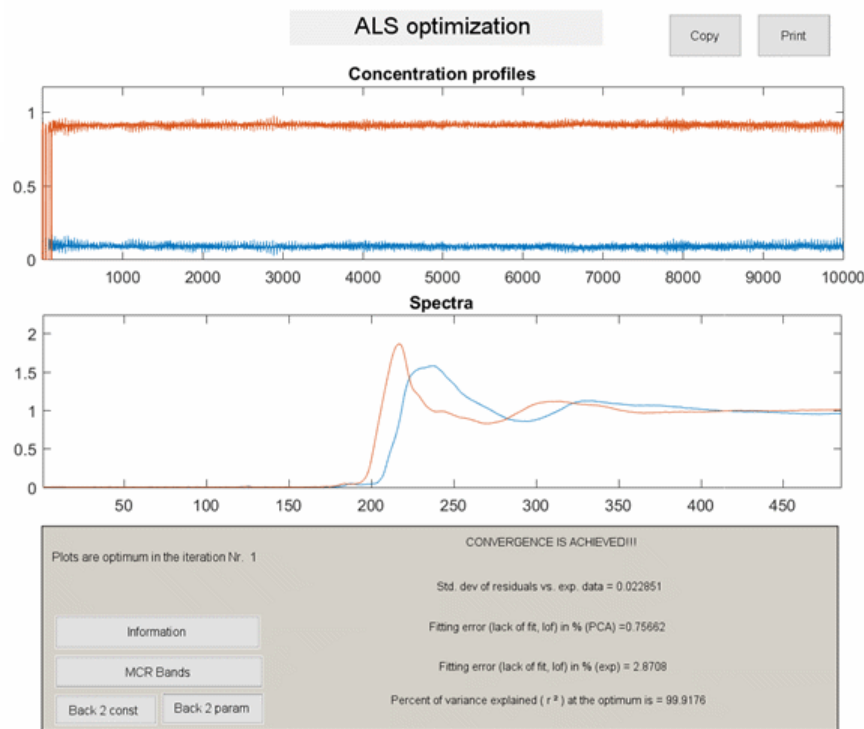
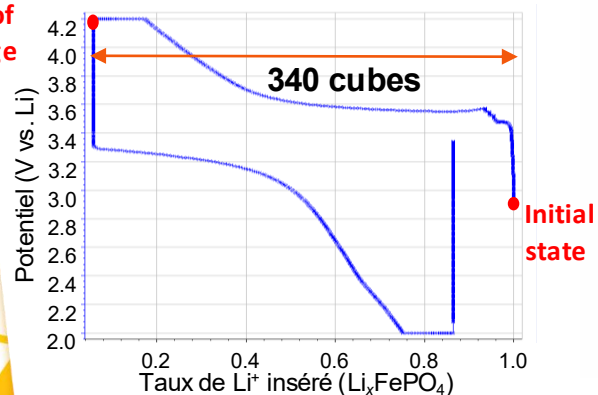
SPATIAL RESOLUTION

ROI : 348 (H) x 2048 (V) pixels
8x8 Binning Pixel : 10.4 x 10.4 μm^2
Field of View : 2.66 (H) x 0.45 (V) mm^2

TIME RESOLUTION

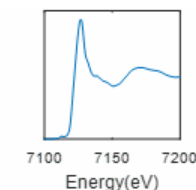
Monitoring of 1C charge
1 cube every 13 s \Rightarrow -0,0036 Li

End of charge

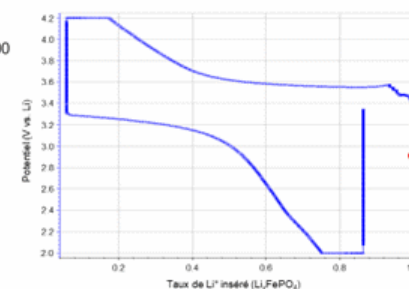
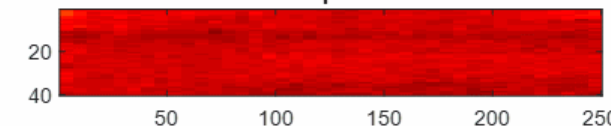


MCR-ALS on flattened spectra

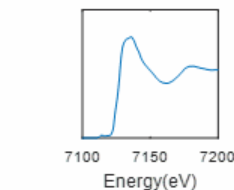
Briois *et al.*, *J. Sync. Rad.*, 2024, 31, 1084-1104



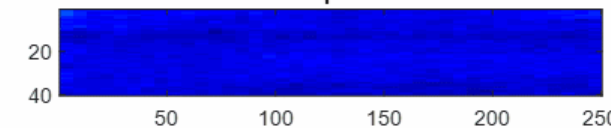
Cube 1 - Cp1 LiFePO4



1C



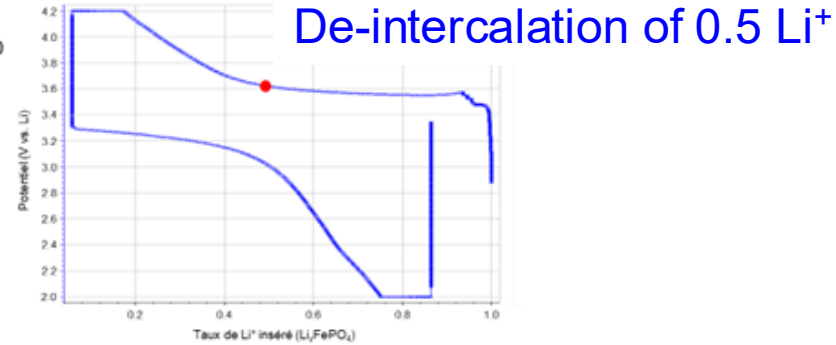
Cube 1 - Cp2 = FePO4



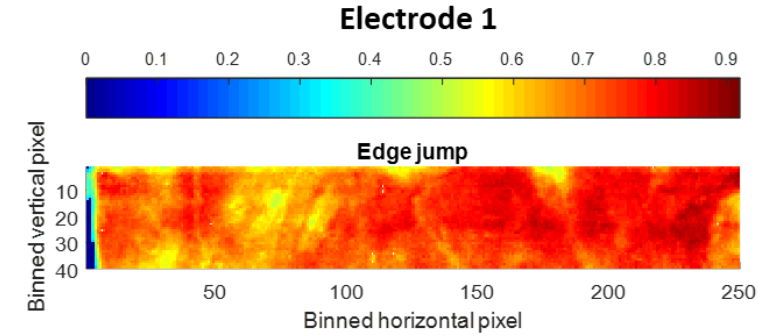
FoV: 2.66 (H) x 0.45 (V) mm^2



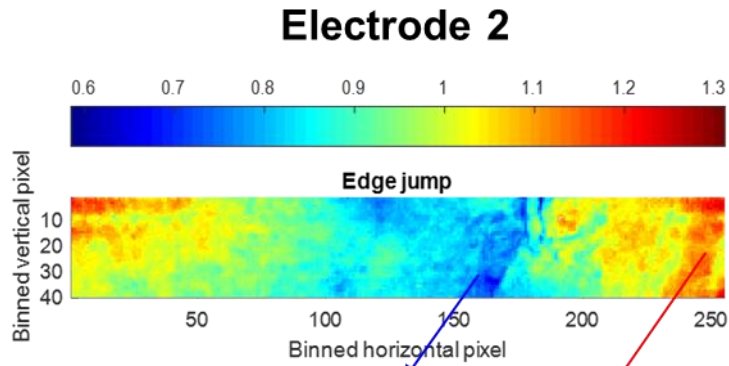
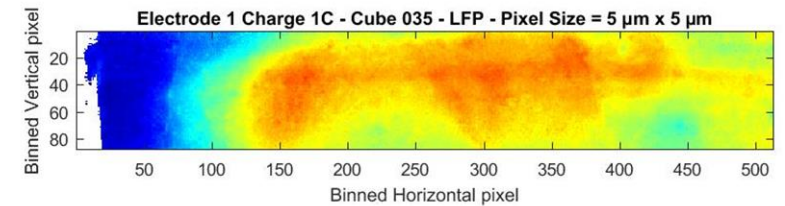
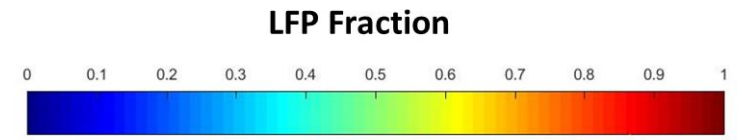
LiFePO₄ Cycling with time- and spatially- resolved imaging



~ homogeneous distribution of Fe within the electrode



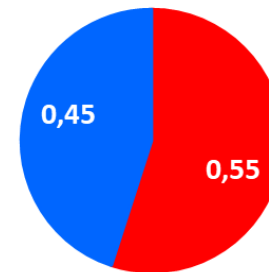
FoV: 2.66 (H) x 0.45 (V) mm²



5.2 x 5.2 μm
2D spatial resolution
13 s/scan

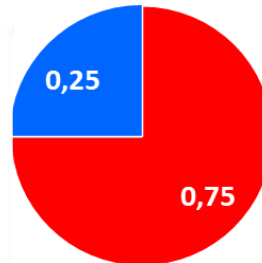
1C

Electrode 1



■ LFP ■ FP

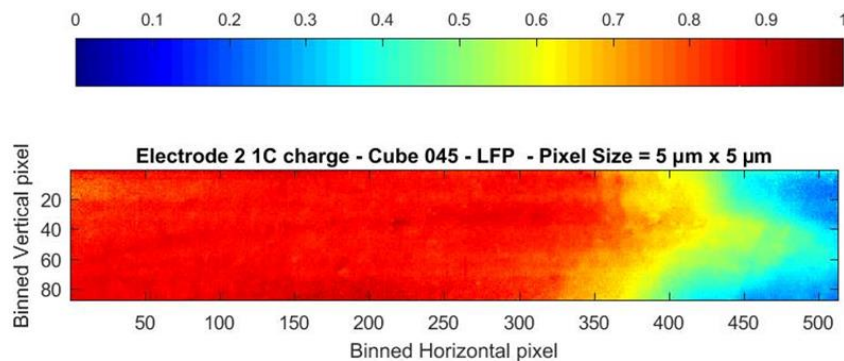
Electrode 2

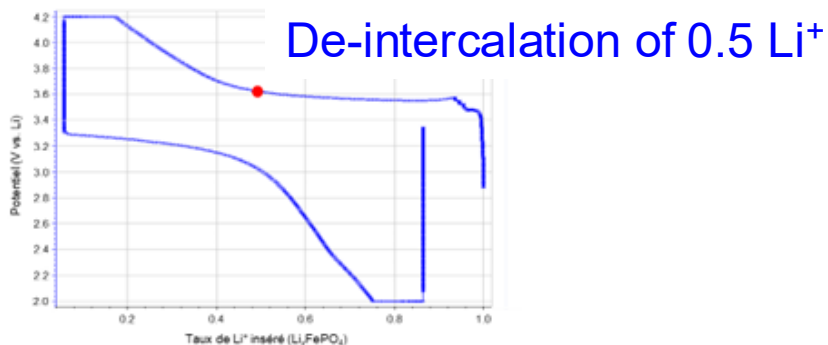


we are not so far from the composition
0.5 FePO₄ - 0.5 LiFePO₄
But with strong spatial heterogeneity

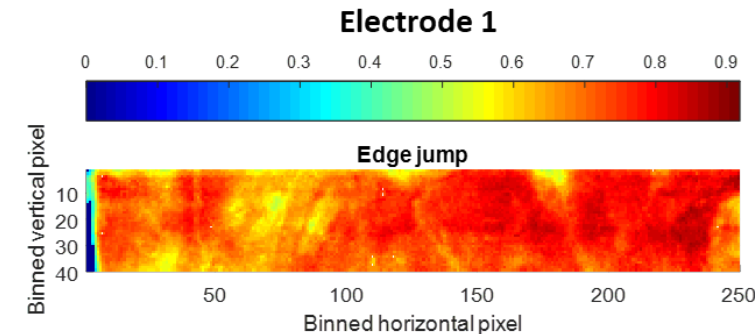
NON - homogeneous distribution of Fe within Electrode 2

Thin area Thick area
LFP Fraction





~ homogeneous distribution of Fe within the electrode



Quantifying Reaction and Rate Heterogeneity in Battery Electrodes in 3D through Operando X-ray Diffraction Computed Tomography

Hao Liu,^{†,▽} Saeed Kazemiabnavi,^{‡,⊕} Antonin Grenier,^{†,⊕} Gavin Vaughan,[§] Marco Di Michiel,[⊗] Bryant J. Polzin,^{||} Katsuyo Thornton,[⊥] Karena W. Chapman,^{*,†,#} and Peter J. Chupas^{*,#,*1}

200 μm in plane
100 μm depth
3D spatial resolution

30 min /scan

C/10

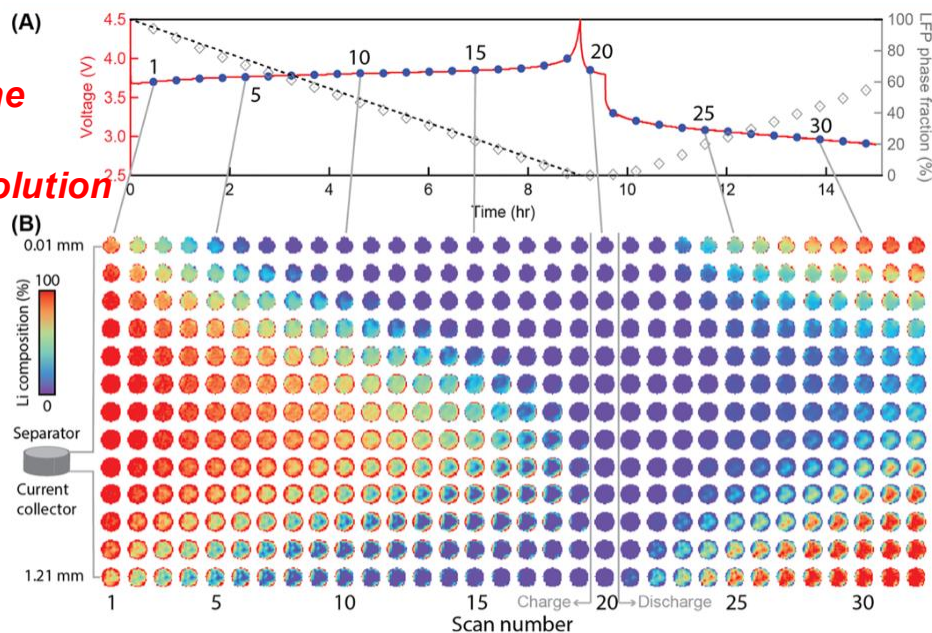
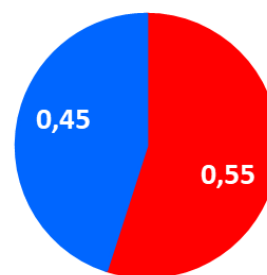


Figure 2. (A) Voltage profile of the thick electrode cycled at C/10 during operando XRD-CT (red curve). Blue dots indicate the time of each tomography measurement. Gray squares indicate the average LFP phase fraction of the entire electrode. The dashed black line indicates the ideal LFP phase fraction during galvanostatic charge. (B) LFP phase fraction, that is, the Li composition, map of different horizontal layers across the electrode during cycling. The Li composition (LFP phase fraction) is represented in color. The separation between adjacent layers is 0.1 mm.

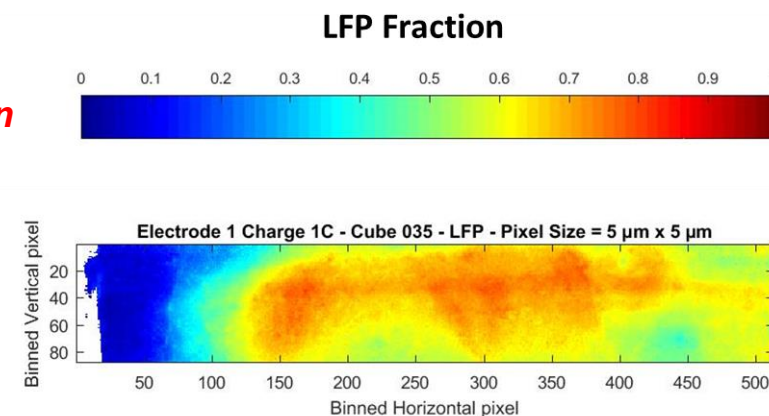
1C

5.2 x 5.2 μm
2D spatial resolution
13 s/scan

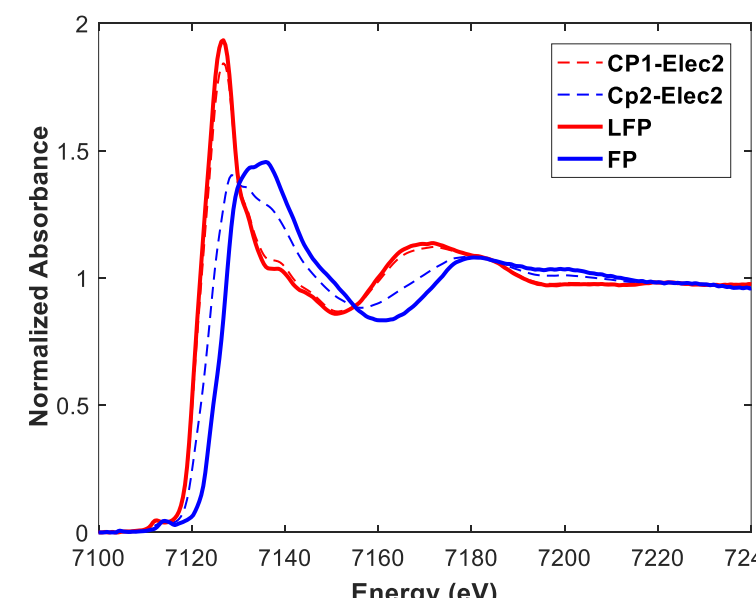
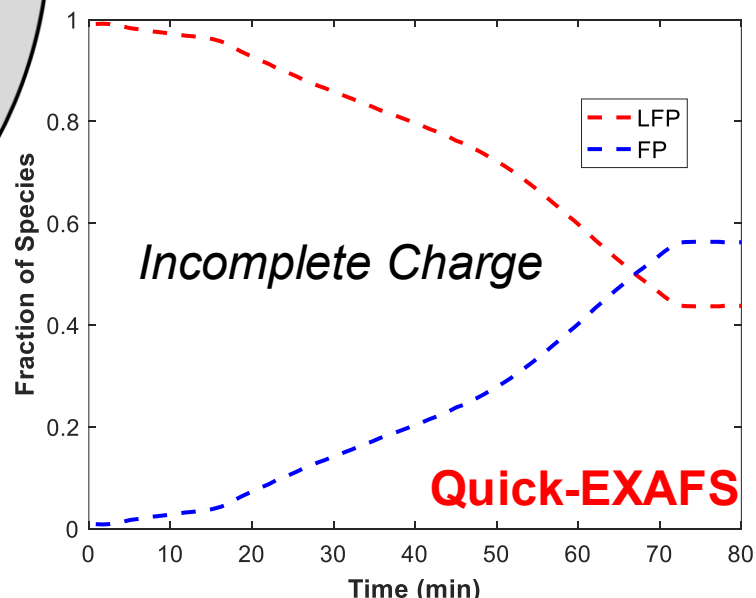
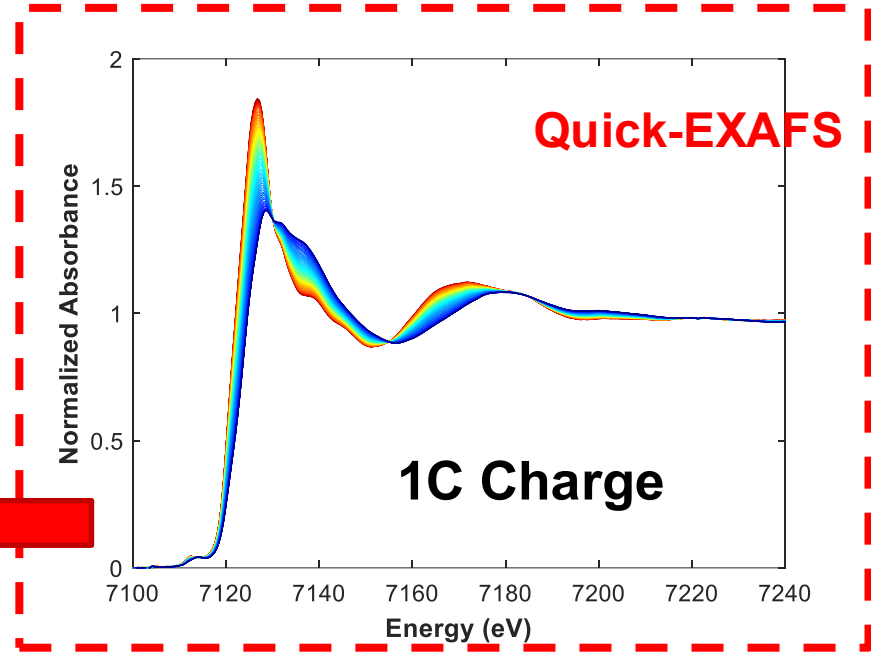
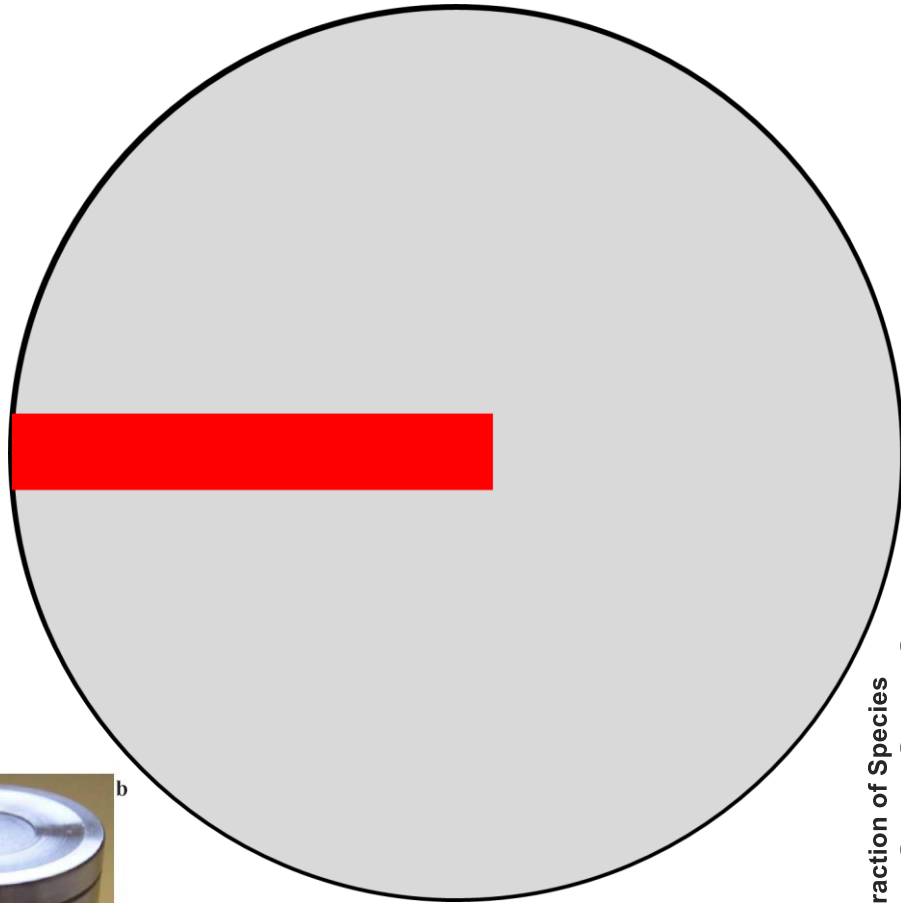
Electrode 1



■ LFP ■ FP

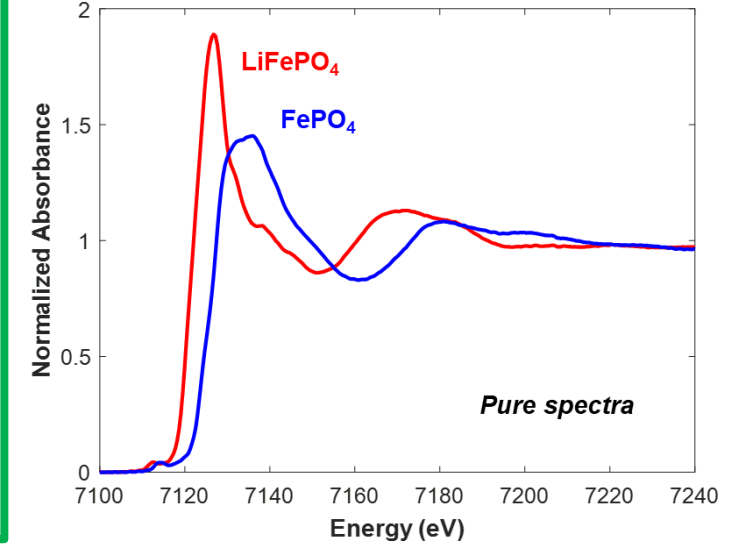
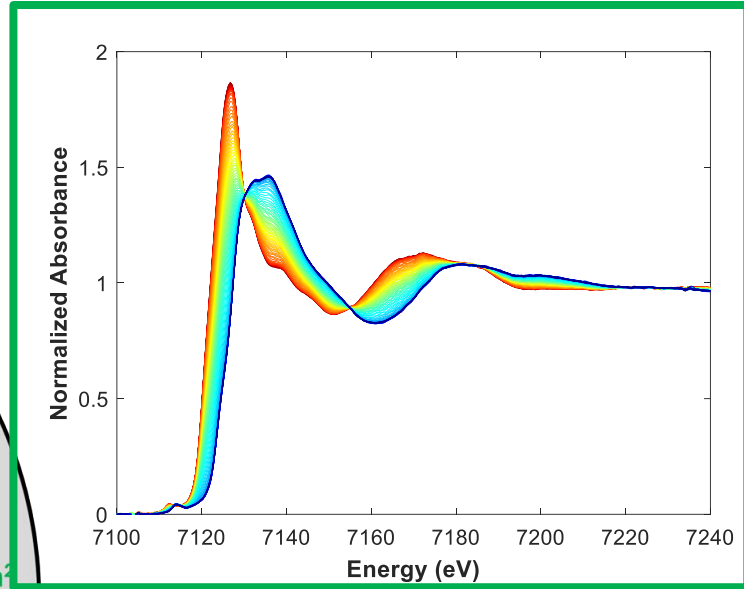
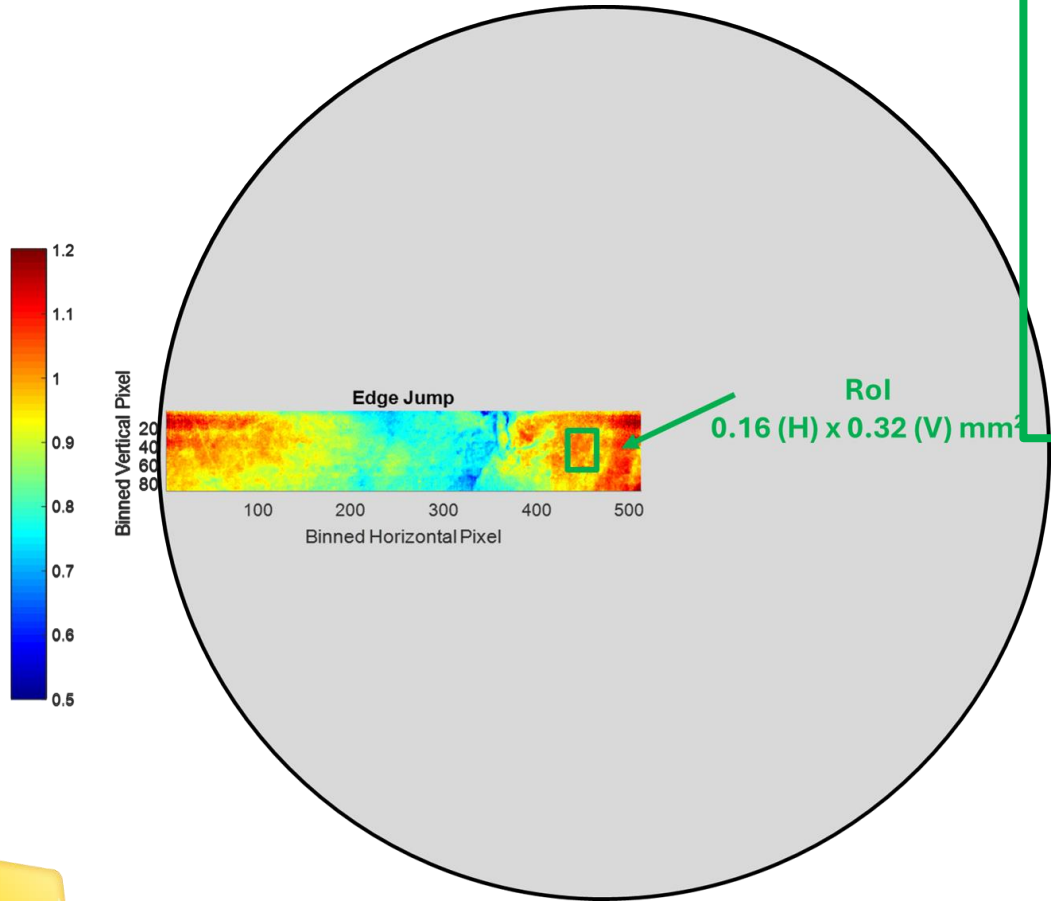


we are not so far from the composition
0.5 FePO₄ - 0.5 LiFePO₄
But with strong spatial heterogeneity

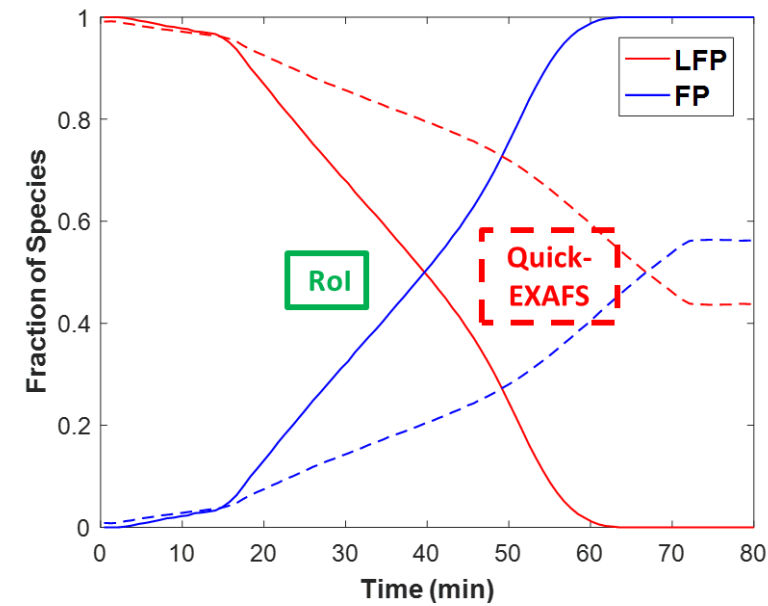


Electrode Φ 6 mm





Looking inside the cell by hyperspectral imaging allows for drawing correct conclusions from the operando experiment



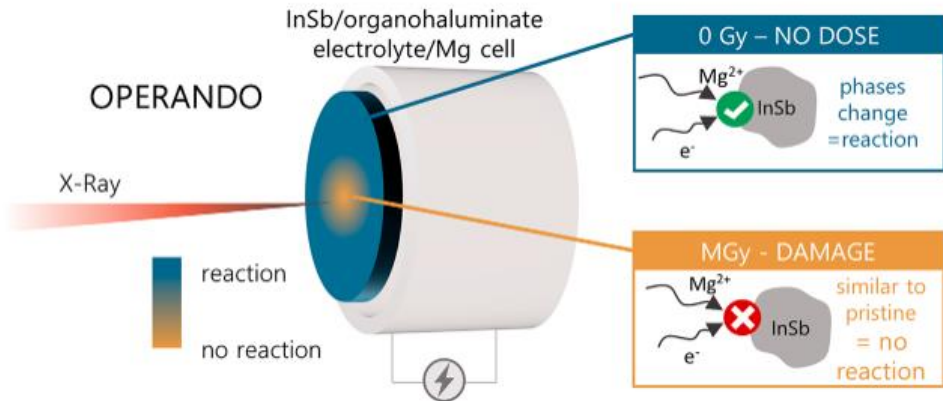
- With a larger beam => decrease of beam damage

Are Operando Measurements of Rechargeable Batteries Always Reliable? An Example of Beam Effect with a Mg Battery

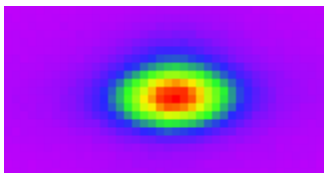
Lucie Blondeau, Suzy Surblé, Eddy Foy, Hicham Khodja, Stéphanie Belin,* and Magali Gauthier*

Cite This: *Anal. Chem.* 2022, 94, 9683–9689

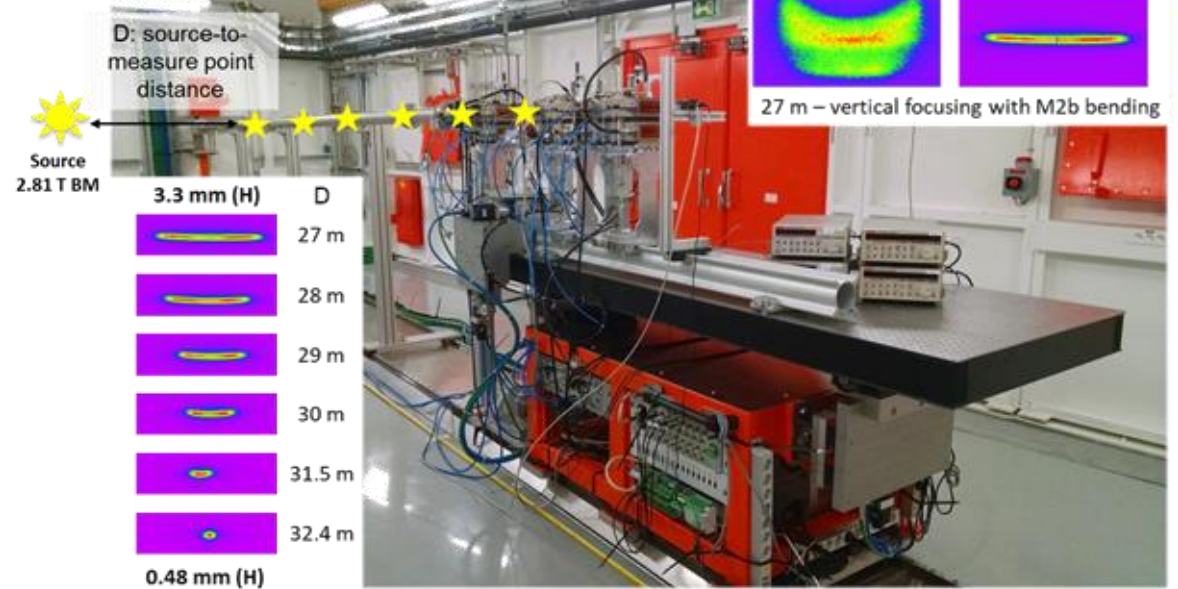
Read Online



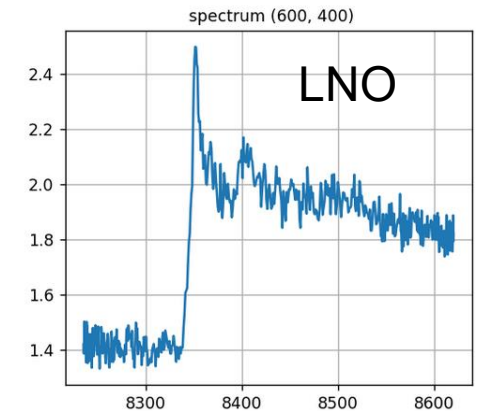
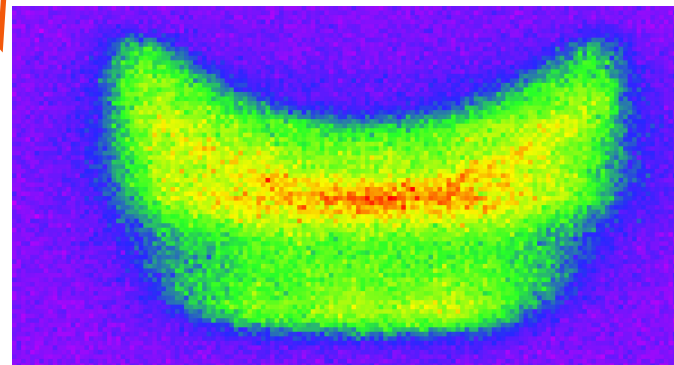
Beam size :
0.68 (H) x 0.28 (V) mm²



X 20: Dose decrease



Field of View :
3,28 (H) x 1.21 (V) mm²



Noisier data but so many => MCR-ALS

X-ray Absorption Spectroscopy

LOCAL PROBE

- Atomic Selectivity
- All kind of materials
- Simplicity of data Collection
- Reaction Process

Which allows for

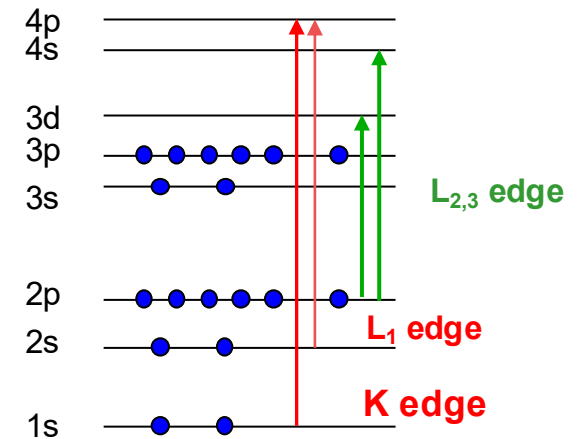
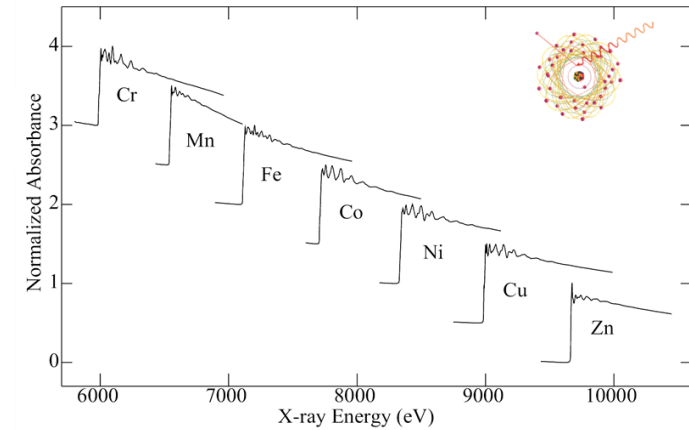
- direct determination of interatomic distances
- determination of electronic structure -
Orbital Selectivity

Limitation :

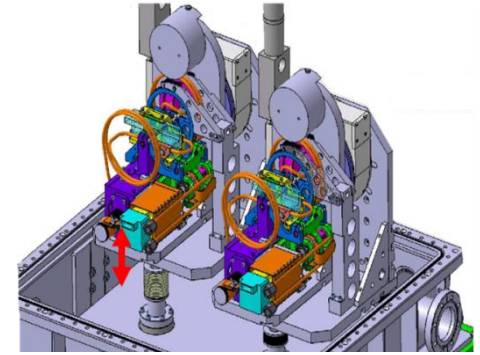
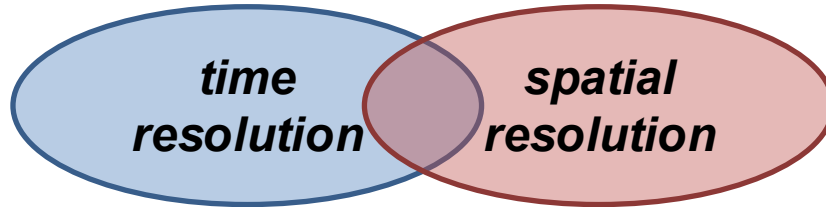
- Not an absolute technique
- High sensitivity to the bulk

XAS is Element Specific

(from S. Webb)



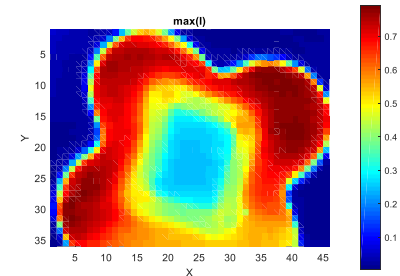
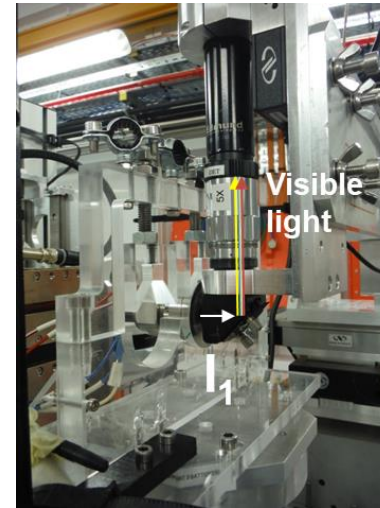
$$\Delta l = \pm 1$$



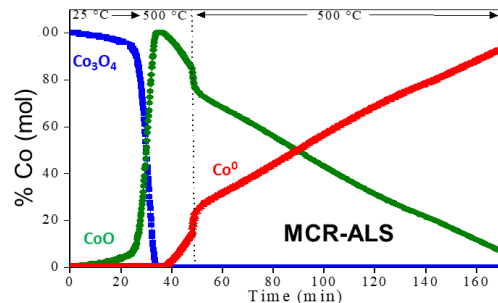
- ✓ Time resolution ranging from 20 ms to 5s
- ✓ + Space Resolution at the micron-meter scale
- ✓ 3 Quick-EXAFS monochromators for 4-43 keV energy range
- ✓ “Edge jumping” capability

✓ Dedicated sample environments for *operando* studies
 In energy storage, heterogeneous catalysis, electro-catalysis, photo-catalysis, nanoparticles nucleation/growth

- Raman spectroscopy
- UV-Vis spectroscopy
- 2D microscopy

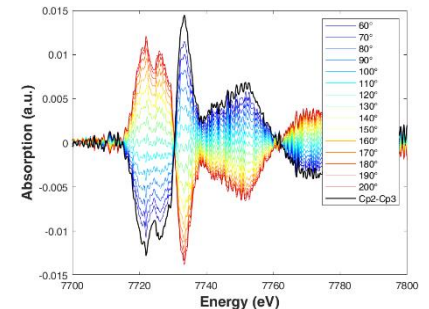


✓ $D = C S^T + E$



✓ Modulation Excitation Spectroscopy

$$A_k^{\phi^{PSD}}(e) = \frac{2}{T} \int_0^T A(e, t) \cdot \sin(k\omega t + \phi^{PSD}) \cdot dt$$





Synchrotron Radiation News (2020) 33(1) 20-25



Beamlines supported by a Consortium for Catalysis and Energy-Related Materials



BIG-MAP



Gilles Moehl



Lucia Pérez Ramírez



Francesco La Porta

Thank You for your Attention

Laurent Barthe

Stéphanie Belin

Anthony Beauvois



Antonella Iadecola (RS2E)



Olga Roudenko





Pionners of XAFS in France: Pierre Lagarde, Denis Raoux, Alain Fontaine, Dominique Chandesris, Georges Calas, Jacqueline Petiau, Michel Verdaguer, Françoise Villain, Christian Brouder

From LURE to SOLEIL : Michèle Sauvage, Jean Daillant

Pleanry Lectures : John Rehr, Yves Joly, Janis Timoshenko

Invited Lectures : Philippe Saintavit, Christèle Legens, Sakura Pascarelli, Laurent Michot, Géraldine Sarret, Jean Susini, Laurence Croguennec, Karine Provost

Oral Communications will be selected on abstracts + Poste session

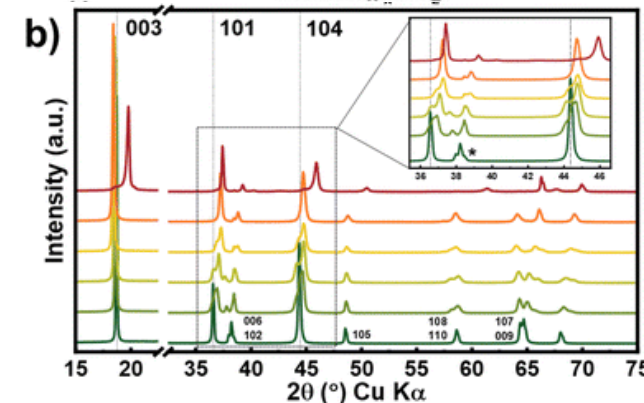
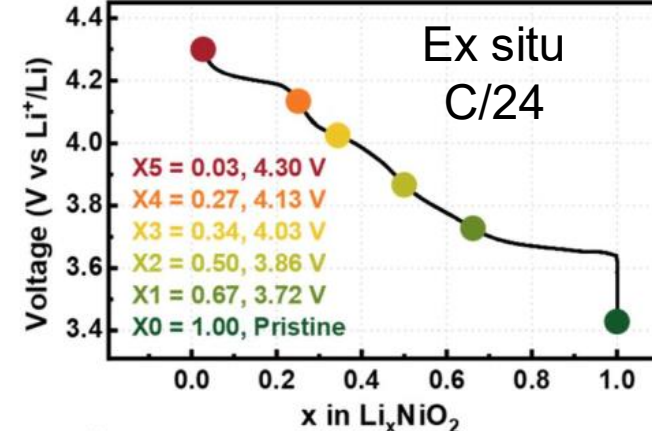
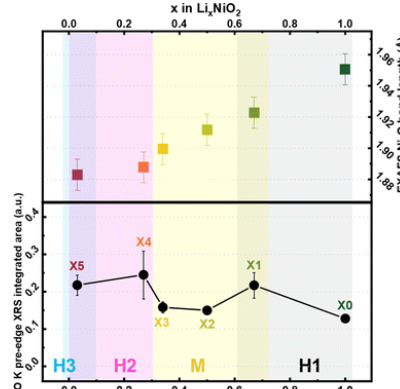
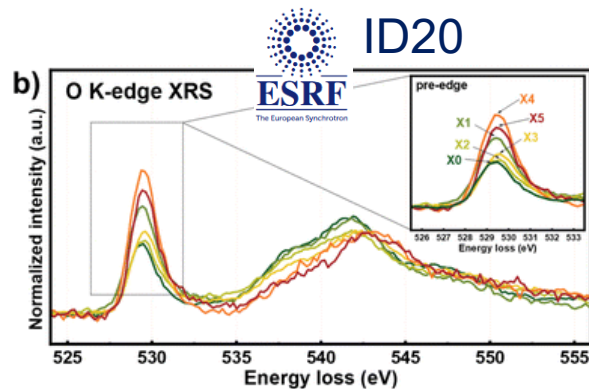
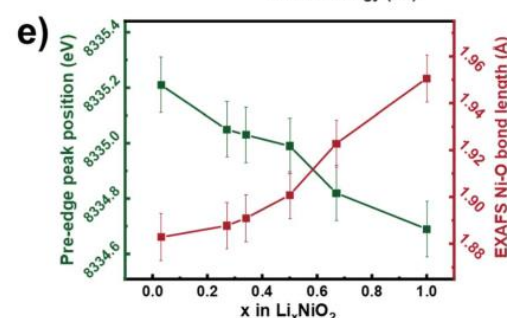
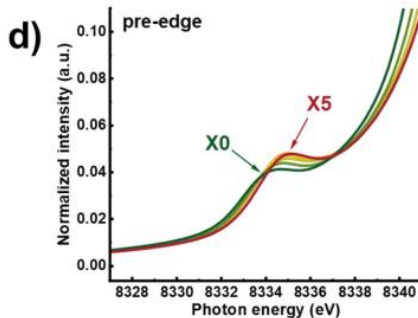
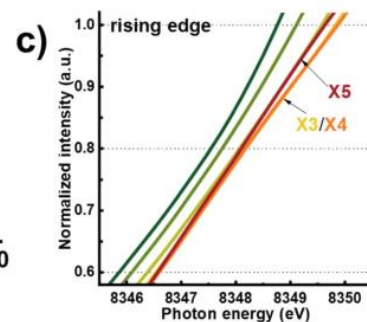
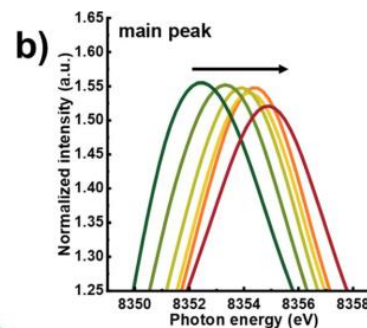
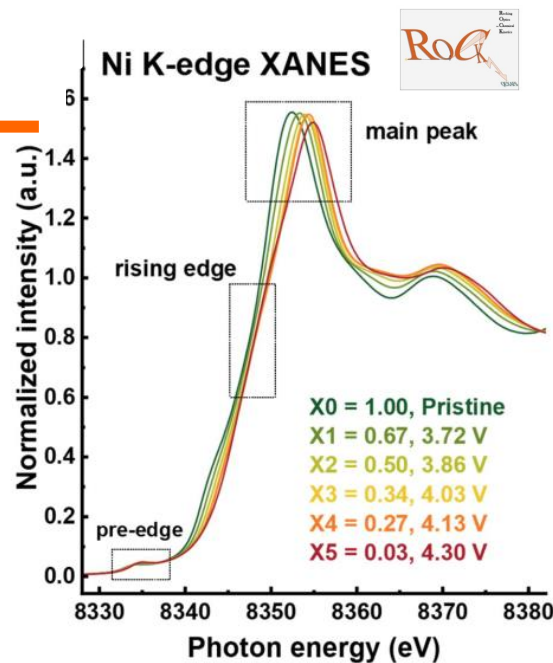
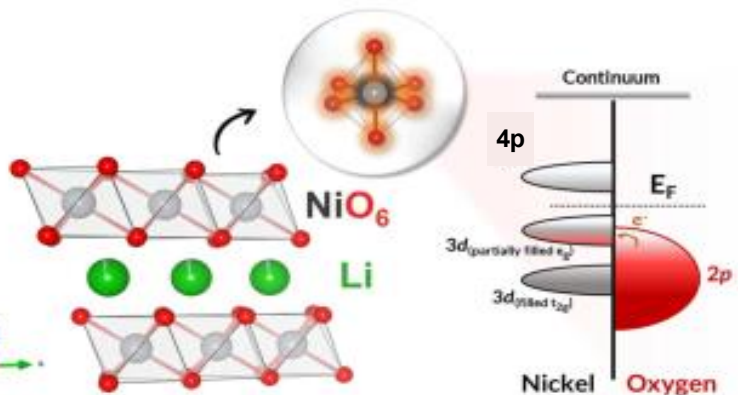
Insights into the role of the covalent Ni-O bonds in LiNiO₂ positive electrodes: a correlative hard X-ray spectroscopy study

Jazer Jose H. Togonon, Jean-Noël Chotard, Alessandro Longo, Lorenzo Stievano, Laurence Croguennec and Antonella Iadecola

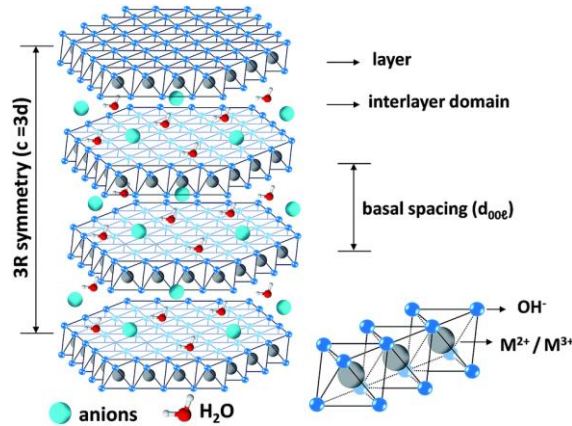
Representative NiO₂ layers composed of edge-sharing NiO₆ octahedra and a Li interlayer from the LiNiO₂ crystal structure, highlighting a local NiO₆ unit. A simplified band diagram illustrates the negative charge transfer from O 2p to Ni 3d states.

J. Mater. Chem. A, 2025, **13**, 28305-28317

<https://doi.org/10.1039/D5TA03573B>



Sample - Formula	Crystal Phase from XRD	EXAFS d _(Ni-O) , Å
X0 - LiNiO ₂	Hexagonal	1.95(1) × 6
X1 - Li _{0.67} NiO ₂	Monoclinic	1.88(1) × 4
X2 - Li _{0.50} NiO ₂	Monoclinic	2.00(1) × 2
X3 - Li _{0.34} NiO ₂	Monoclinic	1.87(1) × 4
X4 - Li _{0.27} NiO ₂	Hexagonal	1.99(1) × 2
X5 - Li _{0.03} NiO ₂	Hexagonal	1.88(1) × 6



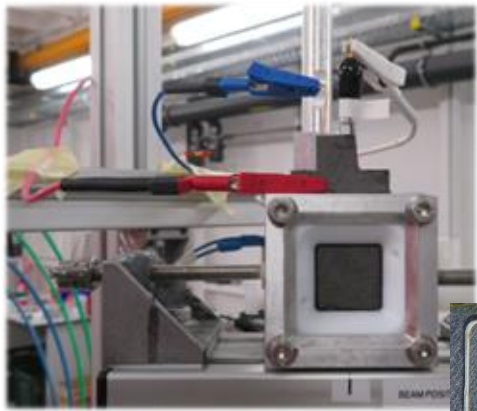
Operando X-ray Absorption Spectroscopy of Co₃Mn–CO₃ LDH: Formation and Structural Evolution under Electrochemical Conditions

Hani Farhat, Christine Mousty, Vanessa Prevot, Anthony Beauvois, Stéphanie Belin, Valérie Briois,* and Claude Forano*

Cite This: *J. Phys. Chem. C* 2024, 128, 21023–21037

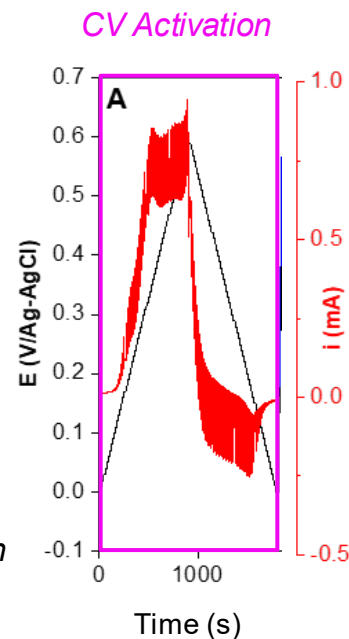
Read Online

Santos et al., *J. Mater. Chem. A*, 2017, 5, 9998



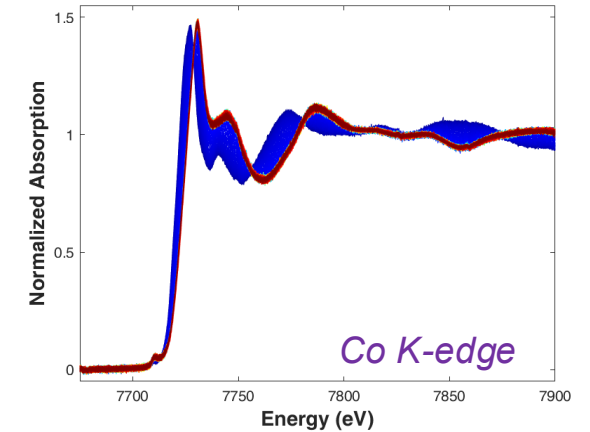
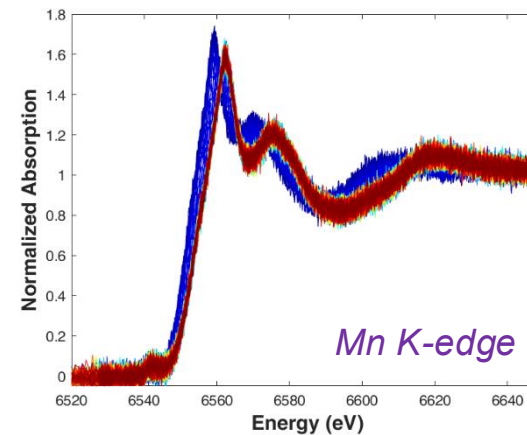
In 0.1 M NaOH

Carbon paper for material deposition

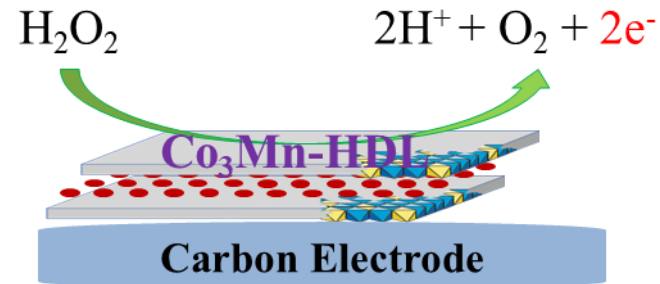
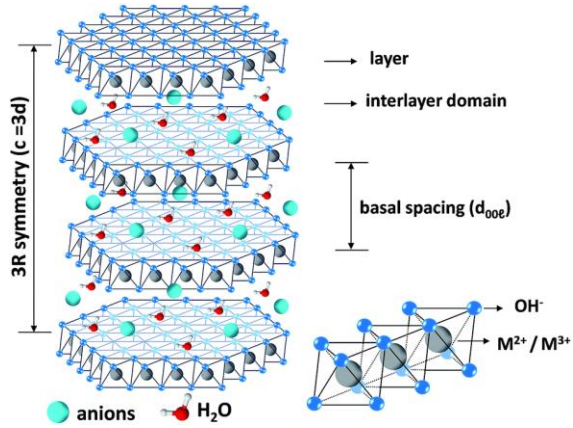


Transformation during electro-chemical CV activation

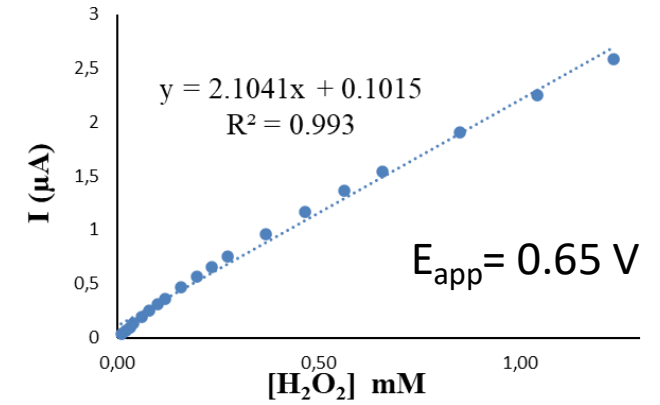
Co(II)₃Mn(III) LDH \longrightarrow Co(III)₃Mn(IV) Layered Species



Co₃Mn-based LDH used as electro-chemical H₂O₂ sensor

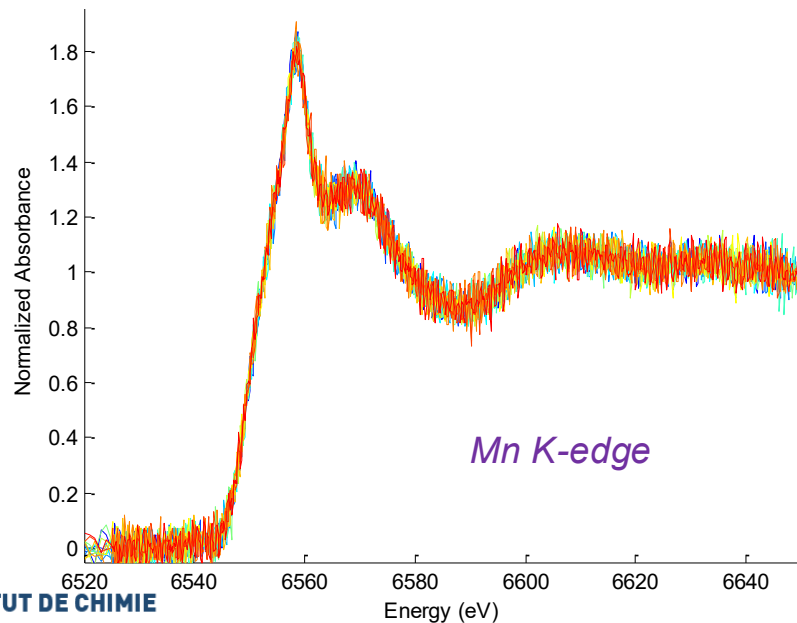


Farhat et al., *J. Phys. Chem. C*, 2020, **124**, 15585–15599



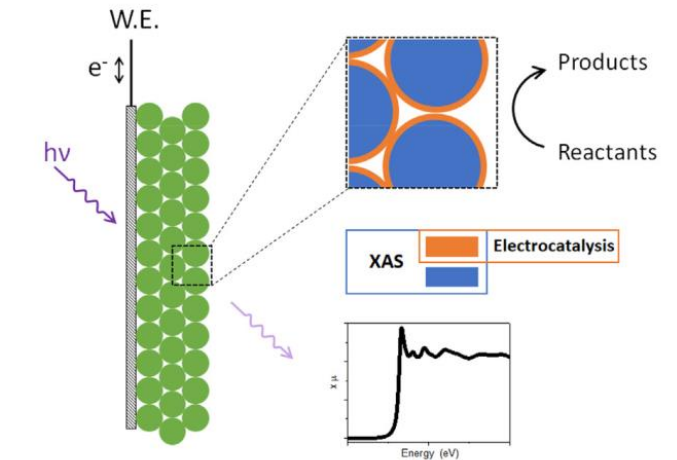
Santos et al., *J. Mater. Chem. A*, 2017, **5**, 9998

Operando : Activated Catalyst + H₂O₂



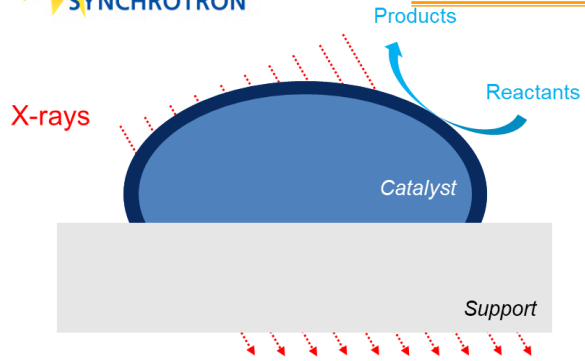
XAS is a bulk technique and electro-catalysis proceeds at the interface liquid-electrode.

How to increase the interface sensitivity?



Current Opinion in Electrochemistry 2021, **27**:100681

⇒ X-ray Absorption Spectroscopy is a bulk technique

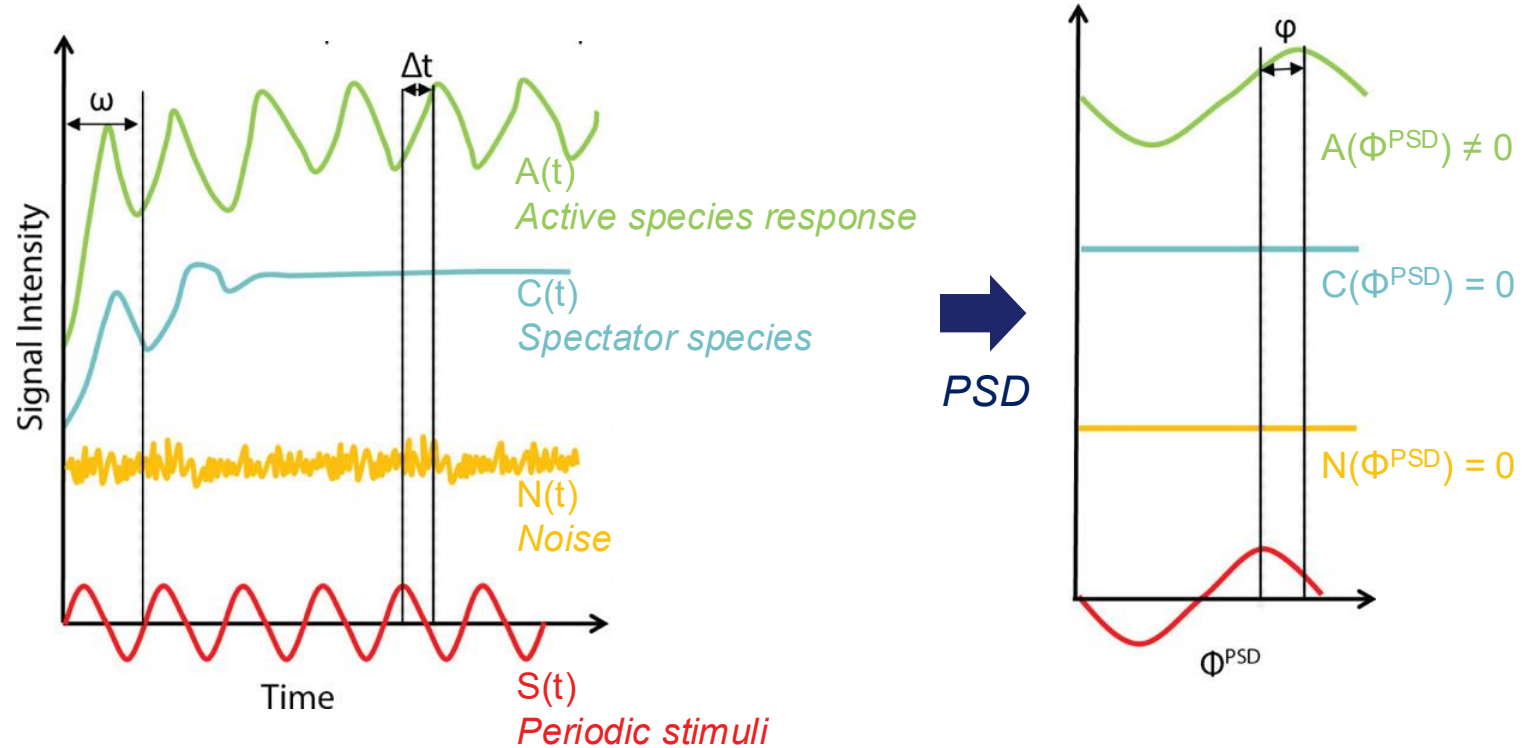


? How to extract the reactive species when they are not the major part of the material of interest?

1. Applying a periodic stimuli to the system

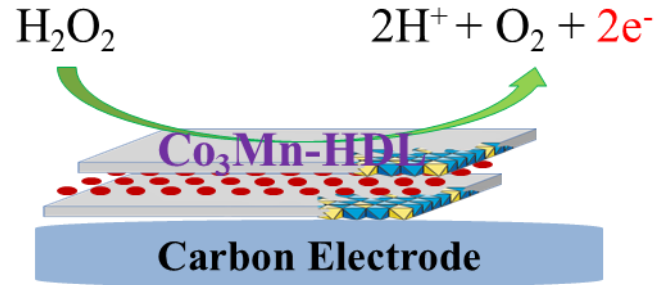
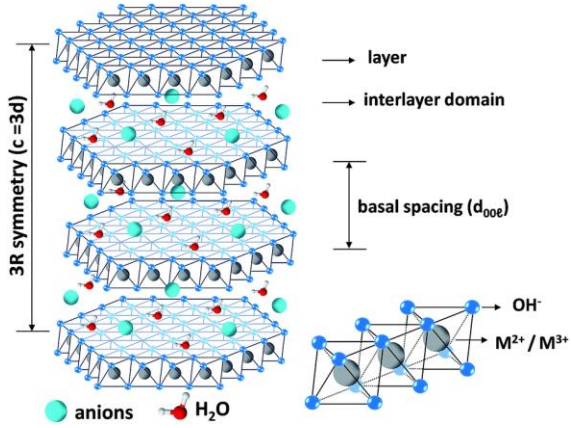
2. Average the periods when the steady state is reached

3. Apply Phase-Sensitive Detection

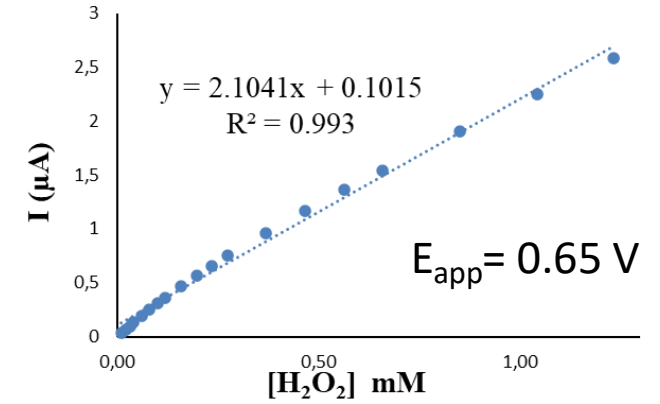


$$A_k^{\phi^{PSD}}(e) = \frac{2}{T} \int_0^T A(e, t) \cdot \sin(k\omega t + \phi^{PSD}) \cdot dt$$

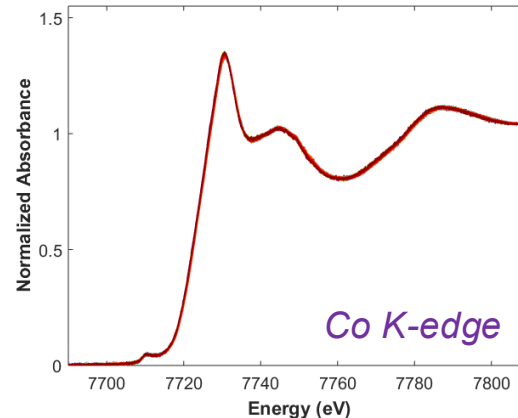
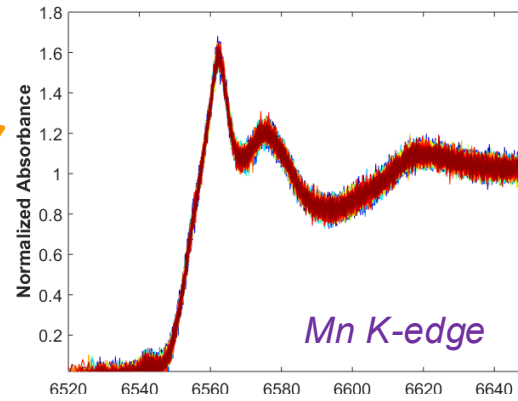
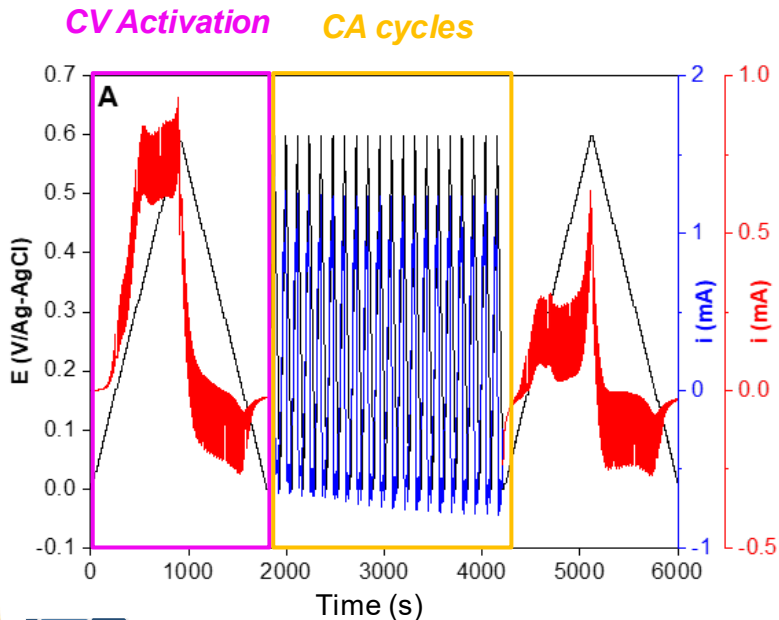
Co₃Mn-based LDH used as electro-chemical H₂O₂ sensor



Farhat et al., *J. Phys. Chem. C*, 2020, **124**, 15585–15599



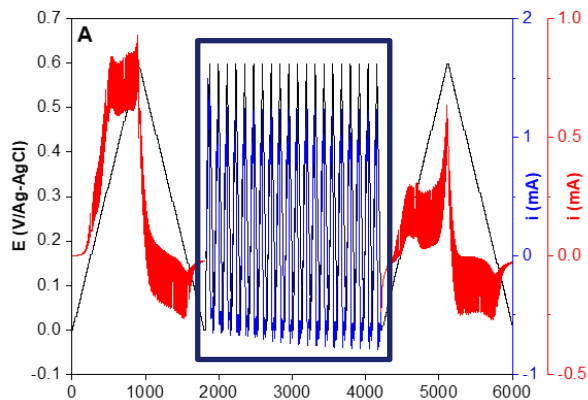
Santos et al., *J. Mater. Chem. A*, 2017, **5**, 9998



Which element is responsible for the electrochemical response of Co₃Mn LDH?
Is their synergic effect between Co and Mn?

Electrochemical behaviour of Co₃Mn LDH

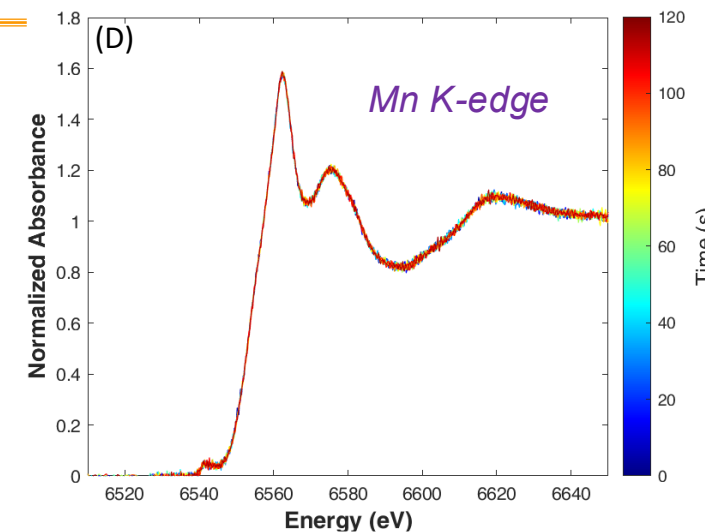
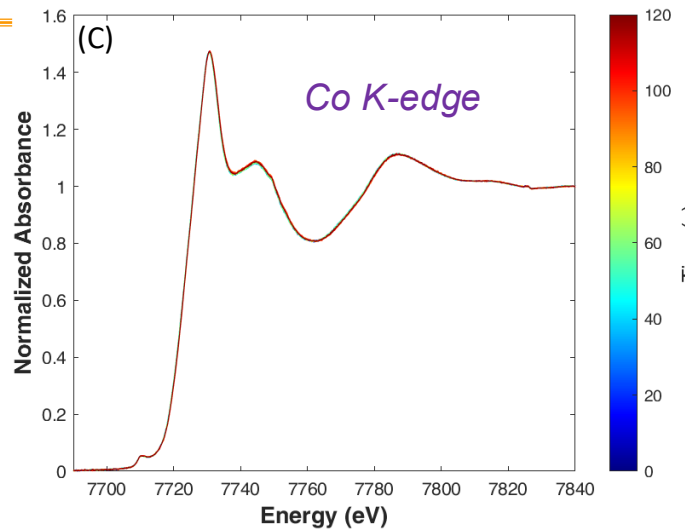
Electrochemical treatment : CA pulses



Focus on CV reversible pulses

1. Average the spectra into 1 period
2. Apply Phase Sensitive Detection (PSD)

$$A_k^{\phi^{PSD}}(e) = \frac{2}{T} \int_0^T A(e, t) \cdot \sin(k\omega t + \phi^{PSD}) \cdot dt$$



PSD



Only the Co is sensitive to CV pulses
2 Co species involved



PSD

